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## **Brief notes taken at a course of lectures on pneumatic chemistry, or the production, medical use, & application of factitious airs.**

[London]: [publisher not identified], 1796

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RY

Chymistry  
*or other*  
*Physicks*

M/2

A manuscript on 146 pages.

<sup>4</sup> Samuel Vaxley was a scientific instrument maker and founder of the Chemical Society, forerunner of the Royal Institution. The anonymous student has taken pains with his notes, which give a very clear picture of the state of Chemistry at the time.

The notes also include a series of lectures by A. Walker delivered in 1797.

*Flight & Williams,*  
 STATIONERS  
 and PAPER HANGERS.  
 No. 318  
 Corner of Southampton Buildings  
 HOLBORN

M.H.A.D.  
166.

Brief - Notes taken at a course of Lectures  
on Pneumatic Chemistry  
or the  
Production, Medicinal Use, & Application of  
Factitious Airs.

By Samuel Varley  
Delivered at Hatton-House, No. 16. Cross Street, Hatton Garden.  
Composed in 12 Lectures, commencing Nov. 24. 1796.

M.H.A.D.

166.

Explanatory & supplementary remarks.

There is often found a little carbonic acid air, as 1 in an 100 pts,  
in y<sup>e</sup>. atmospheric air. This no fire burns or animal breathes near.

GAHM

Prima Lectio.

Atmospheric air consists of 28 pts ~~Hydrogene~~ <sup>Oxygen</sup>, & 72 azotic gas  
Puristley & Scheele discovered Oxygen about same time.

Lavoisier's pneumatical apparatus cost 4 or £5000, but 4.  
simple one of Puristley equally useful, except in measuring  
caloric & weighing gasses.

Hydrogene g. made by putting a few iron turnings into a  
common glass phial, then adding water & sulphuric acid in  
equal parts, an effervescence takes place & 4. hydrogen gas  
is extricated or formed which is to be conducted into any  
other glass phial by means of a glass tube. NB. One end  
of 4. tube is inserted in a cork previously perforated by  
an iron rod, which is applied to 4. mouth of 4. bottle  
containing 4. iron & acid. It is matter of dispute between  
Puristley & Lavoisier whether in this process 4. iron or 7.  
Pneumatic. 100

Apparatus for making hydrogene gas & by burning it with  
 oxygene producing water. see Sect. 9.

A. A common quart bottle with wire filings, sulphuric acid & water. On  
 y<sup>e</sup> top of y<sup>e</sup> bottle is a tin box, to which is adapted a bladder with brass  
 cock. B. The bladder inflated, with a brass pipe inserted or screwed into  
 y<sup>e</sup> cock, this which the gas by pressing the bladder  
 Receiver C placed in y<sup>e</sup> pneumatica lib. & fitted  
 brass cock, y<sup>e</sup> upright one for exhausting y<sup>e</sup> air  
 filling A it with water by creating a vacuum  
 for conveying the gas by means



tube under an  
 by a wire fixed to  
 middle down into  
 from  
 round funnel or such like vessel supported  
 y<sup>e</sup> stand E. The glass D is for the water to  
 y<sup>e</sup> funnel E.

water is decomposed, P-y contends for y<sup>e</sup> former, Lavoisier's experi-  
ments more decisive.

This Hydrogene G. may be confined in any common phial  
by means of a cork; its goodness is ascertained by holding  
y<sup>e</sup> mouth of y<sup>e</sup> phial to a candle, y<sup>e</sup> quicker & louder y<sup>e</sup> exp<sup>n</sup>  
y<sup>e</sup> better.

Oxygen gas procured from Manganese, kept in quart  
bottles, necks downwards with a little water in to prevent  
access of other air. All gasses are very subtle as is known  
from Priestley's experiments with bladders filled with blood  
& placed nigh others filled with hydrogene & oxygene gas  
respectively.

A piece of twisted wire with a piece of cotton ballowed at  
y<sup>e</sup> end & inserted in a phial filled with oxygene gas  
affords a beautiful sight. The phial for this purpose ought  
to have no bottom & set in an hollow saucer (if white  
y<sup>e</sup> better) with a little water in to receive & cool y<sup>e</sup> balls

Pneumatics





of iron  $\frac{1}{2}$  are melted & formed in this experiment. The longer  
 $\frac{1}{2}$  gas supports the combustion of more pure & strong it is.

This oxygen gas is of great use in preventing contagious  
disorders, take one or more earthen pipkins containing  $\frac{1}{2}$  lbs  
of hot sand, place in  $\frac{1}{2}$  middle a cup containing  $\frac{1}{2}$  lbs  
of sulphuric acid to which add  $\frac{3}{4}$  of purified nitre,  $\frac{1}{2}$   
gas so obtained & denominated metric gas purifies  $\frac{1}{2}$   
air by attracting or seizing  $\frac{1}{2}$  putrid exhalations. 27  
pipkins sufficient for an hospital ship & will last 4 hours.

Pneumatics

Secunda — Lectio.

Chymical union attended with several singular circumstances,  
such as change of properties of  $\frac{1}{2}$  ingredients. Thus muriatic acid  
which may be safely taken if only diluted united with mercury  
which is of a bland nature constitutes a strong corrosive substance.  
a change of taste is produced, as is manifest from tasting  $\frac{1}{2}$   
compound result last mentioned. a change of smell which is



shown by taking ammonia & quick lime inodorous substances, joining  
ing them together, their compound (ammonia preparata) is highly  
plungent. Change of color shown by adding aqua ammonia  
to a solution of ~~fine~~<sup>cuprum</sup> vitriolatum, a fine blue color is produced  
which may be destroyed by vitriolic acid.

Synthesis & analysis of salt, an example of chymical affinities.  
Synthesis. Common salt or soda muriata consists of soda & muriatic  
acid, mix these & set them to evaporate slowly you will have  
good salt (in the process of carbonic acid escapes. Analysis.  
Pour sulphuric acid on salt, of muriatic acid escapes in fumes  
which if received in water wd constitute aqua fortis: of compound  
left behind wd be soda vitriolata Glauber's salt. Other examples  
of changes produced by chymical union exemplified by metals.  
Thus zinc & copper rubbed together give a brass color & are called  
brass; which metal differs in hardness from of constituent parts.  
If to brass thus formed more zinc be added you obtain a



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different metal still of extreme hardness so as to bear an excellent polish which makes it fit to be used for reflectors, especially as it never rusts.

The constituent parts of all bodies most probably hard, for water a drop of confined in a glass tube exhausted of air sounds like a stone when shaken.

That metals are susceptible of expansion by heat proved by only rubbing a metallic rod by y<sup>e</sup> hand, it will not enter y<sup>e</sup> same groove it did. Straight wood expands about y<sup>e</sup> 40 part as much as iron.

Passive caloric y<sup>e</sup> same as latent heat.

Mix = weights of sulphuric acid & water, no effervescence takes place yet much heat ensues; y<sup>e</sup> caloric <sup>of y<sup>e</sup> acid</sup> is probably displaced & shews itself. Something is disengaged one w<sup>d</sup>. suppose because y<sup>e</sup> bulk of y<sup>e</sup> compound is not equal to y<sup>e</sup> bulk of the ingredients.



Mix equal weights of sulphuric acid & water pour it on soda vitriolata, y<sup>e</sup> cold produced will be 40 or 50 degrees below y<sup>e</sup> common temperature.

Add to 2 oz of soda vitriolata 11 dr. of ammonia & 10 of lixivium nitratum y<sup>e</sup> cold will be more intense

The doctrine here alluded to further illustrated by covering y<sup>e</sup> bulb of a thermometer with a resin coat, this on y<sup>e</sup> application of warmth melts but y<sup>e</sup> mercury descends.

The double bulbed tube containing colored alcohol applied to this subject; for when y<sup>e</sup> spt is driven out into y<sup>e</sup> remote bulb a degree of cold is felt by y<sup>e</sup> hand.





Pressure of y<sup>e</sup> atmosphere demonstrated by laying y<sup>e</sup> hand on y<sup>e</sup> top of an open receiver & then exhausting y<sup>e</sup> internal air; y<sup>e</sup> hand cannot be taken off.

This is further explained by affixing a piece of wet bladder instead of y<sup>e</sup> hand; at every stroke of y<sup>e</sup> pump y<sup>e</sup> bladder is pressed inwards in a funnel form and at last it bursts by y<sup>e</sup> superincumbent pressure of atmospheric air. The explosion is great owing to repercussion of y<sup>e</sup> air against y<sup>e</sup> plate at y<sup>e</sup> bottom.

An experiment illustrating y<sup>e</sup> doctrine of sounds made by enclosing a drop of water in a bulbous tube exhausted; on y<sup>e</sup> application of y<sup>e</sup> warm hand y<sup>e</sup> water is impelled against y<sup>e</sup> sides of y<sup>e</sup> bulb with y<sup>e</sup> noise of a pebble or shot.

Pressure of y<sup>e</sup> atmosphere shewn by inverting a small



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Bolt head & immersing it halfway in a glass of water which is placed under a receiver; on exhausting y<sup>e</sup>. receiver y<sup>e</sup>. air descends from y<sup>e</sup>. bolthead into y<sup>e</sup>. water & ascends up in bubbles; on restoring y<sup>e</sup>. air into y<sup>e</sup>. receiver y<sup>e</sup>. water is forced up into y<sup>e</sup>. neck of y<sup>e</sup>. bolthead to supply y<sup>e</sup>. place of y<sup>e</sup>. air y<sup>e</sup>. had escaped.

Let y<sup>e</sup>. same bolthead be placed as before only let it be filled with water leaving however a bubble or little space of air at y<sup>e</sup>. top; on exhausting y<sup>e</sup>. receiver this confined air expands itself & expels y<sup>e</sup>. water.

Pressure of y<sup>e</sup>. atmosphere shewn by exhausting y<sup>e</sup>. air out of a receiver, it is then so fast as not to be displaced.

Further illustrated by placing two receivers on y<sup>e</sup>. plate one within y<sup>e</sup>. other; on exhausting y<sup>e</sup>. air, y<sup>e</sup>. external large



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receiver will be rivetted to its place by y<sup>e</sup> atmosphere  
but y<sup>e</sup> small interior receiver not being acted upon by any  
external air / being within an exhausted receiver or vacuum /  
is moveable by shaking; but take away y<sup>e</sup> exterior receiver  
& it is in its turn immoveable

The same doctrine exemplified by taking two hollow brass  
hemispheres, which are made to fit tight by means of  
tallowed leather; there is an hollow neck to y<sup>e</sup> inferior  
one so y<sup>e</sup> being placed over y<sup>e</sup> orifice of y<sup>e</sup> plate y<sup>e</sup> air  
within y<sup>e</sup> sphere may be exhausted & prevented from returning  
by means of a stop cock; the air being exhausted y<sup>e</sup> hemisphere  
cannot be severed by y<sup>e</sup> strength of any one person, but on  
turning y<sup>e</sup> stop cock & readmitting y<sup>e</sup> air are easily separated.  
Put a small bladder fastened at y<sup>e</sup> neck & enclosing a little



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air under y<sup>e</sup>. receiver; on exhausting y<sup>e</sup>. air within y<sup>e</sup>.  
receiver & removing y<sup>e</sup>. pressure, y<sup>e</sup>. air within y<sup>e</sup>. bladder  
dilates so as ultimately to burst it.

The converse of this shewn by means of a bladder within  
a glass, & put under y<sup>e</sup>. receiver; in this case y<sup>e</sup>. more y<sup>e</sup>.  
receiver is exhausted y<sup>e</sup>. more is y<sup>e</sup>. bladder compressed.

The expansive power of y<sup>e</sup>. air when y<sup>e</sup>. pressure of y<sup>e</sup>. atmos:  
phere is removed is further shewn by taking a phial  
part full of water with a siphon in y<sup>e</sup>. connecting it with  
another vessel of water; on exhausting y<sup>e</sup>. receiver y<sup>e</sup>. small  
portion of air left in y<sup>e</sup>. first phial expands & drives y<sup>e</sup>.  
water out by means of y<sup>e</sup>. siphon, but on restoring y<sup>e</sup>.  
equilibrium y<sup>e</sup>. water disembogued returns & ascends  
by y<sup>e</sup>. siphon.





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The last experiment diversified by using a short siphon whose external end did not communicate with  $y$ . water in  $y$ . open vessel, & therefore  $y$ . water discharged did not return by it but only air; a second phial was therefore exhibited with a long siphon communicating with  $y$ . water in  $y$ . open vessel likewise & thro' this  $y$ . water which had been discharged by  $y$ . first siphon was transmitted to a third vessel or phial.

This doctrine further illustrated by artificial fountain.

Air weighed by means of a Florence flask, with a brass cap perforated & covered close by a piece of oiled bladder; this admits  $y$ . air to be exhausted but does not suffer it readily to return.

A piece of cork & lead suspended in equilibrio at  $y$ . ends of a balance, put into an exhausted receiver, shew  $y$ .  $y$ . cork being bulkier & meeting with no resisting medium sinks first.



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A piece of wood put into water & the in an exhausted receiver emits for  $\frac{1}{2}$  space of an hour a prodigious quantity of air. An apple does  $\frac{1}{2}$  same; but after  $\frac{1}{2}$  operation is shrivelled.  $\frac{1}{2}$  wood is heavier having absorbed so much water in lieu of its air; if it had been immersed in mercury it wd. being cut appear full of metallic particles.

A glass of water submitted to  $\frac{1}{2}$  receiver emits an astonishing quantity of air in  $\frac{1}{2}$  form of froth; after  $\frac{1}{2}$  operation it is flat & vapid. This shews  $\frac{1}{2}$   $\frac{1}{2}$  carbonic acid air is  $\frac{1}{2}$  rest of malt liquor.

The nature of  $\frac{1}{2}$  Air Gun explained

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For making oxygen gas from manganese, choose  $\frac{1}{2}$  larger pieces, & take care  $\frac{1}{2}$  it be clean & free from all foreign substances especially inflammable ones as straws &c. Let all the vessels used in the process be very clean. The manganese must be powdered in a glass or marble mortar. Wedgwood one is better than any - An earthen retort, glass tube, water tub & bottle, open furnace with charcoal fire used by Mr. Varley.

This oxygen is the acidifying principle of the universe, as is hereafter shewn by deflagrating sulphur, charcoal, & phosphorous in it; whereby acids are produced.

Oxygen gas may be inspired 5 times before it be reduced to  $\frac{1}{2}$  state of atmospheric air - Is not miscible with water - Heavier  $\frac{1}{2}$  atmospheric air - Mr. O. thinks that after entering  $\frac{1}{2}$  blood it is united with the carbonic acid & passes of thro'  $\frac{1}{2}$  pores as well as lungs, being thus rendered volatile.

It is a question among chymists whether calx of manganese will recover its oxygen by long exposure to  $\gamma$ . atmosphere— Mr. V. thinks it does not.

The calx however is equally serviceable as  $\gamma$ . manganese earth itself in making glass & enamel.

Distinction is to be made between oxygen & oxygen gas. The oxygen is united with  $\gamma$ . manganese earth from which being separated & combined with caloric becomes a permanent elastic fluid denominated oxygen gas.

Oxygen gas inhaled increases the pulse in 4 minutes from 74 to 94 & in 4 minutes more to 124.

Oxygen gas has a great affinity for arctic gas. Hence oxygen gas thus bottled & suffered to remain in the vicinity of a necessity is soon corrupted. Further experiment with bladders of blood. Also by wetting  $\gamma$ . leaves in one end of a book with solution of acetate of lead, & the other end with hepatic sulphuric; the writing made by  $\gamma$ . former appears,  $\gamma$ . acetate of lead discharges its oxygen &  $\gamma$ . lead becomes mineralized or regularized as seen thro' a microscope.

is open at both ends,  $\frac{1}{2}$  of upper end is closed with a piece of blue slate or glass ground called the valve, &  $\frac{1}{2}$  of lower end is placed in a cup of water. The combustion seen over the water ascends into  $\frac{1}{2}$  jar which indicates  $\frac{1}{2}$  nothing has been expelled from  $\frac{1}{2}$  metals & other substances, but something has been absorbed. This phenomenon takes effect when only atmospheric air is used in a small degree.

In making phosphorus W. O. finds that when pure glass is used & the distillation is slow,  $\frac{1}{2}$  best is obtained. Great care necessary in using it, as it ignites at  $40$  degrees of temperature. Many shocking accidents have happened from not using this with due care.

Thick glasses when exposed to any heat crack sooner than thin ones.

It is almost impossible to expell  $\frac{1}{2}$  last portion of water from charcoal, hence its impurity. 72 grains of oxyd muriate of pot ash yield 100 cubic inches of oxygen gas; great care however is to be taken  $\frac{1}{2}$  no inflammable substance come into contact with it or it will explode.

Water is very greedy of carbonic acid air; takes up voluntarily an equal quantity, & by agitation a great deal more.

Test papers to try acidulated liquors, prepared from a solution of lyebris.

Lecko. 5.

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Experiments made by burning phosphorus, charcoal, & sulphur in oxygen air - in the first case phosphoric acid is produced by a brilliant deflagration, in the second, carbonic acid; in the third sulphuric acid attended with a fine blue flame.

A steel watch spring burnt in oxygen air deflagrates with much brilliancy, the sides of the glass are covered with a sort of rust, the red oxide of iron.

A common iron wire is converted into black oxide of iron, but steel which is free from plumbago carbon & which of itself is in the first stage of oxygenation is converted as observed into a red oxide.

If these oxides were weighed they would be found equal to the metal & the oxygen air consumed. Hence it is inferred that instead of something being thrown out from the metal as the Phlogistonians contend, something is absorbed - The acids formed from the simple substances sulphur & phosphorus are equal in weight to the substance & the oxygen gas used.

In oxygenating these substances another circumstance occurs which militates against the Phlogistonian system. The glass jar used on these occasions is



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For making a simple apparatus at a cheap expence for obtaining different gasses; provide yourself with an earthen pipkin, & let it have a stand or shelf within held fast to the side by means of two upright deal sticks tightened by a screw or string. Then procure a few pipe stoppers, let y<sup>e</sup> ends of y<sup>e</sup> pipes be tapered or ground with a scythe rag stone, and let y<sup>e</sup> corks be first perforated with a broach & then rounded & enlarged with a file or hot iron. If they should not be air tight which may be known by keeping them under water & blowing thro' them, take a little wet Turbitch clay, or bees wax. The latter however must not be used, where any heat is present; the former may be got in Long-acre & near St. John's Gate, & should be passed thro' a lawn sieve before used. Thus nitrous gas is prepared. The more crooked the pipe the better



This glass used for breathing in & condensing y<sup>e</sup> air into water; or it may be filled with oxygen air & used for complaints.

a bird confined in a glass half full of water with a wood perforated to stand upon, in a little time grows faint but on y<sup>e</sup>. admission of oxygen recovers.

Air passed thro' y<sup>e</sup>. lungs is impregnated with carbonic acid, proved by blowing thro' a tube into a phial of lime water which becomes precipitated

Air in which a candle has burnt is charged with carbonic acid & precipitates lime water in like manner; when y<sup>e</sup>. candle is just expiring apply oxygen & it recovers.

The goodness of air formerly tried by seeing how long a candle or mouse w<sup>o</sup>. live in a given quantity, but is now tested by uniting nitrous gas to it which absorbs y<sup>e</sup>. oxygen & occupies less room in proportion to y<sup>e</sup>. purity of y<sup>e</sup>. air.

Nitrous gas is produced by pouring nitrous acid diluted with an equal quantity of water on

The residuum after nitrous gas is obtained, may be used for making match papers. Take slips of Brown paper & dip them in  $\frac{1}{2}$  gr. Nitrate of copper, dry them, & they will take fire with almost the warmth of the hand. Useful for igniting any air or other matters in close vessels, by first putting in a slip of this paper, & then holding a hot poker on the outside. After deflagration the red copper appears which may be used for making enamel.

Before  $\frac{1}{2}$  gr. atmospheric, oxygen, or other gas be combined together in the eudiometer, let the two airs be poured into a wide mouth bottle & shaken smartly as they are mixed; then pass them into  $\frac{1}{2}$  gr. eudiometer; this is Mr. Cavendish's method. Instead of marking the eudiometer with a diamond as Dr. Priestley does which mark is not seen when the tube is full of water; Mr. C. applies a common wood scale to the outside. If  $\frac{1}{2}$  gr. nitrous gas has much water to pass thro' (which is the case in long narrow tubes) its strength will be impaired by  $\frac{1}{2}$  gr. absorption of the water.

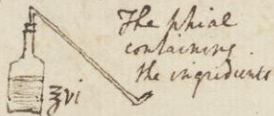
on copper filings <sup>or shavings</sup> in a phial; y: apparatus is simple & continued by tobacco pipes.

Acid obtained by means of hep. sulphur <sup>or sulphate of potash</sup> made of caustic fixed alkali & sulphur which must be kept in close bottles.

The attraction of hep. sulphur is such, y: if you write with a solution of sac. saturni or sulphate of lead on y: beginning of a blank book & wet y: last leaf with a sponge dipped in solution of hep. sulph. it will strike thro' all y: leaves & cause y: writing to be visible. It will even act thro' a stone wall which proves y: nothing much less y: lungs is impervious to y: action of oxygen.

Atmospheric air may be respired 5 times successively & with a little respite 8 or 9 times. It fatigues & oppresses the person very much, but this may be removed by taking a little hydrogen or <sup>hydrog.</sup> carbonic acid gas

Hydrogene gas obtained from zinc or iron filings or clippings immersed in a dilution of vitriolic acid. Iron filings are apt to be greasy, require more acid, & more heat. Zinc requires no heat & only  $\frac{1}{4}$  of acid to  $\frac{1}{2}$  water used. After air is come away, white vitriol will be formed. Hydrogen is  $\frac{1}{8}$  lightest of all airs, & therefore when mixed with oxygen to produce an explosion if it has stood any time, should be shaken together.



In order to expell the air from the bladder, it should be laid flat on the table & rolled; if some hydrogene gas was introduced & then expelled, the bladder would be better prepared. Apply a candle to the end of the glass tube, as the hydrogene comes over & you have a philosophical candle.

Section 7

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Hydrogene gas mixed with  $\frac{2}{3}$  atmospheric air or  $\frac{1}{3}$  oxygen conducted into a bladder by means of a bottle without a bottom (which being immersed in water of air is driven upwards thro' y<sup>e</sup> neck of y<sup>e</sup> bottle into y<sup>e</sup> bladder) is made by pressing y<sup>e</sup> bladder to pass thro' a pipe & inflate soap bubbles which ascend to y<sup>e</sup> top of y<sup>e</sup> room; touch these as they float with a lighted candle & they explode with prodigious noise. I believe a certain proportion of oxygen is mixed with it.

It is remarkable that tho' this air is inflammable when in contact with atmospheric air, it of itself extinguishes flame as is tried by putting a lighted match into a phial full of it. Invert the bottle & it will continue to blaze at the mouth. The flame called y<sup>e</sup> philosophical candle is difficult to blow out, & owing to y<sup>e</sup> carbonic acid breathed from the

A curious phenomenon is exhibited by means of phosphorus mixed with caustic lime. For instance put into a Bulbous phial holding 6 or 8 ℥, a dram or two of Salt of tartar & half the quantity of caustic alk lime, then about an ounce or more of water, & put under a spirit lamp; when you think the atmospheric air in the phial is expelled add 5 or 6 grains of phosphorus, & apply the lamp again. By means of a glass tube crooked at both ends, one inserted in the mouth of the phial, & the other dipped in water, so as leave its surface just at the surface, you will have bubbles of flame come over, accompanied with curious wings of smoke.



lungs, gives a blue flame when blown upon.

An air pistol charged with hydrogen, & fired by means of an electrophorus gives a good report; instead of hydrogen a drop of ether let fall into the pistol will answer. But the most stunning noise ever heard is made by filling the bladder with 2 parts hydrogen & 1 oxygen & by means of a pipe stopper affixed to the end of it, blow up a number of soap bubbles in a wash hand basin, then set fire to them & they all explode at once.

Hydrogen breathed 3 or 4 times over loses its inflammability, owing to its being impregnated with carbonic acid gas. It lowers the pulse very much.

Fulminating silver, explodes with the least motion.



### Lectio 8.

Hydrogene inflamed by electrical fluid in an electrophorus. This gas will preserve its strength 8 or 10 weeks. A bladder is fitted to the head of a bottle containing  $\frac{1}{2}$  ounce of sulphuric acid, which bladder has a stop cock, & is afterwards decembogued into the electrophorus, displacing by force  $\frac{1}{2}$  ounce of water in the lower part or vessel & compelling it to ascend by means of a tube into the superior part— Mix minium or litharge with tallow, put a lump into excavated charcoal & with a blow pipe direct the flame of a candle to it,  $\frac{1}{2}$  ounce of metal will be refined. A wafer held to a candle will exhibit particles of lead. Zinc plate dipped into a solution of acetate of lead will have  $\frac{1}{2}$  ounce of lead precipitated upon it.  $\frac{1}{2}$  ounce of same happens to iron immersed in solution of copper. Or a few drops of lit fall into a bladder on  $\frac{1}{2}$  admission of air converts it into hydrogen. Shewn by fixing a pipe to the mouth of the bladder & then forcing the gas out at the point of a candle, a stream of fire issues from the pipe; if the bladder

has been filled with air from the lungs y<sup>e</sup> flame will be <sup>bluish</sup>.  
Instead of a pipe, a pair of flues may be affixed to the bladder  
which will afford a resemblance of a fire wheel. Hydrogene  
mixed with arot, gives a red flame. Nitrous gas a green.  
Hydrogene which has passed the lungs acquires a little carbonic  
acid, as it precipitates lime water.

### Lectio. 9.

An experiment was made by some Chymists on water. They took  
a gun barrel, laid it in a long furnace, at one end was a retort  
containing 12 lb of distilled water heated by a chafing dish. The  
steam passed y<sup>e</sup> gun barrel, & when condensed weighed as much as  
at first, & was unchanged or decomposed. They then filled the  
barrel with iron filings, the water was by that means decomposed  
& converted into hydrogene gas; the tube was weighed & found  
to have gained what y<sup>e</sup> hydrogene wanted in weight. Then  
instead of iron filings, charcoal was used, it should not be produced

too fine, nor yet be in large pieces, in y<sup>e</sup> former case y<sup>e</sup> charcoal  
w<sup>d</sup> be carried off by y<sup>e</sup> water, in y<sup>e</sup> latter it w<sup>d</sup> pervade the interstices  
without acting on it) here then hydro-carbonate gas, by some  
called carbonated hydrogen, & by others Heavy inflammable air  
was formed. It is a compound of hydrogen & carbonic gas, but different  
from these gases mixed simply. Mr. Barley sets his Gun barrel  
perpendicular, the lower end being fitted with a brass cap or breech  
& a large touch hole, made capable of receiving a retort. Really experiments  
made by displacing y<sup>e</sup> bark of charcoal in oxygen, y<sup>e</sup> bark which is  
bottle, & looks metallic, & crackles when ignited is the best. Water  
formed by causing oxygen & hydrogen to enter y<sup>e</sup> same receiver  
on y<sup>e</sup> opposite sides thro' gun barrels. Mr. B's plan is to fill a  
phial with zinc filings, sulph. acid & water, & conduct the hydrogen under  
a receiver two thirds full of water; y<sup>e</sup> pressure of y<sup>e</sup> water drives y<sup>e</sup> gas  
thro' a cock at y<sup>e</sup> top & thro' a small tube which is bent & turned upwards  
The gas is lighted with a candle & a glass receiver or inverted funnel  
is held over it, the air escapes thro' y<sup>e</sup> top of y<sup>e</sup> funnel, y<sup>e</sup> oxygen  
comes from y<sup>e</sup> atmosphere & uniting with y<sup>e</sup> hydrogen is converted  
into water which trickles down y<sup>e</sup> sides of y<sup>e</sup> funnel.

## Lectio. 10.

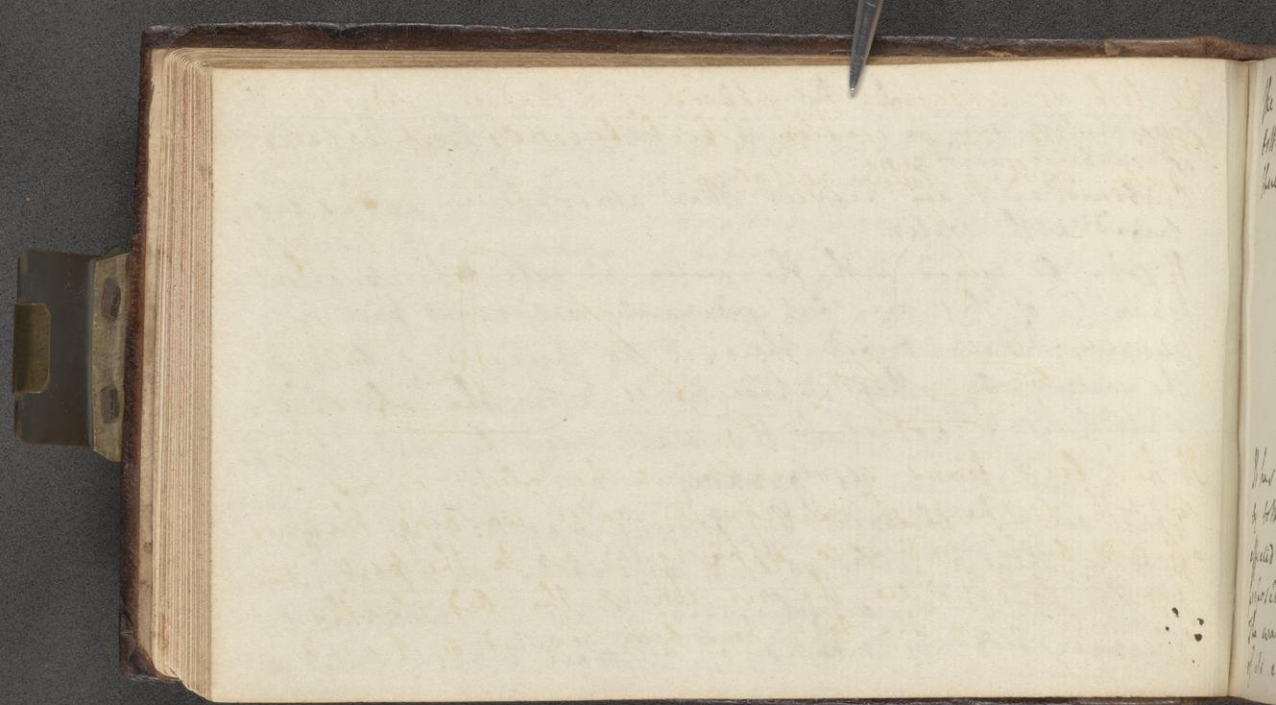
41

Fermentation may be considered as nothing more than the extrication of carbonic acid air.

Carbonic acid air heavier than atmospheric; and is easily mixed with water.

If soda be mixed with the water it will take up a double quantity of this air; but ordinarily only equal part or quantity; when mixed there is no increase of bulk in the water; by other improvements & by the assistance of pressure & agitation the water may be made to take up 4 or 5 times its quantity.

This air is obtained by pouring marble dust into a bottle then a little water & afterwards an equal quantity or thereabouts of vitriolic acid; if air is immediately disengaged & by means of a crooked tube inserted in



The Cork is conducted by means of a bladder under a  
bottle inverted in water, which becomes impregnated  
thereby, with a little agitation. 43



It has been found serviceable in paralytic cases  
by both drinking of impregnated water, washing the parts  
affected by it, and also applying of air to the part; it is  
proposed to introduce the air under the bed cloaths.  
The water after the body has been washed by it is divided  
of its air.



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45  
It has been tried with success in healing ulcers &  
stopping mortifications; in which cases the air is applied  
& confined to the parts by means of bladders -  
Philosopher Walker's wife was cured of a sore breast  
by the air.

The water impregnated is eminently serviceable in cases  
of stone & gravel, & probably the air or water might avail  
in the gout.

When the ulcers have been very foul, the carbonic  
acid air is sometimes decomposed, & y: water turns  
black.

This air extinguishes flame

The water impregnated with this air when bottled  
and with y: addition of a Spoonful of iron filings is a  
fine chalybeate & good in low nervous cases.





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47

There is a singular experiment to be made with charcoal. Heat it well pound it in a heated marble mortar put it into a cup over a gentle heat or lamp for the purpose of extracting & expelling aqueous particles then pour upon it strong nitric acid & the charcoal will become ignited in a brilliant manner. This should be done in the chimney on account of the nitrous fumes that arise.



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## Lecture II.

49

All species of spars or spathose earths yield fluor acid gas. by pounding them to a coarse powder, adding equal quantity of sulphuric acid, & applying a gentle heat as lamp.

'Tis observed however y<sup>e</sup>. coloured spars are most productive. The fluor acid gas has y<sup>e</sup>. singular property of corroding glass with y<sup>e</sup>. same facility as aqua fortis does metals; and it is observable y<sup>e</sup>. in countries abounding with spar the windows of churches have suffered much.

It has been found by Mr. Barley y<sup>e</sup>. in those parts where spar abounds as in y<sup>e</sup>. district between Doncaster & Gainsboro' y<sup>e</sup>. corpses are long before they corrupt after interment.

To check glass let it be coated with bees wax & colcothar, & in 5 minutes after it has been exposed to the action of the f.a. gas it will be marked. The extrication of y<sup>e</sup>. air after y<sup>e</sup>. operation is done may be checked by y<sup>e</sup>. external

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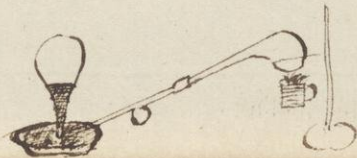


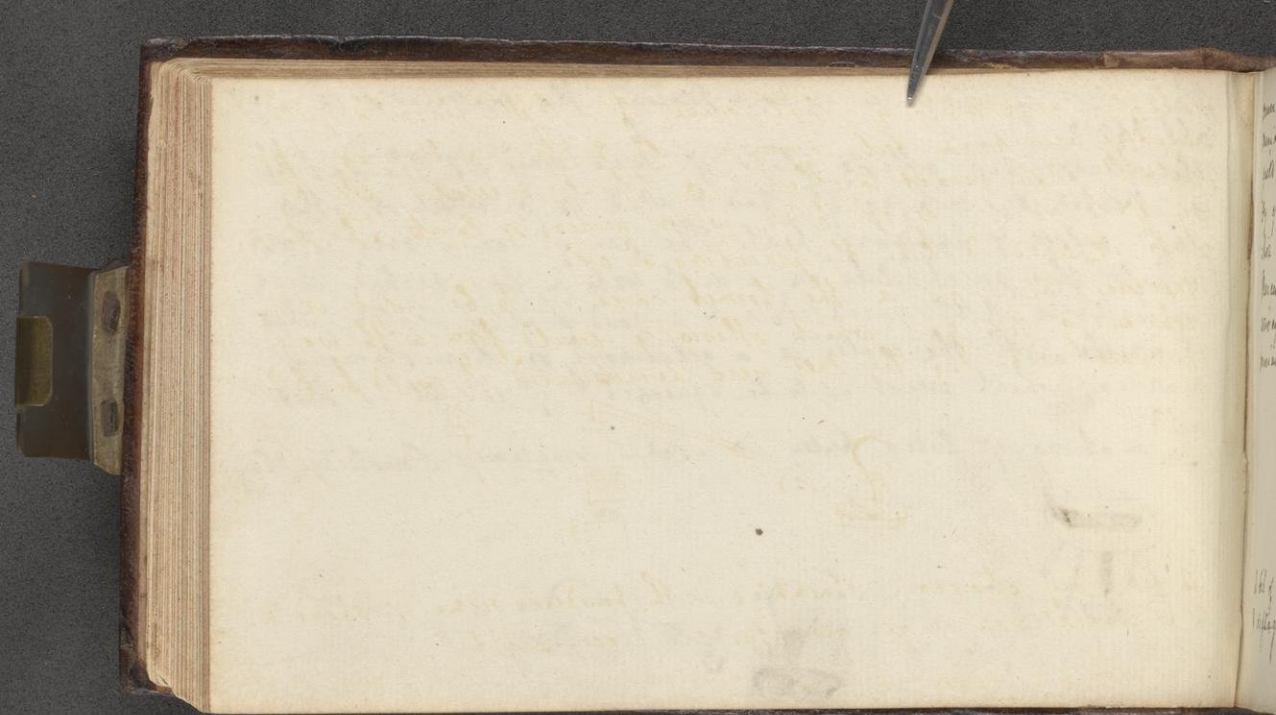
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Application of cold water or even placing the materials in a cold place. (51)

The fluor acid affects glass in the same way as the gas & is obtained by mixing of spar & acid in a retort or other glass vessel & applying heat. The air which comes over is made to pass through quicksilver in a dish into a long necked phial containing a little quicksilver & a good deal of water; when it unites with the water in a gelatinous or crustaceous manner; which crust is to be squeezed & the result is fluor acid.

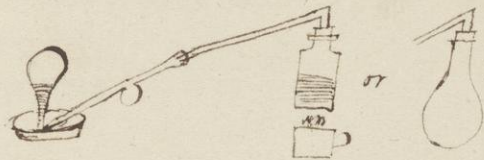
In making glass tubes putty is used - sometimes Stourbridge clay.





53  
muriatic acid with  $\frac{1}{2}$  application of a little warmth yields  
muriatic acid gas - but common salt 2 pts & sulph. acid. 1 pt.  
yields it also mixed together in a phial or retort & assisted  
by  $\frac{1}{2}$  heat of a warm hand, warm water, or lamp. The opera-  
tion is checked like the preceding by cold.

Mercury also as in the former case is to be used in  
conveying the gas which otherwise would form with water  
muriatic acid. If the salt were deceptated it would be better.



A bit of dry charcoal sprinkled with powdered zinc suspended  
& deflagrated in oxygen air affords a pretty sight.





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## Sect. 12.

55

Oximuriatic gas obtained by distillation of salt with sulph. acid with  $\frac{1}{2}$  of substance of manganese. Proportions, Manganese 6 $\frac{1}{2}$  salt powdered 16 $\frac{1}{2}$ . sulph. acid & water aa 12 $\frac{1}{2}$ . These ingredients will of themselves yield  $\frac{1}{2}$  gas, but it is better to assist  $\frac{1}{2}$  operation with a lamp. As glass phials are apt to burst & in this case produce disagreeable consequences Mr. C. contrived lead ones, but they also are liable to melt.



In the large phial is a capillary tube called the tube of safety, which prevents  $\frac{1}{2}$  gas from bursting the phial.  $\frac{1}{2}$  water within  $\frac{1}{2}$  tube is driven up when the gas is violent & by  $\frac{1}{2}$  superincumbent water cannot escape.



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The crooked tubes are inserted loosely into the necks of the phials, & closed with putty which effectually prevents the nitric gas from escaping. If it does however escape, it may be detected by dipping a feather in vol. alkali & applying it to the neck of the bottle, the gas will be precipitated in a white cloud.

This oxemuratic gas is highly serviceable in bleaching destroying colors, washing &c; it will applied to the shirt of a blackamoor turn him white.

If instead of water alone in the large three necked vial, a solution of pot ash be used, a salt will be produced called Salt of Sylvius, which will be succeeded by a second (the first being taken away) called oxyd muriate of pot ash with prismatic fibrous crystals. Some use mild, some caustic, some vegetable, & some mineral alkali for this purpose.



59  
One grain of this oxyd muriate of pot ash, mixed with 2 grains of sulphur, in a marble mortar, with the addition of  $\frac{16}{1}$  grains of phosphorus will by strong friction produce a great explosion. The French intended substituting this mixture for gun-powder, it being much stronger.

60 cubic inches of oxyd muriatic gas in a temperature of 70 will (contained in a bottle) by pouring in powdered antimony produce an agreeable sight. Some precaution is however necessary for as the oxygen is absorbed so the nitric gas is set loose.

Experiment of Boyles. Take a Florence flask, put in 13 of water & a little phosphorous suppose 31. hold it over a lamp, a beautiful green flame arises in the bottle, accompanied by sparks like shooting stars - Every now & then the lamp is



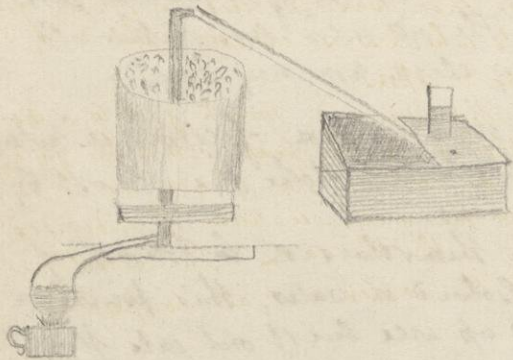
is extinguished, for y<sup>e</sup> purpose of making the Stars shoot better.<sup>61</sup>  
Phosphorus which is not reckoned very pure starts this best,  
as y<sup>e</sup> made from urine answers better than y<sup>e</sup> from bones.

Concentrated sph of nitre poured on oil of turpentine produce  
an instantaneous ignition. The oil used was rather old,  
the nitric acid was not very clear nor very brown

If brown paper be steeped in a solution of copper in nitric  
acid it makes good matches & will take fire almost by  
y<sup>e</sup> application of a warm hand.

If y<sup>e</sup> crystals which form themselves in a solution of  
copper in nitric acid be dissolved in water, this poured on  
tin foil & immediately lapped up will burst out into flame  
& the tin will be oxidified

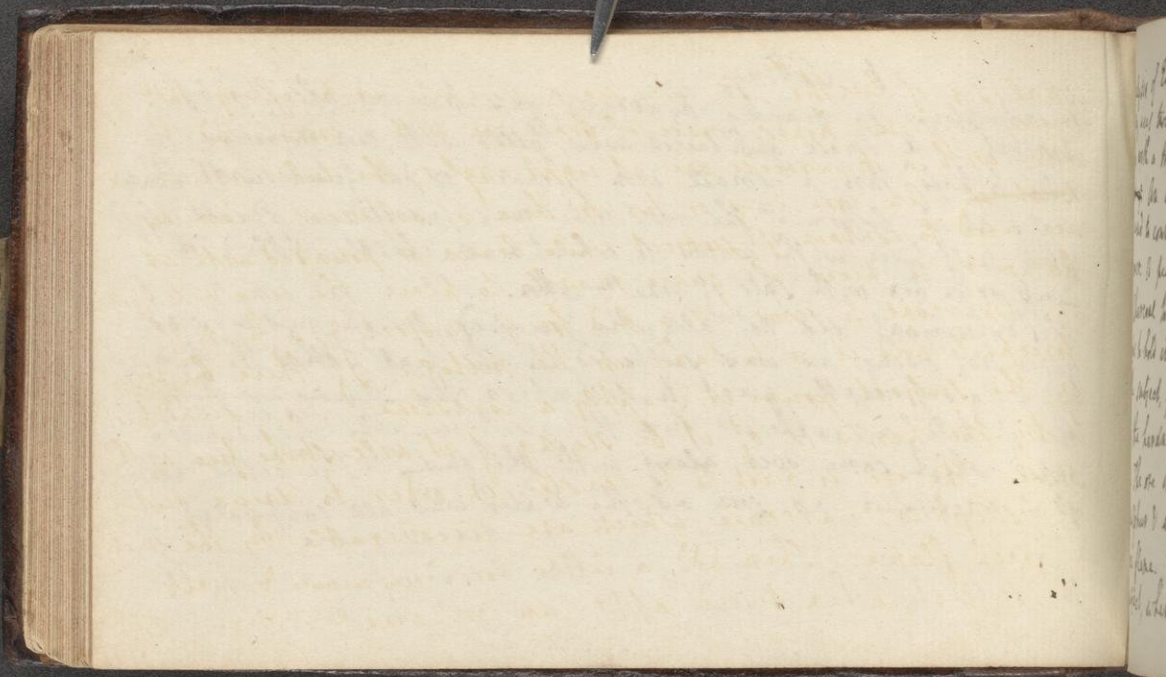




## 1. Sifting.

Was employed in making hydrogen gas from iron assisted by charcoal. A small apparatus was filled with iron turnings & made to pass thro' a small iron furnace or pot filled with char. coal, at y<sup>e</sup> bottom of the iron tube was inserted a small glass retort half filled with water to which was closely applied a lamp or tin box with spls of wire & cotton.

This experiment did not altogether succeed, much gas was not generated; owing as was supposed to the heat within the furnace not being sufficiently great to decompose the steam as it passed thro' y<sup>e</sup> tube, as appeared from the great quantity of condensed vapor which came over along with the gas. The retort broke at y<sup>e</sup> conclusion, as soon as the water was all evaporated.



2.  
 Analysis of Earths. For assaying a blow pipe is necessary. Mr. Valey uses two kinds made of brass, one with a crooked end, the other with a transverse one at right angles; the tubes are conical. ~~beat of~~ An iron ring or two of these of different diameters are used to contain y<sup>e</sup> subject while beaten or pounded with an hammer. A piece of square marble to place the ring & subject on. Charcoal to serve as a bed for y<sup>e</sup> subject, & a pair of forceps to hold it, or to take up the metal. If there be iron in the subject, pound it, & apply a loadstone. 'Tis usual first to try the hardness of it by seeing if it will strike fire with steel. The ore is first to be roasted in order to drive out the sulphur & arsenic, which are discoverable by the smell & green flame. Then add a little borax very minute & melt the subject, when fused apply an iron wire & if there be



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any copper present it will attach itself & be precipitated <sup>67</sup>  
on the iron; if silver be suspected to exist, use a copper wire  
in like manner. A magnifying glass is necessary both  
before & after fusion to examine the shape & color of the  
materials. In some cases a gold silver or platinum spoon  
is used, but the bowl of a tobacco pipe No. 6 works as  
good as ought. In lieu of Borax, mineral alkali & micro:  
cosmic salt is necessary.

Detonating powder. Take 3 grains of oxy muriate of potash  
1/2 gr. of sulphur, & 1/2 gr. of charcoal; grind them all separately  
then mix them; place gr. compound on a piece of paper on an  
anvil & strike gr. anvil with an hammer. The powder will  
both deflagrate & detonate.



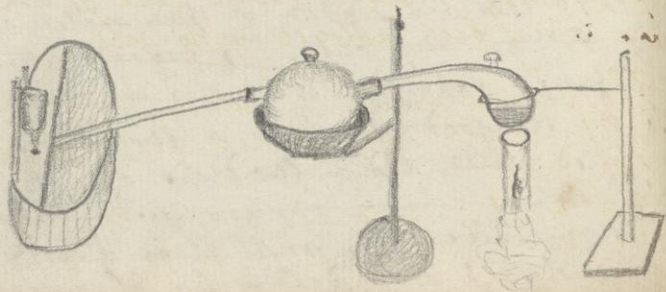
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69

Nitric acid. Take 3iss of nitre powdered, put it into a Stopped  
retort, with a tubulated receiver luted on; to which must be  
added 3 vials connected by crooked tubes half filled with water,  
& terminating with a tube in a pneumatic cistern. Place the  
retort over Argand's lamp & add to ʒi. nitre ʒiiss of sulphuric  
acid. The acid which will be found in ʒi. receiver is very good  
& free of all muriatic & sulphuric acid & amounts to ʒiiss. or  
more. The gas which passes thro' ʒi. several phials & is  
at last received under ʒi. pneumatic cistern is oxygen  
gas but not very pure. To free ʒi. nitric ~~gas~~ acid of its  
acid fumes expose it in an open vessel over Argand's lamp.  
To try its goodness, pour a few drops into a glass of  
distilled water, add a few drops of nitrated silver, if any  
cloud or precipitation ensues, tis a proof that it is not free  
from muriatic acid; for if it be clear & you add a grain  
or two of salt or a drop or two of muriatic acid, you will





71

have y<sup>e</sup>. mixture turn milky. Again to try whether it contains any sulphuric acid, pour a few drops into a glass of distilled water as before, then add a few drops of nitrated barytes, & if a cloud follows tis a sign y<sup>t</sup>. y<sup>e</sup>. sulphuric acid is present; for if it should prove clear, & you drop in a little epsom salt or sulphuric acid y<sup>e</sup>. mixture will be thick & milky.

In making nitrated silver, care sh<sup>d</sup>. be taken y<sup>t</sup>. y<sup>e</sup>. silver has never been alloyed with gold

Phosphorus immersed in nitric gas super-oxygenated (i.e. distilled with manganese) causes a great heat, & smoke, & at last a surprising deflagration.

Fuming nitrous acid was purified by distilling it afresh in a glass retort over Argand's lamp; y<sup>e</sup>. fumes pass'd thro y<sup>e</sup>. retort



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73  
into of receiver, where they lodged; to of receiver was luted a  
glass tube connected at of. Other end with an pneumatic apparatus  
for of purpose of collecting & trying of gas of came over. A great  
deal of gas or air was produced, but it was not oxygen as was  
conjectured. The carbonic acid by this process was rendered very clear.



Wald  
Spul  
Land v  
Gammel  
die g:  
die rde  
had  
die w  
die w  
die w

75

Notes taken at a Lecture on Chymistry by  
A. Walker delivered in Conduit Street Hanover Square  
in April. 1797.

Pound some Pyritis, add water to it, then apply the  
Thermometer; you will find an increase of temperature—  
Some of the warmth of Bath waters accounted for; they  
are percolated thro' mountains abounding with Pyrites—  
The heat proceeds from the sulphur (the mother of vitriolic  
acid) which is contained in the pyrites—add a little  
vitriolic acid to water, heat is generated, i.e. caloric  
escapes.



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Again Pyrites contains Iron, & therefore if water in which Pyrites has lain will by ~~down~~<sup>boil</sup> green tea or a solution of galls turn black.

Let some Iron nails or filings be put into a mixture of vitriolic acid & water, & hydrogen Gas will be produced; an open glass vessel in which this Gas has been received will on if application of a lighted Candle produce a flame with a slight explosion.

Put a hollow or tubular match into if top of a Phial containing if ingredients producing hydrogen gas & hold over it a glass Receiver, an aqueous mist will be formed within. This say if Anti-Phlogistic is water formed by if union of hydrogen & oxygen - but D. Priestley & M. Walker think it is only if water within if Phial raised by if heat into a slight steam or vapor.





Limestone consists of a Calx combined with water & Carbonic Acid, which may be efflued by heat. The Calx is dissoluble in water, & then constitutes limewater; if this solution be acted upon by  $\frac{1}{2}$  pints of breath,  $\frac{1}{2}$  pints of calx is again converted into limestone, & in that state is precipitated to  $\frac{1}{2}$  pints bottom. This is illustrated by putting a little lime-water into a Siphon & blowing into it. It is owing to its rapacious affinity to fixed Air or Carbonic Acid,  $\frac{1}{2}$  pints of lime tears  $\frac{1}{2}$  pints of flesh from  $\frac{1}{2}$  bones of a carcass in 24 hours.

Lime consists of Spirit, water, & Tartar; fill a copper still with wine which heat with spirits of turpentine &  $\frac{1}{2}$  pints of water & spirit will be distilled & come away pellucid, leaving  $\frac{1}{2}$  pints of tartar which is  $\frac{1}{2}$  pints of colouring ingredient at  $\frac{1}{2}$  pints bottom.

Water containing an acid becomes milky by adding a solution of tartar;  $\frac{1}{2}$  pints of acid of sugar produces  $\frac{1}{2}$  pints of same effect where an alkaline quality is present.

## Geology.

The anchor on Seleucian Medals marks  $\gamma$ . coin to be struck at Antioch, where an anchor was found upon digging  $\gamma$ . first foundation of  $\gamma$ . City, tho' at a considerable distance from  $\gamma$ . sea. (Pinkerton's Medals. 1789. v. 1. p. 191.)

Seeing  $\gamma$ . organised bodies are found entombed in  $\gamma$ . hearts of rocky mountain:ains, it follows  $\gamma$ . these mountains must at one time or other have been in a soft state; &  $\gamma$ .  $\gamma$ . world was replenished by vegetables & animals of all kinds previous to  $\gamma$ . formation of these mountains

Now as all bodies owe their softness to  $\gamma$ . agency of fire or water, i.e. to fusion or solution; & as these mountainous masses had they been liquefied by fire w<sup>d</sup> have destroyed  $\gamma$ . texture of most bodies that occur in them & particularly loads which are found now alive, it

destroyed of texture of most bodies  
that occur in them & particularly

follows, y<sup>t</sup>. y<sup>e</sup>. liquefaction of these  
rocky mountains must have been  
effected by moisture & not heat.

From loads being found <sup>else</sup> embedded in  
shore gravels even under y<sup>e</sup>. sea,  
it seems y<sup>t</sup>. they must have been  
incarcerated & excluded from y<sup>e</sup>. air  
when in a torpid state, from which  
they cd. not recover without y<sup>e</sup>. admis-  
sion of air & an alteration of tempera-  
ture; & from y<sup>e</sup>. paucity of vegetables  
found in a petrified state, as well as  
from y<sup>e</sup>. description of those which are  
found, may we not conclude y<sup>t</sup>.  
y<sup>e</sup>. inundation took place here in the  
winter season?

The soil which is found upon secondary  
mountains which contain petrifications  
is of course post-diluvian. True. Is  
any part of y<sup>e</sup>. soil, antediluvian? If  
y<sup>e</sup>. deluge was general, a great deal of  
y<sup>e</sup>. present soil must be antediluvian.  
Otherwise had all y<sup>e</sup>. antediluvian soil  
been perished or been overwhelmed, all

## Geology

vegetable kingdom must have perished along with it, & no substitute left for those land animals which survived. We are told indeed by Scripture that Olive Trees &c escaped y<sup>e</sup> flood; nor was any provision made by Noah to preserve y<sup>e</sup> vegetable kingdom which it was intended therefore sh<sup>d</sup> not suffer in y<sup>e</sup> Catastrophe.

What proportion is there between y<sup>e</sup> antediluvian & postdiluvian coal in point of extent?

Most mountains have their strata horizontal as if formed by deposition; some however as y<sup>e</sup> Westmoreland mountains have their strata oblique & even perpendicular; has not y<sup>e</sup> position of such strata been effected by volcanos subsequent to their first creation?

'Tis said y<sup>e</sup> there are instances of pebbled bodies been found in oblique & vertical strata; if this be true

The said of these are instances of  
stratified rocks being found in all

It may confirm of opinion of such  
mountains have had their strata  
disordered by volcanos subsequent  
to of deluge. This must have happened soon after of  
deluge, else of soil wd. not have covered these mountains or it  
was generally said.

Marine animals are generally found  
in limestone: Diverse. Whether any  
land animals or vegetables are found  
in limestone? I have got a set of  
vegetables from Switzerland in a calcareous  
stone. vid. *Philos. 1. 422.*

From of very few land animals &  
vegetables found in a petrified state  
compared with of number of marine  
bodies, are we to conclude of what  
is land now was formerly of bottom  
of of sea; or of of bottom of of sea  
with its contents were projected by  
subterraneous force on of land where  
they now remain covering of former  
surface?

If all what is now land was formerly  
of bottom of of sea, whence did all  
of present vegetable & animal kingdom  
originate?

## Geology.

Had y<sup>e</sup>. order of y<sup>e</sup>. strata been y<sup>e</sup>. same  
all over y<sup>e</sup>. globe, however they might have  
been disturbed by convulsions, their formation  
might have easily been accounted for  
by repeated precipitation & deposition.  
but not only y<sup>e</sup>. order of y<sup>e</sup>. strata varies  
in every district & in every mountain  
almost, but y<sup>e</sup>. nature too of y<sup>e</sup>. genera  
of y<sup>e</sup>. earths varies. Doe y<sup>e</sup>. strata  
assume different appearances, <sup>at different times</sup> & do  
minerals as well as vegetables & other  
bodies pass from one state into another?  
Some stones we know perish in course  
of time, whilst other earths & substances  
are converted into stones by length of  
time. Stones grow, stones contain  
air, stones suffer decomposition, &  
then enter into new compositions.  
There is much analogy in all y<sup>e</sup>.  
works of nature.

It is possible y<sup>e</sup>. y<sup>e</sup>. Zoology, that of  
elephants &c y<sup>e</sup>. inhabitants of  
y<sup>e</sup>. warmer climates may have  
been transported into these northern  
climates by means of whirlpools

of warmer climates may have  
been transported into these northern

of subterraneous channels effected  
when the fountains of great deeps  
were broken up. B. Duane, whether the bones of the  
mammoth are not better for those of the elephant.

Rejecting this account, it must  
follow that the animals formerly  
inhabited these climates, which  
consequently must have been  
warmer than at present; it also  
follows that the vegetable kingdom  
must have been different in  
these climates from which  
Lithoxyla & Lithophylla seem to  
confirm; it also follows (unless  
each Island was originally replenished  
which neither revelation nor experience  
allow) that England & other Islands  
where the bones of Elephants & Rhinoceros  
have been seen must have been at one time  
contiguous & made part of two  
continents, but what time did the  
separation take place



## Geology.

If 4<sup>e</sup>. separation took place previous to 4<sup>e</sup>. deluge, then 4<sup>e</sup>. deluge was not universal; because had all 4<sup>e</sup>. earth been covered with water & thereby all Land animals destroyed by what means or 4<sup>e</sup>. retreat of 4<sup>e</sup>. waters did these islands get replenished with animals particularly of 4<sup>e</sup>. noxious kind as serpents, spiders, (snice) moths, & were 4<sup>e</sup>. eggs of these animals permitted to escape 4<sup>e</sup>. destruction: & live effects of water?

If 4<sup>e</sup>. separation took place immediately after 4<sup>e</sup>. deluge, this objection will still hold good.

If 4<sup>e</sup>. separation took place at a considerable period after the deluge then 4<sup>e</sup>. occupation of these islands may be accounted for.

It seems however wonderful 4<sup>e</sup>. Noah sh<sup>d</sup>. be able to collect 4<sup>e</sup>. inhabitants of all climates into 4<sup>e</sup>. ark as 4<sup>e</sup>. White Bear & Raven deer from 4<sup>e</sup>. Frigid Zone & 4<sup>e</sup>. Zebra &

ask as if white bear & hair deer  
from the Arctic zone & if there

if and bear from if. Torrid zone;  
& also proper food for them;  
if collected, nothing short of a miracle  
could preserve such different exotics  
in if. same atmosphere.

How were if. Crocodiles, if. Hippopotami,  
& all fresh water fish preserved?

As nature always operates by  
if. most simple & ready means,  
we may suppose if. either if. deluge  
was not universal at least at if.  
same time, or if. most of if.

organized bodies attached to different  
continents have been formed since  
if. general deluge; this process  
seems to require fewer miracles  
to accomplish it than if. other  
of collecting & preserving in or Ark  
all if. animals of every sort & country.

It seems to be a matter of uncertainty  
whether any carnivorous or noxious  
animal of any kind existed before if.  
deluge; have we any fossils of such  
animals? It seems if. more as well

## Geology

as all other animals was previous to 4<sup>th</sup> era gnamivorous; after however 4<sup>th</sup> deluge, these animals of a voracious & noxious kind were produced as well to prevent too great an increase of 4<sup>th</sup> other kinds, as to punish man for his impiety.

Some animals of an offensive nature may likewise have been created subsequent to 4<sup>th</sup> deluge adapted to 4<sup>th</sup> variation of climate which has succeeded 4<sup>th</sup> event or at least preceded it.

Have any Land animal exuvia been found in limestone strata; if not, are we to suppose 4<sup>th</sup> limestone mountains abounding with marine exuvia have been 4<sup>th</sup> bottom of 4<sup>th</sup> sea formerly; and are we to conclude 4<sup>th</sup> strata abounding with terrestrial exuvia have been antediluvian land; are marine & terrestrial exuvia ever found together imbedded; are they

found ever in alternate strata?

Are there any instances of a district of limestone containing pebbles <sup>isolated by</sup> surrounded by land (not touching of sea in any part) if there is & this surrounding land contains exoskeletons of terrestrial bodies then this district of limestone must have assembled by Caspian sea which is very unlikely or must have been collected by a deluge & not have been part of a subterranean ocean. An instance or two of this sort might superintend much light on a subject.

Had a chalk stone mountains been on a soft shale when a deluge was abating, the waters would have carried off with them a great portion of the calcareous earth & thereby left traces of their channels & caving; but a valleys among & near a chalk hills instead of

## Geology

being full of Calcareous earth  
are full of soil; as if y<sup>e</sup>. chalk  
hill had been in a solid state  
during y<sup>e</sup>. time of y<sup>e</sup>. deluge, & y<sup>e</sup>.  
soil only which covered them had  
been washed off. There is reason  
to think these mountains as well  
as all others were prior to y<sup>e</sup>. deluge,  
& y<sup>e</sup>. y<sup>e</sup>. extraneous fossils are y<sup>e</sup>.  
remains of y<sup>e</sup>. globe or planet  
out of which the earth was  
formed a part

As y<sup>e</sup>. greater part of y<sup>e</sup>.  
fossil shells belongs to y<sup>e</sup>. Nautilus  
or Polythalamian sort, as Nautilus,  
Ammonia, Belemnites, which from  
what we know of them are of y<sup>e</sup>.  
pelagian sort; may we not conclude  
y<sup>e</sup>. at first y<sup>e</sup>. sea prevailed &  
very deep, so as to be unfit for y<sup>e</sup>.  
abode of any but y<sup>e</sup>. pelagian  
sort. What reasons are there to

very deep, so as to be  
above of any but 4<sup>th</sup> pelagian  
suppose of such was 4<sup>th</sup> state of  
4<sup>th</sup> globe previous to 4<sup>th</sup> deluge.

Do 4<sup>th</sup> Pelagian fossils prevent  
themselves in 4<sup>th</sup> lowest strata  
in exclusion of others?

~~Supposing~~ however 4<sup>th</sup> Sea to have  
prevailed generally at first & 4<sup>th</sup>  
deep; 4<sup>th</sup> pelagian shells were first  
formed, & 4<sup>th</sup> as 4<sup>th</sup> sea grew shallow  
by depositions, exhalations, or absorptions  
below, 4<sup>th</sup> other shells were formed;  
4<sup>th</sup> afterwards islands or continents  
were formed as we see them, by  
volcanoes, by expansion of 4<sup>th</sup> globe,  
or by violent different depressions  
of 4<sup>th</sup> bed of 4<sup>th</sup> sea in some parts;  
how do we after all this account  
for terrestrial exuvie bees found  
imbedded subsequent to 4<sup>th</sup> formation  
of land?

The exuvie of land animal, which inhabits  
4<sup>th</sup> first islands were washed into 4<sup>th</sup> sea & contributed  
to 4<sup>th</sup> formation of other islands, or were deposited  
where we find them by 4<sup>th</sup> general deluge.

## Geology.

The most comprehensive system  
is this;  $4^{\circ}$  chaotic flood was  
peopled by degrees as it became  
more clear & shallow by shells  
fish adapted to  $4^{\circ}$  different periods;  
but this is not what I intended.  
I meant to say, That in best  
to suppose 'Sea & Land formed  
nearly at  $4^{\circ}$  same time agreeable  
to  $4^{\circ}$  Mosaic Account;  $4^{\circ}$   $4^{\circ}$ .  
Sea was peopled &  $4^{\circ}$  Land also  
by their respective genera;  $4^{\circ}$   
afterwards at  $4^{\circ}$  time & catastrophe  
called  $4^{\circ}$  Deluge,  $4^{\circ}$  Sea &  
Land nearly changed places;  
that was  $4^{\circ}$  bottom or bed of  $4^{\circ}$ .  
Sea forms on present land, &  
 $4^{\circ}$  land  $4^{\circ}$  was sunk down to  
 $4^{\circ}$  present level of  $4^{\circ}$  bed of  $4^{\circ}$ .  
Sea. On this principle we are  
to enter  $4^{\circ}$   $4^{\circ}$  mountains we now

See on this principle we see  
to enter of the mountains we now  
see were formed gradually by  
precipitation & attraction of composition  
at the bottom of the sea. see evidently in  
them of course all the shells fish &c  
which were above them of the sea were dead,  
of the sea. evidence of terrestrial bodies  
were such as had been washed  
away & carried by the reflux of  
the waters into the sea; of some  
places the bed of the sea was such  
as in the channel between the  
French & English coast; of the  
other parts of the land (which we  
may suppose to have been low) was

As the nature of Geology is abstruse &  
hidden as it were within the earth, are we  
to conclude of the knowledge of the formation of  
the earth was meant to be concealed from men?  
Are we to lay it down as a general  
proposition, of that which may be as well  
learnt by observing merely what nature  
has laid before us as by minute experiment  
of our own & laborious researches into  
the secrets of Nature?

It does not indeed appear notwithstanding  
our best information of the nature of things  
of which we entertain more just or more certain  
ideas of their formation than of such Philosophical  
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## Geology.

Amongst all  $\gamma$ . evolutions & convulsions  
which  $\gamma$ . strata seem to have undergone  
we never meet with any of mould or  
soil below  $\gamma$ . surface. Was  $\gamma$ . arid?  
= was earth as  $\delta$ . woodward thinks all  
discoloured by  $\gamma$ . water, previous to  $\gamma$ .  
changes which  $\gamma$ . strata experienced as  
appears by  $\gamma$ . shells &c? We must admit  
this or suppose  $\gamma$ . there was no aridness,  
= was earth, or partly  $\gamma$ .  $\gamma$ . convulsions  
of  $\gamma$ . globe were previous to  $\gamma$ . deluge.  
& formation of soil.

Is it true  $\gamma$ . Char fish is only to be found  
in one small lake in Cumberland, &  $\gamma$ .  
There are several species of fish only to  
be found in certain rivers; how came  
 $\gamma$ . Char in this lake, before or after  
 $\gamma$ . deluge? If before, then this part of  
 $\gamma$ . country had never have been  $\gamma$ . bottom  
of  $\gamma$ . sea, because 'tis a fresh water fish.

Was  $\gamma$ . inundating water fresh or salt?

The fossil shells as they resemble saltwater shells shew of  $\frac{1}{2}$  sea was salt. But there are also shells resembling fresh water shells, hence we may conclude of there was fresh water also - but I am not clear, as to this latter description,

Now if increase of water at  $\frac{1}{2}$  deluge if it was also salt must have destroyed all  $\frac{1}{2}$  fresh water fish; if it was fresh, guess whether  $\frac{1}{2}$  saltwater fish wd not have been destroyed. At least it remains to be proved of salt & fresh water can be mixed in so just a proportion as to serve as an element both for sea & river fish. This may be soon tried.

If no such mixture can be made, then  $\frac{1}{2}$  deluge was not general; or if general, animals have been created since to replenish  $\frac{1}{2}$  waters.

If such a mixture was made,  $\frac{1}{2}$  waters of many lakes & inland seas wd. have partook of this degree of saltness now, but we find them all either very salt, or very fresh.

## Cycology.

The existence of caverns & deep of the globe was not reduced to one soft mass which afterwards was hardened, concreted, or crystallized, without having experienced subsequent evolution. These caverns & deep glass <sup>masses</sup> must have been formed by simultaneous convulsions or earthquakes after the mountains had attained their present degree of solidity.

One remarkable feature in the globe is the gradual elevation of the earth from the sea to the centre of the land, by which means rivers are enabled to flow into the sea. How had the mountains being created by any sudden revolution or the inundation or universal volcano, not one river would have been able to have flowed into the sea. But among the thousands & millions of mountains which occur in most large islands & continents, a way is provided amongst them which thro' often circuitous conduits of fresh waters to the sea. Were it not for these passages, we should have lakes every where surrounded by mountains.

In discussing upon Geology, it seems  
best to compare yr. Inocai Account with  
facts; & sh. it be found unsatisfactory,  
to suppose some other hypothesis - Whitchamit  
which makes yr. globe to have been covered  
with water, & in yr. state inhabited by  
fish, & afterwards divided into land &  
water by volcanos or central fire seems  
to be very probable but perhaps not  
comprehensive enough. Inure, are there  
any fresh water fish found petrified? v. p. 103.

It must be remarked yr. yr. hypothesis of  
Whitchamit does not account for wood  
be being found petrified. But I believe  
I mistake him; he thinks islands were  
first formed gradually, & enlarged, previous  
to yr. grand eruption or deluge, which was

If yr. petrifications in mountains were  
caused by yr. deluge, then yr. mountains  
which crisscross them were created by yr.  
deluge; but how d. yr. deluge make moun-  
tains like yr. Dutch ones, or how d.  
it act so regularly in to beam channels  
from yr. very bottom of them gently declining  
to yr. sea.

## Geology.

Whether is it more probable of  
y<sup>e</sup> isthmus between France & England  
was depressed by a fall of y<sup>e</sup> ground  
of y<sup>e</sup> globe, or y<sup>e</sup> y<sup>e</sup> two continents  
were surounded by a subterraneous expansion  
under each?

The rents in y<sup>e</sup> mountains & between  
adjacent islands in which y<sup>e</sup> strata  
are similar on both sides but have  
different levels shew that convulsions  
have taken place since y<sup>e</sup> strata were  
formed & since y<sup>e</sup> inundation which  
prevalled because these strata is divided  
have many of them petrifications.

Altho y<sup>e</sup> strata do not take place  
according to their specific gravities, yet  
y<sup>e</sup> being formation of them seems to be  
y<sup>e</sup> result of a regular slow process  
& not of an accumulating undulating  
either cause as a deluge.

As it generally happens that a  
great number of shells or of y<sup>e</sup> same  
genus occur in y<sup>e</sup> same place, ex. gr.  
y<sup>e</sup> corals & occur in y<sup>e</sup> Derbyshire  
marble; we may infer y<sup>e</sup> there

Shells have never been removed  
far from the place of their existence.  
For had they been brought from a  
distance by an irregularly sweeping  
torrent they wd. have been entirely  
destroyed & mixed with various other  
masses of a terrestrial & marine  
origin.

As most of the <sup>pelagic</sup> shells found in our sea  
are inhabitants of the Indian Ocean,  
Iq, whether the pelagic shells found  
in the vicinity of that sea correspond  
with the shells inhabiting the temperate  
zone.

Q. Whether the great change which has  
been effected in the climate has been gradual  
or violent.

The Society Islands in the Pacific  
Ocean seem at present to enjoy the  
same temperature which one would  
in this northern ocean; their plants  
seem to correspond with those we find  
pelagic here. Their mountains are  
covered with fern, the principal vegetable  
whose exuvia occur here. They have not  
above 3 or 4 sorts of animals, which  
may account for the paucity of  
terrestrial bodies imbedded.

## Geology.

I once thought that fishes might have most of them been created since 4<sup>th</sup> deluge, & again that no poisonous animals existed before 4<sup>th</sup> deluge; but both those ideas vanish at 4<sup>th</sup> sight of so many shark teeth as are found here especially in 4<sup>th</sup> strata of Shepsh. Crocodiles also have been dug up here.

It is difficult to account for 4<sup>th</sup> preservation of both sea water & fresh water fish at 4<sup>th</sup> time of 4<sup>th</sup> deluge; we must admit 4<sup>th</sup> 4<sup>th</sup> sea-water fish only existed before, & that 4<sup>th</sup> freshwater fish have been created since.

The small quantity of fish bones found imbedded shew that they did not suffer at 4<sup>th</sup> deluge.

Supposing this climate originally enjoyed 4<sup>th</sup> temperature of southern countries, as all 4<sup>th</sup> extraordinary fossils vegetable, marine, & terrestrial confirm; did any reason at that <sup>time</sup> labour under 4<sup>th</sup> disadvantages of a frigid zone, or even of our present climate. Was there a frigid zone? & if there was, were 4<sup>th</sup> seas or lands under it inhabited by animals? What records have we of

such animals as were first introduced, were  
of the same kind, and it is established by

such animals as vegetables as are  
peculiar to the present time? Perhaps  
the same Deer is of the present one.

It seems extraordinary that whilst  
we are continually meeting with the  
fossil bones of elephants, mammoths,  
& perhaps mammoth; we meet with none  
of horses, oxen &c. There are proofs  
again of a change of ordigious animals.

If there was not originally a peopled  
zone, where were derived the vegetables  
& animals ex. gr. of reindeer & white bear  
which now inhabit it. This compels  
us again to resort to a subsequent creation.  
Unless indeed it can be proved (which is not  
improbable) that the present vegetables  
& animals have assumed different charac-  
-teristics from what originally peopled  
the consequence of the effect of cold. It is  
different of Newfoundland Dog is from an  
Polish Greyhound or a French Lap Dog, &  
yet it is of the species of some naturalists  
as Mr. J. Hunter that all dogs are  
spring from the wolf.



## Geology.

As most if not all vegetable petrifications resemble no vegetables now in existence, we may safely attribute their origin to an arid illuvial era. The same observation extends to many animal petrifications - This is a mighty proof that there existed a world prior to this, producing vegetables & animals different from the present.

It is most probable y<sup>t</sup> America was divided from Europe & Asia in the days of Peleg - The cause denounced against the earth for Adam's transgression imparts y<sup>t</sup> noxious weeds are of a subsequent creation. Might not y<sup>e</sup> earth have experienced some of those violent changes which disfigure it at y<sup>e</sup> same time? & might not various noxious animals have been created then also?

The days spoken of dening y<sup>e</sup> creation of y<sup>e</sup> world must have been periods of 24 hours, <sup>not years</sup> because 'tis said the evening & y<sup>e</sup> morning were y<sup>e</sup> 1<sup>st</sup> day & so on. This is further proved from y<sup>e</sup> oblate spheroidal figure of y<sup>e</sup> earth. If y<sup>e</sup> temperate zone included the Arctic circle as some imagine in order to account for y<sup>e</sup> inhabitation of elephants crocodiles &c, where did y<sup>e</sup> white bear y<sup>e</sup> residence & other animal as well as vegetable inhabitants of y<sup>e</sup> rigid zone enjoy existence?

I have an ul. pelvisid. In q. mountains  
of Carre of League from Meadwick, q.  
vertebra of a crocodile 30 feet long, was  
found in a stratum of sandstone. The  
remains of a crocodile were also found in  
a stratum of stone at Bleasheim & are now  
in q. possession of q. learned Mr. Bayart.  
An impression also of a crocodile was found  
in limestone at Ashford by Mr. St. Watson of  
Bakewell. Whitcomb. p. 29. & 184.

As most of q. bones found in this country  
are of q. granivorous class, are we to conclude  
q. the carnivorous animals formerly existed,  
to be remembered q. few if any carnivorous  
animals exist in N. America, whose exvoid  
& those of Britain bear a great affinity.

As to few if any exvoid of horses & oxen  
occur, are we to conclude q. such  
animals did not exist at all, or q. they  
did not exist in the present soils, or  
that being domesticated were driven  
by their proprietors upon the approach of  
the deluge to places of greatest security,  
where they were overwhelmed with  
their owners. This seems probable, because  
we do not meet with human bones no  
more than those of a domestic genus.

### (Non) Minerals

Minerals sh<sup>d</sup>. be divided into primary & secondary ones. All metals, crystallized substances, & minerals are secreted from strata of limestone &c. & therefore may be denominated secondary minerals, or mineral secretions - We find no such substance as wax in <sup>the bodies of</sup> men, trees, or flowers, yet such substances are secreted, so lead ore, spec, &c. are secreted from stony strata, tho' the particles of those bodies do not present themselves in y<sup>e</sup>. strata. In fact no such particles do exist in y<sup>e</sup>. strata, the minerals being created where found on the surface of the strata by means of secretory ducts analogous to the various glands of animal bodies.

It is evident from Whitehurst's own work altho' his inferences are many of them founded on a supposed general uniform order or series of strata that no such order exists in different places. For instance in Derbyshire millstone Grit produces neither minerals nor exuvia. But in Sweden & Norway minerals are found in fissures of quartz, as stone which enters into the composition

minerals are found in figures of quality  
stone which value is the same

of millstone - Shale he says p. 189  
in Derbyshire contains neither animal  
nor vegetable impressions; but p. 54 he  
observed a stratum of shale at Clayby  
in Lincolnshire great number of  
fossil shells, & nodules of ironstone. The  
shale was covered by a stratum of  
limestone, which in Derbyshire lies  
under of shale - He says too that altho  
Shale is not productive of Minerals, a mine  
near Egan contained a vein of lead ore -  
Again, altho he lays it down as a general  
rule that coal is not found under limestone,  
yet he acknowledges an instance to the  
contrary at Etruria & Little Foston near  
Newcastle in Staffordshire. 1793.

1. Stratum. Ratahile or fragments of Stone
2. Limestone 1 foot thick, devoid of ~~vegetable~~ <sup>figured stones</sup>
3. Sand - 4. argillaceous stone - 5. Brick - 6. Coal

This stratum of limestone the author however  
thinks a recent production, which is rendered  
probable by 4- adartitious beds of sand & gravel.

It will seem that copper ore also is found  
in a mine ~~near~~ (shale I expect) near  
Gurdon in Staffordshire. p. 227.

# Minerals

106

Whinstone. In y. county of Chester  
fragments of blue quartzose stone seem to  
be universally mixed with y. soil.  
Whitehurst. p. 65.

In many parts of N. Wales the strata  
are quartzose or whinstone; a substance  
similar to rounded pebbles & boulders, observed  
in the gravel & sand pits in Cheshire &  
other parts of England, where no such  
strata appear. ib. 244.

In y. vicinity of Maghera (N. of Duleek)  
I observed many fragments of lavas  
& also of bluish quartzose stone called  
whinstone. p. 246. (vid. Millstone p. 112)

The generality of these fragments are  
rounded as if were by attrition, or stones  
upon a sea beach, & are usually applied  
to the paving of streets, public roads &c.  
I have observed a stratum similar to  
the above in N. Wales, & upon the  
River in Shropshire. p. 65. supra.

Two miles east of Bally Castle  
the following strata commence. 1. Whinstone  
a quartzose substance, which strikes fire  
with steel. (v. infra 125)

2. The 7<sup>th</sup> stratum is coal - the 10<sup>th</sup> whinstone  
& y. 14. Millstone Grit. p. 260

NB. The Whinstone is coal - the 10<sup>th</sup> Whinstone  
is a 14<sup>th</sup> Whinstone - ditto.

Some Whinstone in some instances are  
so similar as to render them not easily  
distinguishable from each other. *ibid.*

The above stratum of whinstone vicum:  
= bent on coal is y<sup>r</sup>: only instance of  
y<sup>r</sup>: kind which has occurred to me.

[*ibid.* Millstone. 112)

*ibid.*  
v. infra. 125.

At Lady (Ireland) a vein of lead ore has  
been discovered in a stratum of Whinstone.

[*ibid.* Millstone. 112)

Whitcl. 264.

Whinstone, Basalt, & other stones whose hardness  
is such that their angular parts will scratch glass,  
& whose fracture exhibits an appearance so shag  
like that which might be expected from sand ball  
whipped, as to have induced many to suspect them  
of volcanic origin from strata of considerable  
importance from their hardness & quantity in most  
coalmines. They are called cochle in Cornwall,  
& sturdy in y<sup>r</sup>: north of Scotland.

Parkinson's *Byzantia* remains. 1. 170.

## Geology.

Whitehurst never observed any asteria in y. Derbyshire strata; this may intermixed with y. soil. p. 241.

Not fossil bones & teeth, imbedded together with y. shells of fish in limestone <sup>anywhere</sup> strata. p. 57.

Not recollects ever seeing any univalves in Derbyshire limestone. 184.

And whenever bivalve shells <sup>occur</sup> with their valves entire & close, they form select classes. p. 53.

But where y. beds of shells consist of several species, y. valves are not united, but separated from each other as fragments. p. 54.

Not are y. traces of terrestrial animals often if ever found intermixed with marine shells in limestone strata; nor are marine ones often if ever found in argillaceous strata containing vegetable impressions. p. 57. t. 204.

Not did he observe any marks of sand that were totally free from extraneous bodies, or other marks of y. sand having been deposited by y. flowing of water. p. 215.

It is almost evident from inspecting  
the section of the Derbyshire Strata, those  
particularly where leadstone occurs,  
that the Limestone strata had been  
fissured i.e. divided in a vertical direction  
previous to the introduction of leadstone  
not only between its laminae but  
into its fissures which it frequently  
occupies. It is evident also from the  
appearance of the Strata at Darley Hill  
(p. 120) & also where the strata on  
each side of a fracture are nearly  
parallel, as well as from the strata  
at Mallock High Tor where they are  
considerably deformed on one side of  
the fracture, that these fractures  
were subsequent to the introduction of  
leadstone between the strata. It also seems  
from leadstone outcrops at Carleton that  
the argillaceous & coal strata incumbent  
on Millstone Grit have been destroyed since  
or perhaps at the eruption & ascension of  
the volcanic leadstone. It is probable  
that the fractures (which bisect all the  
several strata) were produced by the same  
convulsion which dejected the strata  
incumbent on Millstone. E



## Geology.

Some of Deluge

Many bones have been deposited by Shere, ex. gr. a human skeleton with British beads &c. were dug up in a stone quarry near St. Paul's Widening St. at Blackney in Lincolnshire. Whitehurst. 29

Human bones & armour with human corn & fibula &c. were found in a stone pit in St. Paul's Widening St. at Sturston in Worfolk. Hid- & Baddar's abdy. Phil. Trans. 0.6. p. 144.

Whitehurst observes that altho' shell fish are generally deposited in limestone, bones & teeth of fish occur rarely, but chiefly in gravel & sand. p. 44.

But on St. Paul's Widening St. mountains of Carver (say Sturton) there is a bed of white laminated stone containing between the lamina various fish compressed like fern. ibid.

Ston. The teeth of sharks, have been taken out of a rock in Sturton Widening St. park, near Malton (which I expect is limestone as also the preceding). Louthorp. A. P. T. 0.2. 228.

See page 103 for examples of crocodiles found imbedded in limestone.

Instances too there have been of toads found in limestone.

But we may infer from the observation of Whitehurst as well as others that either shell fish were created before others, or that being left alive they were

rather than the soil were sealed before others, & that being left alone they were

buried & eroded in the place where they existed. And this might be the case before as well as at 4<sup>th</sup> deluge.

As lakes frequently occupy 4<sup>th</sup> places of mountains & towns absorbed by earthquakes (see Port-royal Demeris page, mountain near Port-morant there, S. Euphemia Isle of Malta 1536, Pico in the Induces, vid. Whitehurst. p. 72) we may suppose that these waters are occasioned by 4<sup>th</sup> low situation of the places, or that there is water beneath the crust of the sea, which is the more probable because were the interior of the earth solid, how could mountains & towns subside? There certainly are large caverns below 4<sup>th</sup> surface.

The eruptions of volcanoes will account for the formation of many solid strata, as well as for detached fragments of the strata being found - also for alternate layers of soil & lava - also for beds of gravel, sand, &c - Inundations or rather alternate occupations of sea & land will account for layers of soil, gravel, for 4<sup>th</sup> remains of towns, anchors, & remains of arts.

## Minerals.

112.

Millstone-gate. A coarse sandstone composed of granulated quartz & quartz pebbles. The former retain of sharpness of fragments newly broken, & latter are rounded by attrition as stones upon a sea beach. This stratum [Derbyshire] is not productive of any fossil exoskeletons, or minerals. The quartz pebbles contained in this stratum indicate of preexistent state of a quartzose stratum, as often is found in those of lime-stone.

The quartz pebbles above are white, & colour of quartzose stones from whence produced is various, as black, brown &c.

They are in common use for paving streets, & are frequently variegated with seams of white quartz running thro' them. They are of common gravel stone of Nottingham & Staffordsh. Derbysh. Cheshire &c.

We are told of quartz minerals in Norway & Sweden are contained in quartzose stone.

Whitehurst. p. 182

[vid. Whinstone. p. 106]

Between the laminae or beds of Millstone-gate at Lincoln Hill near Colbrooke Dale Thropshue intervene small strata of quartz pebbles. Whiteh. p. 209. PIII.

Silicious Grit, Sandstone lies in beds,  
differs much in texture & color within a few  
miles, (is composed of Quartz, & Feldspar or pebbles  
with Mica & Steatite) being of different textures,  
it is useful for Millstones, Buildings, Filtrating  
Cisterns &c.

It contains vegetable impressions (vid sup. con.)  
& Iron like ~~of~~ iron argillaceous Grit strata.  
Galena & other minerals are found in it; but  
this circumstance may probably be owing to  
Dykes.

No more petrifications have as yet been observed  
in this stratum: Toads have been taken out of  
it solid rock at great depths, they have also been  
found in limestone blocks, but all dead immediately  
on being exposed to the air.

Granite is found in detached round fragments  
of different sizes in all parts of Derbyshire, but  
where they came is not yet ascertained.

W. Watson Derby. Strata

Minerals. 114.

Shale or Shiver. A black laminated clay, much indurated. It contains neither animal nor vegetable impressions, & is not considered as productive of minerals, as lead ore, & spar &c. The stratum contains ironstone in nodules, & sometimes stratified. The springs issuing from it are of the chalybeate kind; e.g. one near y<sup>e</sup> bridge at Duxton one at Guarndon, & another beyond Matlock bridge towards Chatsworth.

Whitehurst. 183.

The laminae of limestone are separated by shale or shiver, y<sup>e</sup> upper layers are a good black, take a fine polish, & are there called Black marble. The lower layers are rather brown, as may be observed in y<sup>e</sup> rocks at Matlock High Top.

Whitehurst. p. 184

[N.B. The author is speaking of the limestone stratum in Derbyshire which lies under the Shale & Millstone strata. vid. p. infra. 124]

Lead ore has been found in shale at Shaw. Engine near Eyam, & copper ore near Guarndon in Staffordshire. but these are rare instances.

Whitehurst. p. 227.

See p. infra. 125. where Shale is same as Tell.

Shale or slate clay, called also Shiver black metal, & bleas, & by the northern collectors blae or till, often forms the roof of coal mines.

Shale is slate clay, called also freest  
black metal, & blags & by the northern nations  
black metal, & blags & by the northern nations

It is generally bluish, blackish, or of a reddish  
gray. It is frequently marked with impressions  
of vegetables [vid. supra contra] It breaks into  
long thin shivery laminae, & is very friable. It  
readily resolves, on being immersed in water, or  
even on exposure to the air, into a fine soft clay.  
It contains a portion of bituminous matter, & generally  
has an unctuous feel, on which account the  
Scottish colliers term it creesy blaes or greasy blaes.

Parkinson. Org. hem. 1. 170.

Note. When Whitehurst asserts that Shale has no  
vegetable impressions, he is then speaking of the Shale  
Strata which intervene between Millstone Grit &  
Limestone, see p. 120.

Shale, Shiver, Aluminous Stratum is a black  
argillaceous earth, of a laminated texture, which  
easily cleaving when exposed to y. air, decomposes to clay.  
This Stratum is productive of alum, & abounds with  
nodules of Ironstone, some of which contain Petroleum  
in a fluid state; Asphaltum is found in thin beds;  
also balls of ochre & clay.

The shale is generally a cover for y. veins of ore &  
accounted a good criterion; small veins sometimes  
occur but they seem to be formed from y. ochre veins in  
y. adjacent Limestone Stratum.

W. Watson's Derbyshire Strata

It is remarkable that at y. Basement or coming up  
of y. Shale to y. surface & consequently <sup>at y. points</sup> where  
y. Limestone appears there are springs of warm  
water, namely at Honey Giddleton N. 63° Far.  
M. A. H. S. 68°, Bakewell E. 55° & Burton W. 53°  
Den.

vid. p. 120

119

## Geology.

Grays or Definitions or Elements or Principles

1. The sea encroaches on y<sup>e</sup> land in places.  
 cor. The question of y<sup>e</sup> sea being y<sup>e</sup> same, y<sup>e</sup> land must encroach on y<sup>e</sup> sea in places.
2. The land encroaches on y<sup>e</sup> sea in places.  
 New islands, coral reefs, sandbanks, disintegration, alluvions.
3. Lakes & seas are formed sometimes by earthquakes.  
 Earthquakes produce lakes & seas.  
 Volcanic - Coasts of England & France.
4. Volcanoes produce earth & mountains.
5. The mountains were once fluid.
6. Some mountains made fluid by fire.
7. Others by water.
8. Mountains are primary or secondary.
9. Minerals are all secondary productions or secretions, & decompositions.
10. Petrifications their different origin.
11. Stones their different origin.
12. Fossils & corals their origin.

12. *Fagus excelsa* *Thunbergii*.



Toadstone. a blackish substance, very hard; contains bladder holes, like y<sup>e</sup>. scoria of metals, or Iceland lava, & has y<sup>e</sup>. same chymical properties of resisting acids.

Some of its bladder holes are filled with spar, others in part, others quite empty. This stratum is not laminated but consists of one entire solid mass, & breaks alike in all directions. It does not produce any minerals, or figured stones, nor contains any adventitious bodies.

This stratum (No. 4.) is known by different names; at Malloch & Winsten, toadstone, & blackstone; at Money ash & Tidenwell, charnel; at Carlleton, cat dirt; at Ashover, black clay.

Stratum (No. 6. p. 120) free from bladder holes as may be seen at Money near mine near Winsten.

Stratum (No. 8. p. 120) yet more solid, or may be seen at Hubbard all Mine, near Money ash.

Toadstone has no corresponding fissures to those of limestone, & frequently fills up y<sup>e</sup>. fissures in y<sup>e</sup>. strata underneath it.

The stratum of clay lying under No. 6 is apparently burnt, so as not to be distinguished from y<sup>e</sup>. burnt clay or Hey Nor Common, which was burnt by a stratum of coal or fire underneath it.

Whitehurst. 184. 196.

At Tinswell Moor, yr. leadstone or channel  
as there called has been dug 100 fathoms  
deep & no bottom found. On seven other  
mines its thickness is as follows.

	fathoms
Black Hillsch -	100 not cut thro'
Heath Bush -	16
St. Andrews -	2
St. James -	12
Constant -	7
Calvestone -	7
Burkirk -	7
Chap-maiden -	19

This shews that leadstone is extremely  
variable in its thickness. Whitehurst. p. 195.

The spar contained in yr. bladder holes implies  
not only a solution of spar, but also that  
leadstone is pervious to water thus impre-  
gnated descending from the incumbent  
strata; for whenever the leadstone laps  
or has no incumbent strata, the bladder  
holes are void of all extraneous substances.  
Whitehurst. p. 228.

120

Strata

between Grange Mill & Darley Moor.  
Derbyshire.

No.			compare p. 124.
1.	Milstone-Gaet.	421	
		120.	
2.	Shale or Chioer.	120	
3.	Limestone.	50	
4.	Toadstone.	16	
5.	Limestone.	50	
6.	Toadstone.	46	
7.	Limestone.	60	
8.	Toadstone.	22	
9.	Limestone. not cut thro.		

All the strata from No. 3. to. 9 are divided by others called clays, or way-boards. from 1 to 6 feet thick. Their color is lightish blue, with a small tint of green & they all contain pyrites & spar in small nodules; and it has <sup>p. 114.</sup> been observed by Mr. Geo. Fifeington of Leicester that all the springs flowing from them are warm like those of Buxton & Matlock. These clays are calcareous, & may therefore be classed with marles.

Of these strata. No. 2. appears <sup>or extends</sup> on the banks of G. Derwent; No. 5. in Taques pasture; No. 4. 5. 6. 7. & 8. on Boral Moor.

Agass No. 2. appears on  $\frac{1}{2}$ °. valleys of  
Hakewell, Ashford, & Cattleton, & otherwise  
on Mam. Tor - No. 4. forms  $\frac{1}{2}$ ° summit  
of a mountain near  $\frac{1}{2}$ ° Castle & is called  
Cal-dit. No. 5. forms  $\frac{1}{2}$ ° surface of Money  
ark Moor, particularly at  $\frac{1}{2}$ ° Marble quarry  
& likewise at Hubbardale Mine - The quarry  
on  $\frac{1}{2}$ ° Money Ark Moor near  $\frac{1}{2}$ ° road between  
that town & Hakewell produces the Derbyshire  
Marble

The strata No. 4. & 6. are not universal  
they not exist in Hubbardale Mine, near  
Money Ark. nor Hayworm mine, on  
Bosial moor - nor at High cake near  
Tideswell.

Whitehurst. 182-187, 188. P. 1.

By  $\frac{1}{2}$ ° Clays or Way boards the thickness  
of  $\frac{1}{2}$ ° other strata are ascertained, which  
would otherwise be difficult as the limestone  
beds consist of various laminae.

Whitehurst. p. 187.

The strata at Malloch High Tor on each  
side which are similar to those at Grange  
mill, cover a great convulsion; for the  
beds on Mason mountain are elevated about  
100 yds above Malloch High Tor, & yet  $\frac{1}{2}$ °  
same beds are depressed 100 yds below the  
foot of it at the river. Whitehurst. p. 189. P. 11

Strata. 122.

At Alpeton Common.

1. Clay	7-0
2. Katchee, fragments of stone.	9-0
3. Bind. indurated clay	13-4
4. Stone, argillaceous concreted clay.	6-0
5. Bind	8-8
6. Bind	25-0
7. Stone, black color.	5-0
8. Bind	2-0
9. Stone	2-0
10. Bind	5-0
11. Bind	5-0
12. Coal	1-6
13. Bind	1-6
14. Stone	23-0
15. Stone	14-0
16. Bind	7-0
17. Small, a black substance like coal dust	9-0
18. Bind	2-0
19. Stone	20-0
20. Bind	16-0
21. Coal	7-4
	<u>184-4</u>

# Strata at Well Hallam.

See with

- 1. Clay -- 7. 6
- 2. Sand -- 28. 0
- 3. Smeath. 1. 6
- 4. Church - 4. 0 indurated clay
- 5. Sand - 9. 0
- 6. Stone - 2. 9
- 7. Sand - 1. 0
- 8. Stone - 1. 0
- 9. Sand - 9. 0
- 10. Stone - 1. 0
- 11. Sand - 16. 0
- 12. Shell - 2. 0
- 13. Sand - 12. 0
- 14. Shell - 9. 0
- 15. Church. 5 1/2. 0 Stone & sometimes carb.
- 16. Soft coal. 4. 0
- 17. Clay - - - 0. 6
- 18. Soft coal. 2. 0
- 19. Church & Sand. 21. 0
- 20. Coal - - - 1. 0
- 21. Sand - - - 1. 0
- 22. Heavy Sand. 25. 0
- 23. Coal - - - 6. 0

Is observable that Mr. Whitcomb altho he distinguishes fine clay from sand & sand from Church yet in page 202 of his work says that clay, church, & sand are synonymous terms.

222.3 Whitcomb. p. 212

Strata. 124.

at Bagillt - Flintshire - N. Wales.

1. Gravel, sand &c.	<sup>ft.</sup> 45.0
2. Shale	9.0
3. Freestone argillaceous	6.0
4. Shale not generally.	1.0
5. Coal	6.0
6. Clay	9.0
7. Freestone argillaceous	44.0
8. Coal	2.9
9. Freestone	90.0
10. Shale	15.0
11. Coal	15.0
12. Clay	12.0
13. Shale	6.0
14. Freestone	1.6
15. Clay much indurated	9.0
16. Shale	1.6
17. Blackish Stone containing shell fish	1.0
18. Coal	9.0
19. Shale	1.0
20. Clay or Sand	20.0
21. Freestone	40.0
22. Coal	5.0

The shells in No. 17 show that the incumbent strata are  
more recent than 4. Shale or 9. Freestone that covered.  
Therefore a testimony of 4. pre-9. Freestone operations of 4. time

Whitehurst. p. 242.

Strata.

two miles east of Balley Castle. Ireland

1. Whinstone, v. sup. 106.
2. Freestone, argillaceous
3. Till, shale, or laminated clay.
4. Stone, argillaceous
5. Till or shale.
6. Freestone, argillaceous.
7. Coal  $\frac{1}{2}$  ft 6 inch. thick.
8. Stone, argillaceous
9. Till or shale.
10. Limestone, brown, no shells.
11. Sand, indurated clay.
12. Stone, argillaceous.
13. Strata not ascertained.
14. Millstone grit. Whitehurst

Tablet of Derbyshire Strata by White Watson.

1. Argillaceous Grit.
2. Argillaceous Ironstone.
3. Coal
4. Siliceous Grit.
5. Shale.
6. Limestone
7. Sandstone.
8. Limestone
9. Sandstone
10. Limestone
11. Sandstone
12. Limestone



# Geology

126.

It may be laid down as pretty certain principle, y<sup>t</sup>. y<sup>t</sup>. formation of stony strata is y<sup>t</sup>. effect of aqueous deposition, & y<sup>t</sup>. y<sup>t</sup>. transposition or obliquity of them is the effect of igneous earth quakes.

Volcanoes as well as earthquakes are produced by subterraneous fires; but by the former fresh eruptions of matter are produced, by y<sup>t</sup>. latter the existing strata are only dislocated.

Earth, sand, & gravel may be the consequence either of deposition or volcanoes.

Lava is certainly of volcanic origin.

The observations made on toad stone seem to prove y<sup>t</sup>. y<sup>t</sup>. original strata were disturbed when this substance issued from the bowels of the earth. But perhaps on second consideration this may not be a conclusive inference; it being evident from the great fractures y<sup>t</sup>. the strata were most violently disturbed subsequent to the introduction of toad stone. &

The sediment found at the bottom of Tea-Indies as well as the tuffaceous depositions in the neighbourhood of petrifying springs prove the possibility of strata being formed by water.

d. The intrusion of fused basalt  
between the strata might occasion their  
fractures.

## Decomposition.

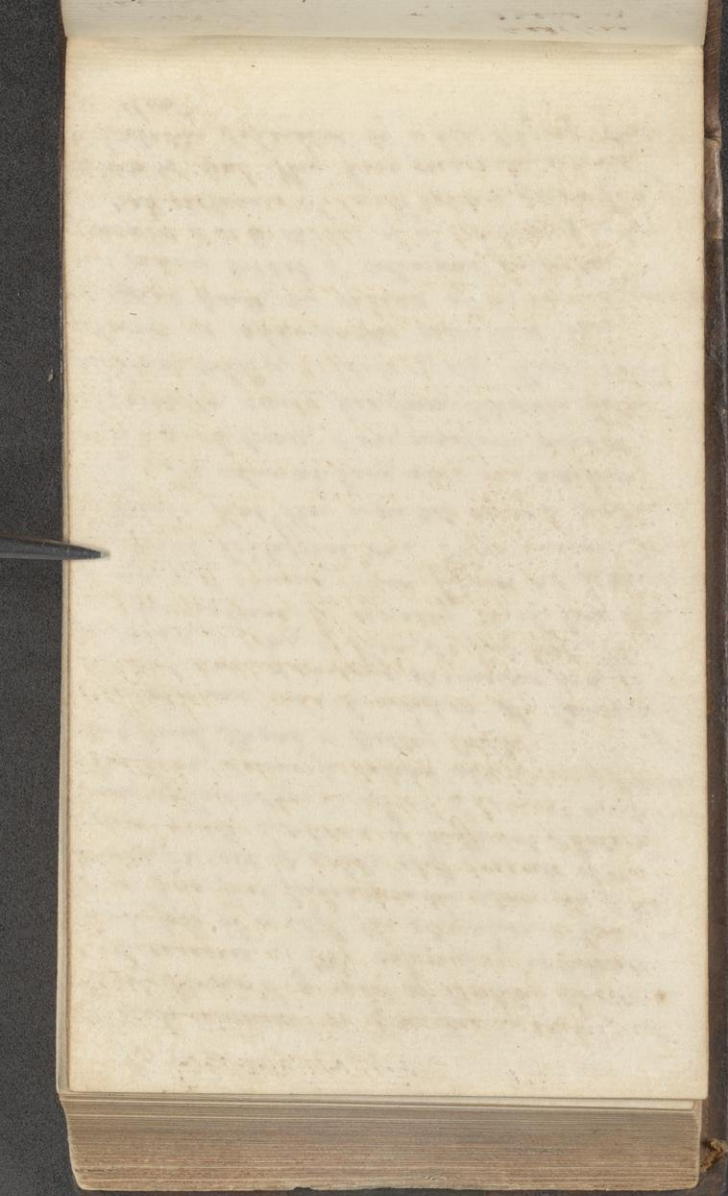
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Whitehurst has observed many instances of dog tooth spar being so uniformly incrustated by calamine as to preserve  $\gamma^c$  true crystalline figure of  $\gamma^c$  spar.

Also instances of the spar thus incrustated being wholly, or only in part decomposed & replaced by calamine still retaining  $\gamma^c$  external figure of  $\gamma^c$  spar - Crop Lake mine, or Shallock High Tor abound with specimens  
Whiteh. 22g.

rottenstone (found 2 miles W. of Bakewell in D. of hullards' lands) is not stratified nor generated in fissures of limestone strata, but is found as broken fragments of stone, imbedded in vegetable soil from  $\gamma^c$  surface of  $\gamma^c$  earth to  $\gamma^c$  depth of 18 inches - It is accompanied with numerous nodules of chert, a silicious substance, generated in limestone strata - Chert & rottenstone being thus blended together manifestly imply the dissolution of a limestone strata in which  $\gamma^c$  chert was generated & that rottenstone is nothing more than  $\gamma^c$  fragments of limestone reduced to a calx.  
Whiteh. 22h.

Many nodulous masses of chert contain the impressions of entrochi; which evidently shew that  $\gamma^c$  entrochi existed anterior to the chert.



Basaltic - Giants causeway 100

Giants causeway is  $\frac{1}{2}$  miles east of Port  
Kush Strand &  $\frac{1}{2}$  west of Balley Castle  
with measure. The causeway is situated  
at  $\frac{1}{2}$  foot of a cliff, the elevation whereof is  
5 or 600 feet perpendicular above the Atlantic  
Ocean; & yet  $\frac{1}{2}$  whole cliff consists of one  
entire mass, composed of different & state of  
lava, apparently  $\frac{1}{2}$  effect of several eruptions.  
This lava seems to extend all  $\frac{1}{2}$  way between  
Port Kush Strand & Balley Castle.

The columns which compose the Causeway  
are vertical, of various diameters from 15 to  
20 inches, & some of them 30 feet long. They  
are prismatic or equally thick from end  
to end, & of various sided figures as pentagonal  
hexagonal, heptagonal &c. These various forms  
indicate that they were not owing to crys-  
tallization. The columns have also one uniform  
color, a dark brown, & one uniform density,  
apparently quite free from bladder holes  
altho'  $\frac{1}{2}$  cliff is replete with them. Each  
column is apparently divided into  
unequal parts, by means of transverse joints  
but many consist of columns partially,  
leaving 5 or 6 inches of  $\frac{1}{2}$  central part solid.  
In some instances  $\frac{1}{2}$  joints extend quite thro'  $\frac{1}{2}$   
columns but they have originally united  
& probably separated by water lodging & freezing  
in them.

The articulations are not flat, but convex & concave, exactly fitted together, but not in any order with respect to the convexity or concavity being upwards or downwards; for in many instances they have been observed in both directions.

From these & other circumstances it seems probable that these columnar bodies were originally in the heart of the mountain & in a state of fusion; which on cooling from must have contracted considerably & gradually. Mr. R. Raspe on extinguished volcanoes says that basalt is commonly found in the vicinity of them.

Mr. Fay Faugas on the same subject produces instances of lava being found near craters wholly & in part columnar - and that in some of the corresponding faces of these prismatic columns are inserted fragments of granite evidently broke & separated by the contraction of the lava on cooling.

That basalt is lava is evident also from the observations of Rev. Dr. Gole in his tour thro' Iceland; for this great island is allowed to be composed principally of lava, & yet various parts thereof abound with basalt columns. Mr. Hodges in his tour thro' India, says, that a cliff called Moutagena is wholly composed of basalt, & that under the cliff is a spacious cavern, which he examined by torch light & observed the same columnar appearances

Basalt & Giant Causeway

as on y<sup>e</sup> exterior thereof, & likewise that charcoal was imbedded in the solid substance of y<sup>e</sup> stone throughout the whole extent of y<sup>e</sup> cavern.

At the foot of y<sup>e</sup> cliff near the Causeway Mr. Whitehurst found a piece of iron ore similar in appearance to the Cumberland ore, oxidized on one side, which is some testimony of y<sup>e</sup> cause whereby some lavas may be magnetical, & likewise that the substances supposed to be lava, have also been in a state of fusion.

Two miles N. of Coleraine on y<sup>e</sup> road to Portrushward he hit a stratum of lava in which a pit had been dug to repair the road. The upper part of y<sup>e</sup> stratum was become totally decomposed & reduced to vegetable soil; tho' it still retained its original dark color. The soil was succeeded by small fragments of lava, y<sup>e</sup> magnitude whereof increased downward to the bottom of the pit, where y<sup>e</sup> stratum became one solid mass. This lava contains many bladder holes either wholly or in part filled up with Zeolite a substance rarely found in England. In some of y<sup>e</sup> holes not completely filled up may be observed y<sup>e</sup> crystallized figure of the Zeolite namely cubical very similar to fluor.

These crystallized bodies show up they  
were not enclosed in y<sup>e</sup> lava while it was  
liquid fire, having been evidently exuded  
thru y<sup>e</sup> substance of y<sup>e</sup> lava; consequently  
lava seems equally as porous to a solution  
of resole, as sandstone or limestone is to a  
solution of mineral substances.

It also has much y<sup>e</sup> appearance of Fluor-  
both as to color, figure, transparency, & hardness.  
Nor will it effervesce like fluor in acids,  
tho' it is said to be soluble in them.

Whitehurst. 248. H.

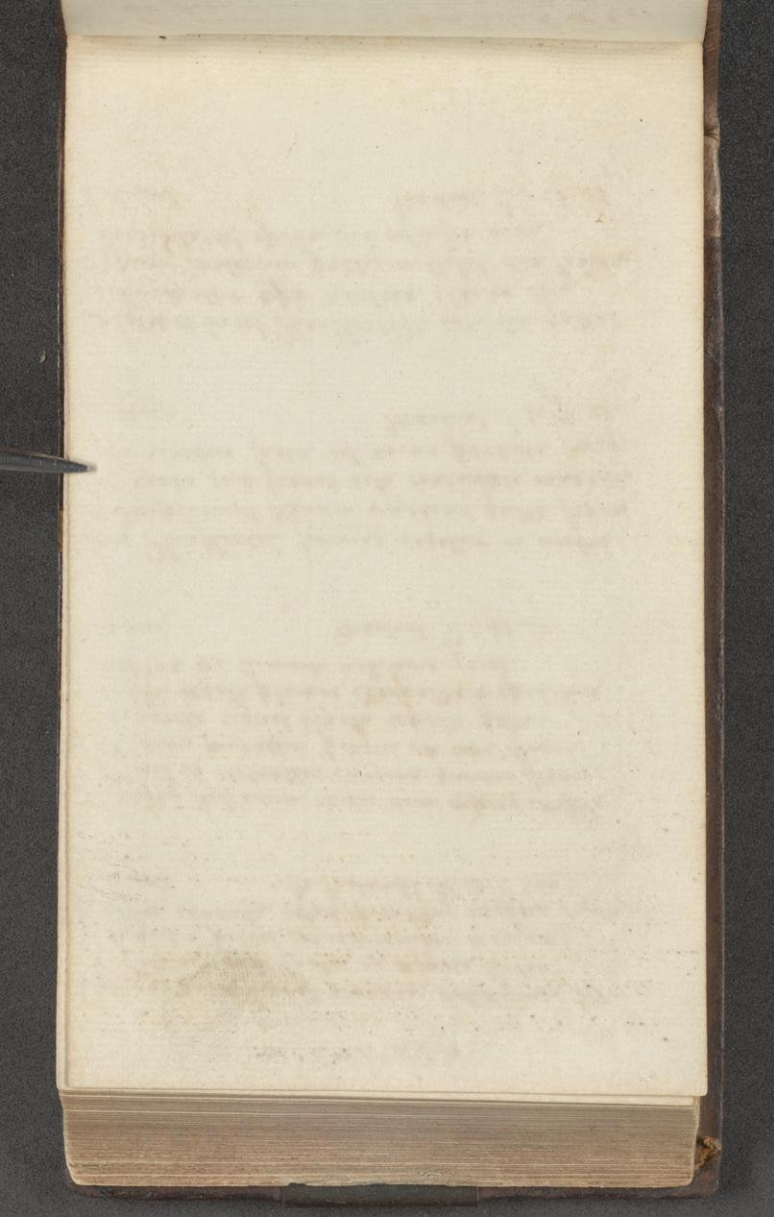


## Medicine.

Spts of Turpentine. Dec. 29. 1804. I took 12 drops of Spirits of Turpentine in a glass of water at bed time, being a little afraid & experimentally inclined, but perceived no other effects than a strong violet scum imparted to y<sup>e</sup> urine. <sup>in addition to the</sup> In the morning I took 20 drops in a glass of water, but had no evacuation whatsoever until evening when I had a small discharge of urine scented as before. The scent continued for

24. whether this medicine might be useful in dropsy or dia- <sup>- Scurvy?</sup>

NO. Spts of Turpentine diluted a little is most efficacious of any thing yet found out in taking away the dental pains occasioned by Hydragyrus. 24. whether it will remove y<sup>e</sup> common tooth-ach.



Geolog. Extract.

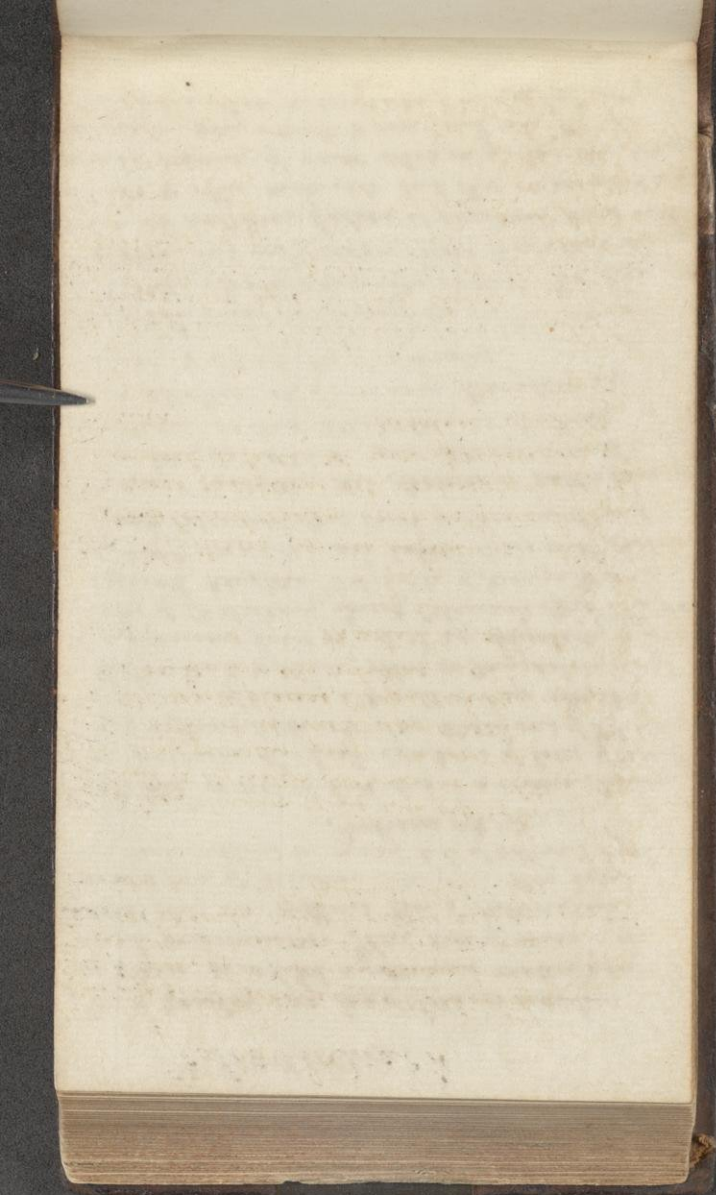
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Vidi ego, quod fuerat quondam solidissima tellus,  
Esse flentem. Vidi factis ex æquore terras;  
Et procul a pelago conchas jacuisse marinas;  
Et octas inventa etiam montibus arborum summis.  
(Persius) Co. Inl. ampl. lib. 15. l. 262.

Flexibus Æliadum carnis dum vipera serpsit,  
Fluxit ex obstantem succina gümma feram;  
Dud, dum miratur, perquisi sibi esse teneri,  
Concreto riguit vincta repente gelu.  
Ne tibi regali placeas, Cleopatra, sepulchro,  
Vipera si tumulo nobiliore jaces.  
(Persius) Martial. l. 4. ep. 59

Sum Pheathontea formica vagatur ex umbra,  
Implicuit teruam succina gutta feram.  
Sic, modo quæ fuerat vitæ contempta maxente,  
Funeribus facta est nunc pretiosa suis.  
(Persius) Martial. l. 4. ep. 15.

Et latet, et licet Pheathontide condita gutta  
Ut videatur apud nectare claua suo.  
Dignum tartorum pectus tulit illa laborum;  
Credibile est ipsam sic voluisse mori.  
(Persius) Martial. l. 4. ep. 32.



# Toadstone.

138.  
from 118.

Trap, rowley rag, toadstone, or washer are stones in which argillaceous matter very much predominates. These from y. masses in which they are diffused thro' y. subterraneous regions, & from y. peculiar structure they perhaps have been supposed by many, to be of volcanic origin.  
Parkinson. D. K. 149.

Toadstone is not in beds, but is a certain stratum of an irregular thickness composed of balls of various sizes, of different textures & colors, totally void of Petrifications; some appearing a Basalt or Trap abounding with Hornblende & Olivin; others, an Amygdaloid, having an argillaceous basis in which are irregularly dispersed nodules of Chalcedony, Quartz, Calcareous Spar, Steatite, Greenish, Keystone (Athyrolite of Kenner) & some of y. Quartz nodules are hollow lined with Crystals of Quartz, Calcareous Spar, Cauck & other substances.

Besides Toadstone this stratum is called Charrel, Blackstone, Cat-dirt &c and appears to be the substance called Whinstone in Scotland, Max delictia in Germany, Basaltus in Ireland, & Trap by y. Swedes.

The veins containing ore in the crumpled Limestone strata generally descend thro' the Toadstone, but are seldom found to contain ore when in toadstone. Galena is sometimes found with Blende & other minerals, but this circumstance rarely occurs; y. veins when in y. toadstone are generally very small & composed chiefly of Calcareous Spar accompanied by Asphaltum;

generally very small & composed chiefly of  
Calcium that accompanied by silica;

when they descend into g. Limestone under  
g. sandstone, they become thicker again &  
frequently as rich in ore - Mountain leather  
sometimes is discovered in their veins, the very  
rarely.

Tools, as g. Heads of Hammers & Axes are found  
made of sandstone, used by g. ancient Inhabitants  
of the County [Derby]

Walton's Derbyshire Strata. p. 21.

## Geology.

The dock yard at Woolwich has been visited by Mr. Churchman accompanied by Mr. Whitby in making observations on the gradual rising of y<sup>e</sup>. tide waters of y<sup>e</sup>. Thames. There were made at this place one whole year's observations commencing on y<sup>e</sup>. 4<sup>th</sup> May 1761 & according to y<sup>e</sup>. register, they were continued during the arrival of our most Gracious Queen at Harwich y<sup>e</sup>. 8. September. The orders from y<sup>e</sup>. Admiralty were to send an account of y<sup>e</sup>. spring & neap tides at this port, after keeping an exact account of y<sup>e</sup>. perpendicular rise from low to high water in each tide, taking notice of y<sup>e</sup>. course of y<sup>e</sup>. wind each day of y<sup>e</sup>. month & whether morning or evening tide. The cell at y<sup>e</sup>. bottom of y<sup>e</sup>. single dock gates was a fixed mark, & y<sup>e</sup>. depth y<sup>e</sup>. water ebbs below it, was added to y<sup>e</sup>. height y<sup>e</sup>. next tide flowed above it, which gave y<sup>e</sup>. perpendicular flow of each tide. Now it is ascertained y<sup>e</sup>. since y<sup>e</sup>. year 1761 there have been 2 new sets of dock gates, on account of y<sup>e</sup>. old ones going to decay; y<sup>e</sup>. first was said to be somewhere about y<sup>e</sup>. year 1766 & y<sup>e</sup>. last in 1783. On paying particular attention to y<sup>e</sup>. cell of y<sup>e</sup>. dock gates at low water it did not appear above surface, but a considerable piece was observed under water, but above y<sup>e</sup>. cell. And altho' y<sup>e</sup>. difference

between high & low water remains nearly  
4<sup>th</sup> same, both high & low water marks  
are about 4 feet perpendicular higher than  
was 4<sup>th</sup> case in 1761; which remains to be  
confirmed by corresponding observations to be  
made in different parts of 4<sup>th</sup> United Kingdom.  
But as there is not any effect without a  
cause, it is hard to dispute 4<sup>th</sup> effect while  
cause remains something of a mystery; but it  
appears rather extraordinary 4<sup>th</sup> what follows  
was inserted in 4<sup>th</sup> voyage of 4<sup>th</sup> celebrated  
Capt. Cook's voyage vol. 3. p. 293 when on the N.W.  
Coast of America. "Mr. Whitty states, 4<sup>th</sup> on his  
2 last excursions, several places were seen  
where 4<sup>th</sup> ocean was evidently encroaching very  
rapidly on 4<sup>th</sup> land, & 4<sup>th</sup> 4<sup>th</sup> low borders extending  
from 4<sup>th</sup> base of 4<sup>th</sup> mountains to 4<sup>th</sup> sea-side, had  
not at any very remote period of time produced  
tall & stately timber, as many of their dead  
trunks were found standing erect, & still  
rooted fast in 4<sup>th</sup> ground at different stages of  
decay, those being 4<sup>th</sup> most perfect that had  
been least subject to 4<sup>th</sup> influence of 4<sup>th</sup> salt  
water by which they were surrounded every  
flood tide - The stumps in some instances at low  
water mark were even with or below 4<sup>th</sup> surface  
of 4<sup>th</sup> sea - Similar observations were made by  
him at Port Chalmers, P. Williams Sound,  
Cook's river, & Gray's Harbour."

London Chronicle. Dec. 27. 1804

x. this effect may perhaps be owing to 4<sup>th</sup> bed of 4<sup>th</sup>  
river being raised by depositions. Now.



Argillaceous Grit.

142.

Lies in beds, is equable & rough to y<sup>e</sup> touch, has  
when first broken an earthy smell, differs much  
in texture & color within a few miles; is chiefly  
serviceable in building & staving, in either of  
which cases, y<sup>e</sup> stones must be placed in y<sup>e</sup> same  
direction as when taken from y<sup>e</sup> rock, otherwise  
they cannot separate & perish in a few years.

This stratum is productive of no mineral but  
Iron, which is observed in caves of shaded ground.  
No mineral Petrifications appear in it, vegetables  
do, chiefly unknown species of *Arundo* & *Euphor-*  
*bia*.

There is generally found under this Gritstone  
where y<sup>e</sup> coal abounds, a thin stratum of a  
compact, silicious matter abounding with  
impressions of an unknown vegetable; this  
stone is called Garnister, Crowstone, Carth,  
Callierde &c & is much used for repairing  
y<sup>e</sup> roads.

W. Watson Derby. *Strata*.  
vid. p. 124

*[Faint, illegible handwriting on aged paper, likely bleed-through from the reverse side of the page.]*

Argillaceous Ironstone. 114.

Is destitute of metallic lustre but weighty, attracts water like Clay, & by torrifaction yields to  $\gamma$ . attractive power of  $\gamma$ . Magnet.

In some places it forms a regular stratum, abounding with petrified mussel shells, incumbent upon which is a ferruginous calcareous clay in curious ramifications, whilst in other places it bears oblate spheroids imbedded in clay.

These nodules are of various textures affording different sorts of Iron; when broken they sometimes contain vegetable & animal Impres-  
sions probably of unknown species. & frequently septaria of Spang Ironstone which is some-  
times crystallized with Galena, Blende,  
& Iron-pyrites.

v. p. 124.

W. Watson Dubyn. Strala.

*[Faint, illegible handwriting throughout the page, likely bleed-through from the reverse side.]*

Principles or Elements  
or  
Introduction to Geology.

Should be composed in the manner of  
Euclid's Elements. First begin with definitions  
of Earth, Minerals, Fluids, Rocks, Mountains,  
Hills, Rivers, Seas, Lakes, Layers, Strata,  
Veins, Fossils, Fire, Metals, Salts, Crystals,  
Nodules, Subidence, Precipitation, Volcanic,  
Alluvial, Detrital, Organic bodies, Forests.

Then arrange propositions in regular succession  
& discourse upon them as, That the Sea  
gains upon the Land in places, That the  
Land gains upon the Sea in places. That  
some Mountains are produced by earthquakes  
or Volcanos. That Mountains were once  
fluid &c. That some Mountains are more  
ancient than others. That Minerals are  
subject to decomposition & transition. &c.

Consult Baccus's Geography for definitions  
& Facts. Keil's Cyclopaedia. Linnich. Gronov.  
Jansson. Lavrier. Hutton. Parkinson.  
Whitehurst. De Lue. Saussure.

What if the work was divided into books  
with Definitions prefixed to each. Suppose  
one on Mountains, 2<sup>d</sup> on Minerals & Mines.  
3<sup>d</sup> on Petrifications. 4. on <sup>Veins</sup> Minerals.

57c on Mountains, 2 on Minerals & Metals.  
on Agriculture. 1. a. 1841.

THE ...



## Definitions

The Earth is that Planet which we inhabit.  
this should be omitted

A Solid is a dry Substance.

A Fluid is a humid Substance.

An Earth is a solid.

Water is a fluid.

Land is a tract of earth ~~uncovered by water.~~

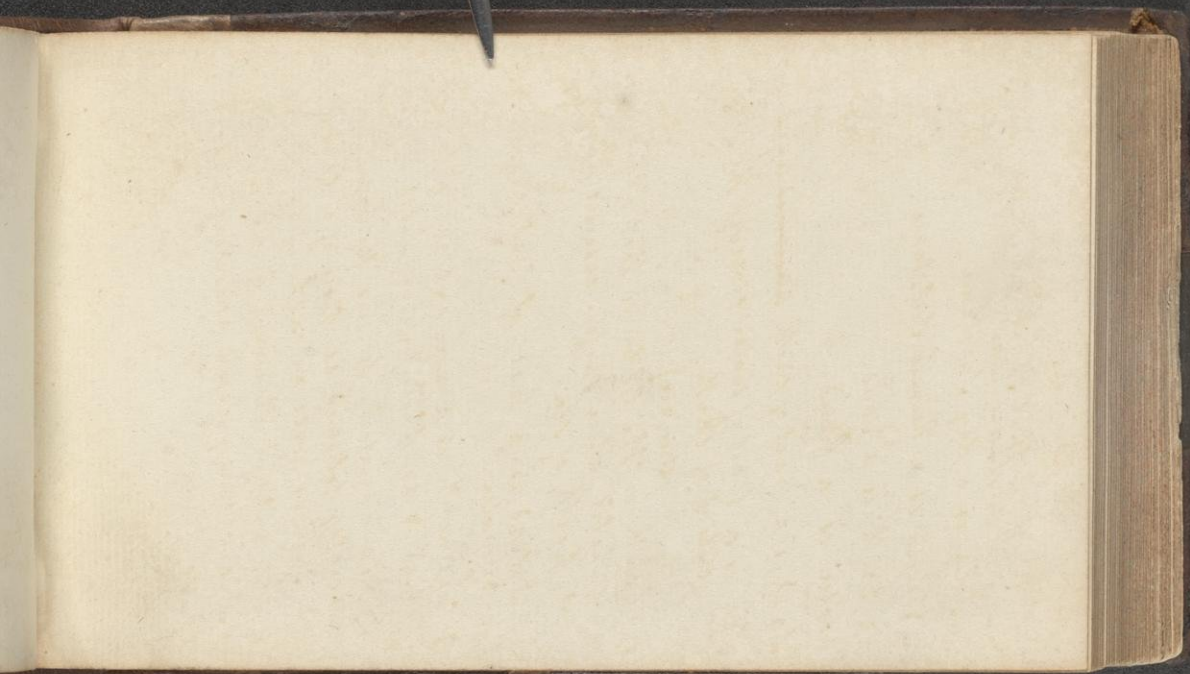
Sea is a tract of water covering the <sup>same</sup> extent of earth.

The Earth or The Globe is that planet which we inhabit, whose <sup>(surface)</sup> surface consists partly of land partly of sea.

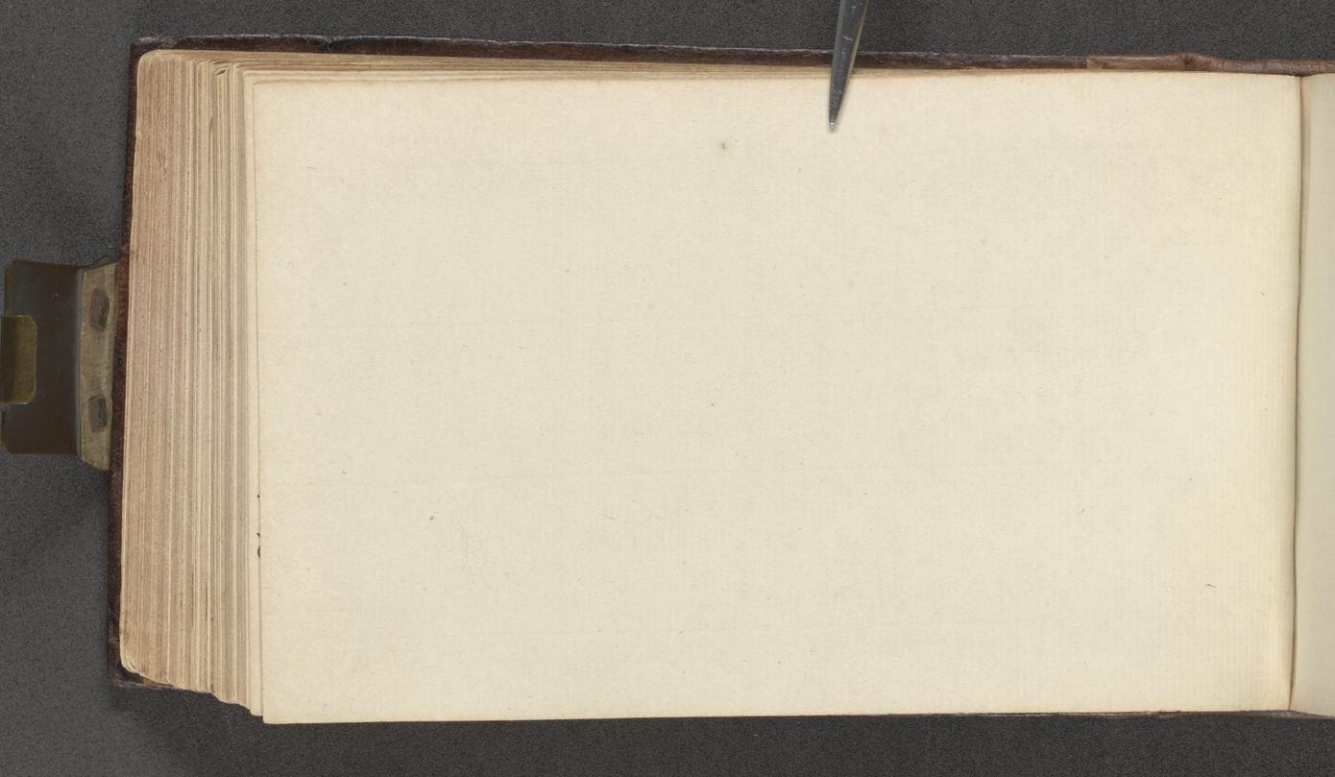
An Island is a tract of land surrounded by sea or water.

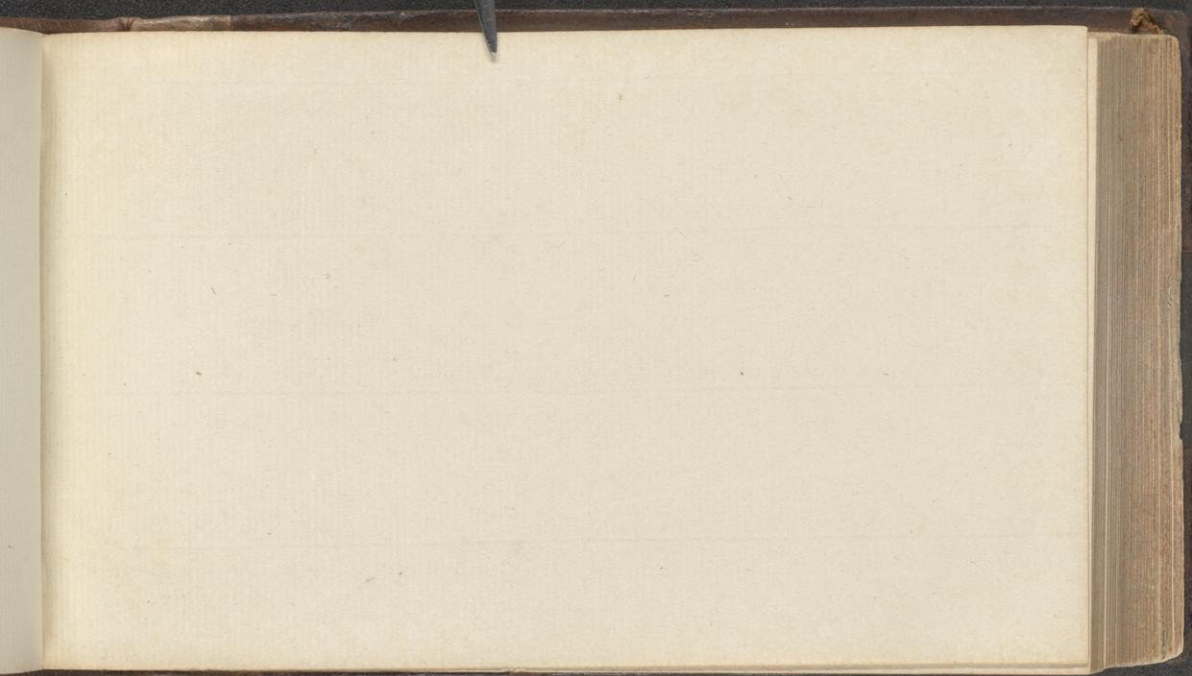
A Continent is a more extensive tract of land & not wholly <sup>explicitly</sup> surrounded by water.

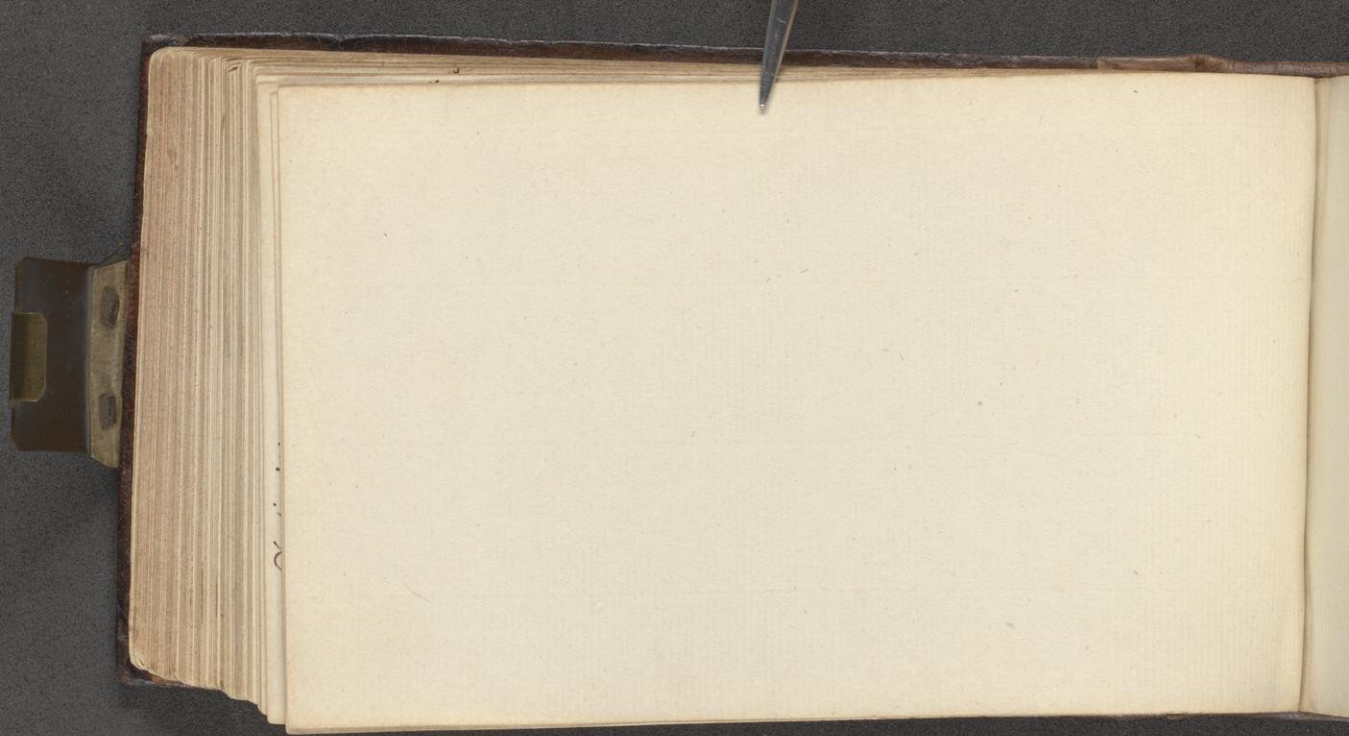
(It is not necessary to use the definitions generally followed as those of a lake & sea, but such as will be adjoined to in the work; and yet it would be as well to explain every term according to its common sense.)

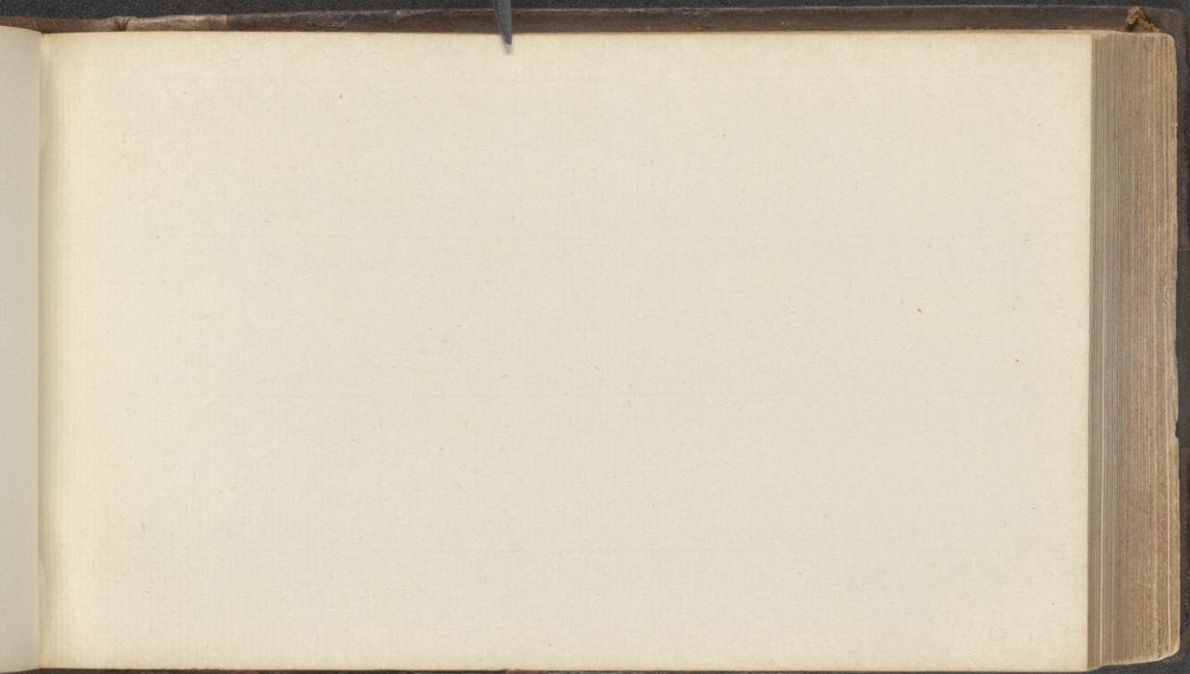








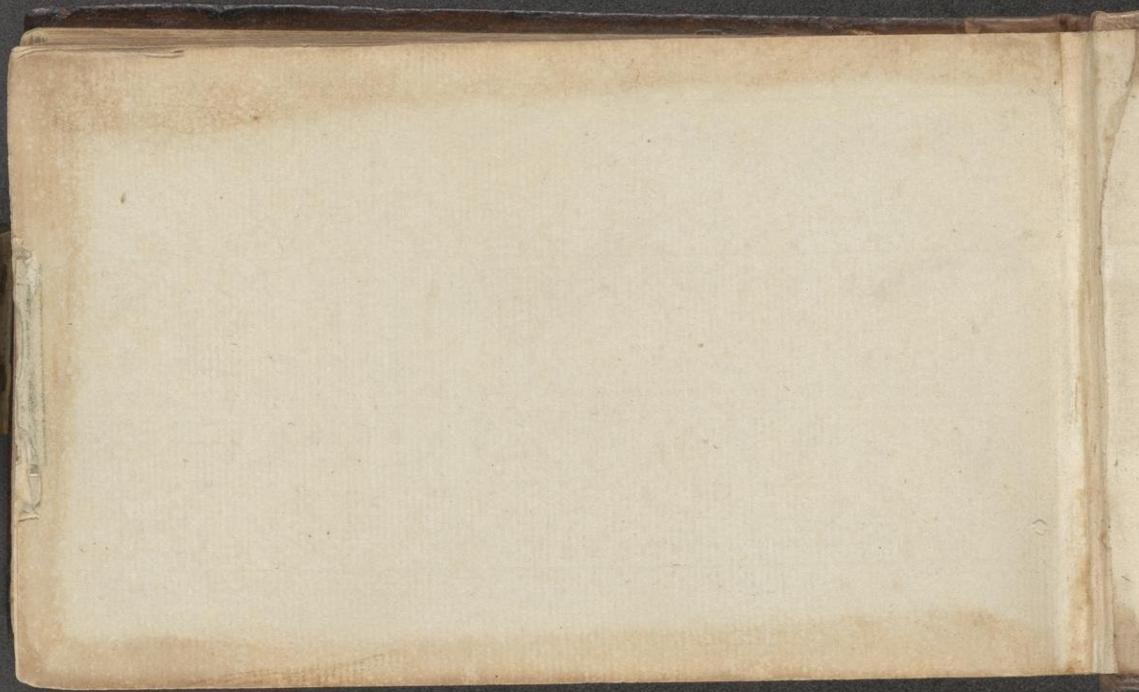




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