

Treatment of cheese processing wastewater by ridge and furrow disposal - nitrogen transformation. [DNR-023] 1985

Boyle, William C. (William Charles), 1936-; Doran, Frederic J. Madison, Wisconsin: Wisconsin Department of Natural Resources, 1985

https://digital.library.wisc.edu/1711.dl/KAA3RICJ66PUE9A

http://rightsstatements.org/vocab/InC/1.0/

For information on re-use see: http://digital.library.wisc.edu/1711.dl/Copyright

The libraries provide public access to a wide range of material, including online exhibits, digitized collections, archival finding aids, our catalog, online articles, and a growing range of materials in many media.

When possible, we provide rights information in catalog records, finding aids, and other metadata that accompanies collections or items. However, it is always the user's obligation to evaluate copyright and rights issues in light of their own use.

Wisconsin Groundwater Management Practice Monitoring Project No. 23



GROUNDWATER Wisconsin's buried treasure

Wisconsin Department of Natural Resources



Treatment of Cheese Processing Wastewater by Ridge and Furrow Disposal-Nitrogen Transformations (Study No. 26)

Investigators:

Title:

Principal Investigator

William Boyle, Professor University of Wisconsin-Madison Dept. of Civil and Environmental Engineering

Graduate Research Assistant

Frederic J. Doran University of Wisconsin-Madison Dept. of Civil and Environmental Engineering

average of 14,000 gpd of processing wastewater.

Objectives:

This project was undertaken to determine the nitrogen transformations in wastewater from two dairy products industries as it percolated from the furrows to the groundwater. Ridge and furrow land treatment effectiveness was evaluated under various soil and loading conditions. Operation, maintenance and accuracy of the monitoring equipment used were also studied.

A ridge and furrow land treatment system consists of a series of ditches

which allow for the distribution, infiltration and treatment of wastewater. Two ridge and furrow systems were studied: a cheese factory in Brodhead, Wisconsin which discharged an average of 39,500 gallons per day (gpd) of wastewater, and a creamery in Mindoro, Wisconsin which discharged an

Background/Need:

Methods:

Groundwater monitoring wells and lysimeters were installed and soil grab samples taken during the initial soils borings. Flow composited influent wastewater samples were collected monthly. Furrow samples were taken during intensive sampling periods at Brodhead in October and Mindoro in November of 1984. Samples were also collected routinely from the wells and lysimeters.

Wastewater, furrow, lysimeter, groundwater and stream samples were analyzed for biochemical oxygen demand (BOD₅), chemical oxygen demand (COD), total suspended solids (TSS), total dissolved solids (TDS), chlorides (Cl⁻), total Kjeldahl nitrogen (TKN), ammonium nitrogen (NH₃-N), nitrate and nitrite nitrogen (NO₃-N+NO₂-N) and pH. Soil and plant samples were also analyzed. Other monitoring included observation of load/rest cycles, reading monthly groundwater and surface water elevations, taking monthly 30-day average wastewater flow readings, cutting periodic grass samples during the growing season to determine nitrogen uptake and performance of infiltration studies to determine unsaturated zone flow rates.

Wastewater nitrogen loss was attributed to denitrification and leaching at both sites. Plant uptake was also a factor for Mindoro. Both Brodhead and

Mindoro had average BOD_5 loading rates over the 100 lb/acre/day Department of Natural Resources limit. COD was greatly reduced as wastewater infiltrated into the groundwater. The nitrogen content of the

Results:

- 55 -

wastewater at both sites was mainly in the organic form. It mineralized to ammonium nitrogen in the settled solids which accumulated in the furrows. The ammonium-nitrogen was oxidized to nitrate-nitrogen as it infiltrated through the unsaturated zone.

Wastewater treatment and disposal in a ridge and furrow system was influenced by wastewater distribution and infiltration, load/rest cycling, winter operation and annual cover crop burning. Grass overgrowth and leaky header gates caused poor wastewater distribution at Mindoro, though this was not a problem at Brodhead. The Brodhead system experienced decreased nitrogen concentrations in the groundwater and improved soil aeration and infiltration due to a short load/rest cycle. Ponding resulted in part from a longer load/rest cycle at the Mindoro system. Annual grass burning enabled a modest nitrogen loss at both locations. Winter operation proceeded adequately at both sites, though Brodhead fared better during subzero temperatures.

Downgradient groundwater concentrations of contaminants were impacted to a greater extent at Brodhead than Mindoro. Nitrogen and COD reductions in the unsaturated zone were similar at Brodhead and Mindoro, though a greater percentage were removed at Brodhead than Mindoro. This difference was attributed to sandy soils at Brodhead which allowed for faster unsaturated travel times than silty loam soils at Mindoro.

Investigators concluded that nitrogen losses around the unsaturated zone were attributable to denitrification at both sites. BOD₅ tests indicate that wastewater loading did not produce an oxygen demand high enough to inhibit denitrification. Nitrogen and COD reductions were dependent on infiltrative capacities. The nitrogen in wastewater applied at both sites was mainly in the organic nitrogen form, which ammonified and eventually diffused into the overlying furrow wastewater. Dissolved ammonium was the primary form of nitrogen in the wastewater applied to the furrows. Surface water remained unaffected from the operation of the ridge and furrow systems at both sites.

Further research is suggested to better determine the impact of loading changes on groundwater quality and to better quantify unsaturated flow times by the installation of tensiometers. Nitrogen loading rates should be met by dischargers to reduce or maintain groundwater nitrogen concentrations. Solids accumulation in the furrows at Brodhead should be reduced with wastewater pretreatment. Chloride concentrations in the Brodhead wastewater should be reduced by brine removal in the plant or prior removal. Annual spring grass burning is suggested for all ridge and furrow systems where feasible. A downgradient well nest should be installed at Mindoro to better define the movement of contaminants off-site. Also suggested is an improved lysimetry method to obtain a more instantaneous sample and allow for winter sampling.

Conclusions:

Recommendations/ Implications: Availability of Report:

This report is available for viewing and loan at:

- 57 -

The Water Resources Center 1975 Willow Drive Madison, WI 53706 (608) 262-3069 Publication 050858

Key Words:

Ammonium-nitrogen, nitrate-nitrogen, ridge and furrow disposal system, wastewater

Funding:

The Wisconsin Department of Natural Resources provided funding for this project through the Groundwater Management Practice Monitoring Program which receives appropriations from the Groundwater Account.

TABLE OF CONTENTS

Chapter		
1	Introduction	1
2	Review of Literature on Nitrogen Removal by Land Treatment	
	A. Introduction	4
	B. Mineralization and Immobilization	10
	C. Nitrification	12
	D. Denitrification	14
	E. Dissimilatory Reduction	19
	F. N Gas Fixation	20
	G. Plant Uptake	22
	H. Volatilization	23
	I. Ammomium Adsorption	24
•	J. Leaching	27
	K. Case Study Summary	28
	L. History of Ridge and Furrow Treatment	30
	M. Ridge and Furrow Design Concerns	31
3	Materials and Methods	
	A. Materials	38
	B. Field Methods	45
	C. Analytical Methods	48
4	Brodhead Results and Discussion	
	A. Site Description	51
	B. Wastewater Chemistry	54
	C. Wastewater Hydraulic Loading	58
	D. Organic Loading Rates	62
	E. Groundwater Elevations and Flow	63

Chapter

napte	r		Page
	F.	Groundwater Chemistry	67
	G.	Unsaturated Zone Flow Rates	83
	H.	Furrow and Lysimeter Chemistry	90
	I.	Grass Uptake of Nitrogen	116
	J.	Sugar River Chemistry	124
	K.	Site Observation	124
	L.	Nitrogen Budget	126
5	Mir	ndoro Results and Discussion	
	A.	Site Description	129
	в.	Wastewater Chemistry	134
	с.	Wastewater Hydraulic Loading	1 <u>3</u> 7
	D.	Organic Loading Rates	140
	E.	Groundwater Elevations and Flow	142
	F.	Groundwater Chemistry	146
	G.	Unsaturated Zone Flow Rates	161
	H.	Furrow and Lysimeter Chemistry	171
	I.	Grass Uptake of Nitrogen	179
	J.	Severson Creek Chemistry	182
	K.	Site Observations	182
	L.	Nitrogen Budget	184
6	Comparitive Discussion of the Brodhead and Mindoro Ridge and Furrow Sites		
	A.	Nitrogen Budget	186
	в.	Wastewater Organic Loading Rates	188
	с.	Nitrogen Transformations	190
	D.	Ridge and Furrow Operation and Maintenance	197
	E.	System Performance	199

Chapter

7	Conclusions	201
8	Recommendations	203

Page

Appendices

- A1: List of References
- A: Well and Lysimeter Logs for Brodhead
- B: Brodhead Soil Analyses
- C: Brodhead Wastewater Chemistry Data
- D: Brodhead Wastewater Flow Data
- E: Groundwater Elevations and Contours for Brodhead
- F: Brodhead Groundwater Chemical Data
 - G: Unsaturated Flow Rate References Bouma
 - H: Furrow Wastewater and Lysimeter Data for Brodhead
 - I: Crop Uptake Data and Calculations Brodhead
 - J: Sugar River Chemistry Data
 - K: Brodhead Nitrogen Budget Calculations
 - AA: Well and Lysimeter Logs for Mindoro
 - BB: Mindoro Soil Analyses

.

- CC: Mindoro Wastewater Chemistry Data
- DD: Mindoro Wastewater Flow Data
- EE: Groundwater Elevations and Contours for Mindoro
- FF: Mindoro Groundwater Chemistry Data
- HH: Furrow and Lysimeter Data for Mindoro
- II: Crop Uptake Data and Calculations Mindoro
- JJ: Severson Coulee Creek Chemistry Data
- KK: Mindoro Nitrogen Budget Calculations

List of Figures

			<u> </u>
	Figure 2.1:	The Nitrogen Cycle	5
	Figure 2.2:	The N-cycle for a Ridge and Furrow Setting	6
	Figure 2.3:	Ridge and Furrow Cell Layouts	34-35
	Figure 2.4:	Furrow and Lysimeter Detail	36
	Figure 3.1:	Typical Well Installation	39
	Figure 3.2:	Typical Lysimeter Installation	41
	Figure 3.3:	Typical Stage Marker	42
	Figure 3.4:	Typical Infiltration Station	43
	Figure 4.1:	Brodhead Topography Map	52
-	Figure 4.2:	Brodhead System Plan View	53
	Figure 4.3:	Brodhead Average Monthly Wastewater Flows	60
÷	Figure 4.4:	Typical Water Table Contours at Brodhead	64
	Figure 4.5:	Well 13A Groundwater Elevations vs. Time	65
	Figure 4.6:	Well 15 and 17 Groundwater Elevation Response to Rainfall Events	67 A
-	Figure 4.7:	Chloride Concentration Contours at the Water Table	70
-	Figure 4.8:	COD Concentration Contours at the Water Table	71
	Figure 4.9:	TKN Concentration Contours at the Water Table	73
7	Figure 4.10:	Ammonium Concentration Contours at the Water Table	74
*	Figure 4.11:	Nitrate Concentration Contours at the Water Table	75
	Figure 4.12:	Well 15 Nitrogen Concentrations	76
	Figure 4.13:	Well 15, 16, and 17 Chloride Concentrations	78
	Figure 4.14:	Well 16 Nitrogen Concentrations	79

۱

.

Page

		Page
Figure 4.15:	Well 17 Nitrogen Concentrations	80
Fugure 4.16:	Well 15, 16, 17 Total Nitrogen Concentrations	82
Figure 4.17:	Barrier Flows - Infiltration Station 1A	84
Figure 4.18:	Barrier Flows - Infiltration Station 1B	85
Figure 4.19:	Barrier Flows - Infiltration station 2A	87
Figure 4.20:	Barrier Flows - Infiltration Station 2B	88
Figure 4.21a:	Cell 1 - Unsaturated Zone Chloride Profile	93
Figure 4.21b:	Cell 2 - Unsaturated Zone Chloride Profile	94
Figure 4.22a:	Cell 1 - Unsaturated Zone Nitrogen Profile	95
Figure 4.22b:	Cell 2 - Unsaturated Zone Nitrogen Profile	96
Figure 4.23a:	Cell 1 Furrow, Lysimeter 1 and 2 Chloride Concentrations vs. Time	98
Figure 4.23b:	Lysimeter 3 and Well 17 Chloride Concentra- tions vs. Time	99
Figure 4.24a:	Cell 1 Furrow, Lysimeter 1 and 2 Total N Con- centrations vs. Time	100
Figure 4.24b:	Lysimeter 3 and Well 17 Total N Concentrations vs. Time	101
Figure 4.25:	Cell 1 Furrow Wastewater Nitrogen vs. Time	102
Figure 4.26:	Lysimeter 1 Nitrogen vs. Time	103
Figure 4.27:	Lysimeter 2 Nitrogen vs. Time	104
Figure 4.28:	Lysimeter 3 Nitrogen vs. Time	105
Figure 4.29a:	Cell 2 Furrow and Lysimeter 5 Chloride Concentrations vs. Time	108
Figure 4.29b:	Lysimeter 6 and Well 15 Chloride Concentra- tions vs. Time	109
Figure 4.30a:	Cell 2 Furrow and Lysimeter 5 Total N Concen-	110

~

•

."

.

.

٠

•

	•	Page
Figure 4.30b:	Lysimeter 6 and Well 15 Total N Concentra- tions vs. Time	<u>Page</u> 111
Figure 4.31:	Cell 2 Furrow Wastewater Nitrogen Concen- trations vs. Time	112
Figure 4.32:	Lysimeter 5 Nitrogen Concentrations vs. Time	113
Figure 4.33:	Lysimeter 6 Nitrogen Concentrations vs. Time	114
Figure 4.34:	Cell 1 Furrow Wastewater Total and Dissolved TKN Concentrations vs. Time	117
Figure 4.35:	Cell 2 Furrow Wastewater Total and Dissolved TKN Concentrations vs. Time	118
. Figure 5.1:	Mindoro Topography Map	130
<pre>> Figure 5.2:</pre>	Mindoro System Plan View	131
Figure 5.3:	Mindoro Average Monthly Wastewater Flows	138
Figure 5.4:	Typical Water Table Contours at Mindoro	143
Figure 5.5:	Well 5, 6, and 7 Groundwater Elevations vs. Time	144
Figure 5.6:	Chloride Concentration Contours at the Water Table	149
Figure 5.7:	COD Concentration Contours at the Water Table	151
Figure 5.8:	TKN Concentration Contours at the Water Table	152
- Figure 5.9:	Ammonium Concentration Contours at the Water Table	153
* Figure 5.10:	Nitrate Concentration Contours at the Water Table	154
Figure 5.lla:	Well 3, 4, and 9 Chloride Concentrations vs. Time	156
Figure 5.11b:	Well 3, 4, and 9 TDS Concentrations vs. Time	157
Figure 5.12a:	Well 2 TDS Concentrations vs. Time	158
Figure 5.12b:	Well 2 Dissolved TKN Concentrations vs. Time	159

.

•

ς.

9

			Page
Figure	5.12c:	Well 2 Chloride Concentrations vs. Time	160
Figure	5.13:	Well 3, 5, and 9 Total N Concentrations vs. Time	162
Figure	5.14:	Well 3, 5, and 9 Nitrate Concentrations vs. Time	163
Figure	5.15a:	Cell 1 Furrow Infiltration Rates	164
Figure	5.15b:	Cell 1 Furrow Infiltration Rates	165
Figure	5.15c:	Cell 1 Furrow Infiltration Rates	166
Figure	5.16a:	Cell 2 Furrow Infiltration Rates	167
Figure	5.16b:	Cell 2 Furrow Infiltration Rates	168
Figure	6.1:	Lysimeter 325 Nitrogen Concentrations vs. Time	193
Figure		Lysimeter 415 Nitrogen Concentrations vs. Time	194
Figure	6.3:	Well 5 Nitrogen Concentrations vs. Time	195

t

•

List of Tables

			Page
Table 3	3.1:	Chemical Analysis Done	49-50
Table 4	la:	Well Specifications at Brodhead Site	55
Table 4	4.1b:	Lysimeter Specifications at Brodhead	56
Table 4	4.2:	Brodhead Wastewater Chemistry	57
Table 4	4.3:	Brodhead Hydraulic Loading	61
Table 4	4.4:	Brodhead Organic Loading Rates	62
Table 4	1.5:	Mean and Standard Deviation of Groundwater Chemical Parameters at Brodhead	68
Table 4	.6:	Brodhead Unsaturated Zone Travel Times	89
Table 4		Brodhead Mean and Standard Deviation of Furrow Wastewater and Lysimeter Chemical Parameters	91
Table 4	.8:	Brodhead Grass Sample Results	119-120
Table 4	.9:	Sugar River Quality	123
Table 4	.10:	Nitrogen Budget Estimate - Brodhead Site	127
Table 5	.1a:	Well Specifications at Mindoro Creamery	132
Table 5	.1b:	Lysimeter Specifications at Mindoro Creamery	133
Table 5	.2:	Mindoro Wastewater Character	135
Table 5	•3:	Mindoro Hydraulic Loading Rates	140
Table 5	• 4 :	Mindoro Organic Loading Rates	141
Table 5		Mean and Standard Deviation of Groundwater Chemical Parameters at Mindoro	147
Table 5	.6:	Average Groundwater Nitrogen Losses at Mindoro	150
Table 5.	.7:	Mindoro Infiltration Rates	169
Table 5.		Mindoro Mean and Standard Deviation of Furrow Wastewater and Lysimeter Chemical Parameters	172

		Page
Table 5.9:	Reductions Along First Mindoro Flow Path	175
Table 5.10:	Reductions Along Second Mindoro Flow Path	175
Table 5.11:	Reductions Along Third Mindoro Flow Path	176
Table 5.12:	Mindoro Wastewater, Header, and Furrow Wastewater Nitrogen Results	178
Table 5.13:	Mindoro Grass Nitrogen Results	180
Table 5.14:	Mindoro-Severson Coulee Creek Quality	181
Table 5.15:	Nitrogen Budget Estimate - Mindoro Site	185
Table 6.1:	Summary of Relative COD Reductions at Brodhead	191
Table 6.2:	Summary of Relative COD Reduction at Mindoro	191

CHAPTER 1: INTRODUCTION

What is a Ridge and Furrow?

For many years the ridge and furrow land treatment process has been a popular and simple method of industrial wastewater disposal. A ridge and furrow system is simply described as a series of interconnected ditches (furrows) which allow for the distribution, infiltration, and treatment of wastewater. Ridges between the ditches support a cover crop which takes up nutrients and water and protects the ditches during the winter. In Wisconsin, there are 83 dairies; four meat packers, a rendering plant, and a pet food manufacturer, which utilize the ridge and furrow process (Rodenberg, 1980). Site areas range from 0.1 to 56 acres.

There are three advantages in selecting the ridge and furrow treatment process. They are 1) ease of operation; 2) cost (capital and operation and maintenance), and 3) year around operation. Design of these systems is based on hydraulic loading rates and BOD₅ loading rates. Nitrogen loading rates are currently not considered. Two or more cells are preferred to allow for loading flexibility. A healthy cover crop is also an important feature of a ridge and furrow site.

Nitrogen Concerns at Ridge and Furrow Systems

Nitrogen, in its organic, ammonium, and nitrate forms, is a major parameter of concern in State groundwater protection programs. The United States Environmental Protection Agency has set a 10 mg/l drinking water

-1-

standard for nitrate-nitrogen. Such standards are set to reduce the occurrence of animal and human disease and to control environmental pollution. These concerns will be discussed in Chapter 2.

Project Description

This report presents the results of a study supported by the Wisconsin Department of Natural Resources (WDNR) and conducted at two ridge and furrow sites treating cheese processing wastewater. One system, operated by Universal Foods in Brodhead, Wisconsin, receives 39,000 gallons per day and is located on 4.7 acres of sandy soil. The other system, operated by the Mindoro Co-op Creamery in Mindoro, Wisconsin, receives 14,000 gallons per day and is located on 3.0 acres of silty loam soil. This is the oldest ridge and furrow system in the state.

The project had four objectives. The primary objective was to determine the nitrogen transformations in the wastewater as it percolated from the furrows to the groundwater. In relation to these transformations, a nitrogen budget estimate was attempted at each site. Other project goals were: 1) to analyze ridge and furrow treatment effectiveness under different soil and loading conditions, 2) to examine the operation and maintenance at these systems, and 3) to evaluate the monitoring equipment used.

To complete these objectives, groundwater monitoring wells and lysimeters were installed and a sampling program was initiated in August of 1983. Until November of 1984, well, lysimeter, furrow-water, wastewater, and bounding surface water samples were taken and analyzed for

-2-

five day biochemical oxygen demand (BOD₅), chemical oxygen demand (COD), total suspended solids (TSS), total dissolved solids (TDS), chlorides (Cl-), total Kjeldahl nitrogen (TKN), ammonium nitrogen (NH₃-N), nitrate and nitrite nitrogen (NO₃-N+NO₂-N), and pH. Periodic analysis for alkalinity, total phosphorus (P), sulfate (SO₄²⁻), potassium (K+), sodium (Na+), magnesium (Mg²⁺), and calcium (Ca²⁺) was also done. Analysis was performed by the Wisconsin State Laboratory of Hygiene.

Other work performed included: 1) changing the load/rest cycles at one site to observe effects, 2) reading monthly groundwater and surface water elevations, 3) taking monthly 30-day average wastewater flow readings, 4) cutting periodic grass samples during the growing season to determine nitrogen uptake, 5) performing infiltration studies to determine unsaturated zone flow rates, and 6) making general site observations.

CHAPTER 2: LITERATURE REVIEW

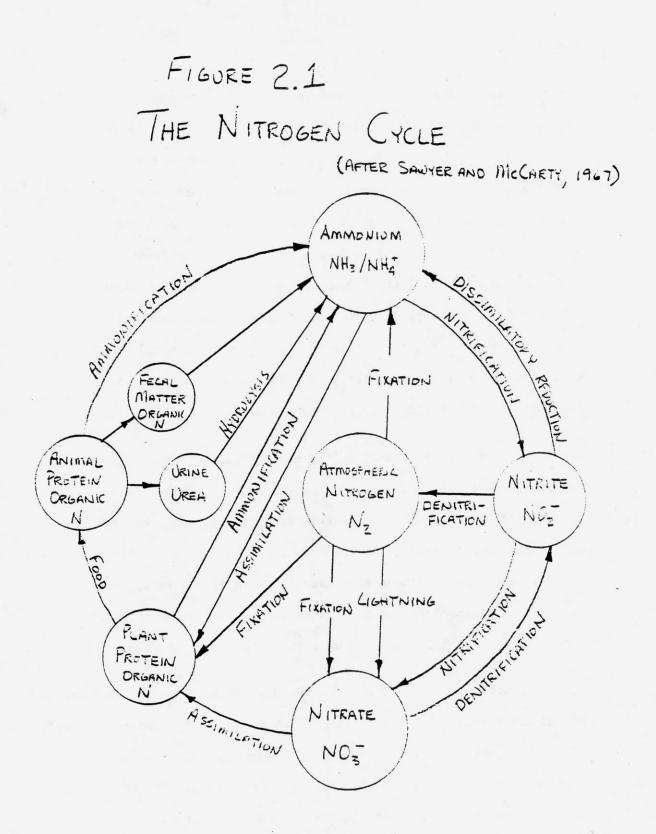
Introduction

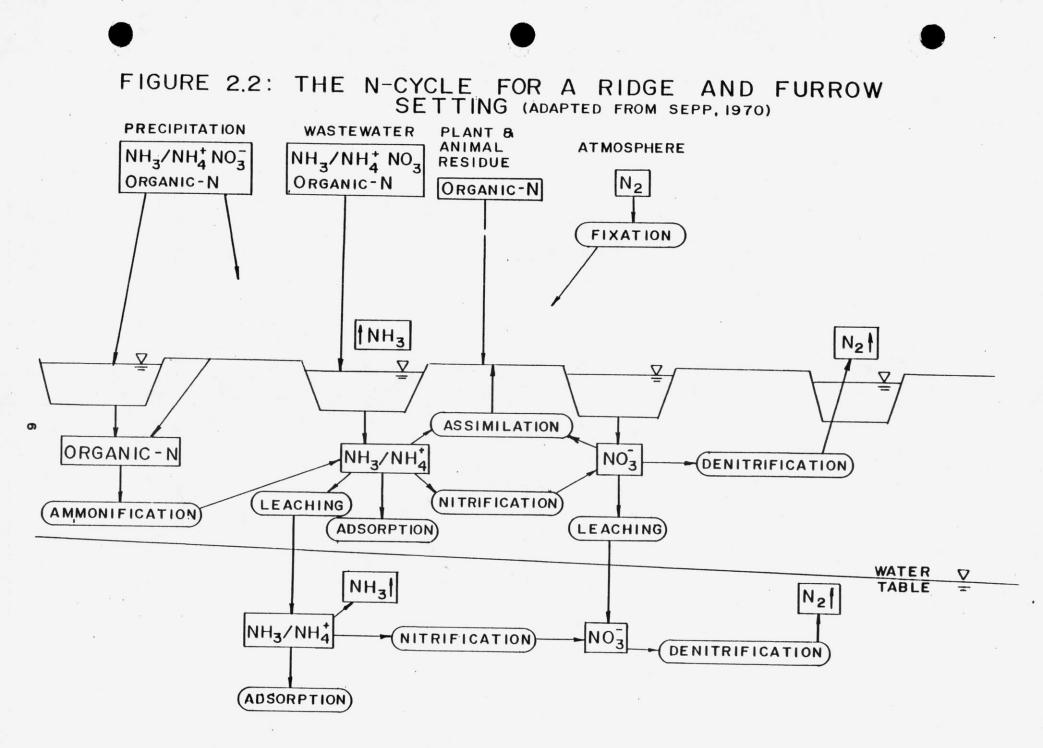
Nitrogen is ubiquitous in our environment. The atmosphere contains 78% nitrogen and topsoils typically contain 4,000 to 200,000 lb N/acre (Winneberger, 1982). Figure 1 illustrates the classical nitrogen cycle indicating the many transformations involved. The nitrogen cycle and its relationship to the soil and groundwater are shown in Figure 2. Nitrogen inputs to the soil system occur through precipitation, waste and fertilizer application, plant residue deposition, and atmospheric nitrogen fixation. System losses occur through ammonia volatilization, plant uptake, and denitrification. Ammonification (or mineralization), assimilation (or immobilization), nitrification, denitrification, and leaching are internal processes.

A large portion (90%) of soil nitrogen is organically bound and contained in the first 40 inches of soil (Tusneem & Patrick, 1971; Winneberger, 1982). Warmer climates favor the breakdown of organic matter and soil N accumulates less rapidly than in cooler climates. Nitrogen and organic matter also increase with effective moisture. Stored N is released when a soil is disturbed (eg. heavy rainfall or fire) with leaching losses ranging from 80 to 4,000 lb N/acre/year.

The inorganic-N soil fraction is in the NO_3^- or NH_4^+ form and constitutes an immediate source for plant uptake (Tusneem and Patrick, 1971). Most of the inorganic-N is water soluble or adsorped on the soil exchange complex. Approximately five percent, and possibly as high as 30%, of soil NH_4^+ may be fixed in the lattice of silicate minerals in a non-

-4-





exchangeable form. Ammonium may be oxidized to nitrate through nitrification. Nitrate in soil may be removed by leaching, denitrification, or plant uptake. It is possible that the reduction of NO_3^- to NH_4^+ could occur but this process is not common (Tiedje, Sorenson, and Chang; 1979).

Inorganic-N in waterlogged soils is high in ammonium and low in nitrate (Tusneem and Patrick, 1971). Waterlogged nitrate concentrations range from 0-3 mg/l. Ammonium in waterlogged soils can be taken up by plants, immobilized in bacterial cells, adsorped on soil particles, and volatilize under alkaline conditions.

The goal of many studies of soil nitrogen is to develop a nitrogen budget at a given site. However, it is not as easy as counting the number of marbles dropped over a given area. It is more like trying to balance a checking account when others have made deposits and withdrawals without giving notice of the transactions. Keeping this in mind, general rate ranges for the various nitrogen transformations have been made and are as follows (Winneberger, 1982):

1. N Fixation - 2 to 500 lb N/acre/yr,

2. Fertilizer Leaching - 0 to 200 lb N/acre/yr,

3. Rainfall Input - 4 to 12.5 lb N/acre/yr,

4. Denitrification - 0 to all 1b N/acre/yr,

5. Storage Losses - 80 to 4,000 lb N/acre/yr, and

6. Plant uptake - few quantifications.

Since nitrification does not lead to a loss or gain of soil nitrogen, it was not considered in the budget.

-7-

There are several undesirable effects of nitrification (Alexander, 1977). Nitrate is an anion very susceptible to leaching, which takes an essential nutrient away from plants. Nitrate also has a role in methemoglobinemia in infants and animals, eutrophication, and the formation of nitrosamines. Wastewater, wastewater sludges, fertilizers, and manure are all potential sources of nitrate. Nitrosamines are not common soil constituents but are in pesticides. They are carcinogenic, mutagenic, and teratogenic.

Nitrogen is a concern in the environment when its nitrate form enters groundwater aquifers. Young infants lack stomach acidity and at pH values greater than four, nitrite formers can exist in the gut. Nitrite, formed from nitrate, then combines with hemoglobin in the blood instead of oxygen, causing oxygen starvation. This illness is termed methemoglobinemia (Winneberger, 1982).

Since 1945, 2,000 cases of methomoglobinemia were reported in North America and Europe with 7-8% fatalities (Winneberger, 1982). There were no fatalities between 1960 and 1972. Possibly only 10% of all cases have been reported. Compared to other causes of infant death, however, methemoglobinemia is rare. In 1975 alone, about 9,000 infant fatalities were reported in the United States, 178 by homocide.

Considering the rarity of methemoglobinemia, the 10 mg/l drinking water standard has received criticism. Nitrate ingestion and methemoglobinemia occurrence may be separate events and setting a NO_3^--N standard may have no effect on the disease rate. Scientists have been limited to correlative studies since direct studies on babies are not ethical.

-8-

The 10 mg/l standard was established using a correlative study. Researchers found that the incidence of methemoglobinemia was insignificant in infants whose drinking water supply contained less than 10 mg/l nitrate. A recent study shows, however, that there is no higher incidence of the disease in babies drinking high nitrate concentrations than in a control group (Winneberger, 1982). The infants had high methemoglobin (NO₂-hemoglobin) levels no matter what the nitrate concentration was in the water source.

Denitrification is related to environmental pollution (Alexander, 1977). Once nitrate passes through the root zone, denitrification is desired to reduce subsurface nitrate concentrations for reasons stated earlier. This process is possible with depth when carbonaceous nutrients are present. This is the case in waste treatment when production of nitrate coexists with available C.

Current data also suggest that microbial release of N₂, N₂O, and NO far exceeds that by human activity (Alexander, 1977). NO reacts with O_3 (ozone) and light to destroy the ozone layer which protects the earth from ultraviolet rays.

With this introduction to the nitrogen cycle and its environmental concerns, this review discusses in more detail the processes of mineralization/immobilization, nitrification, dissimilatory reduction, N_2 fixation, plant uptake, volatilization, leaching, and adsorption. This is followed by sections which summarize current research of the nitrogen cycle as it affects land treatment, which present the history of ridge and furrow systems, and which discuss the design of these systems.

-9-

Mineralization and Immobilization

Mineralization (or ammonification) is the conversion of organic-N to ammonia or ammonium. Immobilization is the assimilation of inorganic-N $(NH_4^+, NO_3^-, and NO_2^-)$ by microorganisms into the nitrogeneous constituents of their cells. The two processes work simultaneously and either net immobilization or net mineralization results. This relationship controls the amount of available N in the soil. In natural systems, mineralization usually exceeds immobilization. (Tusneem and Patrick, 1971; Alexander, 1977).

The decomposition of organic-N in soils is done by general purpose heterotrophs, fungi, and actinomycetes. These organisms use the organic-N compounds as an energy source and produce NH_3 , carboxylic acids, amines, mercaptans, and H_2S (Tusneem and Patrick, 1971). During immobilization, ammonium and nitrate are incorporated into cell amino acids, amino sugars, nucleic acids, and other organic complexes (Alexander, 1977; Paul and Juma, 1979). Immobilization results in depressed plant uptake of nitrogen and decreased plant yield.

Mineralization and immobilization are dependent on several environmental factors including C:N ratio, waterlogging, wetting and drying cycles, temperature, pH, soil moisture, and soil clay content. The ratio of carbonaceous material (energy source) and nitrogen in substances undergoing decomposition usually dictates whether net mineralization or net immobilization occurs. With similar C available, a source rich in N results in net mineralization while a source poor in N results in net

-10-

immobilization (Tusneem and Patrick, 1971). When C:N falls below the 20-30:1 range, net mineralization occurs and inorganic-N will appear (Alexander, 1977).

In waterlogged (anaerobic) soils, less efficient and more restricted bacteria take over (Tusneem and Patrick, 1971). Both mineralization and immobilization are retarded. The features of this anaerobic decomposition are the following:

- 1. incomplete decomposition of carbohydrates into CH_{4} , organic acids, H_{2} , and CO_{2} ;
- lower energy of fermentation leading to less cell production; and
- 3. a low N requirement leading to a more rapid release of ammonium.

If wetting and drying cycles occur, mineralization proceeds at a faster rate upon rewetting than if the soil had been wet all along (Tusneem and Patrick, 1971). This rate declines in later cycles. Also, the longer the drying period, the faster the rate upon wetting. The wetting/drying process may make substrates readily accessible or drying may cause cell disintegration.

Since mineralization is catalyzed by a temperature sensitive enzyme, temperature also affects this process (Alexander, 1977). Mineralization occurs between the temperatures of two to 60°C with an optimum rate between 40 and 60°C. Thawing/freezing action also has a similar effect as wetting/drying.

Soil pH, moisture, and clay content have influences on the mineralization rate as well (Alexander, 1977). Mineralization is favored by a neutral pH environment and increasing soil moisture content. Clay

-11-

minerals have the ability to adsorb cell enzymes, pulling them away from the decomposition process.

In the land treatment of dairy wastes, the mineralization of proteins is a concern. Proteins are broken down into smaller amino acid chains by the extracellular enzyme protease (Alexander, 1977). Once broken down, these acids enter the cell where they serve as N, C, and energy sources. The four paths of amino acid (AA) breakdown are:

1. deamination by direct removal of NH₃,

 $AA \rightarrow RCH=CHCOOH + NH_3$

2. oxidative deamination,

 $AA + 0.5 O_2 --> RCOCOOH + NH_3$

3. reductive deamination, and

 $AA + 2H^+ - RCH_2COOH = NH_3$

4. decarboxylation.

AA \rightarrow RCH₂NH₂ + CO₂

where: AA is RCH_2CHNH_2COOH in Equation 1 and $RCHNH_2COOH$ in Equations 2, 3, and 4.

Nitrification

Nitrification is the biological formation of nitrate or nitrite from reduced N compounds, namely ammonium. Nitrification occurs in two steps, the conversion of ammonium to nitrite and the conversion of nitrite to nitrate. The genera of Nitrosomonas, Nitrosococcus, Nitrosospira, and Nitrosolobus are the principal nitrite formers while Nitrobacter is the principal nitrate former (Alexander, 1977). Heterotrophs and fungi are also capable of oxidizing inorganic nitrogen.

-12-

Nitrosomonas and Nitrobacter are the most frequently encountered nitrifying chemotrophs and they are usually found together. These bacteria typically obtain their energy from the oxidation of inorganic-N. Carbon is obtained from CO_2 or carbonates. Nitrobacter requires low amounts of molybdenum for its metabolism. (Alexander, 1977).

In oxidizing NH_4^+ to NO_2^- , the N oxidation state changes from -3 to +3. The pathway is unclear but is hypothetically as follows (Alexander, 1977):

 $NH_3 \rightarrow NH_2OH \rightarrow HNO \rightarrow NO \rightarrow NO_2^-$

Overall reaction: $NH_4^+ + 1.5 0_2 - NO_2^- + 2H^+ + H_2O_2^-$

Nitrite accumulation is rare. It only results from high alkalinity and high ammonium levels. High ammonium concentrations are toxic to Nitrobacter. Nitrobacter oxidizes N from +3 to +5 yielding two electrons as follows:

 $NO_2^- + H_2O --> H_2O \cdot NO_2^- --> NO_3^- + 2H^+$

Many environmental factors affect nitrification. The process is slow in acid habitats (Alexander, 1977). Rates typically fall at pH values below six and are negligible at pH less than five since the nitrifier population is decreased.

Aeration is essential to nitrification and moisture is also a factor (Tusneem and Patrick, 1971; Alexander, 1977). Waterlogged environments lead to complete suppression of nitrification by limiting oxygen diffusion. On the other hand, the process does not work in arid conditions

-13-

due to a lack of water. Generally, nitrate appears at 1/2 to 2/3 of a soil's moisture holding capacity.

In temperate regions, nitrate formation is most rapid in spring and fall and lowest in the summer and winter (Alexander, 1977). Nitrifier populations are decreased during extreme heat or cold. Rates slow at temperatures less than 4°C or greater than 40°C with an optimum between 30 and 35°C. Soil temperatures are not this warm, however. In-field research is needed to better quantify this assertion.

Nitrification is also affected by the type of crop and the soils cation exchange capacity (CEC) (Alexander, 1977). Roots of some grasses excrete compounds deleterious to the process. Once ammonium is adsorbed to a clay mineral, the availability of this fixed cation to chemoautotrophs is low with less than 25% nitrified within several months.

Denitrification

Denitrification is the bacterial reduction of NO_3^- and NO_2^- with the liberation of N₂O and N₂. During this nitrate reduction, nitrogen fails to enter the cell structure and is lost to the atmosphere. Denitrification is encouraged by a supply of decomposable organic matter, high nitrate supply, and anaerobic (reduced) conditions (Winneberger, 1982; Tusneem and Patrick, 1971). In anaerobic environments, facultative anaerobes use nitrate as an electron acceptor and nitrogen escapes as N₂O or N₂ (Stanford, Vander Pol, and Dzienia, 1975).

Denitrifying bacteria growth is not dependent on nitrate reduction (Alexander, 1977). There presence indicates denitrification potential

-14-

but not that conditions are favorable. Denitrifiers are facultative aerobes which use nitrate as an electron acceptor for growth in the absence of oxygen. Most organisms get less energy using NO_3^- , ie. fewer cells/unit substrate oxidized (Alexander, 1977). Energy conservation is attained by the electron transport phosphorylation (ETP) process (Tiedje, Sorensen, and Chang, 1979).

Denitrification is done by Paracoccus denitrificans, Pseudomonas aeruginosa, and Bacillus licheniformis. Facultative autotrophic denitrifiers, such as Paracoccus denitrificans, use either organic matter or H_2 as an energy source and O_2 or NO_3^- as an electron acceptor. One group of denitrifiers is photosynthetic (Alexander, 1977; Knowles, 1979).

Alexander (1977) and Knowles (1979) present the following overall denitrification reaction:

 $4NO_3^- + 5CH_2O + 4H^+ < --> 2N_2 + 5CO_2 + 7H_2O$

NaR, NiR, NOR, and N₂OR are the catalyzing enzymes nitrate reductase, nitrite reductase, NO reductase, and nitrous oxide reductase, respectively. N₂OR seems most sensitive to low pH and the presence of oxygen, nitrate, and sulfide. In these cases, N₂O will be a significant product. All reductases are repressed by oxygen presence. Small amounts of molybdenum is required by nitrate reductase.

Denitrification is dependent on many environmental factors including soil moisture, aeration, wetting/drying cycles, available organic

-15-

matter, waste loading rates, nitrate concentration, pH, temperature, and the presence of sulfur or acetylene (Standord, Vander Pol, and Dzienia, 1975; Tusneem and Patrick, 1971; Alexander, 1977; Knowles, 1979; Winneberger, 1982).

In well drained soils, nitrogen loss is related to a soil's moisture content with higher denitrification rates occurring at higher water levels. Losses usually do not occur at moisture contents less than 60% of the soil water holding capacity regardless of the carbohydrate or nitrate supply, or the pH. Moisture content governs the diffusion of oxygen to sites of activity. As soil moisture increases, N₂O content will decrease as N₂ is formed. Denitrification is high in waterlogged soils and low in drier soils since inorganic-N is immobilized. Rates can be significant in dry soils as well if water pockets develop. These pockets can create anaerobic micro-environments to promote nitrate reduction.

Aeration is necessary in nitrate production, the basic substrate in denitrification. Oxygen presence must not be so great, however, as to inhibit denitrification. Since total nitrogen losses do occur in aerated soils, the existence of anaerobic micro-environments is again proposed. This theory is also reflected by large nitrogen losses at sites undergoing cycles where oxygen is alternately available and then absent.

Tusneem and Patrick (1971) suggest that moisture fluctuations as a result of flooding and draining create ideal conditions for nitrogen loss. Two layers or zones develop: a surface oxidizing layer and an

-16-

underlying reducing layer. Applied ammonium is nitrified in the oxidizing layer by aerobic bacteria. Nitrate then percolates to the reduced layer and is subsequently denitrified biologically and possibly chemically to gaseous N.

Greenland (1962) reported that nitrification and denitrification could occur simultaneously in wet soil due to aerobic and anaerobic microzones. Russell and Richards (1917) determined that alternate wetting and drying would create an ideal environment for denitrification. Patrick and Wyatt (1964) observed a 20% N loss from this cycle. The frequency of wetting/drying affected total N loss as well as the rate of nitrate reduction in subsequent cycles. Major losses occurred during the first two to three cycles, decreasing as cycles progressed.

Winneberger (1982) believes that denitrification is best facilitated by environments alternately exposed to anaerobic and aerobic conditions. This can be done through loading and resting cycles or by adding energy rich organics to create microanaerobic areas.

Denitrification rates positively correlate with the amount of soil water extractable organic-C (Knowles, 1979). Stanford, Vander Pol, and Dzienia (1975) and Alexander (1977) suggest that low carbon containing soils (eg. sand) support a lower rate of denitrification. At wastewater disposal sites, extra organic matter is supplied as reflected by the wastes BOD or COD.

Waste loading rates also affect denitrification. McMicneal and McKee (1966) spread two feet/day of wastewater and most of the applied N was

-17-

accounted for at depth. Lesser loadings have given high nitrogen removals. Kardos, Sopper, and Myers (1965) reported 68 to 82% N removal by sprinkle irrigation of wastewater at one to two inches/week. In the former case, low nitrogen losses resulted from soil saturation which inhibited the necessary production of nitrate. High nitrogen losses were promoted by the aeration provided by the irrigation and the lower loading rate.

The nitrate concentration in the soil pore water is also a factor in denitrification. According to Knowles (1979), at relatively high NO_3 concentrations, the denitrification reaction is frequently zero order. Depending on the environment of the reaction, the denitrification rate increases linearly to a given nitrate concentration, after which, rates level off and little gaseous nitrogen is liberated.

Many denitrifiers are sensitive to low pH and therefore denitrification rates are positively correlated with soil pH (Alexander, 1977; and Knowles, 1979). An optimum range is between pH 7 and 8. Acidity also governs the abundance of certain gases (Alexander, 1977). N₂O liberation is pronounced in the pH range of 6.0 to 6.5 while NO only appears at low pH. These differences may result from the acid sensitivity of the enzyme system for N₂O reduction.

Rates of nitrate reduction are also temperature dependent. Alexander (1977) stated that increasing the temperature from 2°C enhances denitrification, with an optimum at approximately 25°C. Knowles (1979) found that denitrification is temperature dependent between 10 and 35°C with maximum rates at 60-75°C. Rates have been measured between 0-5°C. Low temperatures reportedly result in relatively more N_2O and NO.

-18-

Sulfur compounds affect denitrification by inhibiting the reduction of NO and N_2O to N_2 (Knowles, 1979). Acetylene also inhibits nitrous oxide reduction.

Dissimilatory Reduction

Dissimilatory reduction is the bacterial conversion of nitrate back to an ammonia form. This reduction could occur in anaerobic habitats since oxygen inhibits enzymes and represses sythesis (Tiedje, Sorensen, and Chang, 1979). Microorganism genera responsible for dissimilatory reduction include Bacillus, Enterobacter, Klebsiella, Erwina, and Clostridia. These organisms are more prevalent than the denitrifiers soil. Anaerobic environments have abundant electron donors and a scarcity of electron acceptors. Therefore, the eight electron reduction to NH_4^+ should be favored over the five electron reduction in denitrification. Also, since this reduction is respiratory rather than assimilatory related, one would expect more NO_3^- to NH_4^+ reduction than assimilatory reduction.

Dissimilatory reduction does not dominate denitrification in most cases, however. Where the oxygen status is more transient or where the redox potential is less reduced, denitrification dominates. Tiedje, Sorenson, and Chang (1979) incubated an organic muck soil (pH 5.7) anaerobically with and without glucose (C) addition. Labeled nitrate and ammonium were added. In the sample without C addition, the dissimilatory reduction rate was 0.3-0.6 ug/g/day while the denitrification rate was 15 ug/g/day. With C addition, the dissimilatory reduction rate was

-19-

1 ug/g/day while the denitrification rate was 25 ug/g/day. The obvious conclusion of the study was that denitrification was the major nitrate reduction process in an organic muck soil.

Nitrogen Gas Fixation

Nitrogen fixation is performed by bacteria or blue-green algae, which use N_2 by non-symbiotic means and by symbiotic associations between microorganisms and a higher plant (Alexander, 1977). Non-symbiotic fixation is performed by actinomycetes, fungi, yeasts, aerobic bacteria, facultative anaerobes photosynthetic bacteria (nonsulfur purple, purple sulfur, green sulfur), and blue green algae.

Azotobacter has been the most intensely studied non-symbiotic N fixer but it is not very common in soils. The dominate anaerobes are Clostridia which proliferate when organic matter is added. They are numerous around plant roots at sites with pH values of five and they are capable of fixing N up to pH nine. Fixing efficiency is low with 2-10 mg N fixed/gram carbohydrate consumed. Blue green algae are common in flooded soils and are stimulated by increasing light intensity. Its fixation of nitrogen is less rapid than in Azotobacter or Clostridia. All photosynthetics are affected favorably by light and inhibited by oxygen. Their rate of assimilation is also quite slow. Non sulfur purples are found in flooded soils, ditches, lake muds, and sea bottoms.

Many factors affect the non-symbiotic fixation of gaseous nitrogen. The presence of nitrate or ammonium can reduce fixation. Nitrogen fixers have the ability to use NO_3^- and NH_4^+ and sometimes prefer them to N_2 .

-20-

Fixation is also dependent on certain metals. Molybdenum, iron, calcium, and cobalt are all critical for the reaction. The availability of energy sources (sugar, cellulose) also limits the rate and extent of fixation by heterotrophs. Typically one to 30 mg of N are assimilated per gram of carbon source.

Bacterial nitrogen fixers are affected by soil acidity. Azotobacter, as well as blue-green algae, are sensitive to pH values less than six. The fixation rate is also determined by soil moisture. Nitrogen assimilation is insignificant when little water is available but activity can be especially great under waterlogged conditions. The optimum water level varies with soil type and quantity or organic matter. Increasing temperature also promotes gas uptake with an optimum of about 35 degrees C. Deliberate vegetation burning seems to promote nitrogen fixation as well and grasslands generally have low activity.

The classic example of symbiotic N fixation is the relationship between leguminous plants and bacteria of the genus Rhizobium. The seat of the symbiosis is in nodules on the plant roots. Rhizobia are gram negative, non-spore forming, aerobic rods which are typically motile. Several carbohydrates are used in its metabolism with occasional acid accumulation. Gas is never liberated. Rhizobia grow readily in media containing a C source, NH_4^+ or NO_3 to supply needed N, several inorganic salts, several B vitamins, and cobalt. Symbiotic N_2 fixation rates range from 65-335 kg N/ha/year. Non-legumenous plants (cg. alder trees) are also capable of nodule formation and N_2 fixation.

-21-

The many environmental influences of symbiotic N₂ fixation include type of legume, inorganic-N content of soil, level of phosphorous and potassium which are essential host nutrients, pH, presence of secondary nutrients, and climate as it affects the host plant. As in nonsymbiotic fixation, symbiotic fixation is inhibited by the presence of inorganic-N. Little nodule formation occurs at pH less than five. This is probably due to iron or aluminum toxicity rather than pH sensitivity. There is some evidence that a calcium deficiency is important in its effects of acidity on fixation. Molybdenum, whose availability is affected by pH, and cobalt also stimulate fixation.

Plant Uptake

As mentioned earlier in this chapter, research findings regarding plant uptake of nitrogen has been limited. A general rule of thumb is that 1 to 4% of a soil's organic-N is released to plants during the growing season in temperate climates (Alexander, 1977). Nitrate is the preferred form of nitrogen taken up by plants. Factors favoring uptake include vigorous root aeration, low initial salt content in plant tissues, high external nitrate concentrations, and an absence of ions which compete for uptake. Unfavorable factors to uptake include low light intensity and a limited carbohydrate level in the plant. Woodmansee, Vallis, and Mott (1979) determined that nitrogen is taken up to above ground plant parts during the growing season and then, during the fall, nitrogen is translocated back to the plant's crown and roots.

-22-

Volatilization

Ammonia is a gas at normal temperatures and pressures and its partial pressure is usually low. Volatilization of ammonia is insignificant, however, when the pH of the soil is less than 7.0 (Alexander, 1977). High concentrations of ammonia with high pH, high temperatures, and low CEC are necessary for NH₃ volatilization (Tusneem and Patrick, 1971).

Ammonia sources include organic-N compounds which decompose to release NH₃ (eg. wastewater), fertilizers, ammonia salts, and urea. Most sources provide ammonium which applies to the following equations:

> $NH_4^+ + OH^- < ---> NH_3 + H_2O$ $K_b = [NH_4^+][OH^-]/[NH_3] = 1.774 \times 10^{-5} \text{ at } 25 \circ C$

 $K_w = [H^+][OH^-] = 1.007 \times 10^{-14} \text{ at } 25 \circ C$

where: $-K_b$ is the dissociation constant of the ammonium reaction, $-K_w$ is the dissociation constant for water, $-[NH_4^+]$ is the ammonium concentration, $-[NH_3]$ is the ammonia concentration, $-[OH^-]$ is the hydroxide ion concentration, and $-[H^+]$ is the hydrogen ion concentration.

When dividing K_b by K_w , one gets a relationship between $[NH_4^+]$, $[NH_3]$, and pH (Freney, Simpson, and Denmead, 1979). Note that pK values are constant with temperature. The relation is:

 $\log ([NH_4^+]/[NH_3]) = (pK_w - pK_b) - pH$

-23-

TABLE 2.1

PH AND PERCENT AMMONIA RELATIONSHIP

рH	\$NH ₃
6	0.0
7	1.0
8	10.0
9	50.0

One can see from Table 2.1, derived from the above relationship, that ammonia concentrations are not significant until pH values are greater than eight.

Since the reaction is pH dependent, the buffering capacity (eg. calcium carbonate content) of the soil play an important role. Since hydrogen ions are released with volatilization, a soil will acidify without buffering. High temperatures also enhance volatilization. Besides increasing the pK's, increasing temperatures decrease NH_3 solubility and increase its diffusion rate which all promote volatilization. Since the concentration of ammonium drives the reaction, volatilization is indirectly affected by plant uptake, leaching, application rate, nitrification rate, mineralization, immobilization, and cation exchange. (Freney, Simpson, and Denmead, 1979)

Ammonium Adsorption

Isomorphous substitution in clay minerals gives clay particles a net negative charge. (Isomorphous substitution is the occupation of a clay matrix position by a cation other than the one normally found, without

-24-

change in the crystal structure.) To preserve electrical neutrality, cations are attracted and held on the surfaces and edges of clay particles. These cations are "exchangeable" since cations of one type may be replaced by cations of another type. The quantity of exchangeable cations required to balance the charge deficiency is called cation exchange capacity and is expressed as milliequivalents per 100 grams of dry soil. (Mitchell, 1976). During this ordinary exchange, larger and high charged cations are preferentially adsorbed as follows: $Al^{3+} > Ca^{2+} > Mg^{2+} > K^+ = NH_4^+ > Na^+$ (Bohn, McNeal, and O'Connor, 1979).

Besides adsorption of cations to clay surfaces, cations can be "fixed" inside spaces within the layering of clay particles. Ammonium takes part in this reaction. Fixed ammonium is defined as the NH₄⁺ ions which are not replaceable by prolonged extraction and leaching of a soil by potassium salt solutions (Nommik, 1979). Tusneem and Patrick (1971) stated that the presence of montmorillinite and illite leads to chemical fixation of NH₄⁺ into a non-exchangeable form. These ions are then withdrawn from entering ordinary exchange and have restricted biological activity. Generally less than one-third of fixed NH₄⁺ is available for nitrification. It has been established that these ammonium ions can slowly be replaced by cations which expand the interlayer (K, Mg, Ca).

Adsorbed ammonium contents in the topsoil are typically 1-25% of the total N; they are 10-90% of the total soil N in lower horizons (Kudeyarov, 1979). The soils capacity to adsorb NH4⁺ depends on the soil's mineral composition, texture, and pH. The amount of fixed ammo-

-25-

nium increases with NH_4^+ concentration in the soil solution as described by the following equilibrium equation:

[NH4+]f <---> [NH4+]ss

where: $-[NH_{4}^{+}]_{f}$ is the fixed ammonium concentration, and

 $-[NH_4^+]_{SS}$ is the ammonium concentration in the soil solution.

There is also a relationship between soil moisture content and the amount of adsorbed ammonium (Kudeyarov, 1979; Nommik, 1979). Increasing the moisture content influences the degree of expansion of the lattice of clay minerals, causing the release and migration of adsorbed $\rm NH_4^+$ from the interlayer space. On the other hand, flooding decreases nitrification and as ammonification continues, the ammonium concentration increases in the soil solution. This increase shifts the equilibrium equation towards fixation.

In non-flooded, non-fertilized soils, the maximum ammonium is adsorbed in early spring and late autumn with minimum fixation at the end of the growing season (Kudeyarov, 1979). In the summer, plants assimilate more mineral N than is produced. Therefore, nitrate and ammonium concentrations decrease in the soil. (It was stated earlier that nitrate is assimilated more rapidly by crops than added ammonium. Perhaps NH_4^+ adsorption delays its uptake.) After the vegetation period, nitrogen assimilation and nitrification decline and, as ammonification continues, a net increase in soil NH_4^+ occurs. Therefore, a seasonal pattern exists where the minimum amount of fixed ammonium is during the period of high nitrification and plant uptake.

-26-

Leaching of Soil Nitrogen

Nitrogen leaching, or migration of N into deeper soil horizons, is serious when rainfall exceeds evapotranspiration (Khanna, 1979). Organic-N, which usually makes up more than 90% of the total soil nitrogen, is considered to have low mobility and leaching potential. Normally 40-50% of total rainwater-N is ammonium. Ammonium leaching is considered unlikely, however, except under heavy rainfall or sewage disposal. The reasons for low NH_{ll}^+ leaching are:

- 1. NH4⁺ adsorption by CEC,
- 2. microbial immobilization,
- 3. nitrification,
- 4. plant uptake, and
- 5. NH₃ volatilization.

Nitrates and nitrites are leached the easiest since their negative change prohibits ion adsorption. Khanna (1979) reported an average spring to autumn mineral-N loss leaching rate of 0.6-1.45 kg N/ha/day. In a clay loam soil, Khanna (1979) reported a nitrate leaching rate of 1.9 mm/day.

Vertical solute movement is described by the following equation (Khanna, 1979): $\frac{\partial C}{\partial t} = D \quad \frac{\partial^2 C}{\partial C^2} - V_0 \quad \frac{\partial C}{\partial Z}$

> where: $C = NO_3$ concentration in mg/l, \overline{D} = apparent mean diffusion coefficient in cm/d, V_o = average pore velocity in cm/d, z = linear flow distance in cm, and t = time in days.

> > -27-

This equation may underestimate vertical flow because macropores, created by plants and animals, will act as direct conduits to flow.

Leaching fluctuates with season (Khanna, 1979). Nitrate concentrations rise in streams in autumn and peak in winter or early spring. There is no relation to individual rainfalls. A drought can often lead to an upward migration of nitrate. Increases in stream and soil water nitrate from leaching have been seen after fires as well.

Land Treatment of Nitrogen: Case Study Summary

Losses of 7 to 94% of applied nitrogen through denitrification were indicated in several lab and field studies. Patrick and Gotoh (1974) observed 67% denitrification loss of applied-N in a silt loam soil mixed with $(NH_4)_2SO_4$ and incubated in the dark for 120 days at 30°C. Up to 68% losses of added nitrogen resulted when Tusneem and Patrick (1971) incubated a silt loam at 30°C in water saturated air. These losses were attributed to denitrification since they were too large to result from volatilization or adsorption. Reddy and Graetz (1981) observed 58 to 71% applied-N losses in aerobic lab conditions and 94% losses in anaerobic lab conditions using a muck soil. These losses resulted from denitrification since pH values were low (less than 9). The same conclusion was reached by Chen and Patrick (1981) during a lab scale overland flow study where 53 to 61% of added ammonium was lost. Ammonium reduction of 93% were witnessed by Olson et.al. (1980) at a sandy rapid infiltration site treating municipal primary effluent. This loss occurred through 22 feet of unsaturated zone. Lab scale lysimeters were used by Leach and Enfield (1983) to determine 30 and 79% denitrifi-

-28-

cation losses in sand and a sand/clay mixture, respectively. Ryden and et.al. (1981) found only 7-9% loss of total applied-N at a secondary effluent disposal field. One-third of this loss was from volatilization since wastewater pH was high and the buffering capacity was low. Soil aeration, low soil nitrate and high redox potential (600 mV) were also reasons for the low losses at this sandy loam site.

Leaching of nitrate or ammonium was also seen as a potential sink for applied nitrogen. Lund et.al. (1981), while studying a loamy sand pasture irrigated with secondary treated wastewater, determined that 51-76% of the applied-N leached to the groundwater. This was a rate of 833 kg NO₃-N/acre/yr. Denitrification was not considered due to aerobic soil conditions and the fast percolation rates. Chen and Patrick (1981) observed 10-30% of applied ammonium in the underflow of a lab scale overland flow system.

Crop uptake was a third major sink of applied nitrogen cited in the literature. King (1982) found that 20-30% of applied nitrogen was recovered in a crop irrigated with wastewater from a fiberboard mill. Chen and Patrick (1981) observed 11-22% uptake of added ammonium during simulated overland flow. Palazzo (1981) determined that 50-85% of applied nitrogen was taken up by orchardgrass planted on a silt loam soil and irrigated with municipal wastewater. Palazzo also stated that highest plant yield and nitrogen uptakes occurred during late spring and concluded that higher loading rates could be applied during this period.

-29-

Many case studies indicated parameters such as C:N ratio, load/rest cycling, soil moisture content, and soil oxygen content which affected nitrogen transformations. Lab studies by Patrick and Gotoh (1974), Tusneem and Patrick (1971), and Chen and Patrick (1981) all determined that nitrogen losses increased with decreasing C:N and immobilization increased with added carbon. Highest nitrogen losses were observed at C:N equal to 15:1.

Tusneem and Patrick (1971), Chen and Patrick (1981), King (1982), and Leach and Enfield (1983) all found that nitrogen losses were stimulated by alternate submergence and drying of soils in both lab and field situations. Leach and Enfield also observed that NO_3^--N concentrations increased in soil pore water during resting and were flushed downward during loading causing nitrate peaks to appear.

Soil moisture content and soil pore air oxygen content also were found to have an effect on denitrification. Ryden et.al. (1981) observed maximum denitrification rates at moisture contents of 15-18% at a secondary effluent disposal area. Patrick and Gotoh (1974) found that nitrogen loss increased with oxygen contents up to 20%. Minimal losses occurred at higher O_2 contents. This indicated that the earth's atmospheric content of 21% is adequate for nitrogen loss.

The History of Ridge and Furrow Treatment

The historical perspective of the ridge and furrow treatment of wastewater was provided in two reports by Schraufnagel (1956, 1962) and one by Monson (1956). The following discussion was based on their findings.

-30-

Ridge and furrow irrigation, in the form of sewage farms in the 1870's, was one of the earliest methods of sewage disposal. Several canneries in Iowa began operation of ridge and furrow sites in the 1930's; during the late 1940's and early 1950's, Minnesota and Illinois canneries also began ridge and furrow disposal. Seasonal flows ranged from 136 to 250,000 gallons per day and site areas range from 2 to 40 acres.

In 1930, a dairy in Phoenix, Arizona, reportedly discharged 60,000 gpd of effluent to eight furrows which were plowed under every other day. A creamery in Minnesota installed a ridge and furrow system in 1950 on a 2.8 acre site divided into three cells and underlain by a line of drain tile. The total cost of the system, exclusive of land, was \$800. The first ridge and furrow system in Wisconsin was developed at the Mindoro Cooperative Creamery in 1954. The three acre site was similar in design to the Minnesota system. A pumping system was necessary and the total initial cost, including \$2000 for land, was \$8000. In 1962, the system treated 23,300 gallons/acre/day and 50 lb BOD₅/acre/day. (It should be noted that the drain tile outlet at this site has since been closed.)

Several meat processing and wood pulping plants have also used the ridge and furrow treatment process.

Ridge and Furrow System Design Concerns

Before operation of a ridge and furrow treatment system, many design concerns must be considered. They include site selection, system size, operation, cell layout, and cover crop (Rodenberg, 1980).

-31-

Site selection is the initial step. A suitable site must be located through a site survey which considers soil classification, topography, and proximity to residences. Relatively permeable soils are needed to provide adequate treatment and hydraulic disposal of wastewater. Low permeability will result in wastewater ponding and high permeability will result in limited contaminant (eg. BOD₅, TKN, etc.) treatment. Soil suitability is quantified through soil borings and percolation tests.

From a construction view point, the cost of ridge and furrow installation is lower if the topography is fairly level. This also limits cuts and fills which could affect the infiltrative capacity of the soil. Control of wastewater flow is also better on flat systems.

In Wisconsin, the minimum separation distance between ridge and furrow systems and residences is 500 feet. A designer must consider this distance closely. Many proposed designs have been delayed or dropped because of homeowner challenges. Sites must also be located 250 feet from water supply wells in Wisconsin.

Hydraulic and BOD5 loading rates are currently used to size a ridge and furrow system (Rodenberg, 1980). Clayey soils should generally have hydraulic rates from 2500 to 5000 gallons/acre/day. Sandy soils may receive up to 10,000 gallon/acre/day. Hydraulic overload has historically been the primary failure at these systems in Wisconsin (WDNR, 1984). A conservative design approach is therefore recommended. Wisconsin Code NR 214 states that BOD5 loading rates at ridge and furrow

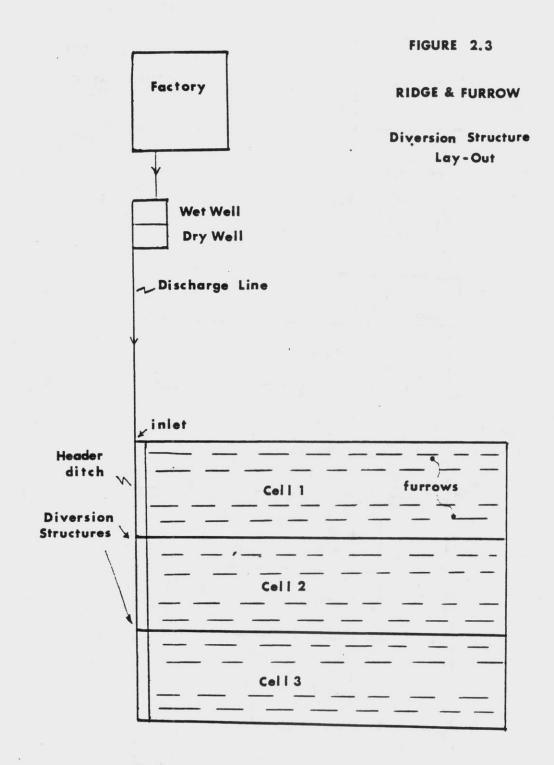
-32-

sites should not exceed 100 lb BOD5/acre/day. In summary, there are two methods to size a potential system provided that wastewater flow and BOD5 estimates are available.

Once the loading rates have been determined, the number, loading schedule, and size of cells (or sections) must be determined (Rodenberg, 1980). It is important to have more than one cell, even in small systems, to allow for alternate loading and resting. Resting helps maintain aerobic conditions beneath the furrows, upholds the adsorptive capacity of the soil, and enhances biodegradation of the wastewater. A cell should not come into service until the previously loaded wastewater has seeped away. A load/rest time period is, therefore, dependent on site soils. Cell size is affected by site topography as well and smaller cells are recommended on steeper slopes. Wastewater distribution efficiency should also be considered in cell sizing. Waste application should be uniform in all loaded furrows.

Figure 2.3 and Figure 2.4 show typical ridge and furrow layouts and furrow construction detail, respectively (Rodenberg, 1980). One typical system layout is to have a header ditch along one side of the site with all furrows, separated eight feet on center, perpendicular to it. Flow is directed to the desired cell by a diversion structure. Another possible layout is to have several header ditches crossing the furrows at right angles within each cell. Flow is directed in valved pipes. Furrows are recommended to be one foot deep, one foot wide at the bottom, and two feet wide at the top. To maintain slope stability, shallower furrow side slopes may be necessary in sandy soils.

-33-



34

.



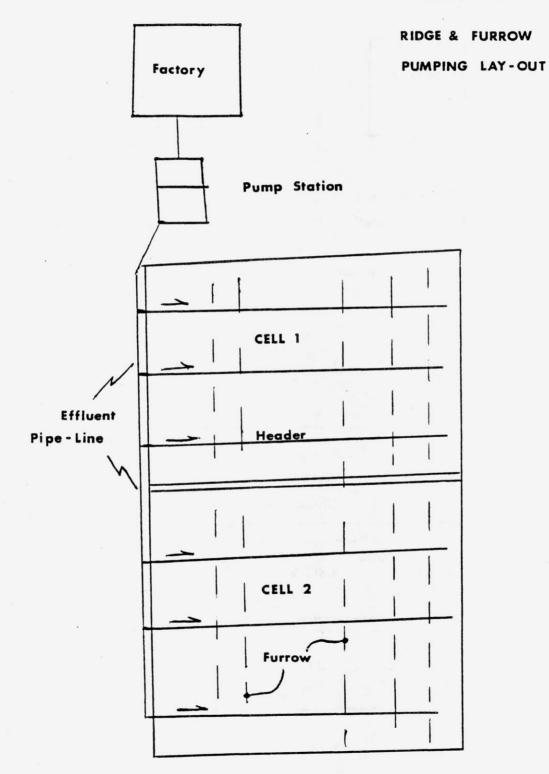
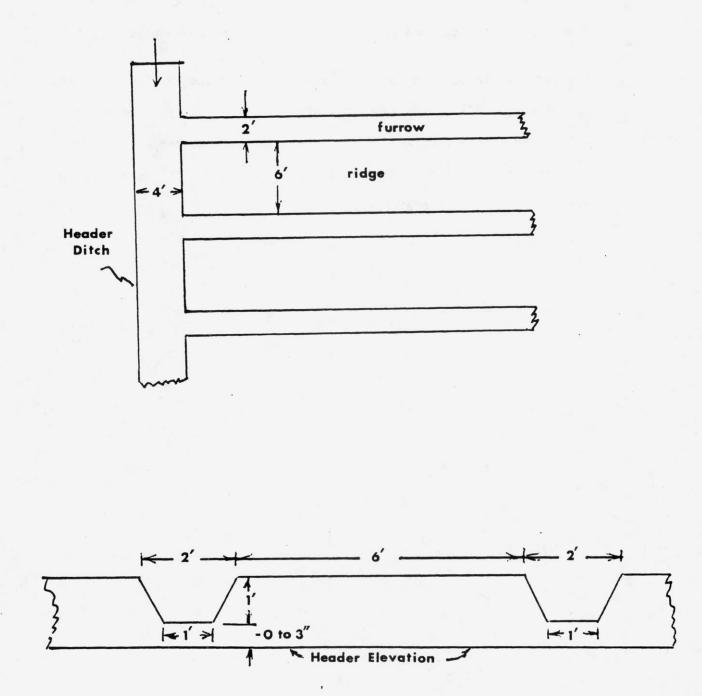


FIGURE 2.4

HEADER & FURROW DETAIL



A ridge and furrow system commonly is surrounded by a berm several feet high and cells within a site are usually separated by an embankment eight feet wide (Rodenberg, 1980). These dikes prevent surface runoff from entering the system, contain wastewater during temporary cell flooding conditions, and permit access of maintenance equipment.

A cover crop is also an important feature of a ridge and furrow system (Rodenberg, 1980). Besides maintaining ridge stability, a cover crop allows for nutrient uptake, evapotranspiration, and odor control. A crop should be able to tolerate flooded conditions. In the Midwest, reed canary grass is preferred.

CHAPTER 3: MATERIALS AND METHODS

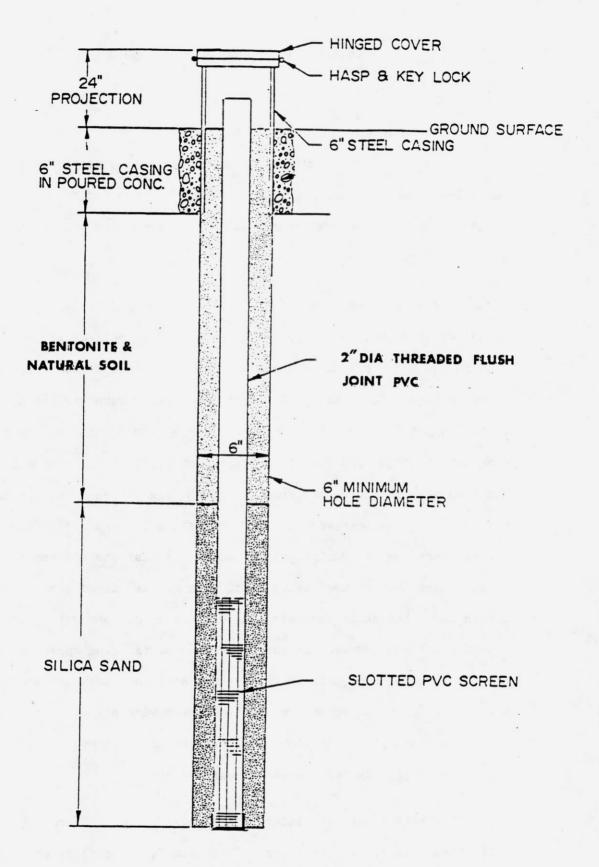
Materials

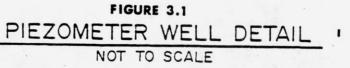
Wells, lysimeters, stage markers, and infiltration stations were installed at each site for use in gathering information for the project. Locations of this instrumentation will be discussed in the respective site chapters.

During August and September of 1983, 2 inch I.D. PVC wells were installed at each site. These wells were used to collect groundwater samples and measure groundwater elevations. A typical installation is shown in Figure 3.1. Six inch diameter boreholes were drilled to the desired depth using a rotary auger drill rig provided by the State of Wisconsin Geologic and Natural History Survey. A six inch hand auger was used at locations not accessible to the rig. Screens of two and one-half and five feet were used (slot width = 0.006 inches); the longer screens were used in shallow water-table installations and shorter screens were used in deep wells. Silica sand was placed around the screen and a bentonite seal was packed about a foot above the screen to retard migration downward of surface contaminants. The upper portion of the borehole was backfilled with natural soil and pentonite and compacted. A concrete cap and protective casing were placed at the surface to secure the well. Elevations were shot on the tops of all wells for reference in groundwater elevation measurements.

Vacuum lysimeters were installed in September and October of 1983 at both sites. These were employed to draw samples of pore water at given

-38-





depths in the unsaturated zone. A typical Teflon lysimeter installation with specifications is illustrated in Figure 3.2. A six inch hand auger was used to drill a borehole to the desired depth. The lysimeter was then lowered into the borehole and a silica flour slurry was poured around it. The silica pack allowed for a continuum between the soil and the lysimeter. After the silica pack hardened, the hole was backfilled with natural soil. A concrete cap and protective casing were installed at the surface to secure the lysimeter.

Stage markers were installed in March and April of 1984 for use in stream elevation measurements. A typical marker is shown in Figure 3.3. A metal stake was driven into the stream sediments and protected by a short length of PVC pipe. Elevations were recorded at the top of the stakes for a reference.

Infiltration stations were constructed in October and November of 1984 to determine the infiltration rate of wastewater into the unsaturated zone. A typical station, with dimensions, is shown in Figure 3.4. A station consisted of two sheets of plywood, hand-driven into a furrow, which isolated a short length of furrow.

Methods

Procedures followed during this ridge and furrow project were based on the objectives listed in Chapter 1 which were:

- 1) to determine the nitrogen transformation in the wastewater during treatment,
- 2) to perform a nitrogen budget at each site,

-40-

FIGURE 3.2

TYPICAL LYSIMETER INSTALLATION

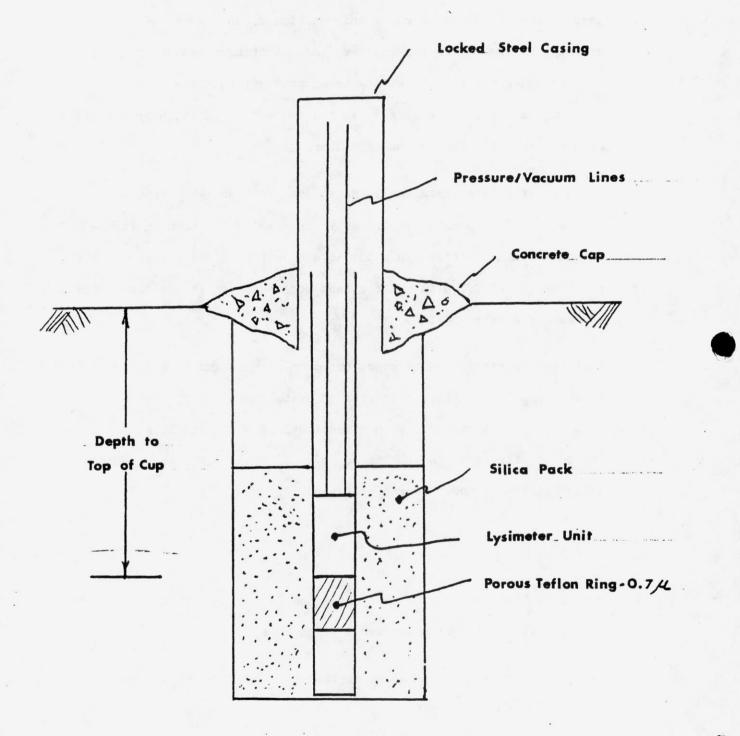


FIGURE 3.3

TYPICAL STAGE MARKER

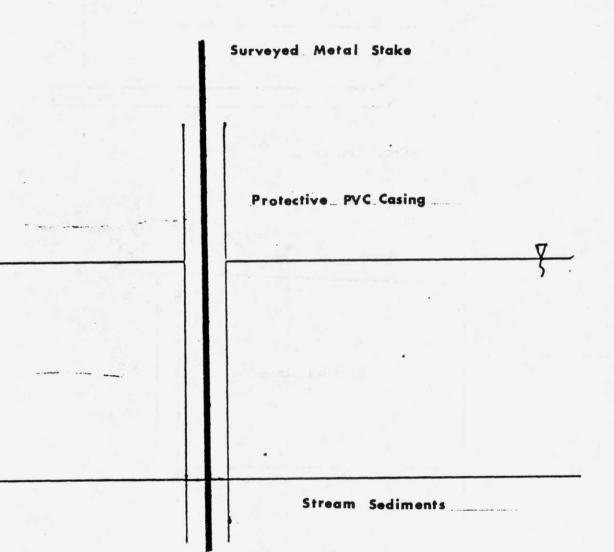
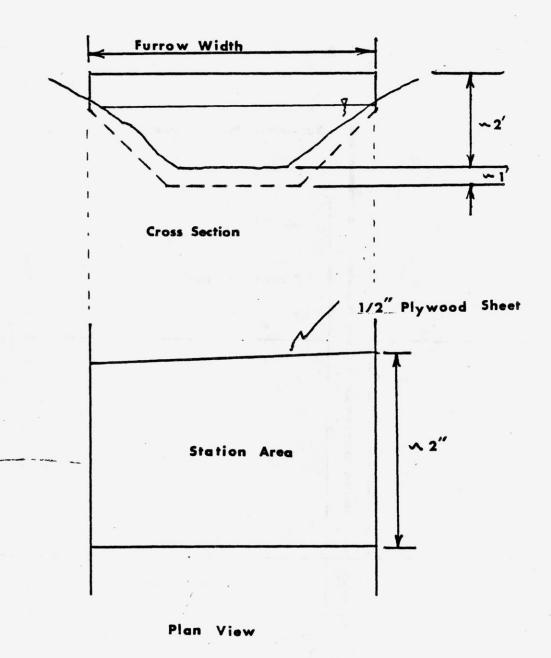


FIGURE 3.4

TYPICAL INFILTRATION STATION



- to analyze ridge and furrow treatment effectiveness under different soil and loading conditions,
- 4) to examine the operation and maintenance at these systems, and
- 5) to evaluate the monitoring equipment used.

This chapter describes these procedures.

The primary task before field work began, was to determine the number and location of well and lysimeter installations. This was done by personnel at WDNR during the summer of 1983. Wells were positioned to detect contamination migrating from each site (shallow and deep) to better define groundwater movement, and to describe background groundwater quality. Additional wells were installed later in the project based on groundwater flow and detected contamination. Lysimeters were placed in centralized areas within site cells which appeared to receive a typical wastewater loading. Background lysimeters were also installed away from the ridge and furrow systems.

Beginning in October of 1983, after initial well and lysimeter installation, a monthly field visit to each site was performed. During a typical visit, groundwater, lysimeter, and wastewater samples were collected, water table elevations were measured, and site observations were conducted.

As the project progressed and questions about the data arose, adjustments in this routine were made. In March and April of 1984, additional wells were installed at each site to better quantify the polluted area. In June of 1984, monthly sampling was discontinued at wells with

-44-

predictable or non-useful chemical trends. At Brodhead, during July-August and October-November of 1984, an intense sampling program was conducted to better quantify nitrogen transformations with depth and time during a loading cycle. July and August sampling occurred during a two week load/rest cycle and the October-November sampling took place during a one-week load/rest cycle. During these sampling periods, wells and lysimeters located inside the system and selected furrows were sampled two or three times per week. An additional well was installed in October 1984 at Brodhead to better define groundwater quality immediately beneath the site.

A complete description of field methods is presented in the following section.

Field Methods

Soil samples were taken during initial borehole drilling. Grab samples were removed from the rotary or hand auger at desired depths. Samples were sealed in labeled plastic bags for travel.

During each monthly trip, a flow composite wastewater sample (typically 24-hour) was collected. Each time the pump would run, a fraction of the flow was diverted through a hose, tapped into the discharge side of the pump, which led to a collection reservoir stored in ambient conditions. Non filtered samples were collected. Nitrogen and COD samples were acidified with sulfuric acid to pH<2; metal samples (when taken) were aci-

-45-

As composite waste samples were collected, wastewater flows to the ridge and furrow systems were also determined. At Brodhead, flow was calculated by subtracting the volume of cooling water discharge (to river) from the total production water pumped into the plant. Both of these pumps were metered. At Mindoro, this calculation was made by dividing the volume pumped by the hours of metered pumping time during sample collection. Also, 30-day monthly wastewater flow averages, as reported to the Wisconsin DNR, were recorded at both sites.

Furrow samples were taken during the intensive sampling periods at Brodhead and during October and November of 1984 at Mindoro. Grab samples taken from selected furrows were field filtered and acidified unless travel time to the lab was short (less than 1 hour).

Field filtering was done with a peristaltic pump, powered by a 12 volt D.C. battery, and pressure filter stand. Samples were filtered through a 0.45 micron filter. As described earlier, all nitrogen and COD samples were preserved with sulfuric acid to pH<2; all metals samples were acidified with nitric acid to pH<2. Samples were packed in ice for transport.

Lysimeter samples were taken monthly with accelerated collection during "intense" sampling periods at Brodhead. Twenty inches of mercury vacuum was applied with a two-way hand pump, to draw a pore water sample into the lysimeter. At Brodhead, an adequate volume of sample was obtained after a 48 to 72 hour vacuum period. At Mindoro, adequate volumes could be obtained from the operating lysimeters within 24 hours. Samples were

-46-

removed from the lysimeter's sample reservoir by pressurization with a hand pump. Samples were field filtered and acidified unless travel time to the lab was short.

Well sampling was performed monthly with accelerated collection during "intense" sampling periods at Brodhead. Wells at Brodhead were purged and sampled with a diaphragm pump or PVC bailer while wells at Mindoro were purged and sampled with a PVC bailer. Apparatus was rinsed with deionized water before purging and sampling. For quickly refilling wells, three well volumes were removed before sampling. For slowly refilling wells, water was purged until the well was dry. The latter wells were located at Mindoro and were sampled the following day. Samples were field filtered and acidified.

Periodically, upstream, midstream, and downstream grab samples were taken of neighboring rivers at each site. Non-filtered samples were acidified in the field as described earlier.

Groundwater elevations were taken each time a well was sampled. This was done with a fiberglass surveying tape with a "popper" attached. The popper was a formed metal cup which produced a pop sound as it contacted standing water within a well. Surface water elevations of neighboring streams were taken periodically after stage markers were installed.

Slug tests, as described by Cooper, Bredehoeft, and Papadopulas (1967, 1973), were attempted at each site to determine hydraulic conductivities in the saturated zone. Briefly, this method involves removing or adding a quantity of water to a well and recording the rise/fall of head with

-47-

time. These curves can be related to type curves to determine conductivity and a storage coefficient. This procedure is presented in Appendix E.

During October and November of 1984, plywood infiltration stations were constructed. A volume of wastewater was taken from a neighboring furrow and added to a dry "station." Water elevation drops were measured with time to determine the flow rate of wastewater through the furrow bottom.

Grass samples were collected during the spring, early summer, and late fall to determine plant nitrogen uptake during the growing season. Cuts were made at about two inches above the ground surface and the area of the sample was recorded. Samples were stored in paper bags for transport.

Site observations were made during each visit. These observations included 1) the extent of freezing conditions during the winter, 2) the extent of plant growth during the growing season, 3) the distribution of wastewater to the furrows, 4) cell loading, 5) the amount of solids build-up in the furrow bottoms, and 6) the operation of the monitoring equipment used.

Analytical Methods

Chemical analysis of wastewater, furrow, lysimeter, groundwater, and stream samples were performed by the Wisconsin State Laboratory of Hygiene. Complete tab procedures are described in the "Manual of Analytical Methods-Inorganic Chemistry Unit" written by the Lab of Hygiene in 1980. As mentioned before, the lab also filtered and

-48-

acidified samples if travel time from a site was short (less than 1 hour). Table 3.1 lists the frequency of analysis and sample source (waste, furrow, etc.) for each chemical parameter. Readings of pH were made in the field with a <u>Tripar Industries</u>, Inc. three-parameter digital meter.

TABLE 3.1

CHEMICAL ANALYSIS DONE

PARAMETER	SAMPLE SOURCE	FREQUENCY *
DISS BOD5	furrow, lysimeter, groundwater	4,4,3
TOTAL BOD5	wastewater, furrow, stream	1,4,4,
DISS COD	furrow, lysimeter, groundwater	2,3,3,
TOTAL COD	wastewater, stream	1,4
TSS	wastewater	1
TDS	wastewater, furrow, lysimeter, groundwater	4,4,4,1
DISS TKN	wastewater, furrow, lysimeter, groundwater	2,2,3,3
TKN	wastewater, furrow, stream	1,2,4
DISS NH3-N	wastewater, furrow, lysimeter, groundwater, stream	1,2,3,3,4
DISS NO2-N+NO3-N	wastewater, furrow, lysimeter, groundwater, stream	1,2,3,3,4
c1-	wastewater, furrow, lysimeter, groundwater, stream	1,2,3,3,4
pH	wastewater, furrow, lysimeter, groundwater, stream	1, 1, 1, 1, 1, 1
DISS ALK	groundwater	4
TOTAL ALK -	wastewater, stream	u, 4

TABLE 3.1 (Continued)

PARAMETER	SAMPLE SOURCE	FREQUENCY #
DISS TOTAL P	groundwater	4
TOTAL P	wastewater, stream	4,4
DISS SO4 ²⁻	groundwater	4
total so42-	wastewater, stream	4,4
DISS Na+	groundwater	4
TOTAL Na+	wastewater, stream	4,4
DISS K+	groundwater	4
TOTAL K+	wastewater, stream	4,4
DISS Mg ²⁺	groundwater	4
TOTAL Mg ²⁺	wastewater, stream	4,4
DISS Ca ²⁺	groundwater	4
TOTAL Ca ²⁺	wastewater, stream	4,4

*Frequencies: 1 - monthly

2 - intense periods only 3 - 1 and 24 - periodically

Soil and plant analysis was performed by the Soil & Plant Analysis Laboratory, University of Wisconsin Extension. Complete lab procedures are described in "Wisconsin Procedures for Soil Testing, Plant Analysis, and Feed and Forage Analysis" (1980). Soil samples were analyzed for percent sand, silt, clay, and total nitrogen; CEC; and pH. Plant samples were analyzed for sample weight; percent ash; percent nitrogen of dry and ash sample; and percent P, K, Ca, Mg, and S; and Zn, B Mn, Fe, Cu, Al, and Na concentrations.

-50-

CHAPTER 4: BRODHEAD SITE - RESULTS AND DISCUSSION

Site Description

The Universal Foods cheese factory is located in Brodhead, Wisconsin in eastern Green County. The plant receives 260,000 pounds of milk per day and discharges 39,500 gpd (average) of processing wastewater which is treated by a 4.7 acre ridge and furrow system consisting of two cells. This treatment system began operation in 1972. Figure 4.1 is a topographical map showing the general location of the system.

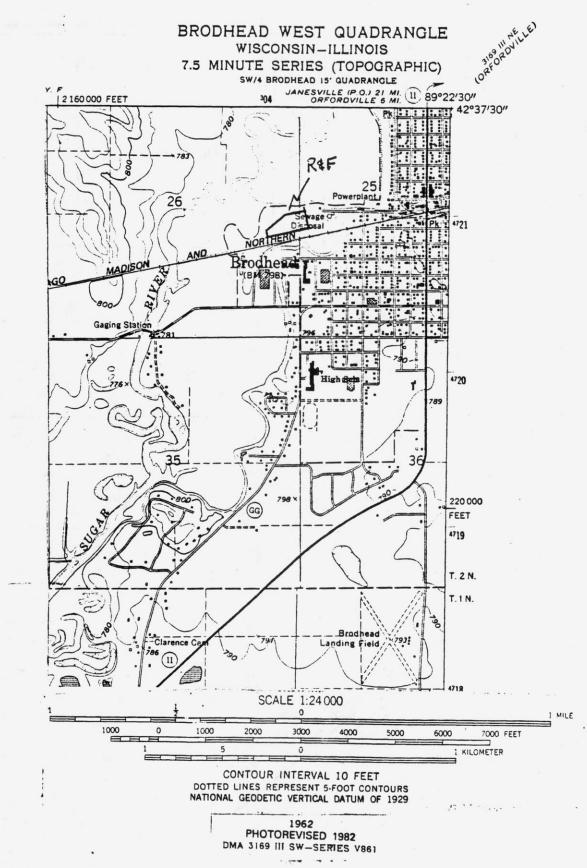
A plan view of the treatment system is illustrated in Figure 4.2. Well, lysimeter, stream stage, and infiltration station locations are indicated as well as cell locations and areas. General information concerning well and lysimeter depths and location is presented in Table 4.1. Complete well and lysimeter logs are given in Appendix A.

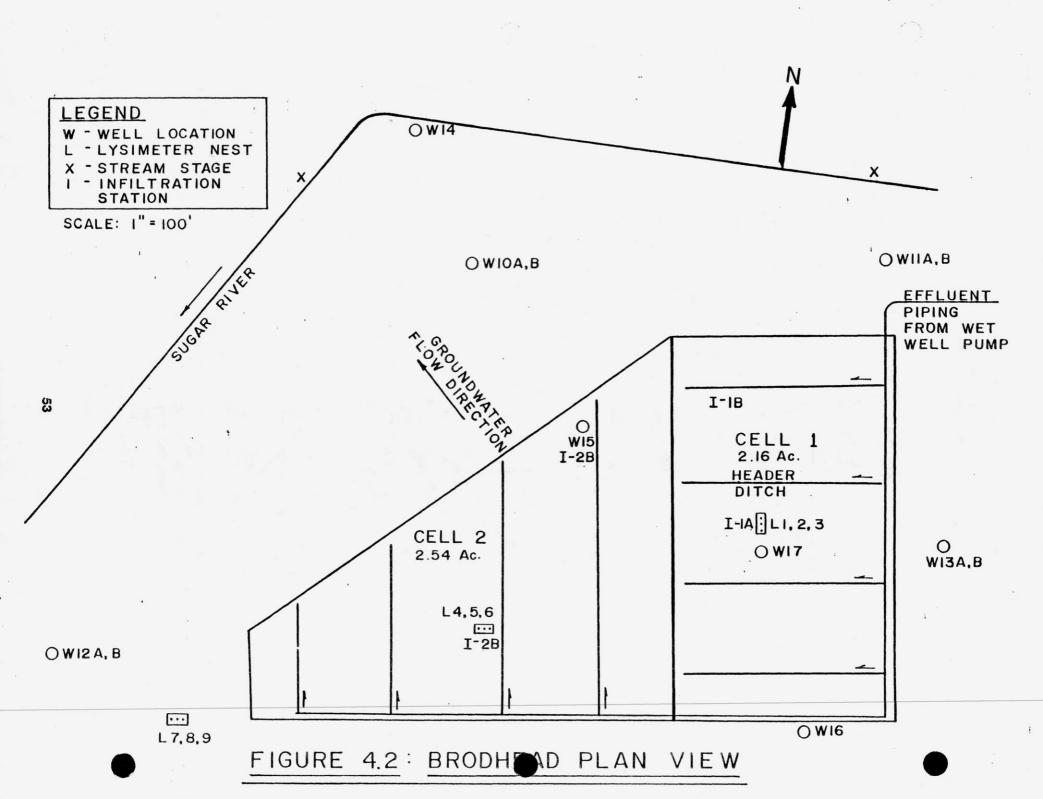
The ridge and furrow site is located on an unconfined aquifer consisting of glacial outwash material along the Sugar River. This sandy material extends approximately 70 feet deep and overlays a sandstone aquifer. The Soil Conservation Service describes this Maumce, Orion soil as a poorly drained sandy loam soil over a fine to medium sand. It is formed on low stream terraces and somewhat poorly drained soils formed in silty alluvium (DNR, 1984).

Results of the soil analysis at Brodhead indicated the following parameter ranges:

-51-

FIGURE 4.1





-sand: 81-99% -silt: 0-16% -clay: 1-5% -Total N: 0-0.11% -CEC: 1-13 meg/100g soil -pH: 6.5-8.6

The higher silt, clay and total N fractions occurred in shallow samples. Also CEC tended to decrease with depth and soil pH tended to increase with depth. Complete soil analysis data are presented in Appendix B. After reviewing the data, it may be concluded that the soil below the treatment system is predominantly sand with an average CEC of 3 meg/100g.

Soil borings during well and lysimeter installation also produced the following information. First, profiles outside the system showed a one to four foot silty topsoil layer overlying the sand. Second, cell 1 borings indicated a dark saturated organic layer beneath the furrows. This layer was absent in the extremities of cell 1 and was about one foot thick in the center of the cell. Finally, cell 2 only had these organic layers around the header ditches. The largest thickness found under this cell was three inches.

WASTEWATER CHEMISTRY

Universal Food's wastewater was strong with an average BOD_5 of 1780 mg/l, COD of 2390 mg/l, TKN of 42 mg/l, and chloride of 930 mg/l. Means, medians, and ranges of these and other chemical parameters of the

-54-

wastewater are provided in Table 4.2. Values of a strong typical domestic waste are provided for comparison. A complete tabulation of data is provided in Appendix C.

TABLE 4.1a

WELL SPECIFICATIONS

AT BRODHEAD SITE

WELL*	WELL TOP ELEVATION	DEPTH (ft)	WELL POINT ELEVATION	APPROXIMATE SURFACE ELEVATION	SCREEN LENGTH(ft)	LOCATION
10A	778.37	13.2	763.37	776.6		
10B	777.86	26.2	750.36	776.6	2.5	Downstream
11A	775.80	8.8	765.80	774.6	774.6 5	
11B	776.41	24.0	750.61	774.6	2.5	Adjacent to cell 1
12A	778.06	9.9	766.06	776.0	5	Background
12B	777.37	26.1	749.87	776.0	.0 2.5	
13A	776.75	9•3	766.75	776.0	776.0 5	
13B	777.40	26.3	749.90	776.0	776.0 2.5	
14	776.29	5.8	769.00	774.8 2.5		Downstream
15	780.15	8.5	769.80	778.3 2.5		Cell 2
16	780.05	11.0	769.99	781.0 2.5		Upstream of cell 1
17	780.18	9.8	770.35	778.7	2.5	Cell 1

*All Wells PVC, 2 inch inside diameter

-55-

TABLE 4.1b

LYSIMETER SPECIFICATIONS

AT BRODHEAD SITE

LYSIMETER	DEPTH BELOW* FURROW (FT)	LOCATION	
1	1.0	Cell 1	
2	3.0	Cell 1	
3	4.8	Cell 1	
4	1.0	Cell 2	
5	1.8	Cell 2	
б	3.6	Cell 2	
7	1.7	Background	
8	3.7	Background	
9	4.7	Background	
•		1	

* Depth is to top of teflon cup

• • •

-56-

TABLE 4.2

BRODHEAD WASTEWATER CHEMISTRY

PARAMETER	# OF SAMPLES	MEAN	MEDIAN	RANGE	SD	STRONG TYPICAL DOMESTIC
Total BOD5	11	1780#	1700	980-3200	650	400
Total COD	9	2390#	2300	2000-3400	440	1000
TSS	10	876	869	464-1570	344	350
Total TKN	11	42	40	28-78	13	85
Dissolved TKN	3	26	21	21-37	9	
NH3-N	11	2.5	2.5	1.4-4.1	0.7	50
NO2-N+NO3-N	11	2.7	2.7	0.2-5.8	1.9	0
C1-	11	930	890	32-2300	700	100
рН 	7	7.5	7.4	6.6-9.4	0.9	

- All units mg/l except pH; typical domestic values from Metcalf and Eddy, . 1979

* - Average contains samples which exceeded detection limit, see Appendix C for specific days

SD - Standard Deviation

In addition to the wastewater's general high strength, four other observations were made. First, the chemical data were highly variable. For example, the average chloride was 930 mg/l but the standard deviation was 700 mg/l and the range was 32-2300 mg/l. These variations, which did but occur seasonly, were most likely due to the changes in the amount of rinse water used in the plant. Days of higher rinsing resulted in lower concentrations and vice versa.

Second, the wastewater was high in chloride and sodium content (930 and 843 mg/l, respectively). The source of this brine was from the salt drippings resulting from cheese block formation and from the water softener. It is suggested that waste pretreatment or an in-plant process change be considered to limit the amount of brine pumped to the ridge and furrow system. As will be discussed later, the chloride concentrations in the groundwater downstream from the site were high.

Third, the pH of the wastewater was slightly above neutral(7.5 average). Since volatilization of ammonia predominant at pH > 9, as stated in Chapter 2, one would expect little loss of ammonia from the wastewater.

Fourth, the nitrogen fraction of the wastewater was principally organic-N. This was expected since the wastewater was derived from milk which contains protein. About one-half of this organic -N was in the solid fraction. This was seen by comparison of total and dissolved TKN.

Wastewater Hydraulic Loading

All flow to the ridge and furrow system was derived from cheese production. Flow was calculated by subtracting the cooling discharge (to the

-58-

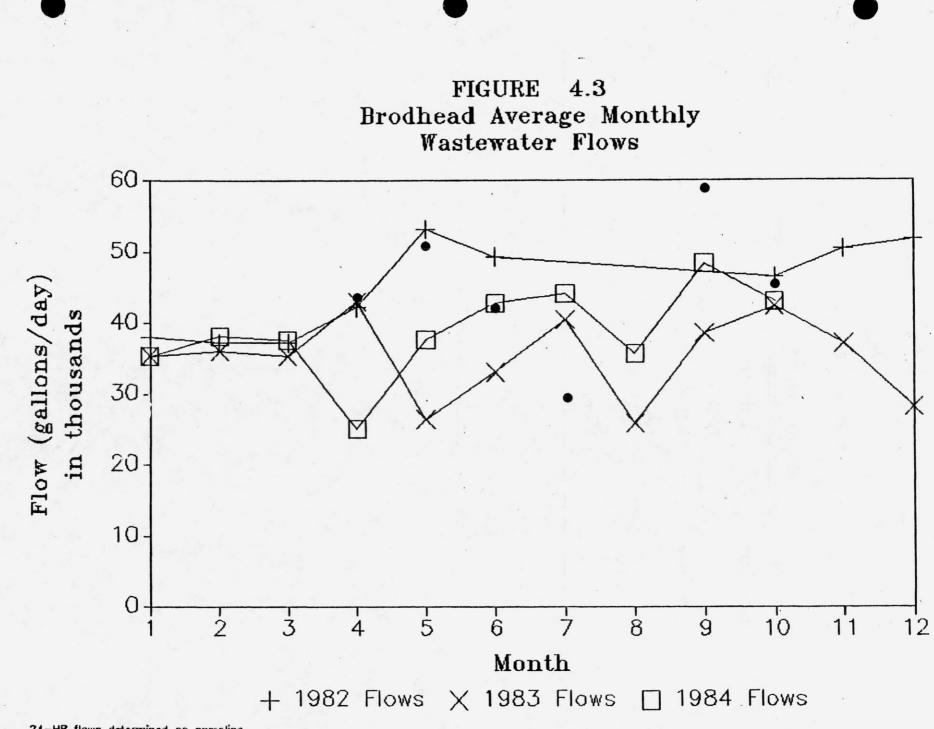
Sugar River) from the total water pumped into the production end of the plant. Both pumps were metered. Drinking and lavatory water was obtained from the city and these wastewaters were discharged to the city treatment plant.

Figure 4.3 illustrates the 30-day average wastewater flow values for the years 1982-1984 as well as 24-hour flows measured on project sampling days. A complete listing of flow data for the past six years is given in Appendix D. The 30-day monthly flows averaged 39,500 gallons per day since 1979. One can see from Figure 4.3 that from 1982 to 1984, flows ranged from 25,056 to 48,414 gallons per day. This was well under the DNR permit flow of 55,000 gpd. Flows during the years 1979 to 1981 were more variable with ranges from 12,700 to 61,500 gpd.

Prior to 1980, cell 1 received most of the wastewater with cell 2 serving as a back-up during high flow periods or when odor problems occurred. After 1980, a flexible 2-week load/2-week rest cycle was employed. A strict 2-week load/rest cycle began on 7/9/84. This scheme was followed until 10/15/84 when a week to week load/rest cycle was instituted. After 11/20/84, cell 2 was loaded for two weeks while cell 1 was loaded for one week. This last change was made based on project observations and will be discussed later. As mentioned in Chapter 2, load/rest cycles help aerate the soil and promote wastewater treatment.

In ridge and furrow systems, as well as other land treatment sites, it is useful to look at hydraulic loading rates in terms of inches per day or gallons/ acre/day. These rates for Brodhead shown in Table 4.3. Using the total site area, the average hydraulic loading rate was

-59-



24-HR flaws determined on sampling days indicated by ${\color{black} \bullet}$

TABLE 4.3

BRODHEAD HYDRAULIC LOADING

FLOW (GAL)	TOTAL AREA LOADED	CELL 1 LOADED	CELL 2 LOADED		
25,056 (min)	0.196	0.427	0.363		
	(5,330)	(11,600)	(9,860)		
39,500 (ave)	0.310	0.673	0.573		
	(8,400)	(18,290)	(15,550)		
48,414 (max)	0.379	0.825	0.702		
	(10,300)	(22,410)	(19,060)		

Cell 1 = 2.16 acres

Cell 2 = 2.54 acres

Units inches/day; or (units) gallons/acre/day

-61-

0.31 inch/day (8400 gpad) with a range of 0.196-0.379 inch/day (5,330-10,300 gpad). This is classified as a high rate system (Rodenberg, 1980). Since single cells were loaded during a load/ rest cycle, single cell hydraulic loading rates are also shown in Table 4.3 for comparison. These rates were about double the total area rates.

Organic Loading Rates (BOD5, TKN)

Code NR 214 of the Wisconsin DNR states that a ridge and furrow system should receive no greater than 100 lb BOD5/acre/day. Using the average flow rate and BOD5 concentration of the wastewater, however, the Brodhead site received 125 lb/acre/day. Using the minimum and maximum hydraulic rates and the average wastewater BOD5 concentration, the BOD5 loading rate range was 79-153 lb/acre/day. These numbers, as well as individual cell BOD5 loading rates, are given in Table 4.4.

TABLE 4.4

BRODHEAD ORGANIC LOADING RATES (1b/day/acre)

LOAD	TOTAL AREA LOADED	CELL 1 LOADED	CELL 2 LOADED
BOD5(min-flow)	79	172	146
BOD ₅ (ave-flow)	125	271	231
BOD ₅ (max-flow)	153	333	283
TKN(min-flow)	1.9	4.1	3.4
TKN(ave-flow)	2.9	6.4	5.4
TKN(max-flow)	3.6	7.8	6.7

Ave. BOD_5 Conc = 1780 mg/l

Ave. TKN Conc = 42 mg/l

Similar calculations for TKN loading rates were performed and are also presented in Table 4.4. There is currently no Wisconsin DNR code for nitrogen loading. Using the range of hydraulic flows presented in Table 4.3 and the average wastewater TKN concentration, TKN loading rates at Brodhead ranged from 1.9 to 3.6 lb/acre/day with an average of 2.9 lb/acre/day. Suggestions for possible loading rates will be made later in this report.

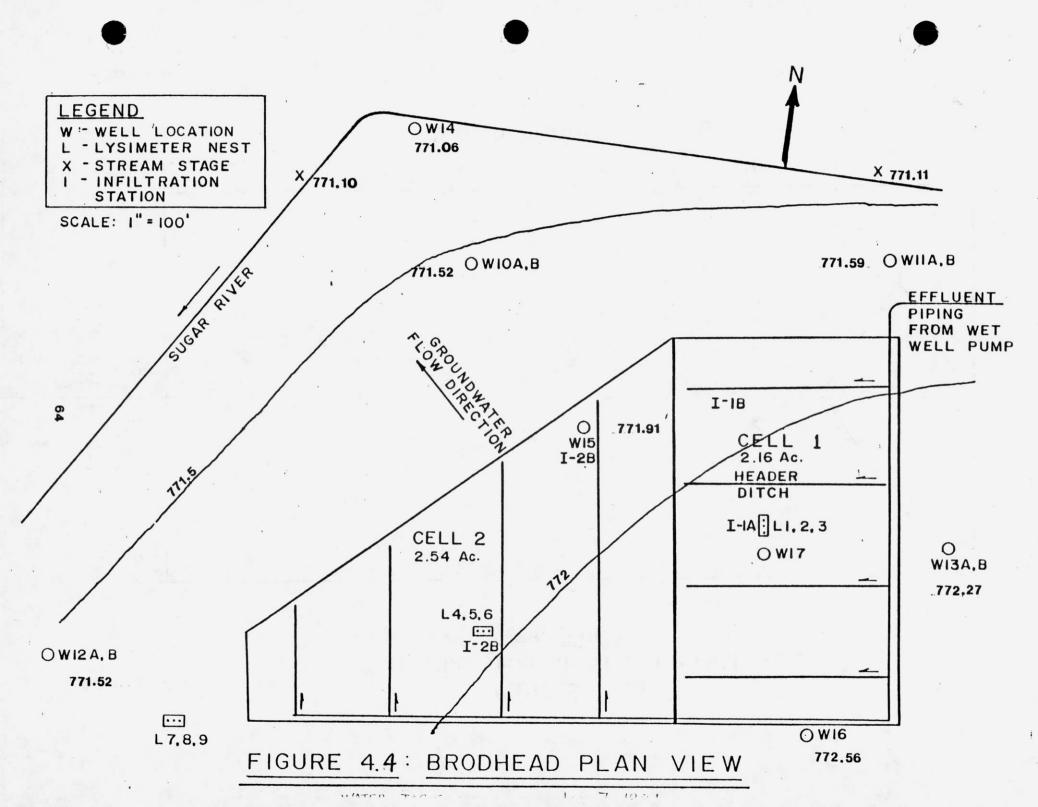
Groundwater Elevations and Flow

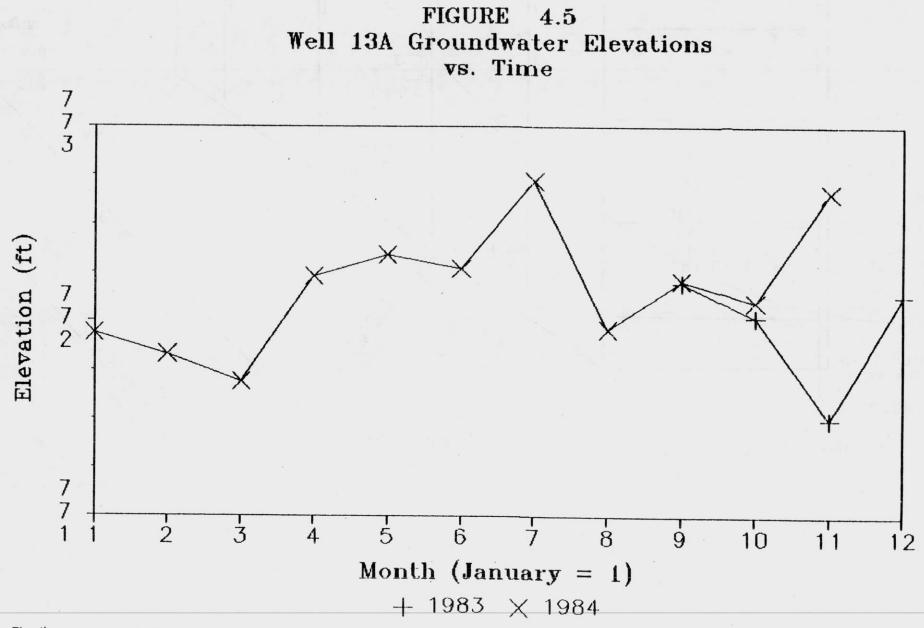
In general, groundwater flow at the Brodhead ridge and furrow was northwesterly toward the Sugar River with a gradual gradient of 0.0025 ft/ft from well 16 to the river. With the exception of well nest 11, no vertical gradients were indicated. Wells 11A and 11B were located within 50 feet of the Sugar River, which was considered a discharge zone boundary for this groundwater flow system. This was verified by the higher head readings in well 11B (than 11A), indicating upward flow gradients into the river.

Since a well nest was not located inside the system, it was not possible to determine whether downward gradients existed due to mounding. Figure 4.4 shows the groundwater contours measured on June 7, 1984. Appendix E contains a complete list of elevation and contour data.

Groundwater (water table) elevations tended to fluctuate with seasonal recharge but not with changes in wastewater flow. Readings decreased during the winter of 1984, increased during the spring (1984) thaw, decreased during the drier late summer months, and increase again during fall rains. This pattern is shown for well 13A in Figure 4.5.

-63-





Elevations mean sea level

The water table response to rainfall events was rapid occurring within one to two days. This is illustrated in Figure 4.6 where head readings in wells 15 and 17 were plotted against time during October and November of 1984. These similar responses were not due to cell loading since groundwater elevations in different cells responded identically to the rainfall. It should also be noted that a two inch rainfall is over three times the average single cell loading rates and that rainfall can also enter the unsaturated zone from the ridges.

Another task in defining the groundwater hydraulic characteristics was to determine horizontal flow velocities. This was done using Darcy's Law:

V = KI/n

where:

V = average linear groundwater velocity (L/T), K = hydraulic conductivity of aquifer (L/T), I = hydraulic gradient dH/dL (L/L), and n = porosity.

Unsuccessful slug and bail tests to determine K were attempted on wells 10B, 11B, and 12B at Brodhead in April of 1984. The wells returned to equilibrium too quickly to acquire meaningful data. Therefore a value of 0.0005 ft/s, obtained from a local pumping test record on file at the Wisconsin DNR, was used for hydraulic conductivity. As mentioned earlier, the aquifer beneath the ridge and furrow is principally sand. For this project, the aquifer was assumed to be homogeneous and isotropic. (This meant that K did not vary in space or direction.) A

-66-

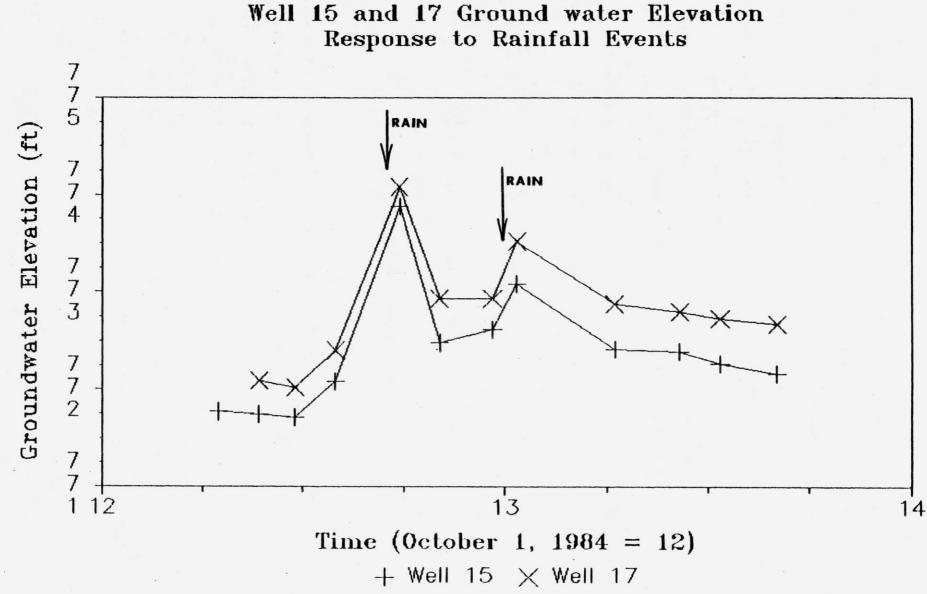


FIGURE 4.6

67A

hydraulic conductivity of 0.0005 ft/s is reasonable for this type of medium (see Freeze and Cherry, 1979).

Two different hydraulic gradients were used. For flow between wells 16 and 10, I was 0.0018 ft/ft; for flow between wells 10 and 14, I was 0.0046 ft/ft. It is reasonable to find that gradients increase near surface waters. A porosity of 0.35 for a typical sand was used (Freeze and Cherry, 1979).

With the above input data, velocities of 0.22 ft/day between wells 16 and 10 and 0.57 ft/day between wells 10 and 14 were calculated. These velocities gave the following travel times:

- from W16 to W17: 2.5 years,

- from W17 to W15: 2.7 years,

- cell 1 boundary to W15: 1.9 years,

- from W15 to W10A: 2.5 years, and

- from W10A to W14: 0.5 years.

One would expect the travel times in this sandy medium to be shorter but since the hydraulic gradient was shallow, the travel times were longer. One should realize, however, that the K value used could realistically be off by an order of magnitude, altering the travel times by a factor of ten.

Groundwater Chemistry

Mean and standard deviations for selected parameters ab each well for Brodhead are listed in Table 4.5. Complete data listings are given in Appendix F. When looking at the chloride data, a good indicator of con-

-67-

TABLE 4.5

MEAN AND STANDARD DEVIATION OF

GROUNDWATER CHEMICAL PARAMETERS

AT BRODHEAD

WELL	DISSOLVED BOD5	DISSOLVED COD	TDS	DISSOLVED TKN	DISSOLVED NH ₃ -N	DISSOLVED NO ₂ -N+NO ₃ -N	C1-	FIELD pH
10 A	7.5 <u>+</u> 4.6(13)*	17 <u>+</u> 2.3(13)	1020 <u>+</u> 163(12)	13 <u>+</u> 2.0(13) 13 <u>+</u> 1.8(13) 0.5 <u>+</u> 0.4(13) * 380 <u>+</u> 92(14)		6.9 <u>+</u> 0.13(8)		
10B	10 <u>+</u> 6.5(13)*	24 <u>+</u> 6.9(13)	1680 <u>+</u> 157(12)	12 <u>+</u> 6.8(13)	$12\pm 6.8(13)$ $12\pm 6.9(13)$ $0.3\pm 0.4(13)$ $630\pm 120(13)$		630 <u>+</u> 120(13)	6.8 <u>+</u> 0.09(8)
11A	4.0 <u>+</u> 1.7(13) *	23 <u>+</u> 7.6(13)	516 <u>+</u> 151(12)	3.3 <u>+</u> 0.58(13)	2.8 <u>+</u> 0.55(13)	0.1 <u>+</u> 0.0(13) *	140 <u>+</u> 63(14)	6.8 <u>+</u> 0.18(8)
11B	6.0 <u>+</u> 5.4(13)#	11 <u>+</u> 16(13)#	434 <u>+</u> 256(12)	0.41 <u>+</u> 0.31(13)	0.27 <u>+</u> 0.24(13)	0.1 <u>+</u> 0.0(13) *	73 <u>+</u> 110(13)	7.1 <u>+</u> 0.15(8)
12A	3.1 <u>+</u> 0.89(8)#	9.5 <u>+</u> 7.0(8)*	373 <u>+</u> 45.0(7)	0.50 <u>+</u> 0.63(8)	0.10 <u>+</u> 0.0(8)	8.6 <u>+</u> 1.6(8)	27 <u>+</u> 3.6(8)	7.0 <u>+</u> 0.15(3)
12B	2.8 <u>+</u> 0.47(9)*	5.4 <u>+</u> 0.73(9)	383 <u>+</u> 42.6(8)	0.20 <u>+</u> 0.0(9)#	0.10 <u>+</u> 0.0(9)*	10.8 <u>+</u> 2.70(9)	38 <u>+</u> 1.4(9)	7.5 <u>+</u> 0.10(3)
13A	2.8 <u>+</u> 0.61(13)*	23 <u>+</u> 6.5(13)	313 <u>+</u> 95.6(12)	0.83 <u>+</u> 0.22(13)	0.18 <u>+</u> 0.07(13)	0.38 <u>+</u> 0.51(13)*	3.8 <u>+</u> 2.0(14)	7.0 <u>+</u> 0.18(8)
13B	2.8 <u>+</u> 0.66(11)*	5.1 <u>+</u> 0.3(11)	317 <u>+</u> 15.4(10)	0.20 <u>+</u> 0.0(11)*	0.10 <u>+</u> 0.0(11)*	4.2 <u>+</u> 0.33(11)	35 <u>+</u> 2.3(11)	7.2 <u>+</u> 0.16(6)
14	5.2 <u>+</u> 5.2(8) *	18 <u>+</u> 7.4(8)	648 <u>+</u> 111(8)	1.8 <u>+</u> 0.64(8)	1.4 <u>+</u> 0.83(8)	1.4 <u>+</u> 1.6(8)	190 <u>+</u> 68(8)	7.1 <u>+</u> 0.08(7)
15	51 <u>+</u> 55(12) *	57 <u>+</u> 50(19)	1640 <u>+</u> 216(12)	16 <u>+</u> 21(23)	15 <u>+</u> 20(23) *	17 <u>+</u> 21(23) *	570 <u>+</u> 110(23)	6.4 <u>+</u> 0.12(8)
16	57 <u>+</u> 57(12) *	83 <u>+</u> 62(19)	1600 <u>+</u> 436(12)	5.0 <u>+</u> 2.5(20)	3.9 <u>+</u> 2.6(20)	0.10 <u>+</u> 0.0(20) *	580 <u>+</u> 180(20)	6.4 <u>+</u> 0.10(7)
17	34 <u>+</u> 37(2)*	54 <u>+</u> 5.8(7)	2540 <u>+</u> 643(2)	31 <u>+</u> 2.6(11)	30 <u>+</u> 1.8(11)	0.92 <u>+</u> 0.27(11)*	650 <u>+</u> 77(11)	

means contain data that was above or below a detection limit; limit was used in average

All values mg/l except pH; () indicates # of observations.

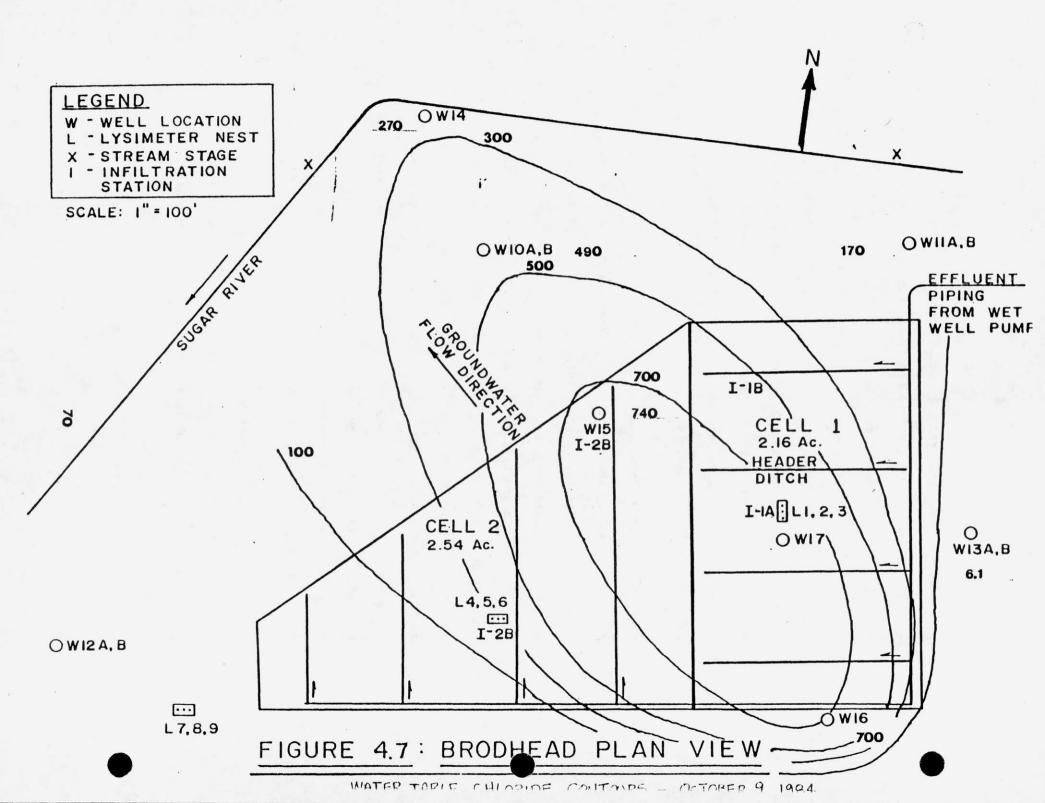
-68-

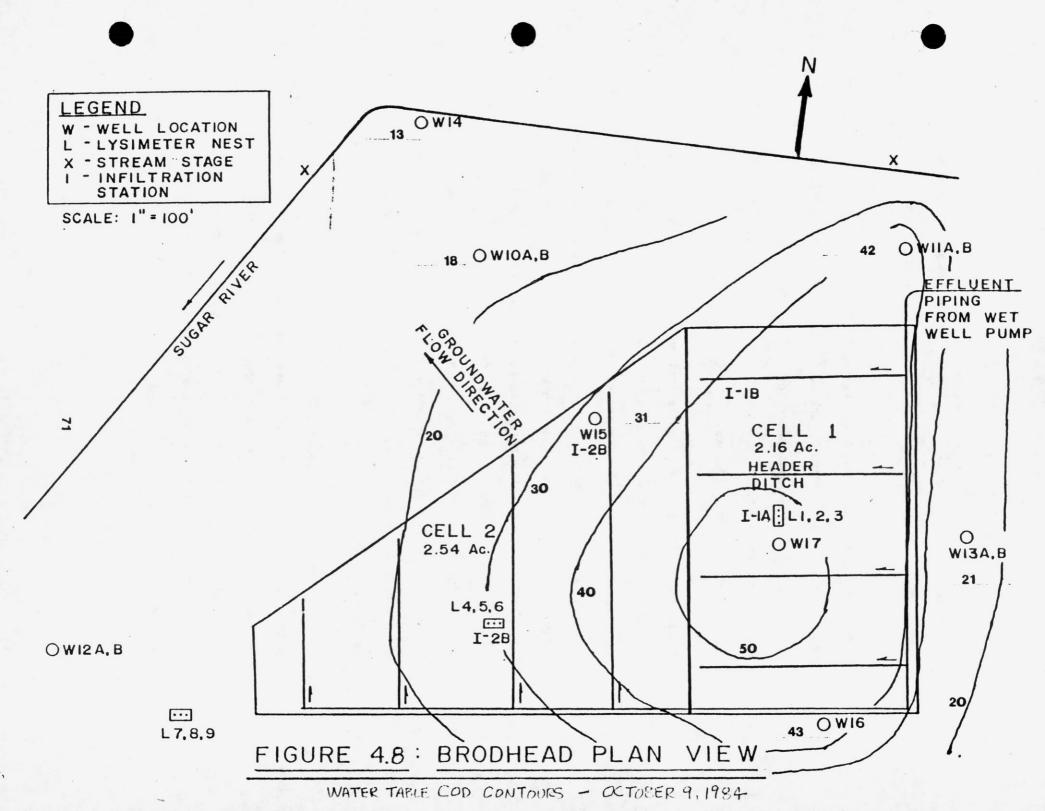
tamination, one can catagorize the wells into three types. Wells 12A, 12B, 13A, and 13B were not influenced by the ridge and furrow system; wells 11A, 11B, and 14 were moderately affected by the site, and wells 10A, 10B, 15, 16 and 17 were highly impacted by the system. This grouping matched the observed flow pattern. It should be realized that averages and standard deviations of well data at this site do not fully describe the contamination. Parameters were variable with time which resulted in high standard deviations (eg. W15 TKN). Averages were only used for relative comparisons.

Figure 4.7 presents a plan view of chloride contours at the water table on October 9, 1984. The contours are approximate but they generally represent the contaminated area. Well 16, which was installed on the south berm, exhibited high chloride concentrations throughout the project. Even though the well was upgradient of the system, it was close enough to be affected by dispersion (or diffusion). Also, in well 10B, at 21 feet depth below the water table, chloride concentrations were about 1.5 times greater than well 10A concentrations at the water table during the study. Well 10B and well 15 values were of the same relative magnitude, indicating that the contaminant was sinking as it traveled downstream due to the density of the plume. This was reasonable since the linear groundwater velocity was slow.

A similar contour pattern existed for COD concentrations at the water table and is shown in Figure 4.8. Again, note the elevated value for well 16 and the deep well 10B concentration relatively higher than the surface well 10A.

-69-



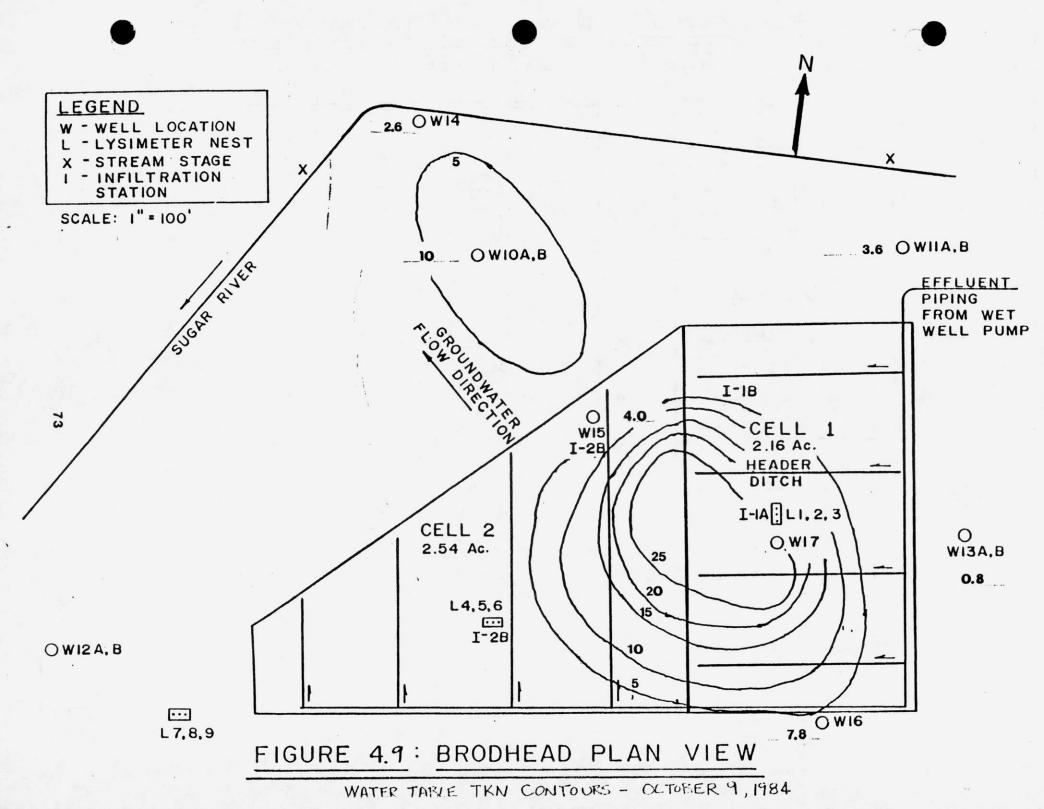


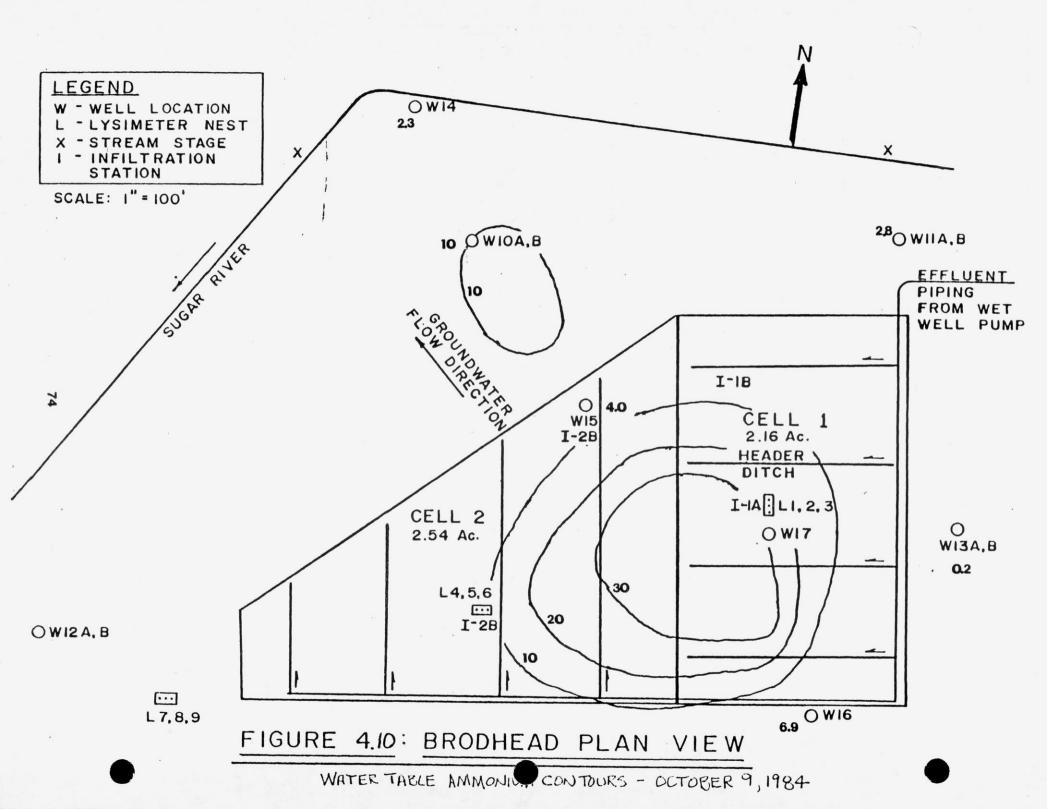
Plan views for the parameters TKN, NH_3-N , and NO_2-N+NO_3-N (from 10/9/84) are shown in Figures 4.9, 4.10, and 4.11, respectively. The TKN and NH_3-N maps are quite similar with high concentrations underneath the middle of the system, a sag area near well 15, and relatively high concentrations near well 10A. Figure 4.11 shows high nitrate values in the vicinity of well 15. Again, note that TKN and NH_3-N values in wells 16 and 10B were relatively high but wells 15 and 10B do not correspond as they did for COD and chlorides.

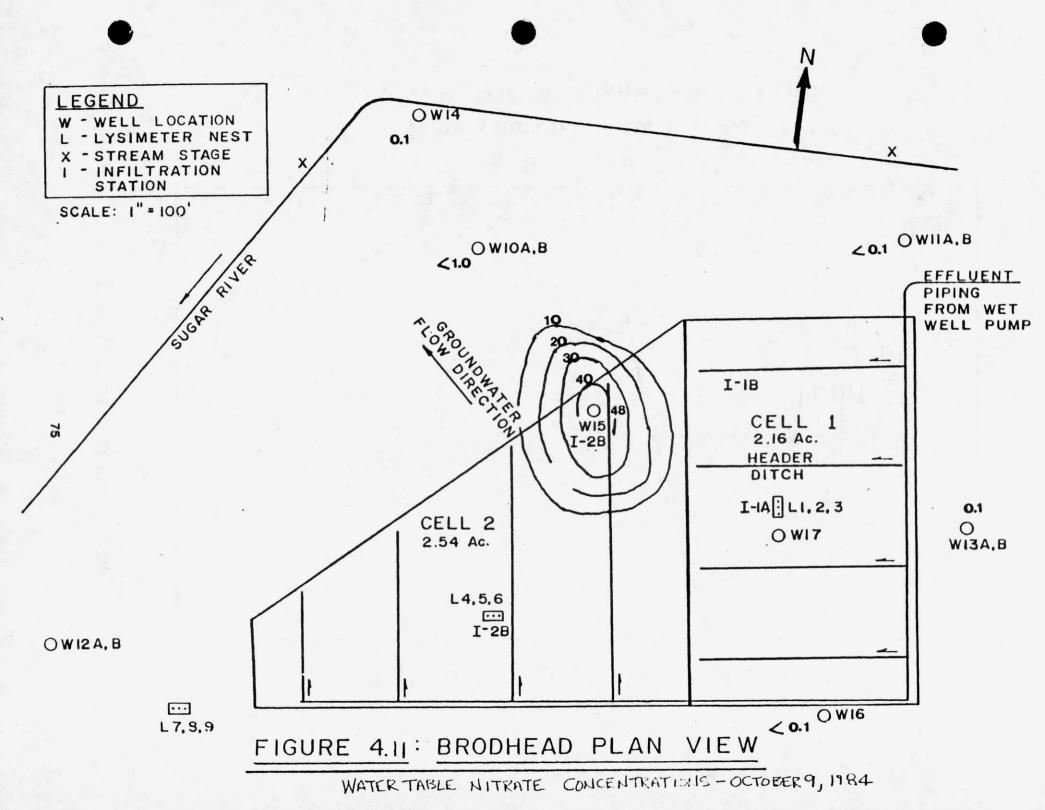
Previous to 10/9/84, nitrate concentration in well 15 were low and ammonium was the principal form of nitrogen. In early October (1984), nitrification of this ammonium in the unsaturated zone began, as depicted by the nitrate contours on Figure 4.11. These contours were estimates and more wells would have been necessary to completely define the nitrifying area. This nitrification is also responsible for the sag areas shown in Figures 4.9 and 4.10. Cell 2 was loaded (to allow cell 1 to completely dry) for four weeks prior to 10/9/84 and this wastewater flushed out a majority of soil pore water nitrate into the groundwater. Nitrate concentrations subsequently decreased at well 15 in late October and November (see Figure 4.12).

The pH in the background wells (12A, 12B, 13A, 13B) was slightly above neutral, ranging from 7.0 to 7.5 on average. Both wells 15 and 16, located within the system, had a pH of 6.4 on average. Downgradient wells (10A, 10B, 11A, 11B, and 14) exhibited pH values nearing neutrality, indicating dilution of this groundwater by water of background quality.

-72-







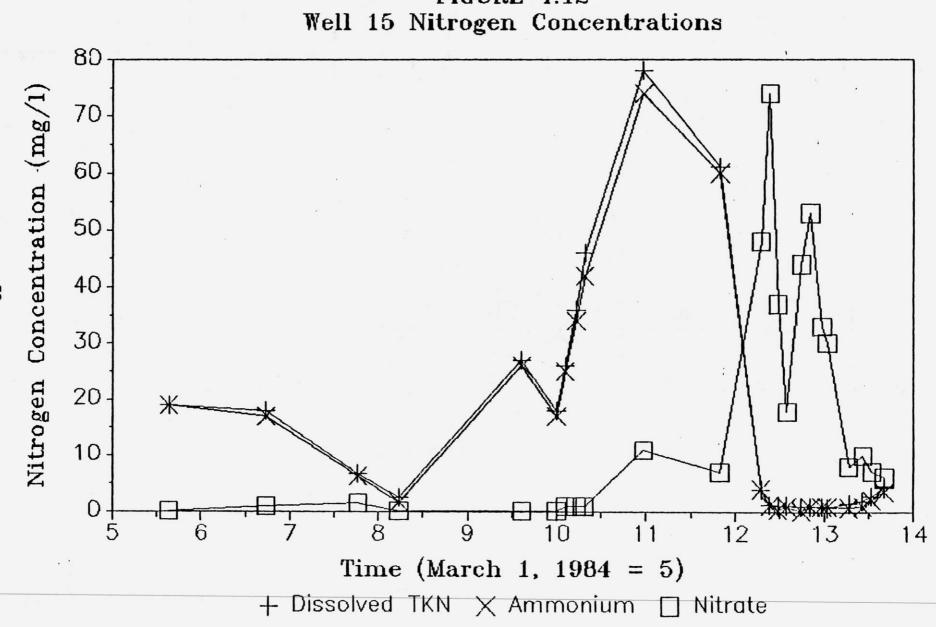


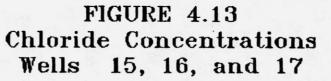
FIGURE 4.12

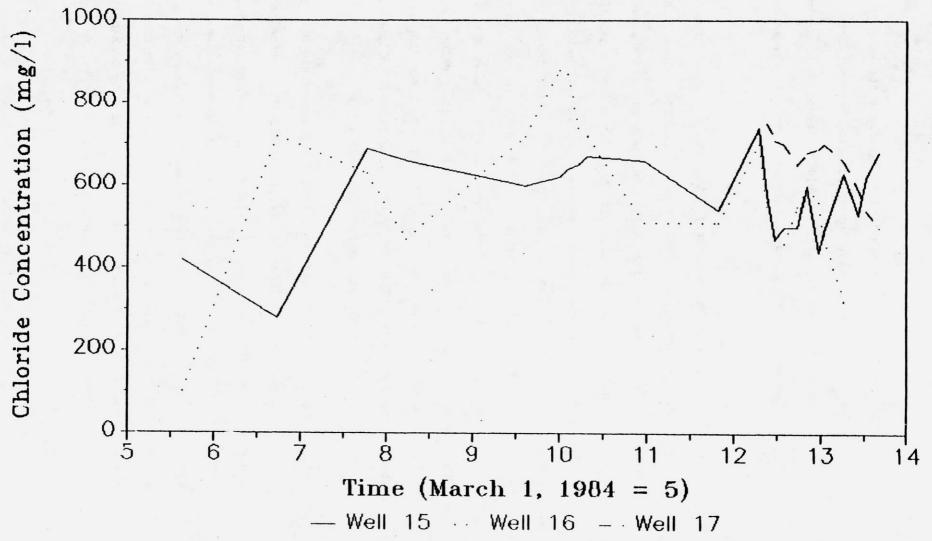
An attempt was made to match chemical changes in space with time between wells. Due to the slow groundwater velocities, however, this was not possible. A longer sampling period would be needed to properly evaluate the data. General observations of temporal trends in the groundwater follow.

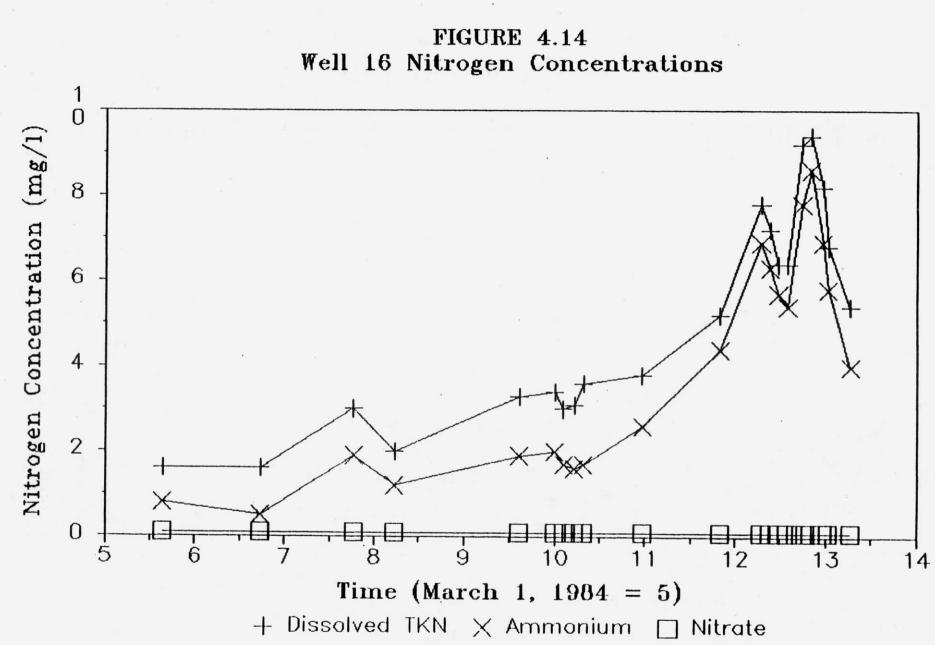
Wells 16 and 17 responded similarly with time. This was expected since both wells were directly affected by cell 1 loading. Direct increases in C1⁻ or N concentration with cell 1 loading were not detected, however. Chloride concentrations with time for wells 16 and 17 appear in Figure 4.13. Nitrogen concentrations with time for well 16 appear in Figure 4.14; Figure 4.15 presents nitrogen concentrations with time for well 17. Chloride data from wells 16 and 17 both had concentration decreases in late October (1984) and early November (1984) with a peak on November 1. These decreases were the result of heavy rains on October 18 and October 31. Note that well 16 values were lower than well 17 due to background dilution. TKN and NH₃-N concentrations followed a similar pattern. Little nitrate was present in these wells.

Well 15 (cell 2) chloride and nitrogen concentrations did not respond in the same fashion as well 16 and 17 values. Well 15 chloride concentrations are plotted in Figure 4.13; nitrogen values for well 15 are presented in Figure 4.12. The most obvious difference between well 15 and wells 16 and 17 was the nitrogen transformation which was occurring in well 15 during October of 1984. Well 15 nitrogen concentrations, which were quite high, changed from primarily NH_3-N to primarily NO_3-N . Nitrogen values then reduced to concentrations of relatively good

-77-







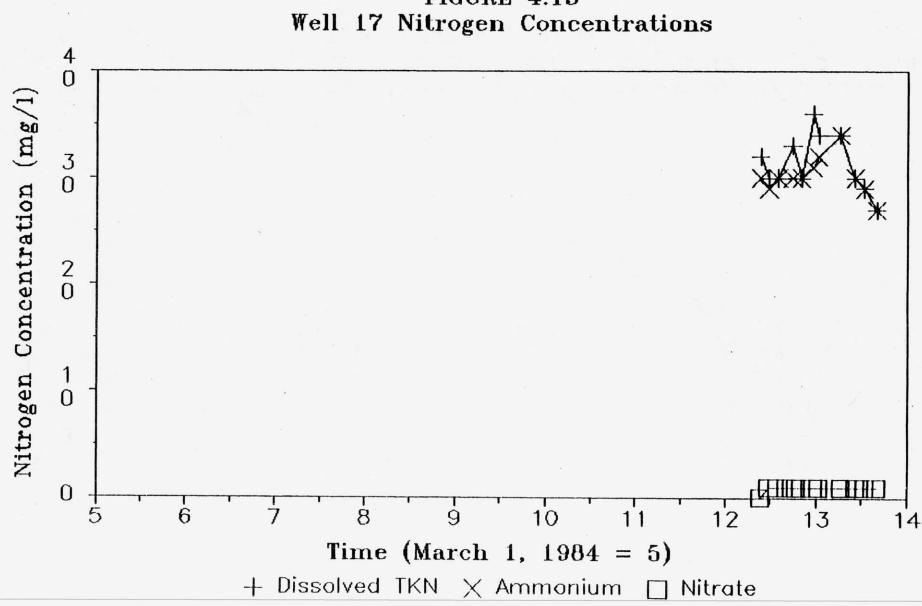


FIGURE 4.15

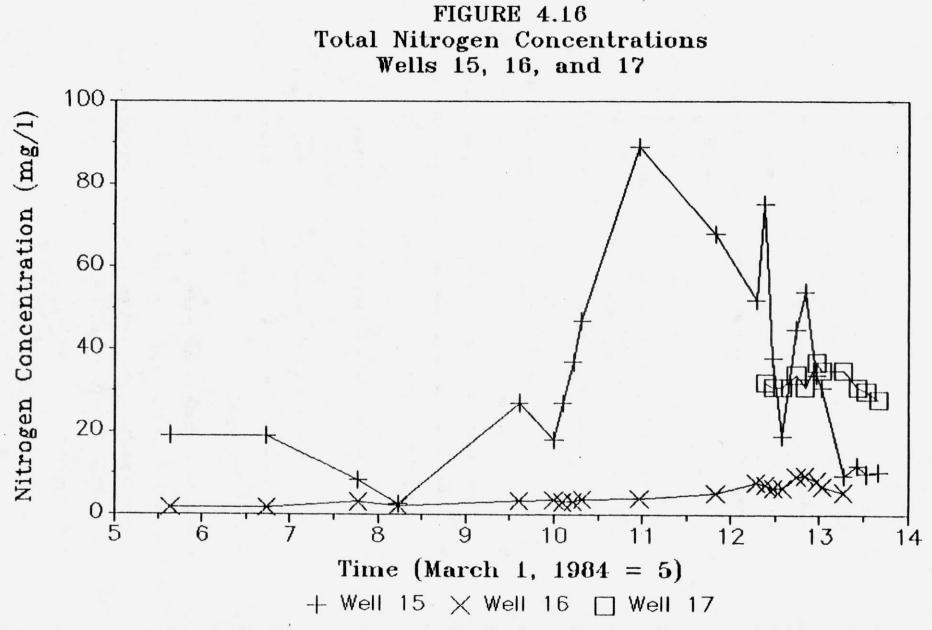
quality (less than 10 mg-N/l). The most likely cause of this transformation was the loading schedule changes initiated in July of 1984. The two week, and later, one week load/rest cycles allowed for aeration of the percolate and in turn, nitrification of the ammonia fraction. As mentioned earlier, the nitrate was flushed out by a four week cell 2 loading while cell 1 was allowed to dry in late September (1984). It was not understood why well 17 did not respond similarly after loading continued in October. Further sampling is necessary to determine if well 15 nitrogen concentration would remain low or whether large concentration variance would continue.

Nitrate decreases in well 15 occurred in late October (1984) and early November (1984) during the periods of heavy rain. When comparing the nitrate decreases to the chloride decreases, however, the nitrogen declines were relatively larger than the chloride declines. This was most likely a result of denitrification losses.

When looking at Figure 4.13, one can see that the relative magnitude of chloride concentrations for wells 15, 16, and 17 was similar. When studying Figure 4.16, which plots total nitrogen concentrations for these three wells, it is clear that the relative magnitude of nitrogen concentrations was not similar.

One final point should be made concerning the groundwater chemistry at Brodhead. The possibility of the contaminant plume sinking exists. Groundwater quality at 30-50 feet below the water table at well 14 is likely similar to quality in well 10B. A deep well near well 14 would be needed to make this conclusion.

-81-



.

Unsaturated Zone Flow Rates

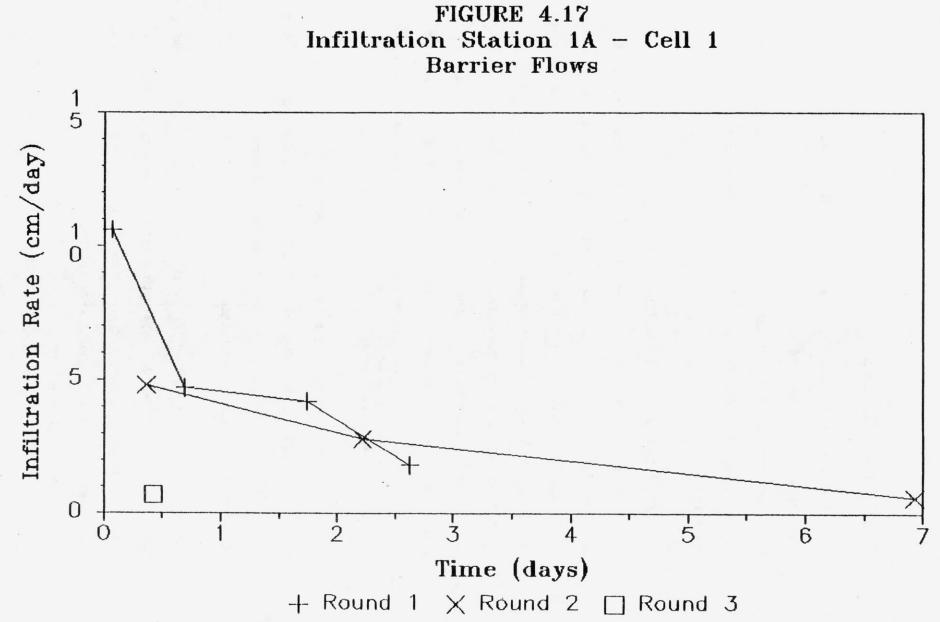
Since the furrows contained up to six inches of settled solids, vertical unsaturated flow rates were divided into two parts: percolation through the furrow "organic barrier" and travel through the underlying unsaturated zone. Furrow side wall flow was considered part of the "barrier" flow.

The infiltration stations, installed in October (1984), were used to determine barrier flow-through times. Barrier thicknesses ranged from less than one inch in cell 2 and outer cell 1 furrows, to two inches in inner cell 1 furrows, to six inches in the header ditches. Station 1A (see Figure 4.2) was located near cell 1 lysimeters in a two inch barrier region. Station 1B was located in the outer northwest corner of cell 1 in a 0.5 inch barrier region. Station 2A was located near cell 2 lysimeters in a one inch barrier region. Station 2B was located near well 15 in cell 2 in a one inch barrier region. Rates were measured three times (or three rounds).

Figure 4.17 illustrates the barrier infiltration rates at station 1A. Steady-state was reached after about 1.5 days with a flow rate of about 4 cm/day. With a two inch barrier at this rate, a barrier travel time of 1.3 days was calculated at this location.

Figure 4.18 presents barrier infiltration rates at station 1B. Steadystate was reached after 0.5 days with a rate of 6 cm/day. With a one inch barrier at this rate, a travel time of 0.4 days was calculated at this location. Round 1 results were omitted since furrow dryness caused the rate to be nearly infinite.

-83-



%

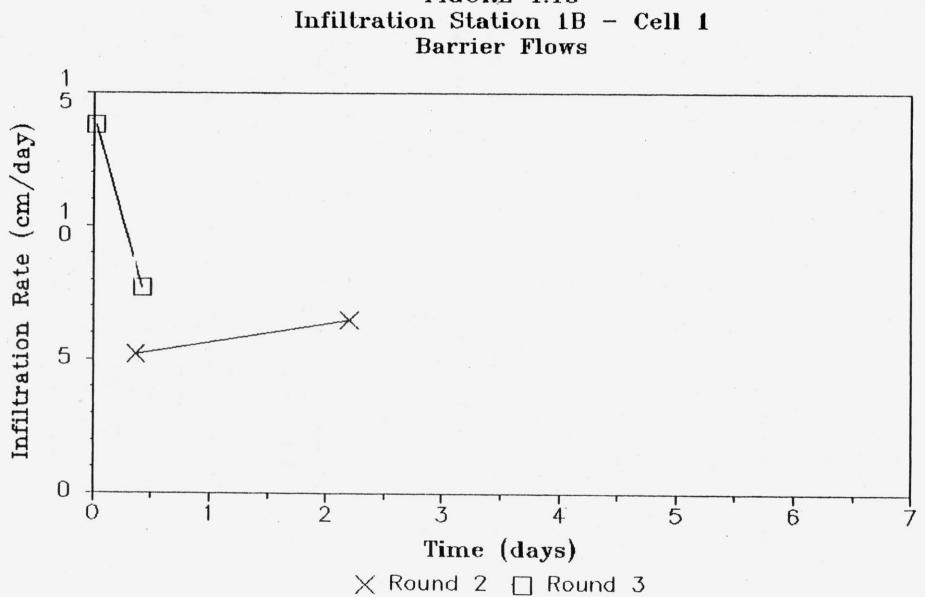


FIGURE 4.18

Station 2A infiltration rates are plotted in Figure 4.19. Steady-state was reached after about two days with a percolation rate of 2 cm/day. With a one inch barrier, a travel time through the barrier of 1.3 days was calculated at this point.

Figure 4.20 illustrates barrier infiltration rates at station 2B. Steady-state was achieved after approximately two days with a rate of 2 cm/day. With a one inch barrier, a travel time of 1.3 days was determined at this location.

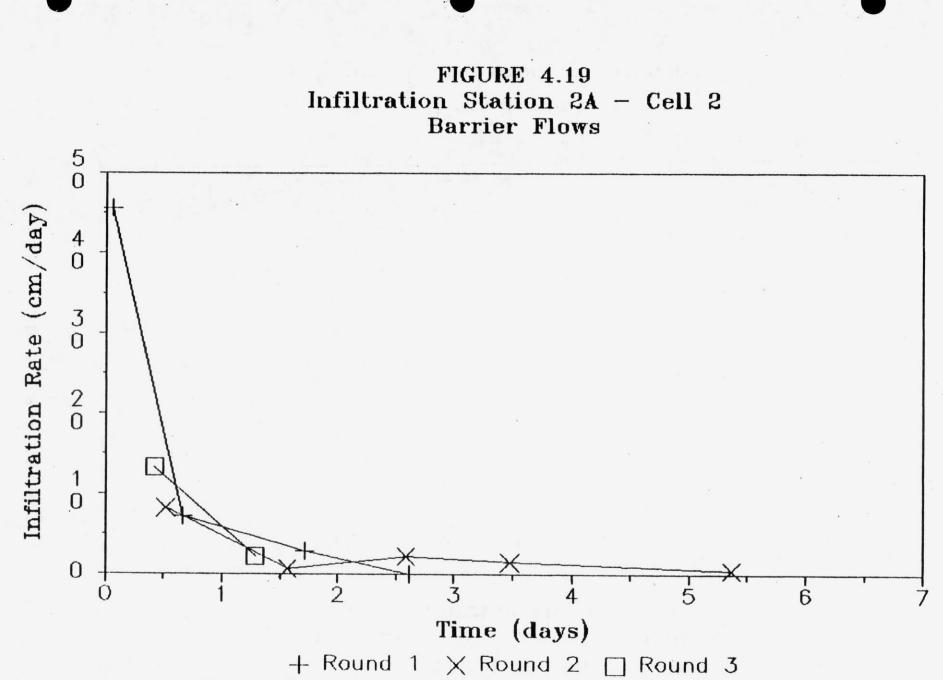
Assuming that an infiltration rate of 2 cm/day was applied to the header ditch, travel through the six inch organic barrier would be 7.6 days.

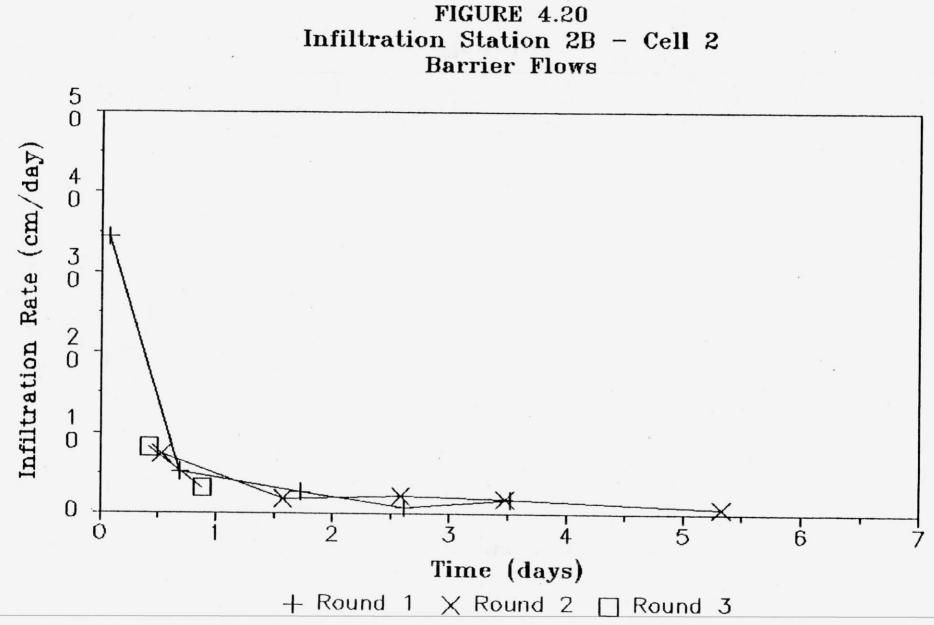
When considering these barrier travel times for the furrows and header ditches, an approximate time range of 0.5-8.0 days was developed for infiltration through the barrier.

Once wastewater percolated through the furrow barriers, it traveled approximately 5.5 feet through the unsaturated zone to the groundwater. Methods by Bouma (1975) were used to calculate these travel times. A graph and table (from Bouma, 1975), used to determine hydraulic conductivity and soil moisture tension, are presented in Appendix G.

From Table 2 of Appendix G, a typical soil moisture tension, in a sandy soil below a seepage system barrier, of 20 cm H_2O was estimated. It was also assumed that this tension was constant with depth. Using a 20 cm (H_2O) soil moisture tension, a hydraulic conductivity of 100 cm/day was estimated from Figure 1 (Appendix G) for a sandy soil. This K value was used to calculate travel times through the unsaturated zone to the

-86-





.

TABLE 4.6

LOCATION	DEPTH BELOW FURROW(ft)	BARRIER(a) TRAVEL TIME(Day)	UNSAT(b) ZONE TRAVEL TIME(Days)	TOTAL TRAVEL TIME(Days)	
Lysimeter 1 Cell 1	1.0	1.3	0.3	1.6	
Lysimeter 2 Cell 1	3.0	1.3	0.9	2.2	
Lysimeter 3 Cell 1	4.8	1.3	1.5	2.8	
Lysimeter 5 Cell 2	1.8	1.3	0.6	1.9	
Lysimeter 6 Cell 2	3.6	1.3	1.1	2.4	
Groundwater	5.5	1.3	1.7	3.0	

BRODHEAD UNSATURATED ZONE TRAVEL TIMES

(a) Based on infiltration tests: site 1A represented flow near

cell 1 lysimeters

site 2A represented flow near

cell 2 lysimeters

(b) Based on Bouma (1975), Appendix G.

depths of the various lysimeters. These results are tabulated in Table 4.6. Travel of percolate in the unsaturated zone to the groundwater took 1.7 days as calculated by the latter method.

Results shown in Table 4.6, combining barrier and unsaturated zone flow times, indicated that it took approximately three days for furrow wastewater to reach the groundwater near the lysimeters. This matched the water table response time to precipitation recharge that was presented in Figure 4.6. Considering the range of barrier flow through times discussed earlier, the range of combined (barrier plus unsaturated zone) travel time at the Brodhead system was 2.2 to 9.7 days. In future work, it is recommended that soil moisture tension instrumentation be used to better define hydraulic conductivity with unsaturated depth.

Furrow Wastewater and Lysimeter Chemistry

Mean and standard deviations of chemical parameters for wastewater at each furrow sampling point and lysimeter are listed in Table 4.7. Complete data listings are given in Appendix H.

When comparing chloride averages, one can see that all lysimeter points were contaminated by the ridge and furrow wastewater except lysimeters 8 and 9, which were intended to be located in an area of background quality. Large data variations existed in furrows and lysimeters affected by cell loading. This resulted in large standard deviations. Therefore, it was realized that these average values do not fully describe the contamination and were only used for relative comparisons.

The chloride averages from cell 1 indicate no dilution from the furrows to the depth of lysimeter 3, and yet considerable reduction of other

-90-

TABLE 4.7

BRODHEAD MEAN AND STANDARD DEVIATION OF

FURROW WASTEWATER AND LYSIMETER CHEMICAL PARAMETERS

(Units in mg/l except pH)

LOCATION	DISS BOD5	TOTAL BOD5	DISS COD	DISS TKN	TOTAL TKN	DISS NH3-N	DISS NO ₂ +NO ₃	C1-	TDS	FIELD pH
Furrow-Cell 1	330 <u>+</u> 310	660 <u>+</u> 490	490 <u>+</u> 290	20 ± 12	31 ± 18	14 <u>+</u> 12	0.4+0.5*	560 <u>+</u> 190	1780+280	7.1+0.56
Lysimeter 1	(4)	(5)	(16)	(20)	(9)	(20)	(20)	(19)	(3)	(2)
(1' depth)			55+12	22+8.6		16 <u>+</u> 10	0.6+0.4*	700 <u>+</u> 170	2130 <u>+</u> 184	7.1 ± 0.30
-	1 1.0 0		(14)	(18)	·	(22)	(22)	(16)	(2)	(3)
Lysimeter 2	4.1+0.0		51 ± 6.6	12 <u>+</u> 10		9.8 <u>+</u> 10 *	8.6 <u>+</u> 9.3 *	690 <u>+</u> 100	2040 <u>+</u> 191	7.2 <u>+</u> 0.33
(3' depth)	(1)		(21)	(27)		(30)	(30)	(17)	(2)	(4)
Lysimeter 3			56 <u>+</u> 4.2	2.7 ± 0.80		2.5 <u>+</u> 3.9 *	6.3 <u>+</u> 4.7	780 <u>+</u> 100		7.6 <u>+</u> 0.0
(4.8' depth)			(2)	(8)		(17)	(17)	(10)		(1)
Well 17 0	34 <u>+</u> 37 *		54 <u>+</u> 5.8	31 <u>+</u> 2.6		30 <u>+</u> 1.8	0.92 <u>+</u> 0.27*	650 <u>+</u> 77	2540 <u>+</u> 643	
	(2)		(7)	(11)		(11)	(11)	(11)	(2)	
Furrow-Cell 2	580 <u>+</u> 120 #	2400 <u>+</u> 0.0	91 <u>0+</u> 530	26 <u>+</u> 14	70 <u>+</u> 100	14 <u>+</u> 10	1.5+2.9*	620+200	2630+127	6.7+0.0
	(3)	(1)	(12)	(16)	(11)	(16)	(16)	(15)	(2)	(1)
Lysimeter 5	3.0 <u>+</u> 0.0*		32 <u>+</u> 11	0.9 <u>+</u> 0.3		0.3+0.4#	11+9.0	400+170		7.4+0.0
(1.8' depth)	(1)	1	(16)	(22)		(22)	(22)	(14)		(1)
Lysimeter 6	3.3 <u>~</u> 0.69*		20 <u>+</u> 4.0	0.7+0.2		0.4+0.4#	21+10	390+150	1400+162	6.8+0.0
(3.6' depth)	(11)		(14)	(19)		(21)	(21)	(19)	(10)	(1)
Well 150	51 <u>+</u> 55 *		57 <u>+</u> 50	16+21		15+20*	17+21*	570+110	1640+216	6.4+0.12
	(12)		(19)	(23)		(23)	(23)	(23)	(12)	(8)
Lysimeter 8	3.4+0.47*		12+1.6	0.4+0.06		0.1+0.0*	0.1+0.1*	1.2+0.29	267+19.9	
(3.7' depth)	(4)		(9)	(11)		(13)	(13)	(11)	(6)	
Lysimeter 9	3.2+0.37	6+0.0	24+3.7	0.6+0.2	0.6+0.2	0.1+0.0*	0.1+0.02#	1.3+0.34	323+24.9	6.8+0.07
(4.7' depth)	(7)	(1)	(14)	(17)	(2)	(19)	(19)	(16)		
			、···/	· · · · /		(77)	(197	(10)	(8)	(2)
	I		<u> </u>							

.

e Well data added to observe transformations of groundwater; () - # of observations.

* Mean includes data above or below a detection limit, limit was used in average.

-16

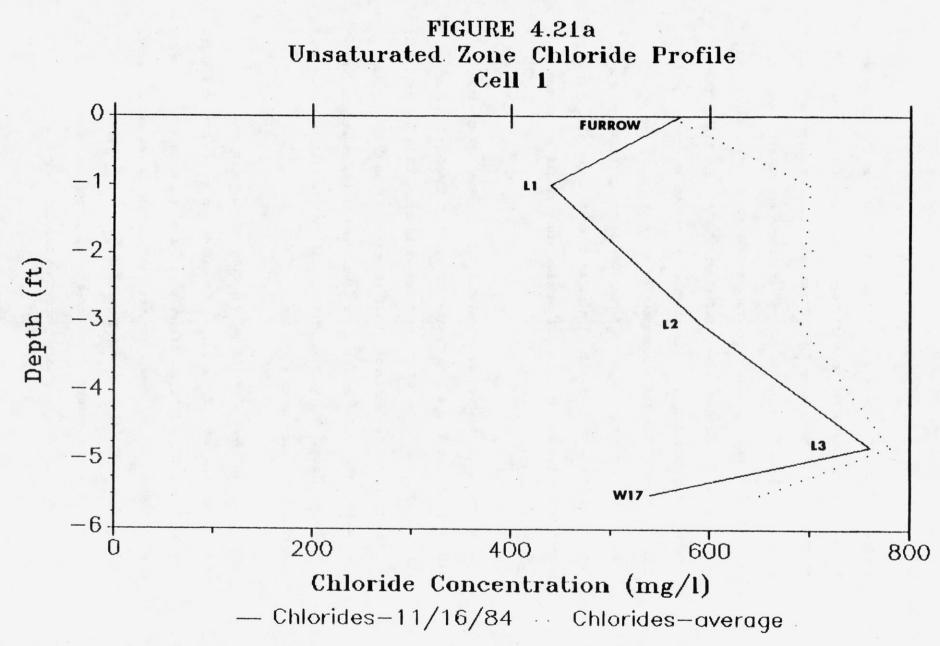
chemical parameters was seen at this depth. Dissolved COD values were reduced 88% and dissolved total nitrogen (TKN + NO_3 + NO_2) concentrations were reduced by 56%. Groundwater quality underneath cell 1, as shown by well 17 data, did not reflect these reductions, however. Dissolved COD concentrations were reduced 89% at the water table, but, dissolved total nitrogen values in the groundwater were higher than furrow concentrations. This was attributed to past overuse of cell 1 prior to 1980. Since the groundwater velocity was slow, the zone beneath cell 1 did not flush out during the course of this project, as the region around well 15 did. A similar discussion for cell 2 average concentration reductions could not be made since values varied too greatly in well 15.

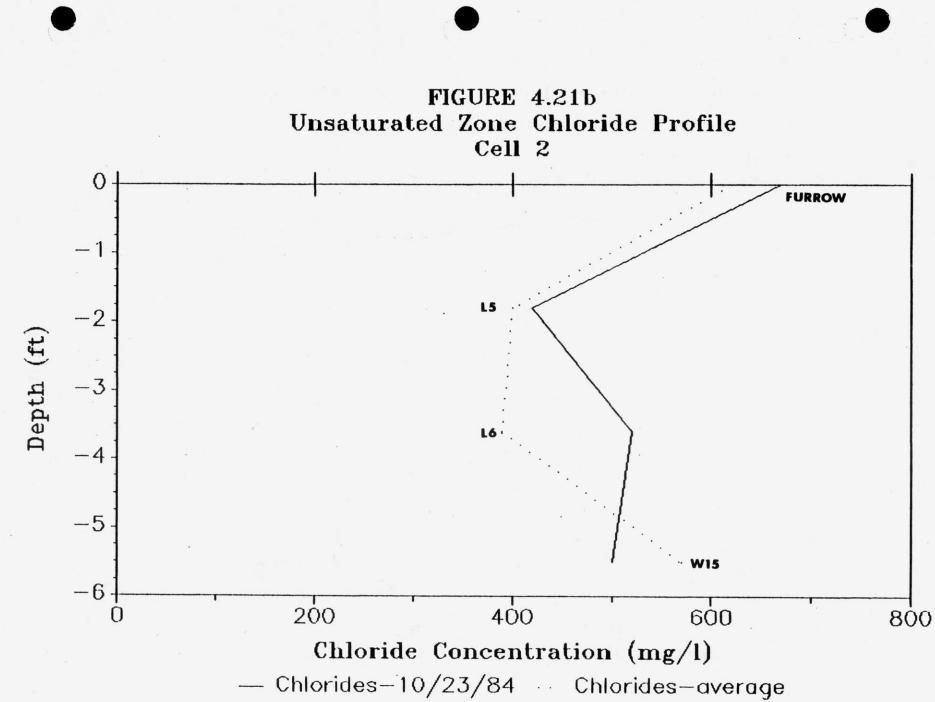
Another important observation from Table 4.7 concerned the transformation of organic-N in the wastewater to NH_3 -N in the furrow water. Ammonium comprised only 10% of the dairy wastewater TKN while the TKN in the furrow wastewater contained 54-70% ammonium. From Table 4.5, one can see that TKN values in the wells downstream of the ridge and furrow system were almost 100% NH_4 +-N. This transformation will be discussed in more detail later in this chapter.

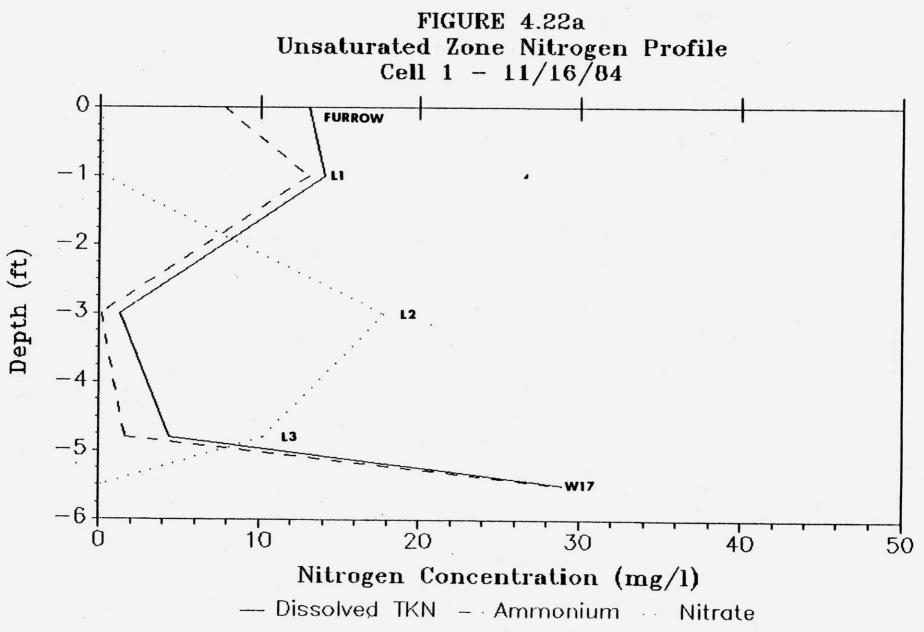
Figures 4.21 and 4.22 show typical chloride and nitrogen profiles, respectively, in both of the cells. Concentration movement in the unsaturated zone could not be correlated with travel time since the travel times presented earlier were less than the time period between sampling. The profiles do illustrate the following points:

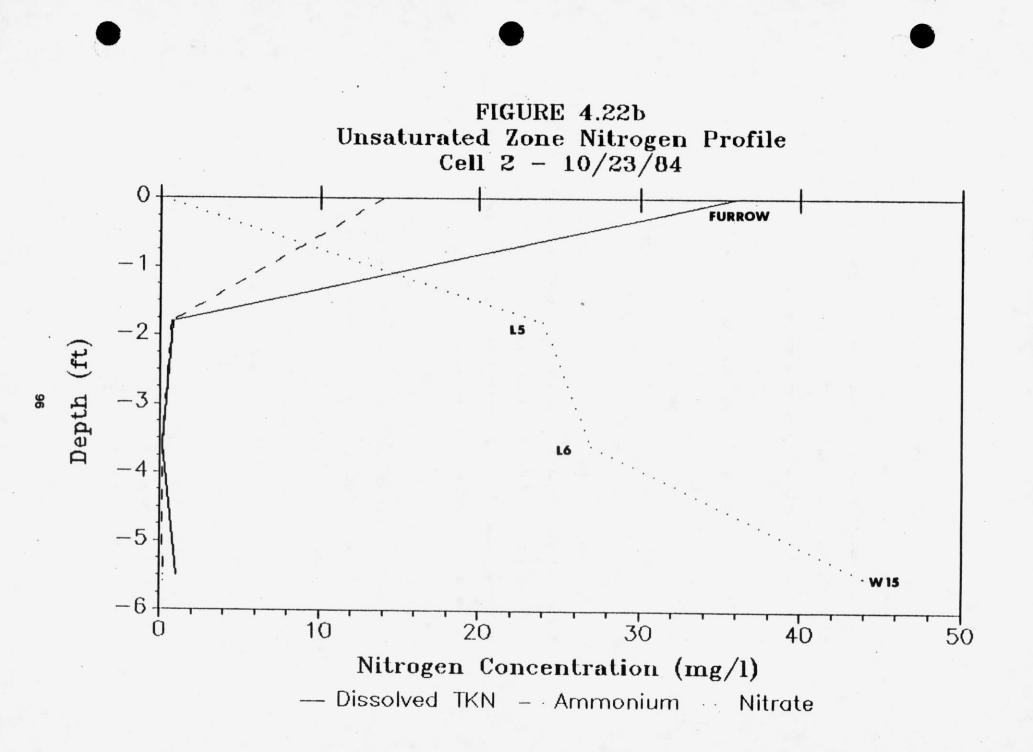
no dilution of wastewater occurred as it percolated beneath cell 1,
 slight dilution of percolate occurred beneath cell 2,

-92-









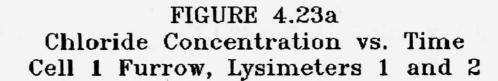
- 3) TKN was transformed to NH_4^+-N after one foot depth in cell 1 and two foot depth in cell 2,
- 4) nitrification occurred beneath 1 and 3 feet depth underneath cell 1,
- 5) nitrification occurred along the whole profile underneath cell 2,
- 6) possible denitrification occurred between the 3 and 5 feet depth underneath cell 1, and
- 7) high TKN and NH_{4}^{+} -N concentrations were observed at the water table beneath cell 1.

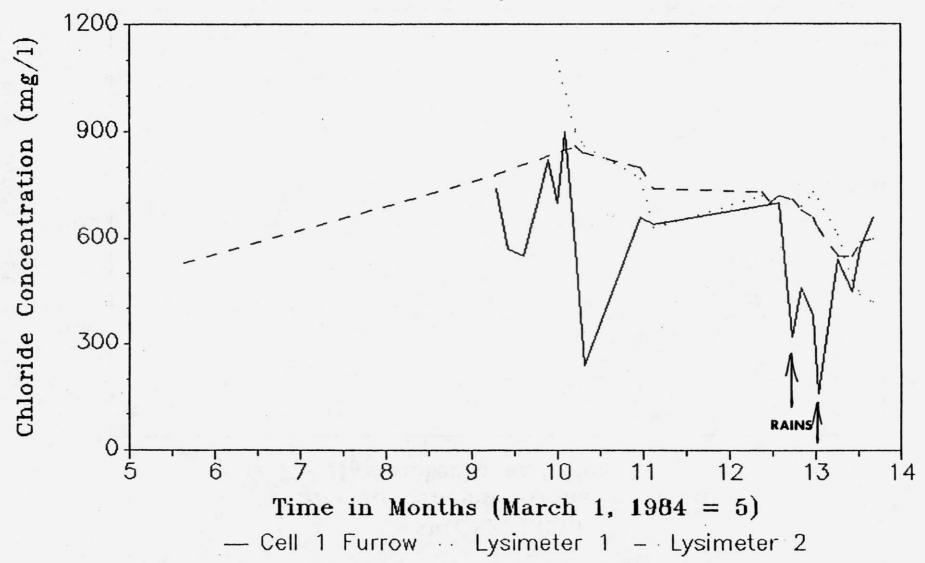
Furrow and unsaturated zone pH values (Table 4.7) ranged from 6.7 in a cell 2 furrow to 7.6 in lysimeter 3. These values are somewhat higher than the average pH values seen in wells 15 and 16, immediately underneath the system. The lower pH values (about 6.4) in the contaminated groundwater were attributed to the past use of the system when overloading caused anaerobic conditions and lower pH values. The low groundwater velocity has delayed the flush out of this zone.

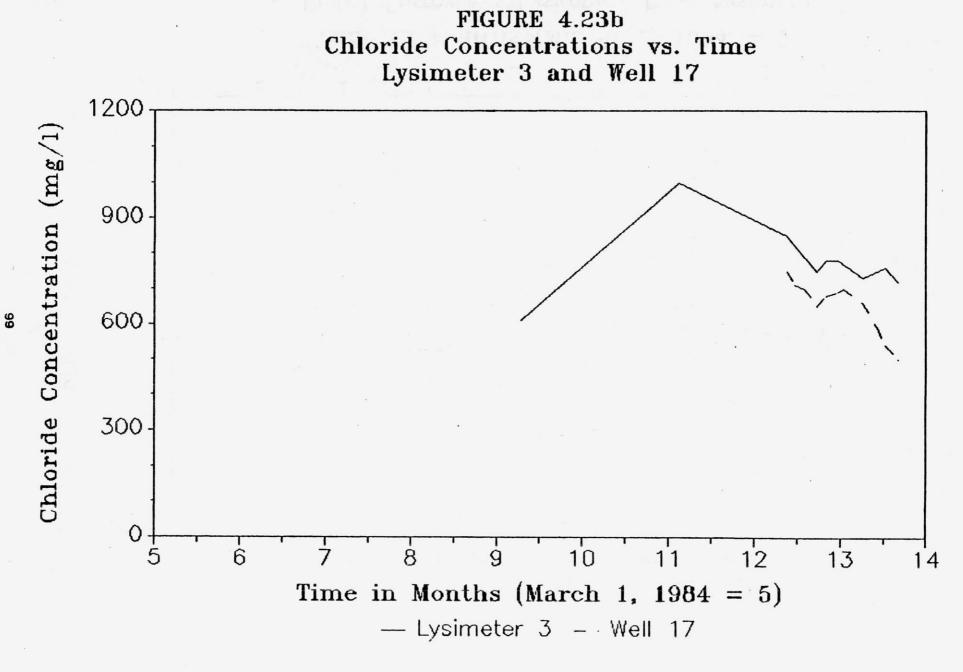
Chloride and nitrogen time plots for furrow, lysimeter, and well data were compared with depth in both cell 1 and cell 2 at Brodhead. Figure 4.23 illustrates chloride concentrations versus time for cell 1 furrow wastewater; lysimeters 1, 2, and 3; and well 17. Figure 4.24 presents total nitrogen concentrations versus time for these respective sampling points. The reader is referred to Figures 4.25 to 4.28 and 4.15 for TKN, NH₃-N, and NO₃-N plots of respective points.

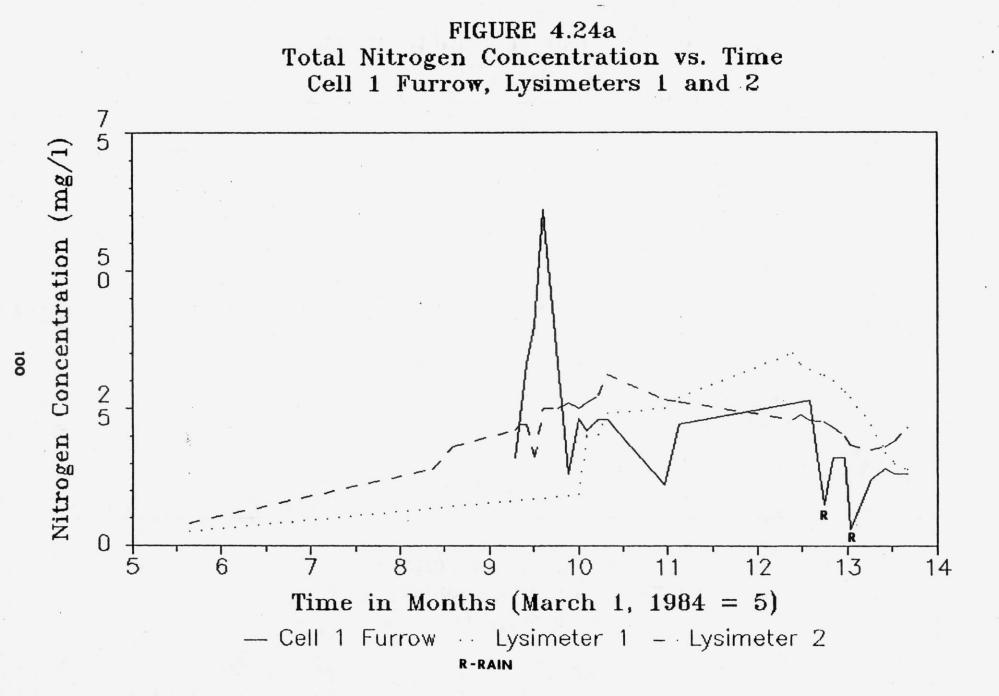
These cell 1 chloride and nitrogen plots were compared to determine if nitrogen decreases were mainly due to dilution (as seen by comparing chlorides) or if unsaturated zone nitrogen losses were actually greater

-97-









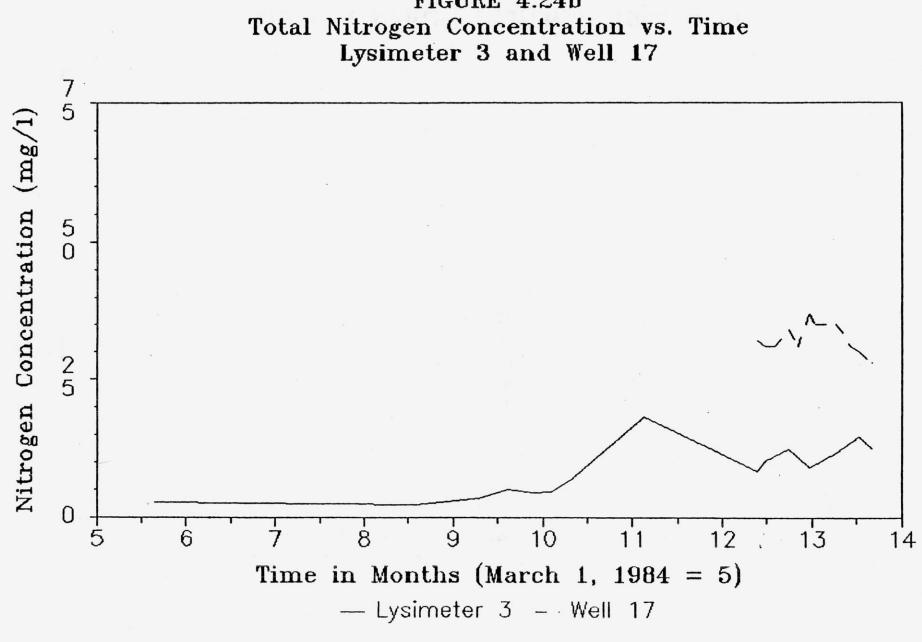
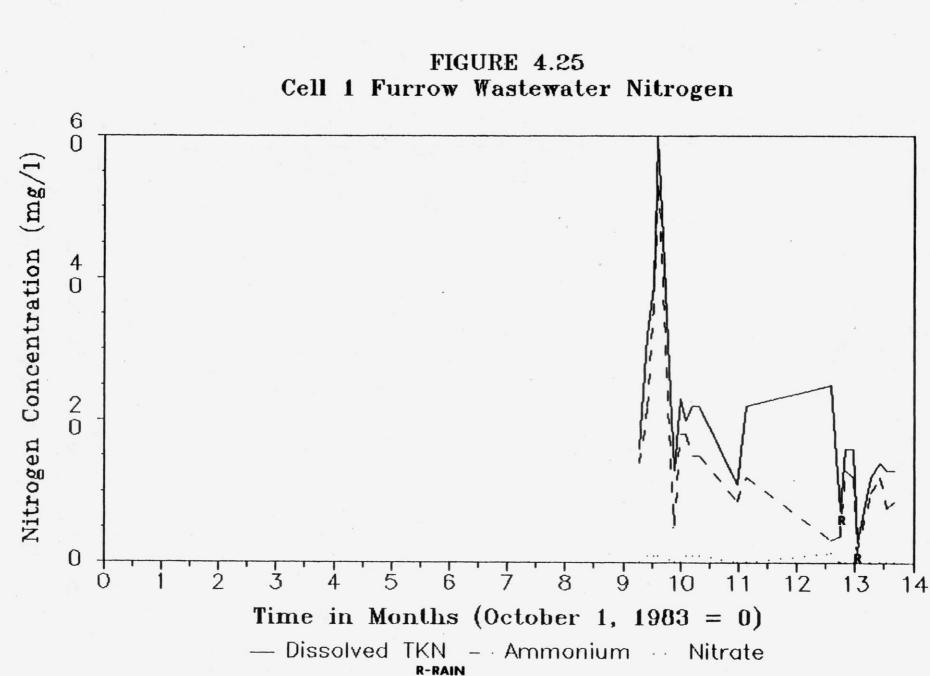


FIGURE 4.24b

ī



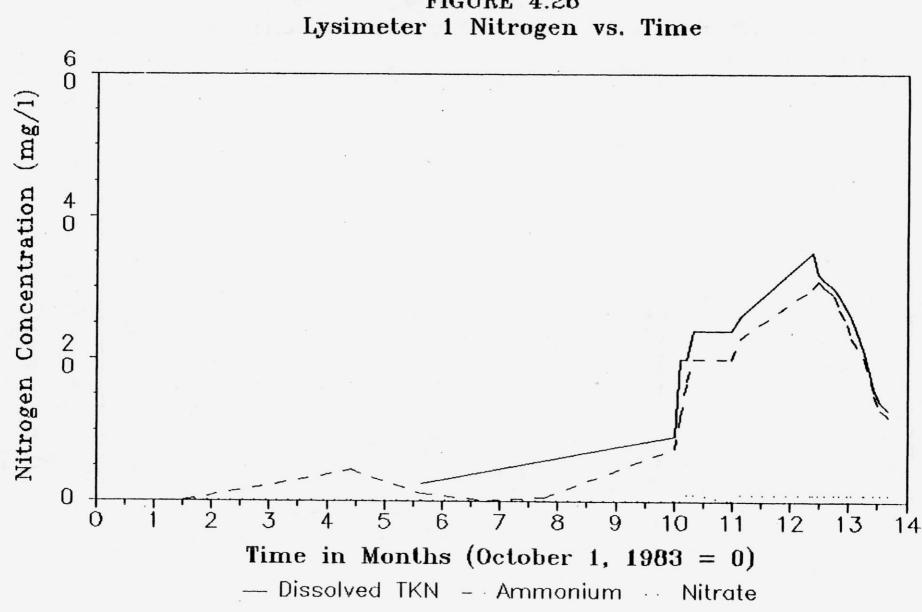
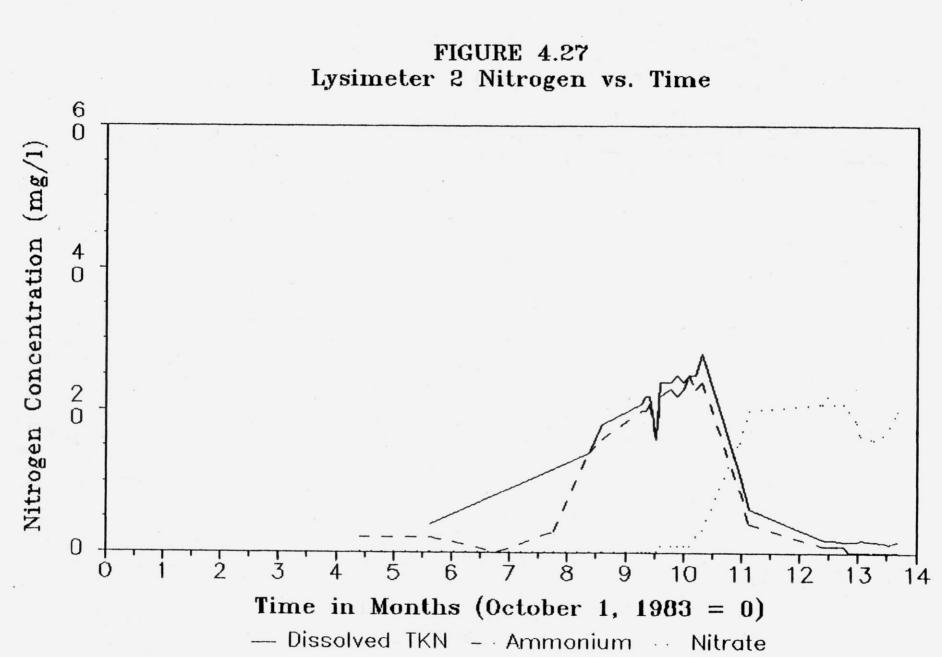


FIGURE 4.26



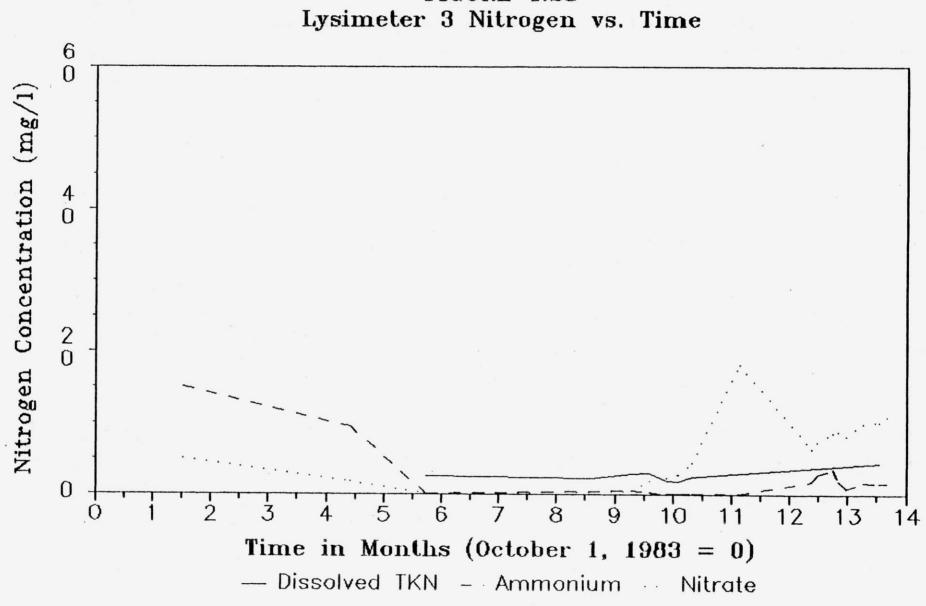


FIGURE 4.28

than chloride decreases. Late October and November (1984) was the best period to compare chloride concentrations with depth in cell 1. As mentioned earlier, heavy rains occurred on October 18 and November 1. This was reflected by a decrease in furrow chloride concentration on these dates as shown in Figure 4.23. After the November 1 rain, a 42% drop in chloride values in lysimeter 1 occurred. A 24% decrease with a subsequent 8% increase in chloride concentrations occurred in lysimeter 2 during the same period. A 28% drop which began November 1, was observed in well 17 as well. Figure 4.23 shows these declines. These results complement the unsaturated zone travel time of less than 3 days calculated earlier. It was not known why a decrease in chloride concentration was not seen in lysimeter 3.

Similar decreases were observed in the total nitrogen concentrations of cell 1 furrow wastewater; lysimeters 1 and 2; and well 17 as shown in Figure 4.24. Decreases in furrow nitrogen occurred on 10/18 and 11/1 as expected. A 52% drop in total nitrogen was observed in lysimeter 1 during the same time period as a 42% chloride decrease. It was not feasible to attribute the additional 10% loss to denitrification since no nitrate was present in the first one foot of depth. Plant uptake of NH4⁺-N was possible since lush growth was occurring on furrow fringes in late November. In lysimeter 2, total nitrogen concentrations decreased by 27% (similar to chloride dilution) and then increased by 20% (compared to a 8% chloride increase). This 12% addition was attributed to nitrate production. Total nitrogen losses in well 17 declined by 20% during this period (compared to 28% for Cl⁻). Lysimeter 3 nitrogen values increased 20% during this period. A decreasing denitrification

-106-

rate could have accounted for this since nitrate values increased and chloride concentrations were fairly level.

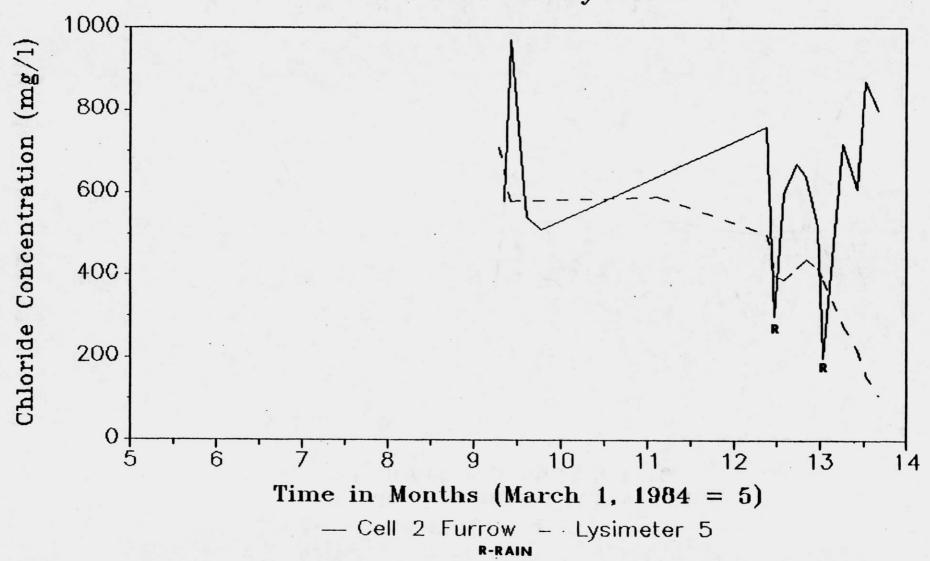
Figure 4.24 indicates that nitrogen losses, most likely through denitrification, occurred from lysimeter 2 to lysimeter 3 after September 1. Prior to this time, both nitrification and denitrification occurred between these two points (see Figures 4.27 and 4.28) since total nitrogen decreased in lysimeter 3 but little nitrate was present.

After September 1, nitrification occurred between lysimeters 1 and 2 as seen in Figures 4.26 and 4.27. Prior to 9/1, however, lysimeter 2 samples contained high nitrogen values (in ammonia form) which did not appear in lysimeter 1 samples. Since nitrogen can not be created, this phenomenon may have been the result of the horizontal positioning of the lysimeters. Since the lysimeters were located adjacent to one another (2 to 3 feet apart), the possibility of the device receiving wastewater of different initial quality existed. Local channeling in the unsaturated zone created by organic barrier heterogenetities may also have caused these anomalies.

As in cell 1, late October and November (1984) was the best period to compare chloride concentrations with nitrogen concentrations through the unsaturated depth of cell 2 to analyze dilution and possible denitrification losses. Figure 4.29 presents chloride data versus time for cell 2 furrow wastewater; lysimeters 5 and 6; and well 15. Figure 4.30 illustrates total nitrogen values versus time for these respective sampling points. TKN, NH₃-N, and NO₃-N plots for these points appear in Figures 4.31 to 4.33 and 4.12, respectively.

-107-

FIGURE 4.29a . Chloride Concentrations vs. Time Cell 2 Furrow and Lysimeter 5



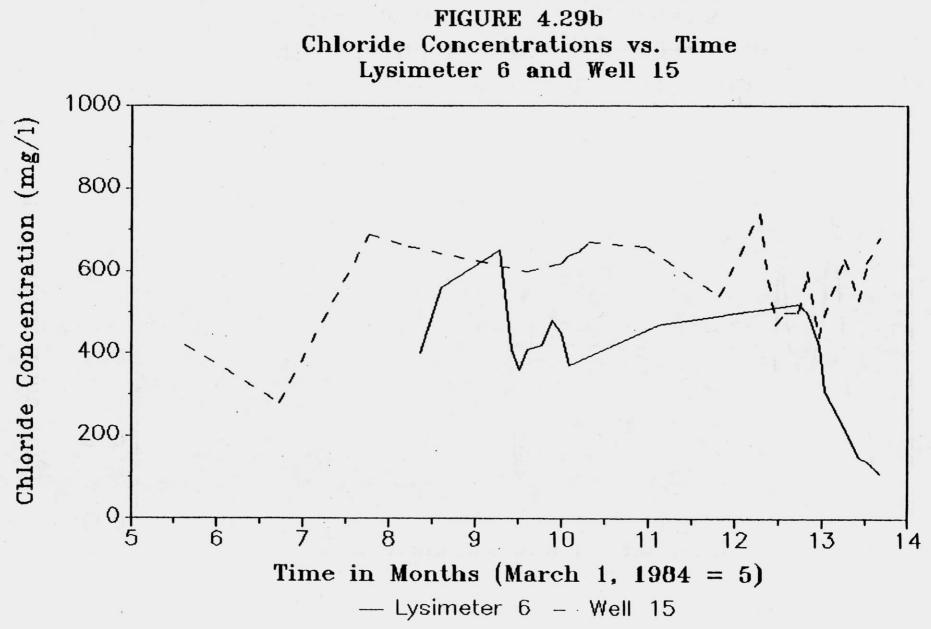
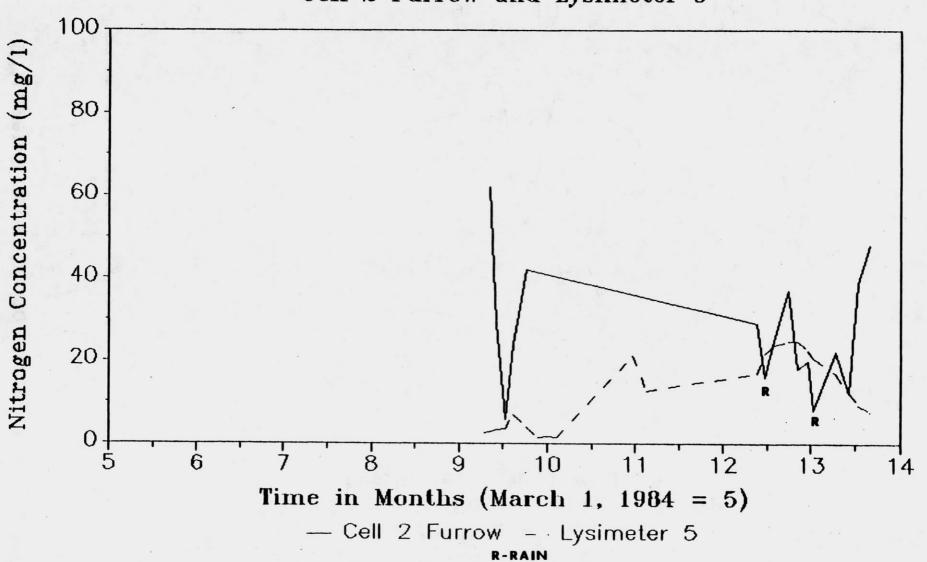
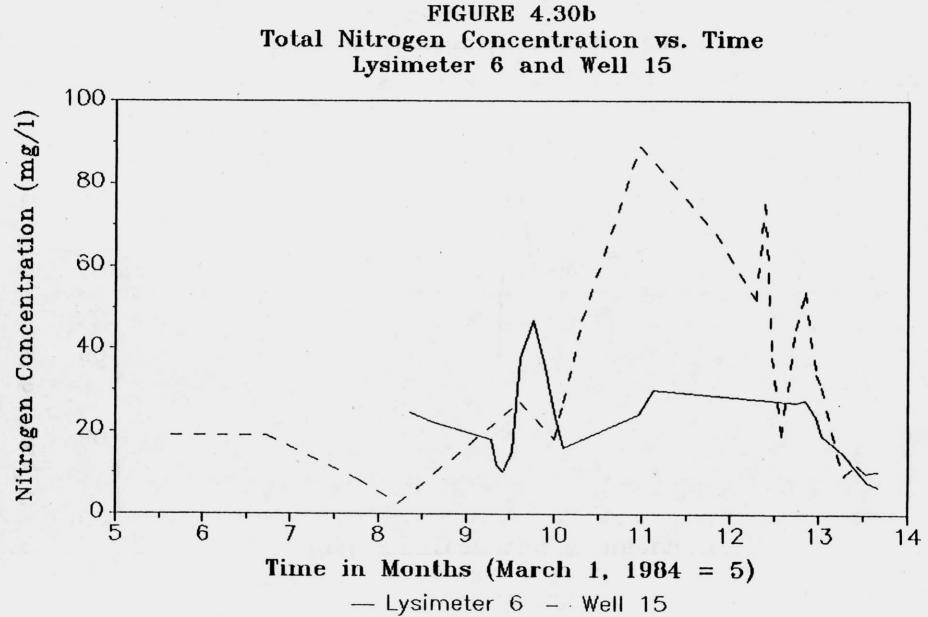
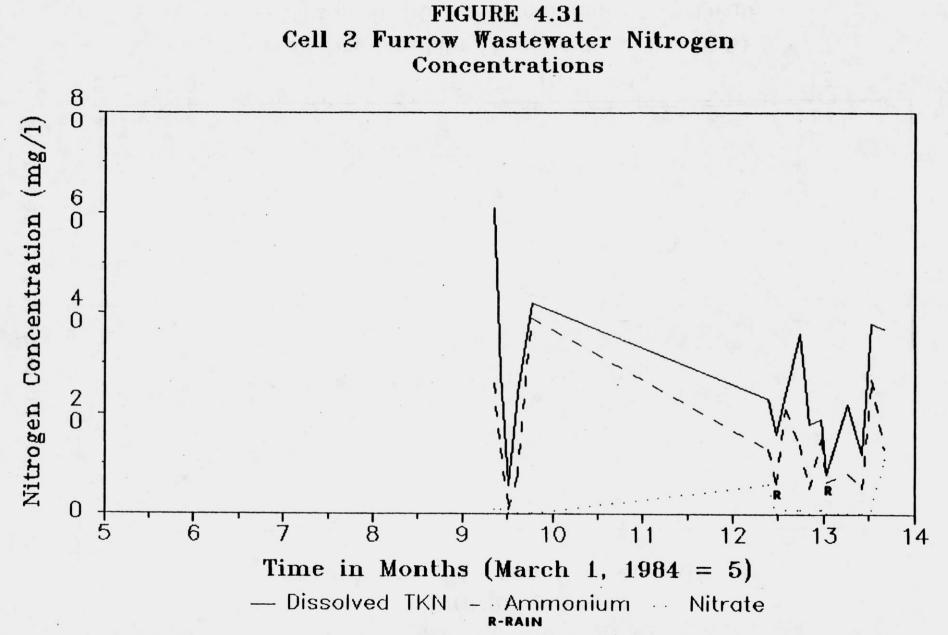


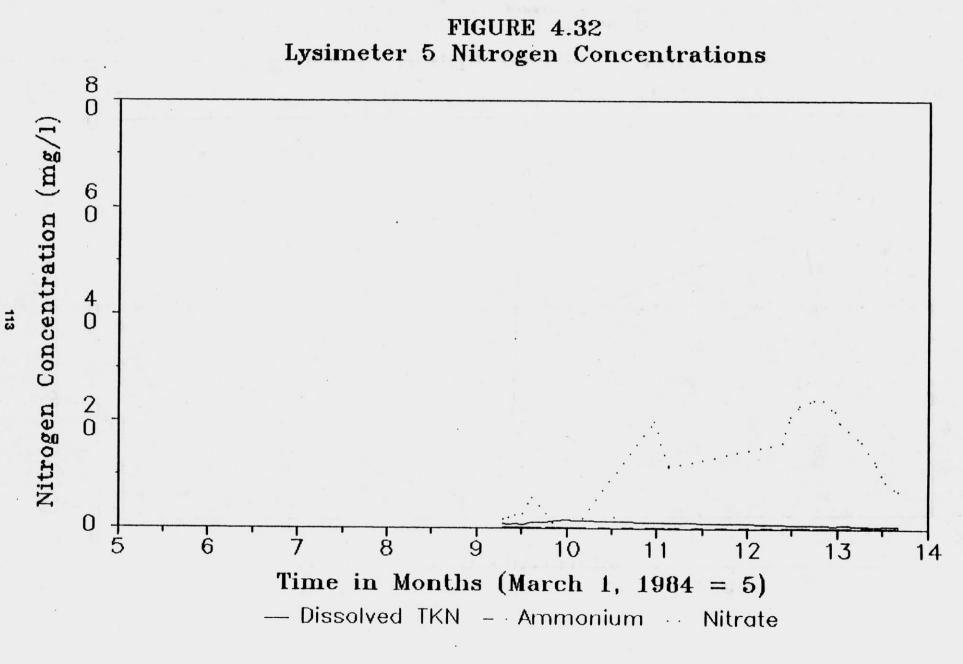
FIGURE 4.30a Total Nitrogen Concentration vs. Time Cell 2 Furrow and Lysimeter 5

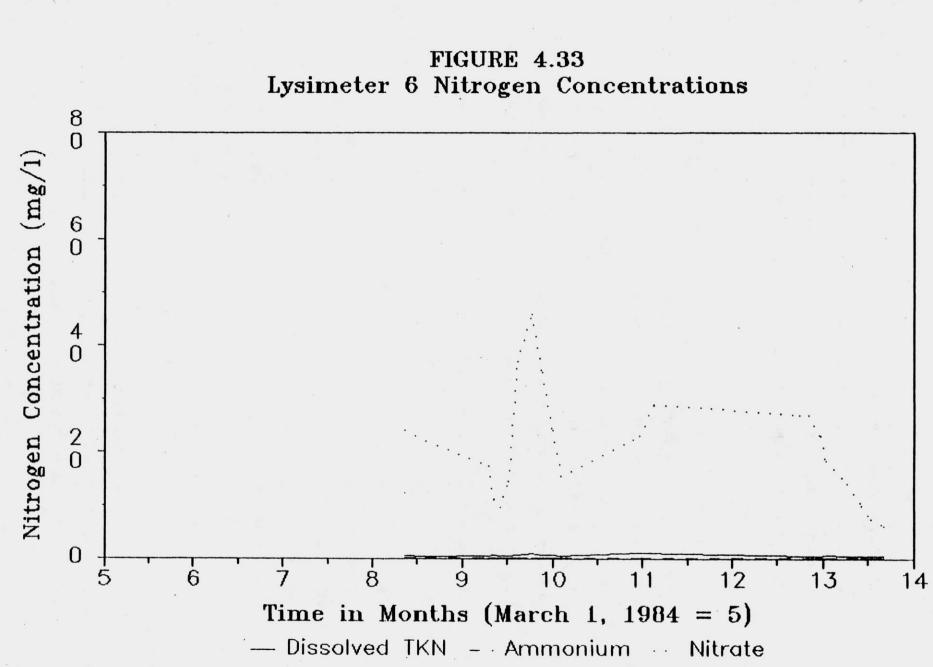




Ħ







Cell 2 furrow chloride values decreased sharply on rainy days just as they did in cell 1. A 75% decrease in chloride concentrations was observed in lysimeter 5 after October 26. Lysimeter 6 chloride values declined 79% after October 23. Similar decreases were not observed in well 15 during this period.

Similar declines during this time were witnessed in total nitrogen concentrations for lysimeters 5 and 6, as well as, well 15 (see Figure 4.30). Dilution (75% chloride decrease) accounted for most of a 71% drop in total nitrogen concentrations in lysimeter 5 after October 6. A 77% decrease in lysimeter 6 N values was also caused by the dilution rainwater. During this period, an 88% decline of N concentrations occurred in well 15. These losses were likely the result of denitrifi-

cation since well 15 Cl⁻ concentrations did not decrease. Underneath cell 2, all denitrification losses occurred below 3.6 feet depth.

Figure 4.30 gives the illusion that nitrogen concentrations did not decrease but actually increased. There were three possible reasons for this. First, lysimeters 5 and 6 are located about ten feet apart and may not have received an identical quality waste. Second, local channeling resulting from unsaturated zone heterogeneities (eg. organic barrier) may have occurred. Third, since groundwater flowed from underneath cell 1 past cell 2, water quality in well 15 was directly affected by the past water quality underneath cell 1.

Nitrate peaks were observed in lysimeter and well data for both cells. These peaks (see lysimeter nitrogen series plots) appeared either late in a loading cycle or just after. Increased vertical flow from loaded

-115-

wastewater tended to flush out nitrate which accumulated in unsaturated pore spaces during rest cycles.

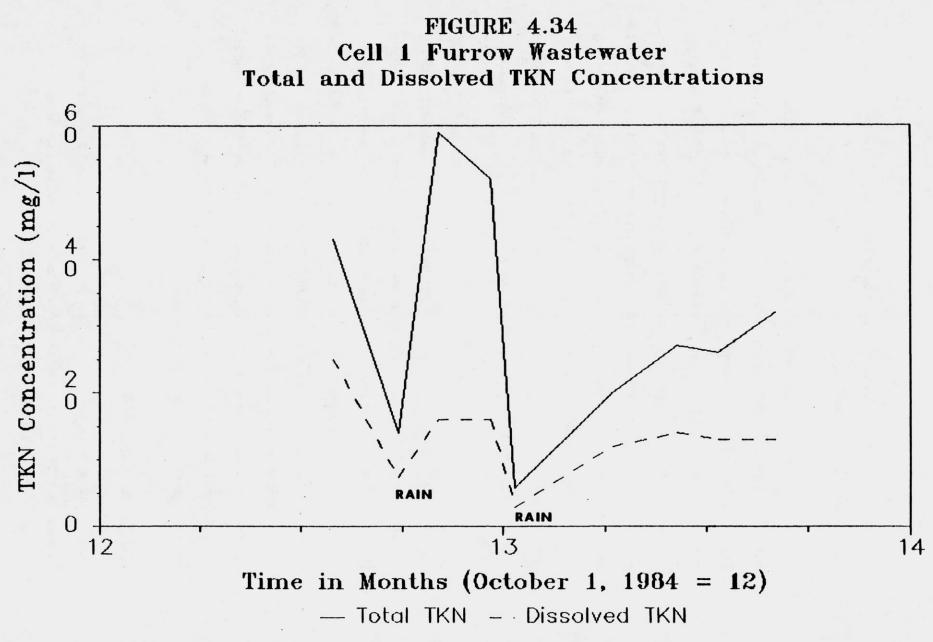
Figures 4.25 and 4.31 present TKN, NH₃-N, and NO₃-N concentrations versus time for cell 1 and cell 2 furrow wastewater samples, respectively. Figures 4.34 and 4.35 present total and dissolved TKN versus time for furrow samples in cells 1 and 2, respectively. The following sequence of furrow nitrogen transformations were established from these results. As stated previously, the wastewater discharges contained nitrogen primarily in organic-N form. Thirty to 90% of this organic-N was tied up in wastewater solids. This was seen by comparing total and dissolved TKN results (Table 4.2). Depending on furrow mixing, total TKN concentrations were quite variable and of the same magnitude as wastewater TKN's.

Dissolved TKN showed a distinctive pattern, however, especially during the dry period of July and August (1984). During initial cell loading, dissolved TKN's were lowest since the solids contained much of the nitrogen. Toward the end of a loading cycle and during the rest cycle, dissolved TKN concentrations increased. This was possibly the result of mineralization of settled furrow solids and diffusion of ammonium into the overlying water column. Reddy and Graety 1981) recognized a similar occurance in their study. These transformations were not as apparent during shorter load/rest cycles.

Crop Uptake of Nitrogen

Seven grass samples were cut at the Brodhead ridge and furrow system during the 1984 growing season. One sample was collected in April

-116-



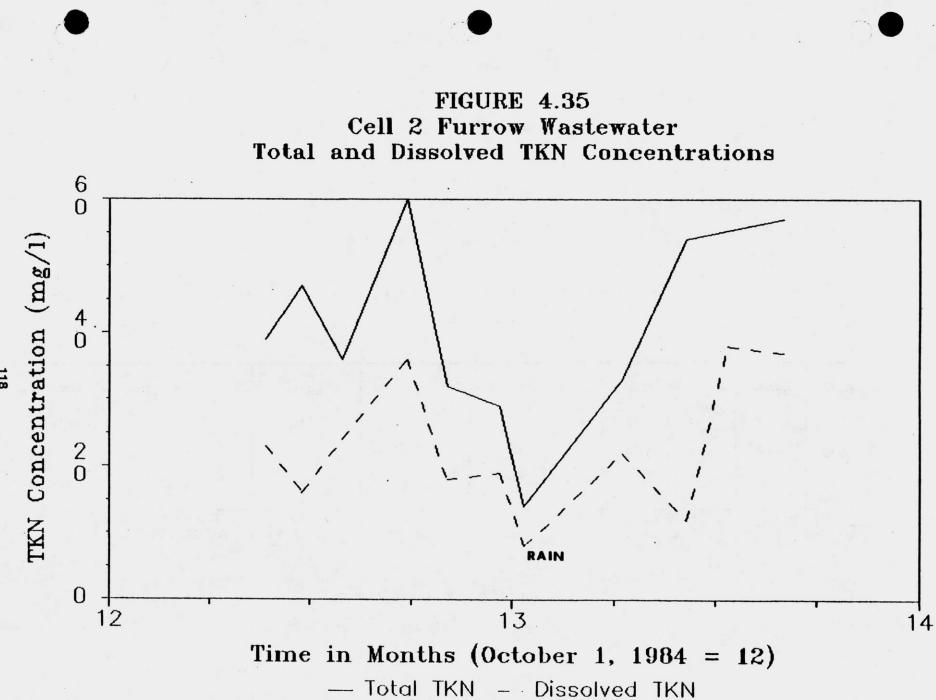


TABLE 4.8

BRODHEAD GRASS SAMPLE RESULTS

DATE	TOTAL DRY WT OF GRASS ON SITE(1b)	\$ N DRY WT-CELL 1	% N DRY WT-CELL 2	\$ ASH-CELL 1 AFTER BURNING	% ASH-CELL 2 AFTER BURNING	% N CELL 1	OF ASH CELL 2
4/24/84	11,646		1.44		10.1		2.64
7/13/84	22,847	1.28			5.9		0.48
9/25/84	9,171	2.70	3.50	5.5	5.6	0.72	0.50
11/20/84	23,282	1.59	1.41	3.9	3.1	0.45	0.49

- weights extrapolated to total site area from weights corresponding to sample areas

DATE	TOTAL N ON SITE BEFORE BURNING(1b)	TOTAL N ON SITE AFTER BURNING(1b)	% N LOST BY BURNING
4/24/84	168	31.0	81.5
7/13/84	292	6.5	97.8
9/25/84	276	3.2	98.9
11/20/84	347	3.7	98.9

TABLE 4.8 Cont.

- weights extrapolated to total site area from weights corresponding

to sample areas

before growth began, two were taken in July at peak growth, two were cut in September during declining growth, and two were taken in November during rejuvenated growth. Results of these analyses are shown in Table 4.8. Calculations are presented in Appendix I. These results were used for the following three purposes: 1) to determine the plant nitrogen uptake during the growing season, 2) to calculate the effect of grass burning on nitrogen losses, and 3) to estimate a nitrogen plant uptake value for a nitrogen budget.

When looking at the total weight of nitrogen on site before burning in Table 4.8, plant uptake of nitrogen appeared to be highest in late spring, slightly declining in summer and slightly increasing in late fall. Heavy rainfall in late October and early November stimulated new growth in November.

The nitrogen taken up over the total area of the ridge and furrow system, by the cover crop during the growing season was 347 lbs. This corresponded to the grass nitrogen content on 11/20/84. Since grass from the previous year died and was immobilized in the soil, the April site grass nitrogen content of 168 lb was not considered a loss to the treatment system nor was it subtracted from the final grass nitrogen weight (347 lb.). Since the Brodhead system operator did not burn site grasses, all grass nitrogen returned to the soil. This meant that grass uptake losses were not a part of a system nitrogen budget. Some operators of land treatment systems do burn the site grasses in the spring to eliminate dead grass accumulation and stimulate new growth. Besides stimulating new luxerient growth, Woodmansee and Wallach (1979) stated that nitrogen losses and nitrogen transformation rate increases occur

-121-

after fires. Nitrogen losses depend on plant biomass and elemental composition, fire intensity and duration, and winds. Since spring fires at land treatment sites are of low intensity, plant roots are not killed and growth will continue. Dead grasses contain most of their N in the root zone, so nitrogen losses are lowest when these grasses are burned.

Knowing this information, grass samples were burned and analyzed for percent ash and percent N in ash. Total pounds of ash-nitrogen were highest in April at 31 lbs. as shown in Table 4.8. The July ash sample indicated 6.5 lbs. of nitrogen on site after burning, the September sample indicated 3.2 lbs., and the November sample indicated 3.7 lbs. A range of 82%-99% nitrogen loss by burning was observed when comparing total nitrogen in the site grasses before and after burning. The lowest loss occurred in April.

After studying the data, one could conclude that it would be best to burn the cover crop of a ridge and furrow site in late fall since grass N losses of 99% would be seen. This would not be advisable, however, since the cover crop provides a vital function in the winter. Dead grasses insulate the soil from freezing thereby providing for wastewater percolation during the winter, protect the furrows from direct exposure to cold weather, and provide ridge stability during the spring thaw. It must also be realized that the grass burning may not be practical at all treatment sites. The operator must be able to access the site with fire fighting equipment in case burning gets out of control.

TABLE 4.9

SUGAR RIVER QUALITY

LOCATION	TOTAL BOD5	TOTAL COD	C1-	TKN	NH3-N	NO3-N	TSS	FIELD pH
Upstream	3.4 <u>+</u> 0.49 *	14 <u>+</u> 6.4	23 <u>+</u> 2.0	1.0 <u>+</u> 0.21	0.3 <u>+</u> 0.1	3.8 <u>+</u> 0.07	382 <u>+</u> 0.0	7.9 <u>+</u> 0.14
	(2)	(2)	(3)	(2)	(2)	(2)	(1)	(2)
Midstream	3.1 <u>+</u> 0.0	14 <u>+</u> 4.9	25 <u>+</u> 6.8	0.8 <u>+</u> 0.0	0.2 <u>+</u> 0.07	3.6 <u>+</u> 0.07	380 <u>+</u> 0.0	7.9 <u>+</u> 0.14
	(2)	(2)	(3)	(2)	(2)	(2)	(1)	(2)
Downstream	3.5 <u>+</u> 0.71 *	14 <u>+</u> 4.9	19 <u>+</u> 1.4	0.8 <u>+</u> 0.28	0.1 <u>+</u> 0.0	3.4 <u>+</u> 0.07	366 <u>+</u> 0.0	8.0 <u>+</u> 0.07
	(2)	(2)	(3)	(2)	(2)	(2)	(1)	(2)

- All values mg/l except pH

-123-

- () is # of observations

* Mean includes data below detection limit, limit used in average

Sugar River Chemistry

Mean and standard deviations of upstream, midstream, and downstream Sugar River samples are tabulated in Table 4.8. Complete data are located in Appendix J. All values were typical of a river of this size and were uniform upstream to downstream. The Brodhead ridge and furrow system did not adversely affect the Sugar River. It must be remembered, however, that the contaminant plume could be sinking below the river due to density effects (see discussion in groundwater chemistry section).

Site Observations

During the course of the project, the following site observations were made at the Brodhead ridge and furrow facility: 1) wastewater distribution and solid build-up, 2) cover crop, 3) winter operation, and 4) monitoring equipment performance.

Wastewater distribution at Brodhead was 100%. That is, during a cell loading, furrows and headers contained wastewater. The grid pattern of headers, crossed by furrows, was a very efficient distribution design. At times during the study, however, furrows became flooded to the point where the water level was as high as the ridge tops. This was especially true after heavy rainfalls. A lower hydraulic loading rate (ie. more system area) may be necessary if wastewater flows increase in the future.

Settled solids build-up was also a problem. Equipment was used to clear the furrows in the spring of 1983 and within a year, one to six inches

-124-

of solids build-up had reoccurred. The need to remove suspended solids at the plant was discussed earlier.

The cover crop of canary grass and weeds at Brodhead flourished with the additional water and nutrients provided by the wastewater. During early spring, grasses were brown and knocked down to about knee high. Grasses grew hip high and weeds grew head high during late spring and early summer. In July and August, grasses and weeds began dying and were blown over. Regrowth along the furrow edges occurred during fall rains. The cover crop did not protect the furrows from the elements that well during the winter at Brodhead since the furrows were wide (about four feet).

Winter operation at the Brodhead ridge and furrow facility was not a problem since wastewater effluent temperatures were about 90°F. No matter what the ice or snow conditions were in the furrow, once cell loading began, the ice would melt and infiltration would begin. No change in hydraulic loading occurred during winter conditions. Ice conditions ranged from no ice near header inlets, to two inches at the extremities of a loaded cell, to completely frozen in resting furrows.

The monitoring equipment used at this site (wells, lysimeters, bailers, etc.) was quite adequate. The only problems which occurred concerned the teflon lysimeters. To obtain a sample of sufficient volume (greater than 50 ml) for chemical analysis, a 20 inch (mercury) vacuum was applied to a lysimeter and a two to three day vacuum period was used. This technique was used on lysimeters 1, 2, 3, 5, 6, 8, and 9. Remembering that the flow times through the unsaturated zone were about 3

-125-

days, a lysimeter sample was not really instantaneous as was assumed in the previous analysis.

Lysimeters did not operate in shallow, drier regions (about one foot) at Brodhead as well. Lysimeters 4 and 7 never provided samples during the study. This was most likely the result of easy air entry form the surface, after vacuum application, and low soil moisture in these particular locations. Teflon is hydrophobic and, therefore, at low soil moisture contents, a high, continuous vacuum may have been necessary.

Winter conditions also caused trouble with lysimeter operation. During sub-freezing weather, ice droplets would form in the tygon tubing and the tubing itself would contract. The tubing near the clamps actually closed. The pressure provided by the hand pump typically could not overcome these blockages. Lysimeter samples were obtained on sunny, 20°F days after working the tygon tubing open. A wider diameter tubing (one quarter inch) and a different type of tube closing valve could help overcome these problems.

Brodhead Nitrogen Budget

A nitrogen balance was estimated for the unsaturated zone of each cell at the Brodhead ridge and furrow system. Wastewater flow readings, waste nitrogen data, plant uptake results and, deep lysimeter nitrogen data were used in this estimate. The balance was on a total pounds per year basis.

Additions to the budget came from applied wastewater nitrogen. Total TKN plus NO₃-N was used assuming all N would have eventually entered the soil. Nitrogen fixation and rainfall N addition were assumed to be

-126-

negligible. In chapter 2, a large input estimate of 12.5 lb N/acre/yr was given for rainfall. This quantity would have been added by the wastewater in about four days. Nitrogen losses in the balance were by plant uptake and leaching. The difference between these additions and losses was accounted for by denitrification. It was further assumed that precipitation and evaportranspiration water volumes canceled out each other (ie. all leaching vertical flow was from wastewater), no volatilization occurred (pH < 9), and all NH4⁺-N adsorption sites were saturated (ie. no soil storage). Results of the budget are shown in Table 4.10; calculations appear in Appendix K.

TABLE 4.10

NITROGEN BUDGET ESTIMATE - BRODHEAD SITE

ADDITION/LOSS	CELL 1		CEL	L 2	TOTAL		
		% of		% of		% of	
	lb/yr	added N	lb/yr	added N	lb/yr	added N	
Wastewater	2687	100	2687	100	5375	100	
Net Plant Uptake ^a	0	о	0	o	0	o	
Leaching	541	20	1304	49	1845	34	
Denitrifica- tion	2146	80	1273	· 51	3529	66	

(4.7 acres)

a: net uptake loss = Grass N before burning - Grass ach N after burning (no burning at Brodhead) With this procedure, a denitrification loss of 80% of applied N calculated for cell 1 and 51% for cell 2 as shown in Table 4.10. Leaching accounted for 20% of applied N loss in cell 1 and 49% of applied N loss in cell 2. Plant uptake was zero because of reasons discussed earlier. Considering the total Brodhead site area, leaching accounted for 34% of applied-N and denitrification accounted for 66% of nitrogen losses.

The difference in denitrification losses between the cells made sense considering the past use of the system. Prior to 1980, cell 2 was used only as a backup. This cell still provided good aeration to percolating wastewater and probably contained fewer anaerobic microzones than cell 1. These zones, as described in Chapter 2, would have enhanced denitrification.

CHAPTER 5: MINDORO SITE - RESULTS AND DISCUSSION

Site Description

The Mindoro Co-op Creamery is located in Mindoro, Wisconsin, in northern La Crosse County. The plant produces 10,000 lb of colby cheese per day with 14,000 gpd (average) of processing wastewater being treated by a 3.0 acre ridge and furrow system consisting of three cells. As mentioned in Chapter 2, this system was the first such treatment installation in Wisconsin beginning operation in 1954. A topographical map showing the general location of the system appears in Figure 5.1.

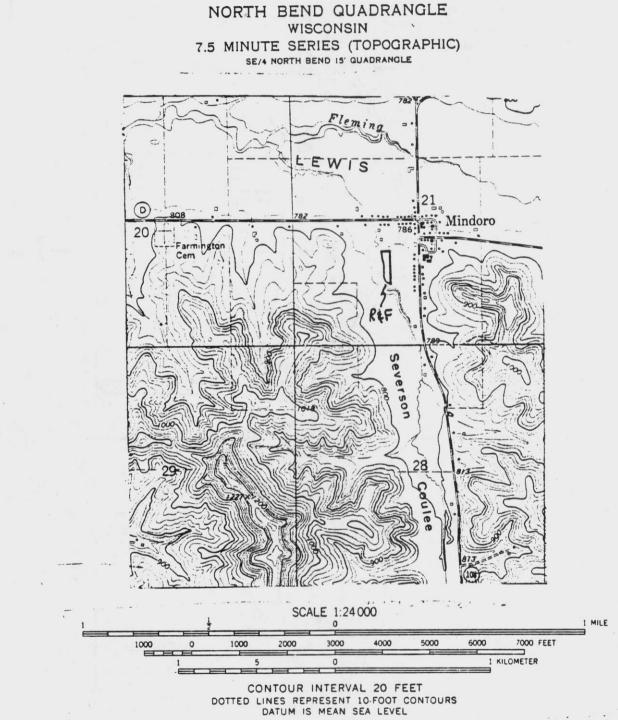
A plan view of the treatment system is illustrated in Figure 5.2. Well, lysimeter, stream stage, and infiltration station locations are indicated as well as cell locations and areas. General information concerning well and lysimeter depths and location is presented in Table 5.1. Complete well and lysimeter logs are given in Appendix AA.

The ridge and furrow site is located over about 12 feet of silt loam soil underlain by a sand and gravel aquifer next to Severson Coulee. This sand and gravel material extends to a depth of approximately 85 feet and overlays a sandstone containing shale seams. The Soil Conservation Service describes the overlying loam as a Toddville loam which is a deep, well to moderately well drained, silty soil formed on stream terraces (WDNR, 1984).

Results of this study's soil analysis at Mindoro indicated the following parameter ranges for the overlying silt loam:

-129-

FIGURE 5.1



1.00

. . .

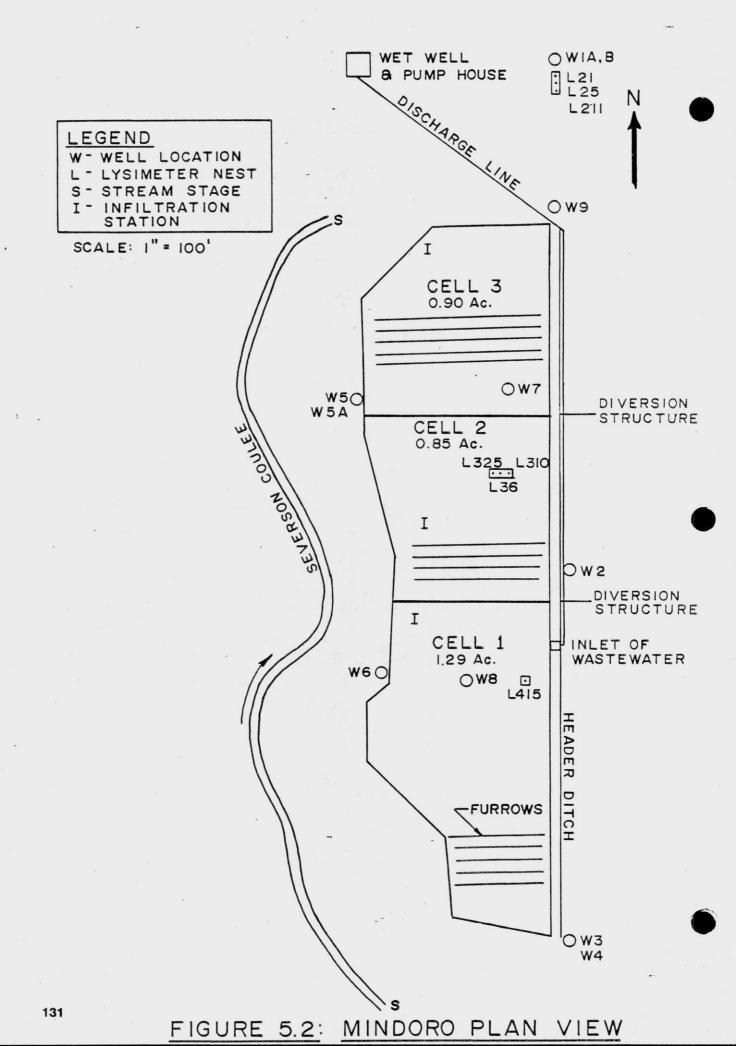


TABLE 5.1a

WELL SPECIFICATIONS

AT MINDORO CREAMERY

.

	Well Top	Depth	Well Point	Approximate	Screen Length	
Well*	Elevation	(ft)	Elevation	Surf. Elev.	(ft)	Location
1A	781.59	13.0	766.59	779.6	5	Background
1B	781.36	35.5	744.16	779.7	2.5	Background
2	783.61	13.2	768.61	781.8	5	Adjacent to Cell 2
3	785.32	15.2	768.49	783.6	5	Adjacent to Cell 1
4	785.28	35.7	747.78	783.4	2.5	Adjacent to Cell 1
5	782.63	15.0	765.80	780.8	5	Downstream of Cell 3
5A	782.31	31.9	749.01	780.9	2.5	Downstream of Cell 3
6	785.69	17.5	765.79	783.3	5	Downstream of Cell 1
7	782.00	14.0	766.80	780.8	2.5	Cell 3
8	783.85	9.5	772.05	781.6	2.5	Cell 1
9	779.68	12.2	765.98	778.2	2.5	Adjacent to Cell 3

#All wells PVC, 2 inch inside diameter.

TABLE 5.1b

LYSIMETER SPECIFICATIONS

AT MINDORO CREAMERY

Lysimeter	Depth Below Furrow (ft)*	Location
21	1.0	Background
25	5.0	Background
211	11.0	Background
325	2.5	Cell 2
36	6.0	Cell 2
310	10.0	Cell 2
415	1.5	Cell 1

*Depth is to top of Teflon cup

į

- sand :	11-15%
- silt :	64-68%
- clay :	19-25%
- total N:	0.01-0.31%
- CEC :	11-22 mg/100 gram soil
- pH :	5.5-7.9

The higher silt and total N fractions occurred in shallow samples. Also CEC tended to decrease with depth and soil pH increased with depth. Complete soil analysis data are presented in Appendix BB. In summary, the soil immediately below the treatment system is primarily silt with some clay. The silt loam had an average CEC of 14.5 mg/100 gram of soil.

Soil borings during well and lysimeter installation also brought out the following five points.

- 1) moist, sticky clays appeared deeper in bore holes,
- 2) some holes had a blue-gray clay just above the sand layer
- 3) the top of the sand tended to be greenish-blue in color,
- 4) in-cell borings had 5-10 feet of gray mottled clay, indicating a flucuating water table, and
- 5) in-cell borings near the north end of cell 3 had groundwater elevations within 1 to 2 feet of the surface.

Wastewater Chemistry

Mindoro Creamery's wastewater had mild strength when compared to the Brodhead site. Mindoro's average wastewater BOD5, COD, TKN, and chloride

-134-

TABLE 5.2

MINDORO WASTEWATER CHARACTER

Parameter	Samples	Mean	Median	Range	SD	Strong Typical ^a Domestic
TOTAL BOD5*	74	830	860	430-1300	280	400
TOTAL COD	12	1200	1300	660-1900	360	1000
TSS	14	262	194	80-616	163	350
TOTAL TKN	14	32	34	14-52	11	85
DISSOLVED TKN	2	24	24	16-31	11	
NH3-N	10	0.9	1.1	0.1-1.8	0.6	50
NO3-N+NO2-N#	10	0.4	0.2	0.1-1.6	0.5	0
C1-	14	100	91	70-210	34	100
рH	8	7.5	7.8	5.1-9.5	1.4	

(all units mg/l except pH)

a from Metcalf and Eddy, 1979

- Mean contains data below or above a detection limit; the limit was included in average

SD - Standard Deviation

concentrations were 830 mg/l, 1200 mg/l, 32 mg/l, and 100 mg/l, respectively. Means, medians and ranges of these and other chemical parameters of the wastewater are provided in Table 5.2. Values of a strong typical domestic waste are provided for comparison. A complete list of data is given in Appendix CC.

There were three general observations made concerning the Mindoro wastewater chemistry. First, the results were highly variable and this variance was not seasonal. For example, the average total suspended solids value was 262 mg/l but the standard deviation was 163 mg/l and the range was 80-616 mg/l. These variations, as mentioned in Chapter 4, were most likely due to the variation in the amount of rinse water used in the plant.

Second, the average pH (7.5) of the wastewater was slightly above neutral. Since major volatilization of NH₃ occurs at pH greater than nine, one would expect negligible gaseous loss of ammonia from this wastewater.

Finally, the nitrogen fraction of the wastewater was primarily organic-N. This was expected since the wastewater was derived from cheese production. Unlike the Brodhead wastewater, however, only around 25% of this organic-N was tied up in the solid fraction. This was seen by comparing total and dissolved TKN concentration averages (Table 5.2).

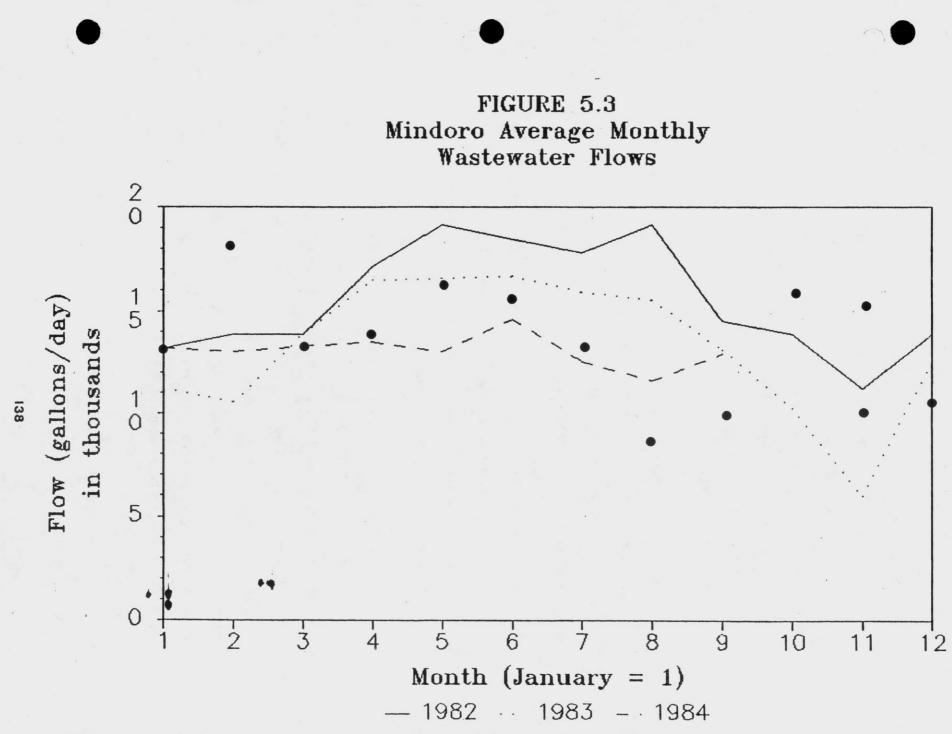
-136-

Wastewater Hydraulic Loading

All flow to the Mindoro ridge and furrow system was derived from cheese production. A low and a high speed pump, alternated periodically, were used to convey flow to the system. Discharge was calculated two ways. During sampling days, the flow was found by first dividing the average pump time, in minutes, into the total wet well volume discharged (260 gallons) per pump run. Then, using the pump's hour meter, this value was multiplied by a pumping hours per day factor and a time conversion to obtain the gallons per day value. For example, the average pump time to remove 260 gallons from the wet well was 3.09 minutes. If the pump ran 2.59 hours during the 24 hour sampling period, the wastewater flow during that day was 13,100 gallons (ie. 260 gal/3.09 min $x_{2.59}x$ 60 min/hr = 13,100 gal). Wastewater volume flowing into the wet well during pumping was neglected. Thirty-day monthly averages were obtained by relating the monthly metered pumping time to a pump time versus discharge curve.

Figure 5.3 presents the 30-day average, monthly Mindoro wastewater flow readings for the years 1982-1984 as well as 24-hour flows measured on project sampling days. A complete listing of flow data for the past three years is given in Appendix DD. Thirty-day monthly flows averaged 14,000 gallons per day since 1982. One can see from Figure 5.3 that from 1982 to 1984, flows ranged from 6006 to 19,140 gallons per day. These discharges were well under the WDNR permit flow of 25,000 gpd.

-137-



--24-HR flows determined on sampling days indicated by

A seasonal trend existed in flow volumes to the Mindoro system. Flows were high from April to July with values decreasing during the hot summer months, fall, and winter. Cheese production at Mindoro was more dependent on local milk supply than Brodhead. The most productive milking period is in spring and early summer. Summer flow readings in 1984 were lower due to a process change made in March which allowed for recycling of cooling water.

A flexible load/rest cycle was followed at this site. The intent was to load a cell for one month and rest it for two months. Each month, the longest rested cell was placed on line. It was not uncommon during this study, however, to see the same cell loaded for three to four months.

Due to inefficient header ditch flow control and blocked furrow openings, wastewater distribution at the Mindoro site was poor. Typically, 1/3 of cell 1, 1/5 of cell 2, and 1/3 of cell 3 were loaded no matter which cell was technically being loaded.

Total area and actual area hydraulic loading rates were calculated and are presented in Table 5.3. Using the total site area, the average hydraulic loading rate was 0.172 inches/day (4670 gpad) with a range of 0.074-0.235 inches/day (2000-6380 gpad). This is classified as a fairly medium rate system (Rodenberg, 1980). A high rate classification would result if actual hydraulic loadings were considered. These values were over three times greater than total area calculations and almost two times greater than average Brodhead system rates.

-139-

TABLE 5.3

MINDORO HYDRAULIC LOADING RATES

units: in/day (gal/acre/day)

Flow (gal)	Total Area Loaded	Actual Area Loaded
6006 (min)	0.074	0.246
	(2000)	(6670)
14,000 (ave)	0.172	0.573
14,000 (2007	(4670)	(15,600)
19,140 (max)	0.235	0.783
·) ; · · · · · · · · · · · · · · · · ·	(6380)	(21,300)

Total Area = 3.0 acres Actual Area = 0.9 acres

Organic Loading Rates (BOD5, TKN)

Code NR 214 of the Wisconsin DNR states that a ridge and furrow system should receive no more than 100 lb BOD5/acre/day. Using the average total area hydraulic loading rate and the average BOD5 concentration of the wastewater, the Mindoro site received 32 lb BOD5/acre/day. Using the minimum and maximum total area flow rates and the average wastewater BOD5 concentration, the BOD5 loading rate range was 14-44 lb BOD5/acre/ day. These numbers were well under the code requirement.

If the actual area hydraulic loading rates were used, however, higher BOD₅ loading rates resulted. The average rate was 108 lb BOD₅/acre/day with a range of 46-147 lb BOD₅/acre/day. These results are presented in Table 5.4.

-140-

Similar calculations for TKN loading rates were performed and are also presented in Table 5.4. There is currently no Wisconsin DNR code for nitrogen loading. Using the range of total area hydraulic flows presented in Table 5.3 and the average wastewater TKN concentration, TKN loading rates at Mindoro ranged from 0.53-1.7 lb N/acre/day with an average of 1.2 lb N/acre/day. Using the range of actual area hydraulic flows and the average wastewater TKN concentration, TKN loading rates ranged from 1.8-5.7 lb N/acre/day with an average of 4.2 lb N/acre/day. Suggestions for possible loading rates will be made in Chapter 6.

TABLE 5.4

MINDORO ORGANIC LOADING RATES

(Units are lb/day/acre)

Total Area Loaded	Actual Area Loaded
14	46
32	108
44	147
0.53	1.8
1.2	4.2
1.7	5.7
	Loaded 14 32 44 0.53 1.2

Ave BOD5 Conc	=	830	mg/l
Ave TKN Conc	=	32	mg/l
Total Area	=	3.0	acres
Actual Area	=	0.9	acres

Groundwater Elevations and Flow

In general, groundwater flow at the Mindoro ridge and furrow was north northwesterly, following in the direction of Severson Coulee. An approximate gradient in the sand layer from well 4 to well 5A was 0.006 ft/ft. Vertical gradients were observed between wells 1A and 1B, 5 and 5A. and 3 and 4 and mounding of groundwater within the system was detected during borings. When standing water was present in well 1A, its elevations were higher than elevations in well 1B, indicating downward flow from the silt to the sand layer. Elevation differences between wells 5 and 5A were variable during the study. If cell 2 was loaded, the groundwater beneath the northern section of cell 2 would mound, and a downward gradient would result. As cell 2 rested, the mound dissipated and upward gradients would begin. The gradients near wells 3 and 4 were unexplainable. The southern section of cell 1 was continually ponded during the project, yet, oscillating elevation differences between these two wells indicated both upward and downward gradients. Downward gradients at Mindoro ranged from 0.01 to 0.04 ft/ft. A well nest within the system would have better defined these gradients. Figure 5.4 shows the groundwater contours on November 5, 1984. Appendix EE contains a complete list of elevation and contour data.

Groundwater (water table) elevations varied 0.7 to 4.7 feet during the project. Time plots for the elevations of wells 5, 6, and 7 are graphed in Figure 5.5. High level values in September (1983) with a subsequent decrease were most likely the result of high summer wastewater flow

-142-

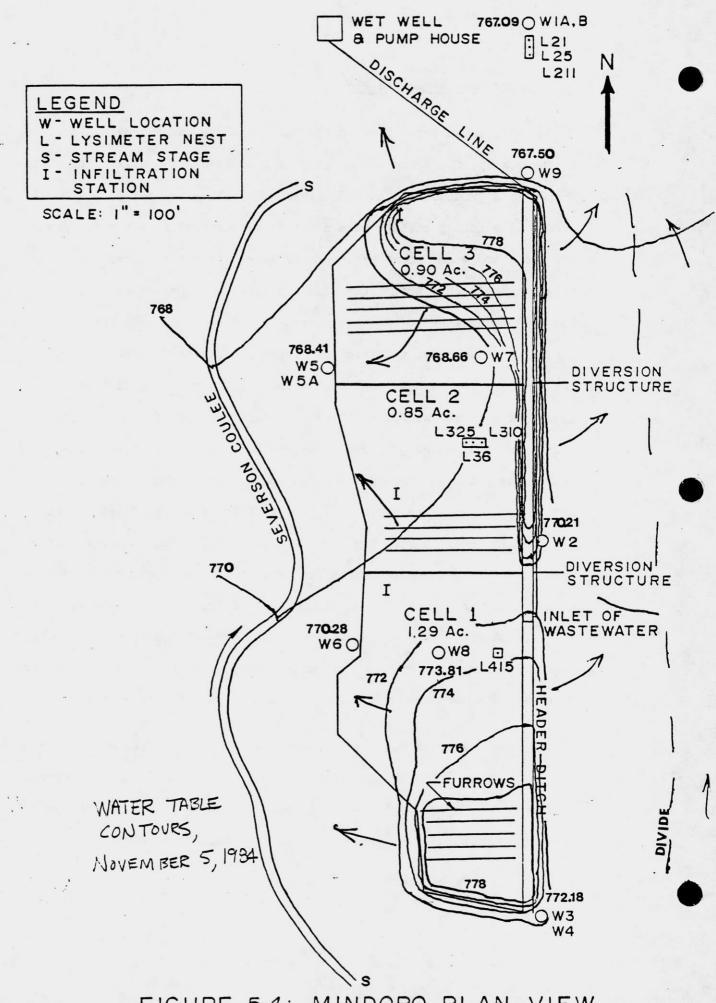
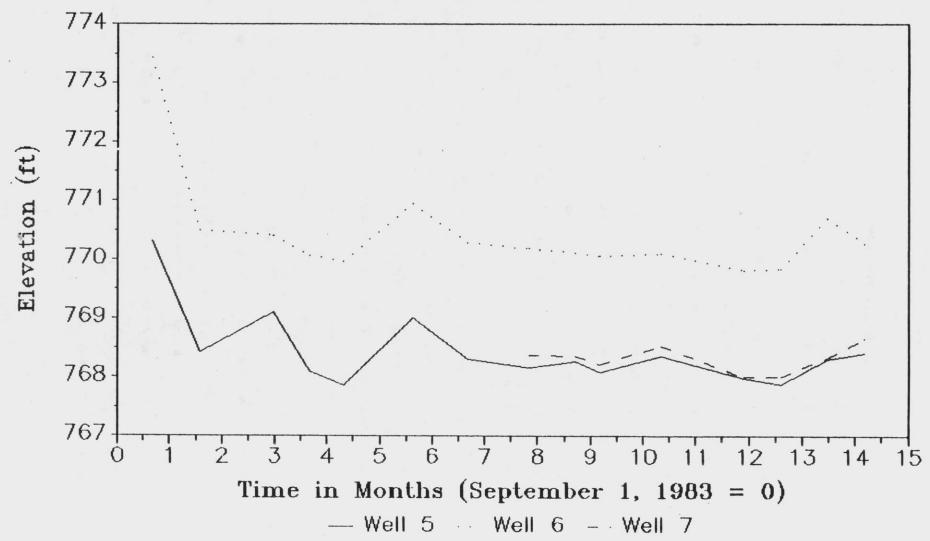


FIGURE 5.4: MINDORO PLAN VIEW

FIGURE 5.5 Mindoro Groundwater Elevation vs. Time for wells 5, 6, and 7



Elevations are mean sea level

followed by lower fall waste flows. Elevation changes could not be correlated with wastewater flow after this period though. No seasonal pattern of water table fluctuation was observed.

Once percolated wastewater entered the saturated zone at Mindoro, its flow regime was complex. In the silt loam layer, flow within the system was principally vertical with some horizontal movement away from the mound. Flow in the silt and upper sand layers beneath the site was totally derived from recharge (wastewater) to the system. This meant a flow divide was created east and south of the site, separating background groundwater flow from wastewater derived flow. This is depicted on Figure 5.4. Upon entering the sand layer, the groundwater flow was horizontal to the northwest.

Determination of flow velocities through the silt layer and the sand layer was difficult and only estimates were made. Darcy's Law (V=KI/n) was used to find the average vertical velocity in the silt loam layer. Slug tests were performed in November (1984) to determine the hydraulic conductivity near wells 1A, 2, and 8. Calculations are presented in Appendix EE. Results of the tests provided K values of 7 x 10⁻⁷ ft/s, 2 x 10⁻⁵ ft/s, and 2 x 10⁻⁷ ft/s for wells 1A, 2, and 8, respectively. It must be realized that these values were for horizontal hydraulic conductivity. Therefore, it was assumed that the silt loam was homogeneous and isotropic. A spatially averaged vertical K of 7 x 10⁻⁶ ft/s was used. An average vertical gradient of 0.025 ft/ft (from 0.01-0.04 ft/ft range) and a typical porosity for silt of 0.4 (Freeze and Cheng, 19;9) were also assumed.

-145-

With the above input data, a vertical velocity of 0.04 ft/day was determined. Assuming that the sand layer is one to 10 feet below the water table, travel times through the saturated silt strata of between 25 and 250 days were calculated. These values were reasonable.

Unsuccessful slug and bail tests to determine K in the sand layer were attempted on well 1B in April (1984). The well returned to equilibrium too quickly to acquire meaningful data. Since this response was similar to the slug tests attempted at Brodhead, an estimated hydraulic conductivity of 0.0005 ft/s was used. Imputing the gradient of 0.006 ft/ft and a porosity of 0.35 (for sand) into Darcy's Law provided a horizontal linear velocity of 0.7 ft/day in the sand layer. At this velocity, it would take about three years for groundwater to flow from the south end to the north end of the site.

Groundwater Chemistry

Chemical analyses of groundwaters for each well at Mindoro are listed in Table 5.5. Complete data listings are given in Appendix FF. When reviewing chloride data, a good indicator of contamination, one can categorize the wells into two types. Wells 1B and 4 were not affected by the ridge and furrow system whereas wells 2, 3, 5, 6, 7, 8, and 9 were influenced by the system. Except for well 2, all of the affected wells contained 77 to 92% of the wastewater chloride concentration on average and demonstrated low standard deviations. This verified the assertion made earlier that the groundwater below the system was primarily derived from applied wastewater. The average wastewater concentration was 100 mg/l. Well 2 chloride values decreased about 70%

-146-

TABLE 5.5

.

•

MEAN AND STANDARD DEVIATION OF
GROUNDWATER CHEMICAL PARAMETERS
AT MINDORO

	Dissolved	Dissolved		Dissolved	Dissolved	Dissolved		Field
Well	BOD ₅	COD	TDS	TKN	NH3-N	NO2-N+NO3-N	C1-	рH
14					0.1 ± 0.0 (1)	1.9 ± 0.0 (1)		
1B	3.3 <u>+</u> 1.1* (8)	5.2 <u>+</u> 0.71 (8)	350 <u>+</u> 11.2 (8)	0.2 <u>+</u> 0.1 [#] (8)	0.1 <u>+</u> 0.0* (8)	3.3 <u>+</u> 0.61 (8)	9.3 <u>+</u> 0.54 (3)	7.4 <u>+</u> 0.15
2	9.1 <u>+</u> 14# (10)	18 <u>+</u> 20 (11)	537 <u>+</u> 127 (10)	1.5 <u>+</u> 0.88 (11)	0.3 ± 0.2 (13)	0.1 <u>+</u> 0.06 * (13)	59 <u>+</u> 20 (9)	7.5 ± 0.20
3	3.0 ± 0.56 (13)	11 <u>+</u> 3.1 (13)	692 <u>+</u> 51.2 (13)	0.6 <u>+</u> 0.2 (13)	0.1 ± 0.06 (13)	0.1 <u>+</u> 0.06# (1 <u>3</u>)	77 <u>+</u> 7.3 (13)	6.9 ± 0.17
4	3.1 ± 0.28 (13)	5.0 <u>+</u> 0.04 (13)	278 <u>+</u> 5.34 (13)	0.2 <u>+</u> 0.07 * (13)	0.1 <u>+</u> 0.0# (13)	0.2 <u>+</u> 0.09 (13)	2.1 <u>+</u> 0.19 (13)	7.3 ± 0.20
5	$\begin{array}{c} 11 + 7.71 \\ (13) \end{array}$	31 <u>+</u> 9.5 (12)	715 <u>+</u> 42.4 (13)	4.0 <u>+</u> 0.74 (13)	3.0 <u>+</u> 1.0 (1 <u>3</u>)	0.1 <u>+</u> 0.0* (13)	91 <u>+</u> 2.3 (13)	6.7 <u>+</u> 0.14 (8)
6	4.5 <u>+</u> 2.2* (13)	25 <u>+</u> 3.6 (12)	737 <u>+</u> 26.8 (1 <u>3</u>)	1.8 <u>+</u> 0.66 (13)	1.0 <u>+</u> 0.40 (13)	0.1 <u>+</u> 0.0# (1 <u>3</u>)	77 <u>+</u> 3.7 (1 <u>3</u>)	6.8 <u>+</u> 0.13 (8)
7	7.6 ± 6.9 (7)	18 <u>+</u> 10 (7)	690 + 88.6	2.5 <u>+</u> 0.50 (7)	1.9 <u>+</u> 0.42 (7)	0.1 <u>+</u> 0.0 * (7)	85 <u>+</u> 5.7 (7)	$6.6^{+} + 0.21^{-}$
8 ·	3.2 ± 0.42 (7)	25 + 4.3 (7)	774 <u>+</u> 201 (7)	2.5 <u>+</u> 1.1 (7)	0.3 <u>+</u> 0.16 (8)	0.2 <u>+</u> 0.1 * (8)	$\frac{84}{(7)}$ $\frac{+}{(7)}$ 0.1	6.8 <u>+</u> 0.21 (3)
9	16 <u>+</u> 12≇ (7)	$\frac{47}{(7)} + \frac{28}{7}$	607 ± 89.9	2.8 ± 0.73	1.5 <u>+</u> 0.34 (7)	0.1 <u>+</u> 0.04 (7)	92 $\frac{+}{(7)}$ 12	6.7 <u>+</u> 0.14 (7)

--->All values mg/l except pH; () indicates # of observations

147.

* - means include values above or below a detection limit; the limit was included in average

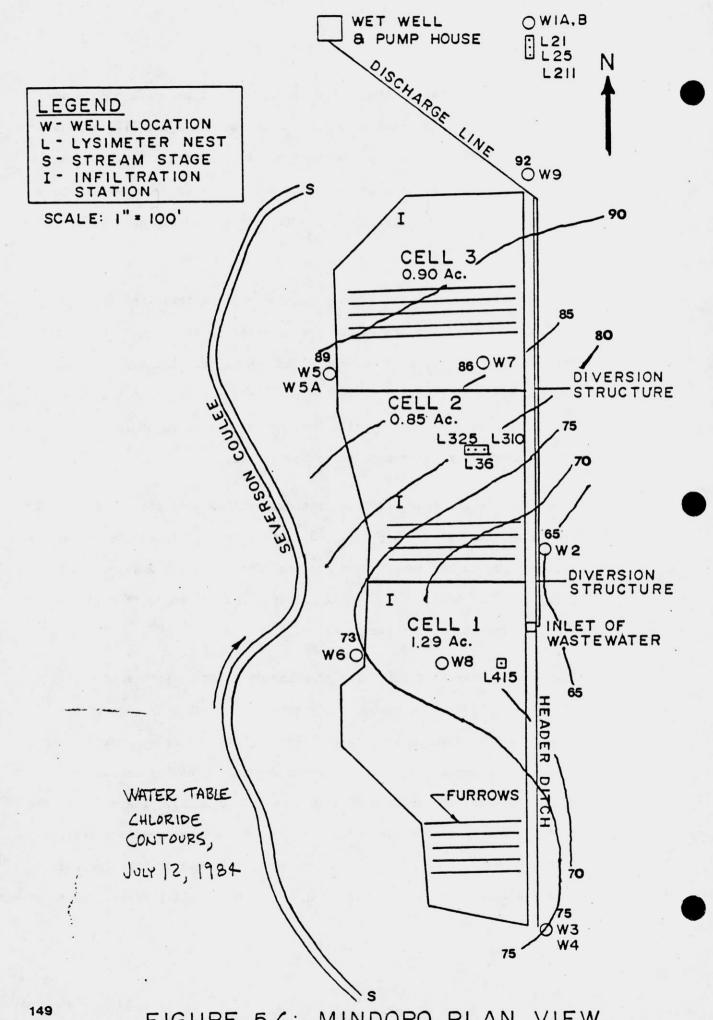
during the last five months of the project. This may have been caused by a shift of the groundwater divide to the west (Figure 5.4), diluting the region of well 2 with groundwater of background quality. It should be realized that averages and standard deviations of well data at this site do not fully describe the contamination and should only be used for relative comparisons.

The pH in wells 1B, 2 and 4 was slightly above neutral, ranging from 7.3 to 7.5 on average. Wells which were affected by the Mindoro ridge and furrow system (3, 5, 6, 7, 8, and 9) had pH values slightly less than neutral, ranging from 6.6 to 6.9. This was the same trend that was witnessed at Brodhead. Since wells were not placed north of the system, it is not known if pH increased again downstream.

Using average chloride and total nitrogen $(TKN+NO_3-N)$ concentrations for the wastewater and wells 5, 6, 7, 8, and 9, one can calculate the well nitrogen losses not caused by dilution. These results are presented in Table 5.6. In these wells, total nitrogen losses ranged from 71-83%. A sample calculation is shown in Appendix FF.

Figure 5.6 presents a plan view of chloride contours at the water table on July 12, 1984. The contours are approximate but they generally represent the contaminated area. These results indicated a high chloride contamination in the southern section of cell 1 (about 80 mg/l) and in the northern section of cell 3 (about 90 mg/l). These areas were ponded with wastewater throughout the project. Concentrations decreased along the eastern boundary of the system as groundwater traveled away from the mounded area and mixed with background quality flow. Chloride

-148-



MINDORO PLAN VIEW FIGURE 5.6:

concentration for wells 1B and 4 were considered background quality. Since wells were not installed northwest of the system, chloride values were unknown downstream of the ridge and furrow system.

TABLE 5.6

AVERAGE GROUNDWATER NITROGEN LOSSES

C1-	Total N	% Reduction of Chloride	<pre>% Reduction of Total N</pre>	% N Losses
100	32.2	0	0	о
91	4.1	9	87	78
77	1.9	23	94	71
85	2.6	15	92	77
84	2.7	16	92	76
92	2.9	8	91	83
	100 91 77 85 84	C1- N 100 32.2 91 4.1 77 1.9 85 2.6 84 2.7	Cl- N of Chloride 100 32.2 0 91 4.1 9 77 1.9 23 85 2.6 15 84 2.7 16	Cl- N of Chloride of Total N 100 32.2 0 0 91 4.1 9 87 77 1.9 23 94 85 2.6 15 92 84 2.7 16 92

AT MINDORO

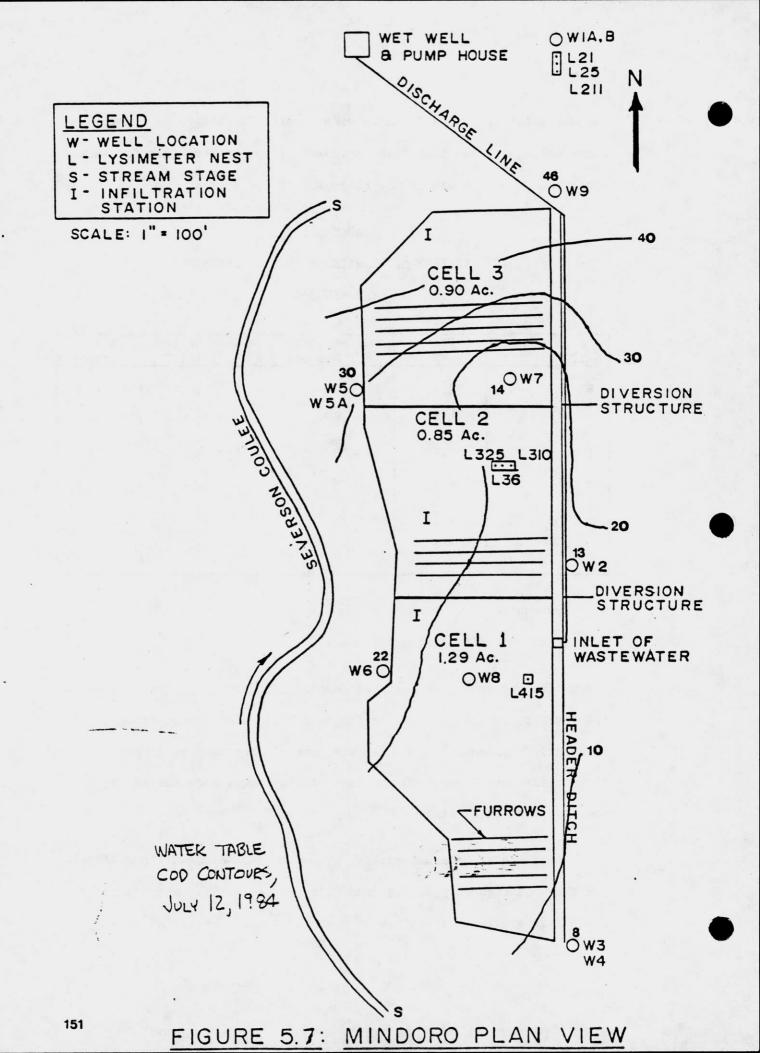
WW = Wastewater

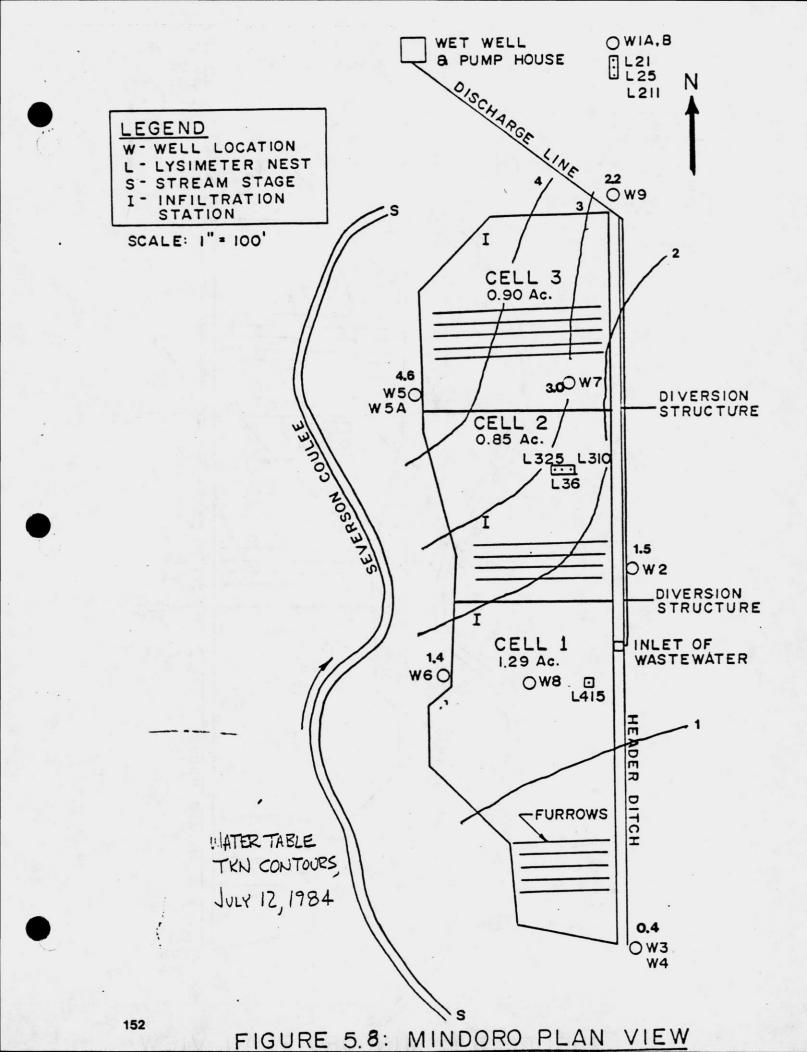
- chloride and total nitrogen in mg/l

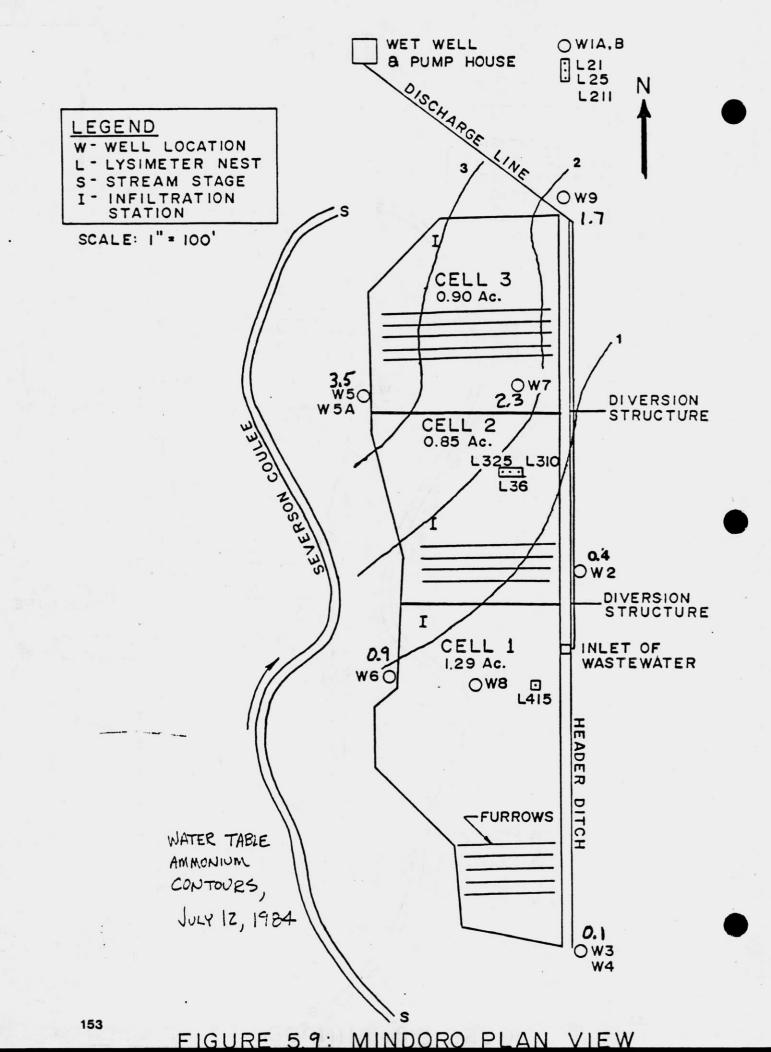
A similar contour pattern existed for COD concentrations (on 7/12/84) at the water table and is shown in Figure 5.7. Again, concentrations increased from south to north in the general direction of groundwater flow in the sand layer. COD concentration changes were not known downstream of the system.

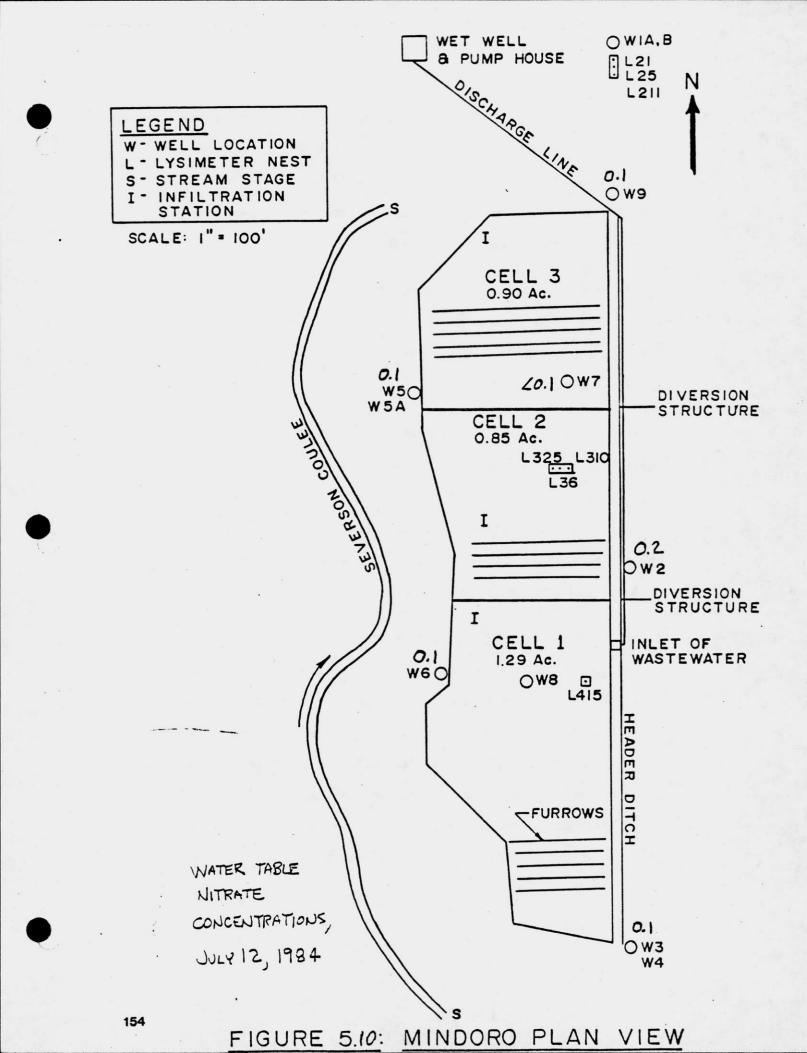
Plan views for the parameters TKN, NH_3-N , and NO_2-N+NO_3-N (from 7/12/84) are shown in Figures 5.8, 5.9, and 5.10, respectively. The TKN and NH_3-N maps are quite similar with concentrations increasing to the

-150-









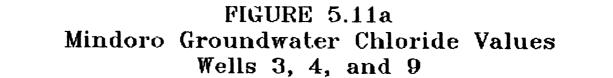
northwest in the groundwater beneath the system. Ammonium values were 75% lower than TKN values on this day. The nitrate concentrations, as shown in Figure 5.10, were low ($\leq 0.2 \text{ mg/l}$) everywhere in the shallow groundwater. Again, nitrogen concentrations downstream of the Mindoro ridge and furrow system where unknown.

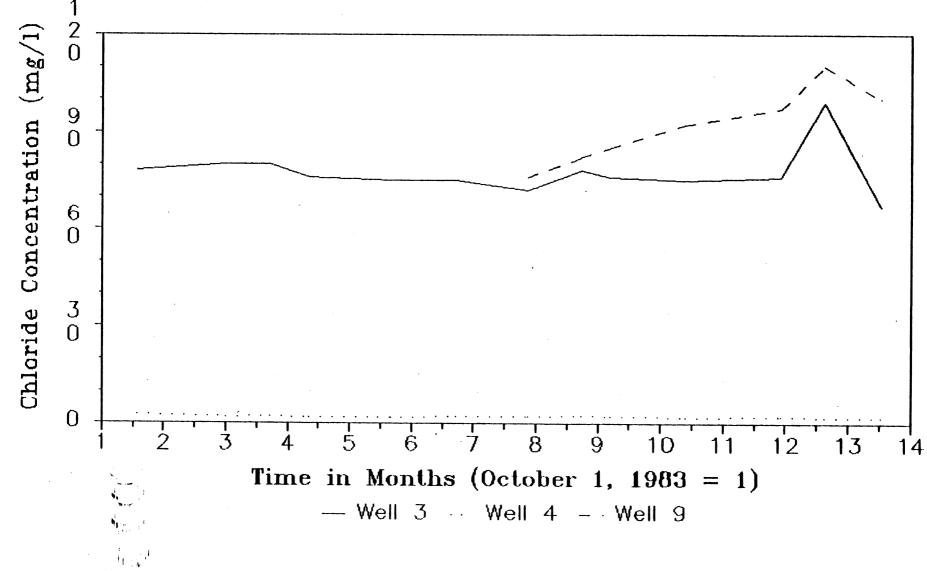
An attempt was made to match chemical changes in space with time between wells. Due to the complexity of the groundwater flow regime, however, this was not possible. General observations of temporal trends in the groundwater follow.

First, chloride and TDS concentrations in the wells were not variable during the Mindoro study. This can be seen in Figure 5.11a and b where chloride and TDS data are plotted versus time for wells 3, 4, and 9, respectively. This graph also shows that downstream well chloride values were of the same magnitude as the wastewater and that no dilution occurred along the groundwater flow path (well 3 to well 9).

Second, well 2 TDS, TKN, and chloride concentrations decreased from June (1984) to October (1984). These plots are illustrated in Figure 5.12a, 5.12b, and 5.12c, respectively. TDS concentrations declined 42%, TKN's decreased 44%, and chlorides dropped 73%. These declines were attributed to the cell loading schedule. Cell 1 had not been loaded since September of 1983. With no wastewater recharge to the cell, the groundwater divide mentioned earlier shifted west and background quality groundwater was able to penetrate the well 2 region, diluting contaminant concentrations.

-155-





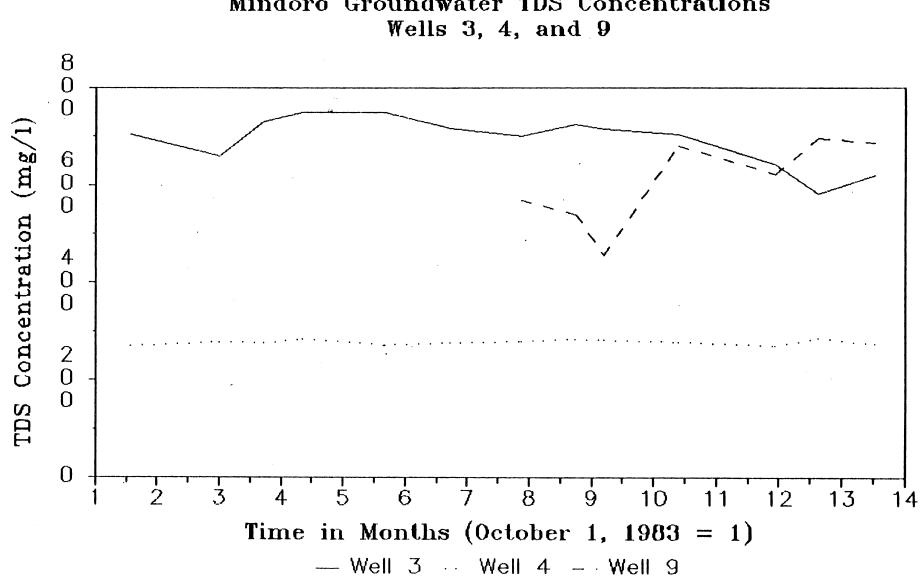
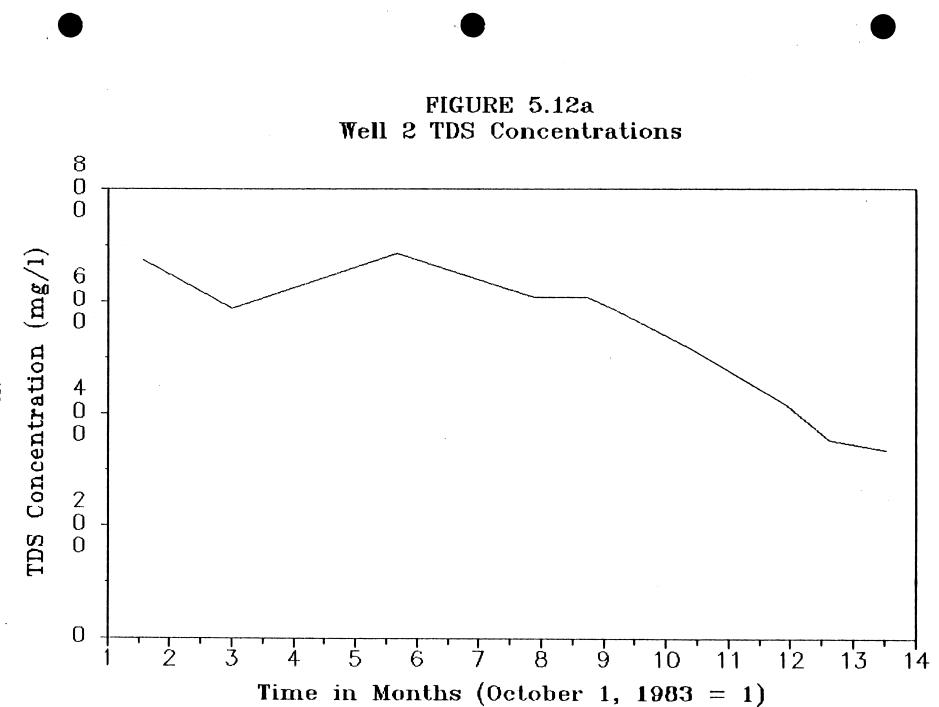


FIGURE 5.11b Mindoro Groundwater TDS Concentrations



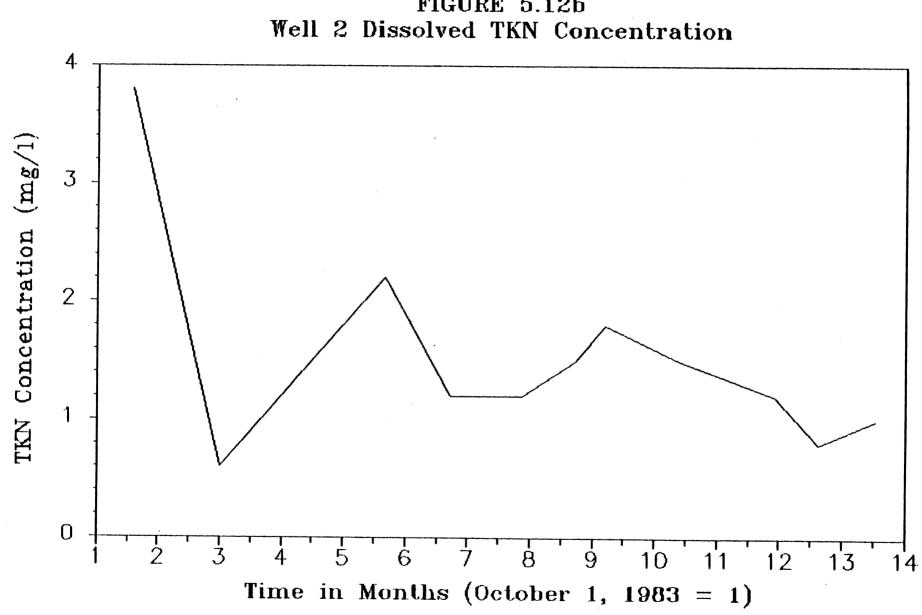
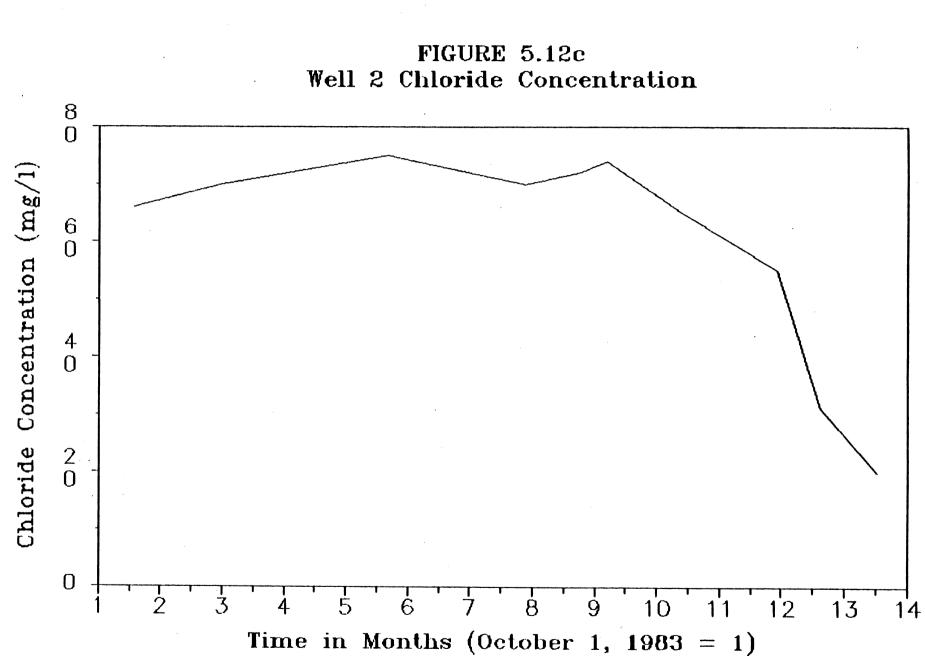


FIGURE 5.12b



Finally, nitrogen concentrations were more variable than chlorides values. This can be seen in Figure 5.13 where total nitrogen data are plotted versus time for wells 3, 5, and 9. This plot also shows that nitrogen values increased downstream. Nitrate time plots shown for these three wells in Figure 5.14 indicated that little NO_3 -N was present in the groundwater and no changes in time were apparent.

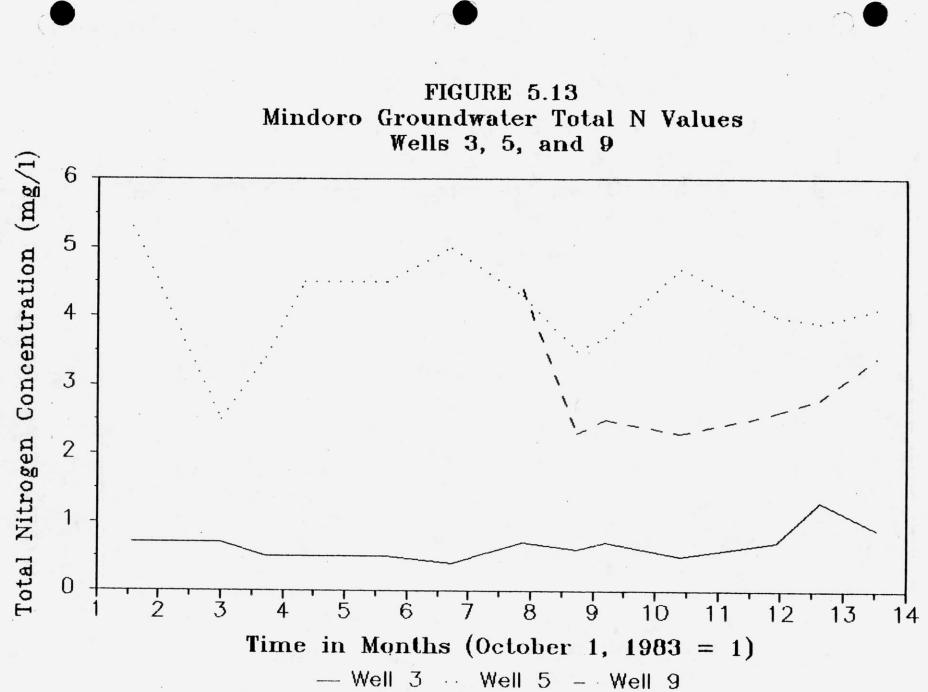
One final comment on the groundwater chemistry at Mindoro is in order. To completely define the chemistry at this site, the installation of a well nest is recommended downstream of the ridge and furrow. This would indicate whether there was nitrification occurring or if a sinking plume existed.

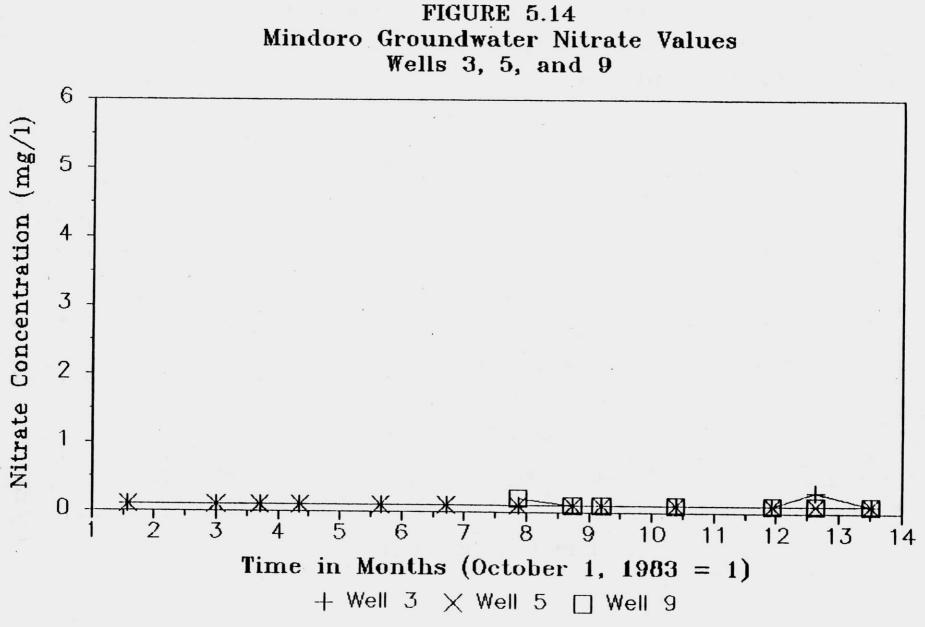
Unsaturated Zone Flow Rates

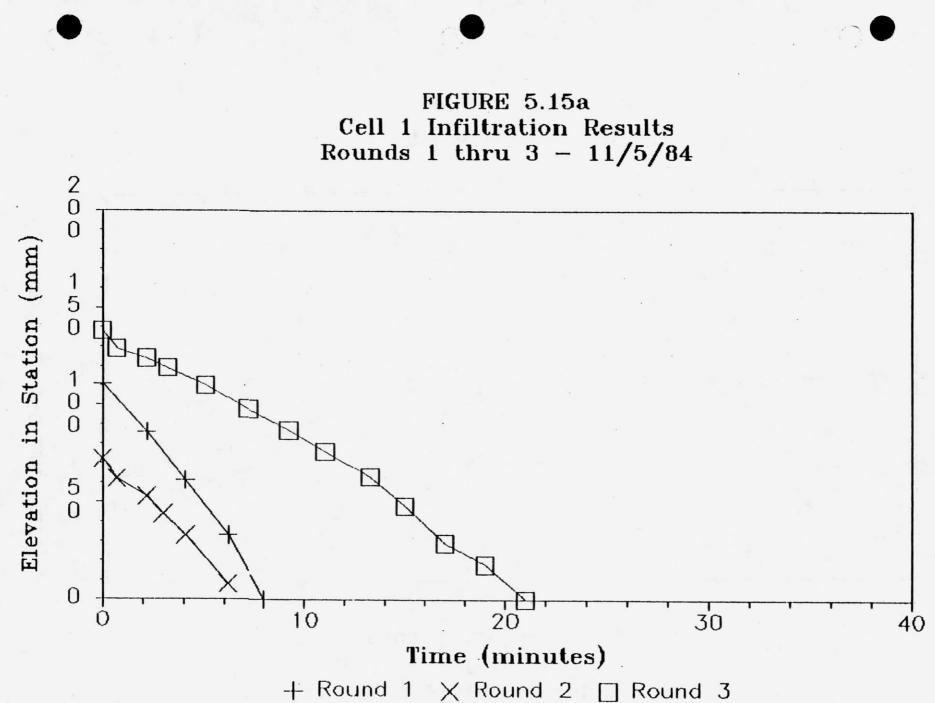
Since furrows at the Mindoro ridge and furrow did not contain a solids barrier, one vertical unsaturated flow rate range was estimated. Two methods were used for this calculation; infiltration station results and the methods of Bouma (1975).

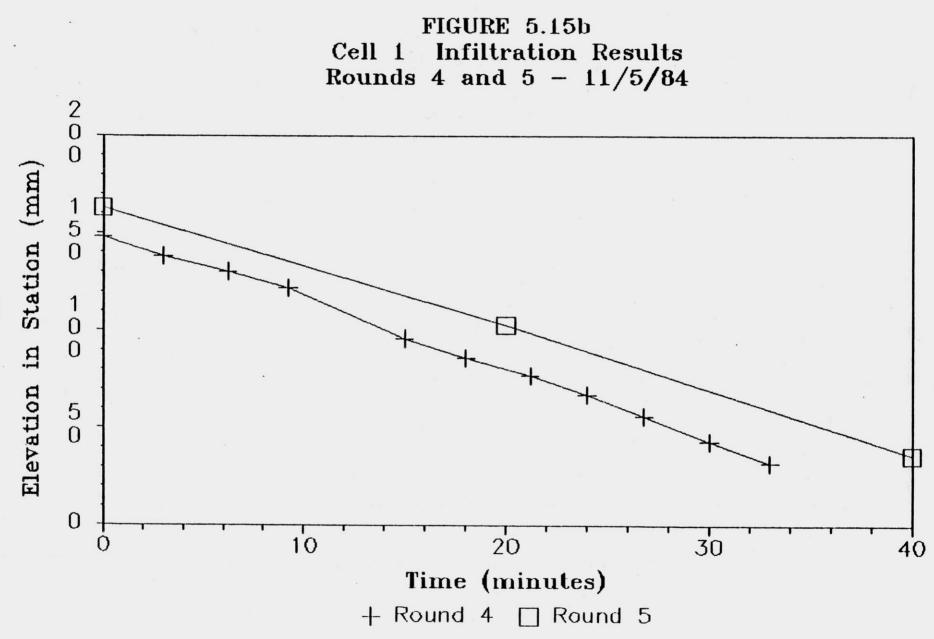
Infiltration stations, installed in November (1984), were used as one method of determining unsaturated zone flow-through times. Results of this work for cells 1 and 2 are presented in Figures 5.15 and 5.16 and summarized in Table 5.7. Cell 3 was loaded in November and water head in this infiltration station did not decline. Cell 1 and cell 2 stations were located in areas that were not loaded during the project. The cell 3 station was representative of unsaturated hydraulic conductivities in a heavily loaded area.

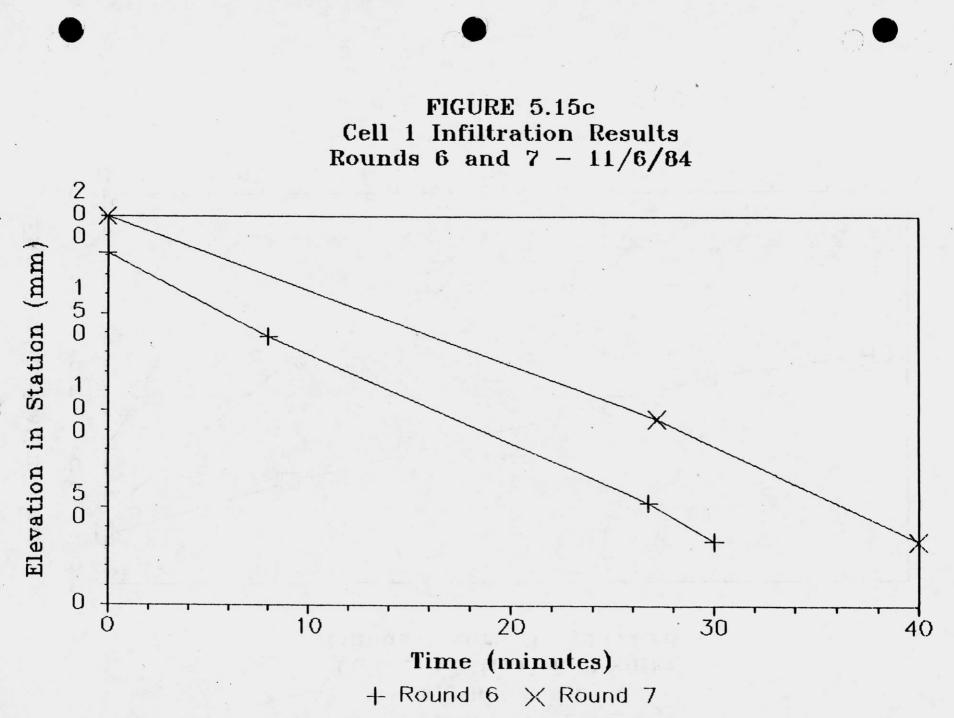
-161-

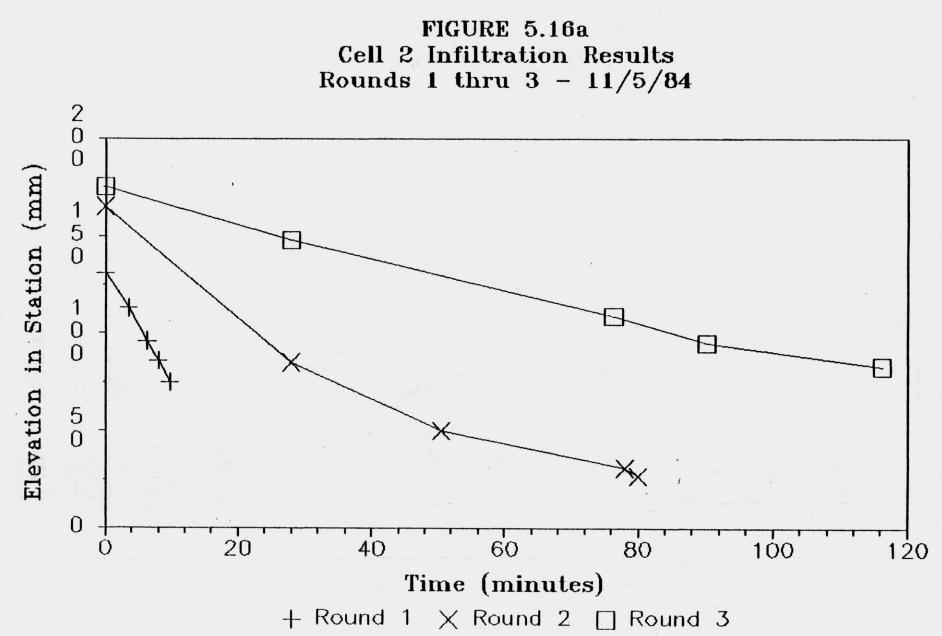












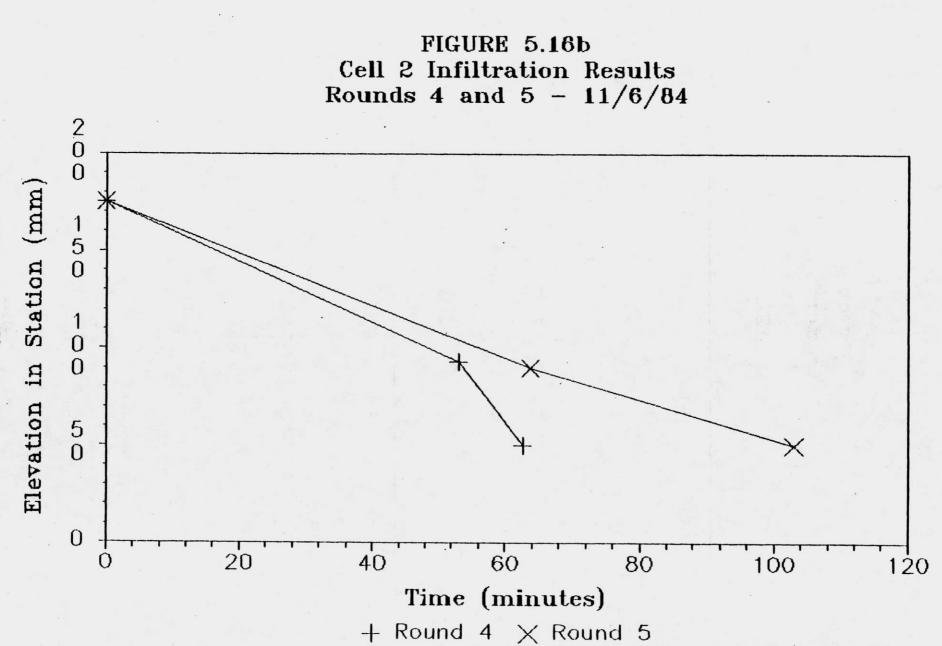


TABLE 5.7

MINDORO

INFILTRATION RATES

CELL 1

Round	H (cm)	t (min)	cm/d	
. 1	11.1	8	1998	
2	6.4	6	1536	
3	13.8	21	946	11/5/84
4	11.6	33	506	
5	12.7	40	457	
6	14.8	30	710	11/6/84
. 7	16.7	30	802	•

CELL 2

Round	H (cm)	t (min)	cm/d	
1	8.4	10	1210	
2	20.6	80	371	11/5/84
3	13.9	116	173	
4	18.8	62.5	433	
5	18.8	103.5	262	

Results plotted in Figures 5.15 and 5.16 show that head decline with time was linear and that infiltration rates decreased with each new test (round). Rates were faster in cell 1. These rates would have continued to decrease with time (to a rate similar to cell 3) had time allowed for more tests. Since a lower steady-state infiltration rate was not found, the lowest rate for each cell was used to calculate flow through times. It should be realized, however, that these times were on the high side.

For cell 1, with an unsaturated flow rate of 457 cm/d, the following travel times were determined:

- Between the furrow and lysimeter 325 (at 2.5 feet depth)
 4 hours,
- 2) between the furrow and a water table at 5 feet depth- 8 hours,
- 3) between the furrow and a water table at 12 feet depth- 19 hours.

For cell 2, with an unsaturated flow rate of 173 cm/d (Table 5.7), the following travel times were determined:

- between the furrow and lysimeter 415 (1.5 feet depth)
 6 hours,
- 2) between the furrow and a water table at 5 feet depth- 21 hours,
- 3) between the furrow and a water table at 12 feet depth51 hours

A technique based on work by Bouma (1975) was the second method used to determine unsaturated zone flow rates. Assuming a constant soil tension of 20 cm H_2O and using Figure 1 (Appendix G) of Bouma (1975), a flow rate of 1.5 cm/day was found. This value was considerably lower than

rates determined in the field, but it was representative of a heavily loaded area such as the ponded regions of this system. Use of this infiltration rate, therefore, provided low travel time estimates.

Using a 1.5 cm/d flow rate, the following travel times were determined:

- between the furrow and lysimeter 415 (1.5 feet depth)
 30 days
- 2) between the furrow and lysimeter 325 (2.5 feet depth) - 51 days,
- 3) between the furrow and a water table at 5 feet depth - 102 days, and
- 4) between the furrow and a water table at 12 feet depth
 244 days.

Based on these calculations, a realistic infiltration rate would have occurred between 1.5 and 457 cm/d and a realistic travel time to a 12 foot deep water table would have been between 2 and 244 days. Loaded areas at this site, however, most likely had unsaturated zone travel times (for 12 feet) on the order of hundreds of days. Installation of tensiometers and further infiltration station work would have better defined unsaturated flow rates at Mindoro.

Furrow and Lysimeter Chemistry

Mean and standard deviations of chemical parameters for each Mindoro furrow wastewater sampling point and lysimeter are listed in Table 5.8. Complete data listings are given in Appendix HH. When comparing chloride averages, one can see that all sampling points in Table 5.8 were contaminated by the ridge and furrow wastewater. Lysimeters 325 and 415 were the only lysimeters providing samples. The reasons for the

-171-

TABLE 5.8

MINDORO MEAN AND STANDARD DEVIATION OF FURROW WASTEWATER AND LYSIMETER CHEMICAL PARAMETERS (unit mg/l except pH)

LOCATION	DISS BOD ₅	TOTAL BOD5	DISS COD	DISS TKN	TOTAL TKN	DISS NH ₃ -N	DISS NO ₂ +NO ₃	C1-	TDS	Total pH
Lysimeter 415 (1.5' depth)	4 <u>+</u> 0.6* (3)		10 <u>+</u> 1.3 (4)	1.6 <u>+</u> 0.66 (5)		0.1 <u>+</u> 0.0# (5)	8.0 <u>+</u> 8.0 (5)	53 <u>+</u> 21 (8)	491 <u>+</u> 43.8 (2)	6.2 <u>+</u> 0.1((4)
Well 8 0	3.2 <u>+</u> 0.42 (7)		25 <u>+</u> 4.3 (7)	2.5 <u>+</u> 1.1 (7)		0.3 <u>+</u> 0.16 (8)	0.2 <u>+</u> 0.1 * (8)	84 <u>+</u> 8.2 (7)	744 <u>+</u> 201 (7)	6.8 <u>+</u> 0.2 (3)
Header-Cell 2		1400 <u>+</u> 0.0 (1)	3600 <u>+</u> 0.0 (1)	30 <u>+</u> 12 (2)	86 <u>+</u> 32 (2)	11 <u>+</u> 0.0 (1)	0.1 <u>+</u> 0.0 (2)	100 <u>+</u> 27 (2)		6.1 <u>+</u> 0.1# (2)
Furrow-Cell 2	220 <u>+</u> 0.0 (1)		320 <u>+</u> 0.0 (1)	16 <u>+</u> 0.0 (1)		16 <u>+</u> 0.0 (1)	1.0 <u>+</u> 0.0 * (1)	99 <u>+</u> 0.0 (1)	796 <u>+</u> 0.0 (1)	6.6 <u>+</u> 0.0 (1)
Lysimeter 325 (2.5' depth)	4 <u>+</u> 0.7* (2)		35 <u>+</u> 33 (5)	2.0 <u>+</u> 1.3 (5)		0.4 <u>+</u> 0.6 (5)	3•9 <u>+</u> 3•3 (5)	92 <u>+</u> 16 (4)	588 <u>+</u> 33.9 (2)	6.6 <u>+</u> 0.1'/ (4)
Well 50	11 <u>+</u> 7.71 (13)		31 <u>+</u> 9.5 (12)	4.0 <u>+</u> 0.74 (13)		3.0+1.0 (13)	0.1 <u>+</u> 0.0 * (13)	91 <u>+</u> 2.3 (13)	715 <u>+</u> 42.4 (13)	6.7 <u>+</u> 0.1# (8)
Header-Cell 3		250 <u>+</u> 0.0 (1)	1200 <u>+</u> 0.0 (1)	15 <u>+</u> 2.8 (2)	42 <u>+</u> 25 (2)	12 <u>+</u> 2.8 (2)	0.7 <u>+</u> 0.4" (2)	82 <u>+</u> 0.71 (2)		6.4 <u>+</u> 0.0 (1)
Furrow-Cell 3		94 <u>+</u> 0.0 (1)	160 <u>+</u> 0.0 (1)	15 <u>+</u> 1.4 (2)	20 <u>+</u> 3.5 (2)	14 <u>+</u> 2.1 (2)	0.6 <u>+</u> 0.6 * (2)	77 <u>+</u> 1.4 (2)		6.6 <u>+</u> 0.0 (1)
Well 90	16 <u>+</u> 12 * (7)		47 <u>+</u> 28 (7)	2.8 <u>+</u> 0.73 (7)		1.5 <u>+</u> 0,34 (7)	0.1 <u>+</u> 0.04 (7)	92 <u>+</u> 12 (7)	607 <u>+</u> 89.9 (7)	6.7 <u>+</u> 0.14 (7)

() indicates # of observations

-172-

April 1

they

@ Well values added for comparison of furrow & lysimeter concentrations to downstream well concentrations

* Mean contains values above or below a detection limit; limit used in average

inability of the other lysimeters to provide samples will be discussed later in this chapter. Large data variations existed in the furrow wastewater and lysimeters affected by cell loading, resulting in large standard deviations. Therefore, it was realized that these numbers do not fully describe the contamination and the averages were only used for relative comparisons.

To determie unsaturated zone COD and total nitrogen losses in comparison to chloride dilution, the flow at the Mindoro ridge and furrow system was broken into three paths. They were:

- 1) from the inlet, to lysimeter 415, to well 8;
- 2) from the inlet, to the cell 2 header ditch, to the cell 2 furrows, to lysimeter 325, to well 5; and
- 3) from the inlet, to the cell 2 header ditch, to the cell 3 header ditch, to the cell 3 furrows, to well 9.

Along the first flow path, chloride concentrations were diluted 16% from the wastewater (100 mg/l) to well 8. COD and total nitrogen averages declined 98 and 92 percent, respectively, along the same path. This implied that actual average COD and total nitrogen reductions were 82 and 76 percent, respectively. Reduction along this flow path are completely listed in Table 5.9.

From Table 5.8, one notices that the lysimeter 415 chloride average was relatively lower than other furrow and lysimeter averages and its stan-

-173-

furrows adjacent to lysimeter 415 did not receive fresh wastewater during the study. Chloride concentrations decreased in lysimeter 415 from 82 mg/l on 5/23/84 to 35 mg/l on 9/19/84.

Along the second flow path at Mindoro, chloride concentrations were diluted 9% from the wastewater to well 5. COD and total nitrogen averages declined 97 and 87 percent, respectively, along the same path. This meant that actual average COD and total nitrogen reductions were 88 and 78 percent, respectively. Reduction along this flow path are completely listed in Table 5.10.

Along the third flow path at Mindoro, chloride averages were diluted 8% from the wastewater to well 9. COD and total nitrogen averages decreased 96 and 91 percent, respectively, along the same path. This implied that actual average COD and total nitrogen reductions were 88 and 83 percent, respectively. Reductions along this flow path are completely listed in Table 5.11.

Nitrogen transformations observed in the applied wastewater at Mindoro were similar to trends observed at Brodhead. As at Brodhead, Mindoro's wastewater nitrogen was principally in organic form. While this waste traveled in the header ditch, the organic-N mineralized to NH_4^+ -N. In the cell 2 header ditch, the average TKN concentration contained approximately 40% NH_4^+ -N and in the cell 3 header ditch, the average TKN value contained 80% NH_4^+ -N. As mentioned in the previous discussion, N losses occurred alorg the header ditch. These were attributed to plant uptake of NH_4^+ since pH values were not conducive to NH_3 volatilization and little NO_3 -N was produced in the header ditch to promote denitrification.

-174-

TABLE 5.9	
-----------	--

LOCATION	C1-	COD	TOTAL N	% C1 REDUCTION FROM WASTE	<pre>\$ COD REDUCTION FROM WASTE</pre>	RELATIVE COD LOSS	% TOTAL REDUCTION FROM WASTE	RELATIVE N Loss
Wastewater	100	1200	32.4	0	0	0	0	0
Lysimeter 415	53	10	9.6	47	99	52	70	23
Well 8	84	25	2.7	16	98	82	92	76

REDUCTIONS ALONG FIRST MINDORO FLOW PATH

TABLE 5.10

REDUCTIONS ALONG SECOND MINDORO FLOW PATH

C1-	COD	TOTAL N	% C1 REDUCTION FROM WASTE	<pre>\$ COD REDUCTION FROM WASTE</pre>	RELATIVE COD LOSS	\$ TOTAL REDUCTION FROM WASTE	RELATIVE N Loss
100	1200	32.4	0	0	0	0	0
100	3600	30.1	0			7	7
99	320	17	1	73	72	47	46
92	35	5.9	8	97	89	82	, 74
91	31	4.1	9	97	88	87	78
	100 100 99 92	100 1200 100 3600 99 320 92 35	100120032.4100360030.1993201792355.9	C1- COD TOTAL N FROM WASTE 100 1200 32.4 0 100 3600 30.1 0 99 320 17 1 92 35 5.9 8	C1- COD TOTAL N FROM WASTE FROM WASTE 100 1200 32.4 0 0 100 3600 30.1 0 99 320 17 1 73 92 35 5.9 8 97	C1- COD TOTAL N FROM WASTE FROM WASTE LOSS 100 1200 32.4 0 0 0 100 3600 30.1 0 99 320 17 1 73 72 92 35 5.9 8 97 89	C1- COD TOTAL N FROM WASTE FROM WASTE LOSS FROM WASTE 100 1200 32.4 0 0 0 0 100 3600 30.1 0 7 99 320 17 1 73 72 47 92 35 5.9 8 97 89 82

- average concentrations used

-175-

- % Relative loss = % total reduction - % chloride dilution

TABLE 5.11

REDUCTIONS ALONG THIRD MINDORO FLOW PATH

	01-	000	momar N	S C1 REDUCTION	S COD REDUCTION	RELATIVE COD	\$ TOTAL REDUCTION	RELATIVE I
LOCATION	<u>C1-</u>	COD	TOTAL N	FROM WASTE	FROM WASTE	LOSS	FROM WASTE	LOSS
Wastewater	100	1200	32.4	0	0	0	0	0
Cell 2-Header	100	3600	30.1	0			7	7
Cell 3-Header	82	1200	15.7	18	0		52	34
Cell 3-Furrow	77	160	15.6	23	87	64	52	29
Well 9	92	47	2.9	8	96	88	91	83

- average concentrations used

-176-

- % Relative loss = % total reduction - % chloride dilution

.

Once wastewater entered the furrows from the header ditch at Mindoro, this ammonification to NH_4^+-N was completed. Furrow sample TKN's collected contained 85 to 100% ammonium. Nitrogen losses, as previously discussed, were also observed in furrow samples as compared to the applied wastewater. Since furrow pH values were low and no nitrate was produced, these N losses were most likely the result of plant uptake of NH_4^+-N . Nitrogen losses were higher in cell 2 furrows than in cell 3 furrows on average.

As the wastewater percolated from the furrows, through the unsaturated zone, to the groundwater, simultaneous nitrification and denitrification were found to occur at Mindoro. Both lysimeter 325 and 415 samples contained nitrate (see Table 5.8) and total N values were lower in these lysimeters than the overlying furrow samples. Samples from wells 5, 8, and 9 indicated a further decrease in total nitrogen averages, as well as nitrate losses, in the unsaturated zone. Nitrification occurred in the upper aerated section of the unsaturated zone. This section also contained anaerobic microenvironments which facilitated denitrification losses. Plant uptake of nitrate and ammonium could also have accounted for N losses in this upper layer. As flow continued to the water table, denitrification losses continued. Ammonium adsorption was neglected assuming that all exchange sites were saturated.

By comparing total and dissolved TKN results for wastewater, header, and furrow samples, the same nitrogen solids dissolution pattern that was observed at Brodhead was established at Mindoro. These data are presented in Table 5.12. As mentioned earlier, the Mindoro wastewater con-

-177-

tained nitrogen primarily in organic-N form. Twenty to 45% of this organic-N was tied up in wastewater solids. Total TKN concentrations in the header ditch were quite variable and were usually higher than wastewater values. In the cell 2 header, 40-80 percent of the TKN was tied up in solids. In the cell 3 header, 45-70 percent of the TKN was tied up in solids.

TABLE 5.12

MINDORO WASTEWATER, HEADER, AND FURROW WASTEWATER NITROGEN RESULTS

DATE	TOTAL TKN	DISS TKN	DISS NH3-N	DISS NO3-N+NO2-N
		WASTE	EWATER	
10/16/84	40	31	0.9	0.8
11/6/84	29	16	1.4	0.6
		HEADER -	- CELL 2	
10/16/84	108	21	11	0.1
11/6/84	63	38		0.1
		HEADER -	- CELL 3	
10/16/84	59	17	14	0.4
11/6/84	24	13	10	<1.0
		FURROW -	- CELL 3	
10/16/84	17	14	12	0.1
11/6/84	22	16	15	

-178-

Due to the poor distribution of wastewater at Mindoro, suspended solids settled in the header ditch. Ammonium concentration in the header ditch were the result of ammonification of these settled solids and diffusion of NH_4^+ -N into the overlying water column. This was the same event that was observed at Brodhead. Due to plugging of furrow inlets, flow entering the furrows was analogous to the overflow of wastewater over the weir of a clarifier. Furrow samples were low in solids and only 20-30 percent of the total TKN was tied up in solids. Also, about 70% of the TKN was in ammonium form.

A final comment on furrow and lysimeter chemistry at Mindoro concerned pH. Even though the mean wastewater pH was 7.5, header, furrow, and lysimeter pH values were consistently between 6.1 and 6.6. Since negligible nitrite was found in header and furrow samples, this pH lowering was not the result of nitrification. Tusneem and Patrick (1971) attributed a similar pH reduction to volatile acid production. In aerobic and anaerobic degradation of organic matter, organic acids are the final product of glycolysis. In anaerobic environments (eg. header and furrow water), terminal oxidation is suppressed and organic acids accumulate. This process was most likely the reason for lower pH in downstream wells at Mindoro and Brodhead as well.

Grass Uptake of Nitrogen

Six grass samples were cut at the Mindoro ridge and furrow system during the 1984 growing season. Two samples were collected in April before growth began, two were taken in July at peak growth, and two were taken

-179-

in November during declining growth. Average results of these analyses are shown in Table 5.13. Calculations are presented in Appendix II. These results were used for the following purposes: 1) to determine the plant nitrogen uptake during the growing season; 2) to calculate the effect of grass burning on nitrogen losses, and 3) to estimate a plant nitrogen uptake value for a nitrogen budget.

TABLE 5.13

MINDORO GRASS NITROGEN RESULTS

Date	lb of Dry Grass on Site (1b)	% N Dry Wt Basis	% Ash After Burning	% N of Ash	lb N On Site Before Burning	lb N On Site After Burning	% N Lost Due to Burning
4/26/84	22,255	1.56	17.1	2.49	347	95	72.6
7/12/84	29,962	2.15	8.5	0.64	644	16.3	97.5
11/6/84	44,439	1.0	8.9	0.31	444	12.2	97.2

- weights are for total site area (3 acre) extropolated from sample area results

The mass of nitrogen on site calculations indicate that plant uptake of nitrogen was high in late spring and early summer. Plant nitrogen content decreased in the fall as plants died and nitrogen moved to the root zone. At the end of the growing season, 444 lb nitrogen was contained on site in the standing cover crop.

The operator of the Mindoro ridge and furrow system burns site grasses in the spring to eliminate dead grass accumulation and to stimulate new growth. Collected grass samples were burned and analyzed for percent ash and percent N in ash to determine the effect of grass burning.

-180-

Total pounds of ash-nitrogen on site were highest in April at 95 lbs as shown in Table 5.13. The July ash sample indicated 16.3 lbs of nitrogen on site after burning and the November sample indicated 12.2 lbs. A range of 72-98% nitrogen loss by burning was observed when comparing total nitrogen in the site grasses before and after burning. The lowest loss occurred in April.

Since site grasses were burned in early spring, the 95 lbs of nitrogen on site after burning was immobilized into the soil. Since this mass did not leave the material balance, it was not a loss or an addition to the budget. Grass nitrogen losses during the growing season studied (4/26/84-11/6/84) were those lost from burning in April or 252 lb. This mass was used in the nitrogen budget estimate.

TABLE 5.14

Location	Total BOD ₅	Total COD	C1-	TKN	NH3-N	NO3-N	Field pH
Upstream	3 <u>+</u> 0.0*	6 <u>+</u> 0	7.4 <u>+</u> 0.71	0.2 <u>+</u> 0.0	0.1 <u>+</u> 0.0	0.8 <u>+</u> 0.07	8.0 <u>+</u> 0.0
	(2)	(2)	(2)	(2)	(2)	(2)	(1)
Midstream	3 <u>+</u> 0.0*	5 <u>+</u> 0*	7.8 <u>+</u> 1.2	0.2 <u>+</u> 0.0	0.1 <u>+</u> 0.0	0.8 <u>+</u> 0.0	8.0 <u>+</u> 0.0
	(2)	(2)	(2)	(2)	(2)	(2)	(1)
Downstream	3 <u>+</u> 0.0*	7 <u>+</u> 0.0	7.6 <u>+</u> 0.56	0.4 <u>+</u> 0.07	0.1 <u>+</u> 0.0	0.8 <u>+</u> 0.0	8.0 <u>+</u> 0.0
	(2)	(2)	(2)	(2)	(2)	(2)	(1)

MINDORO: SEVERSON COULEE CREEK QUALITY

- all values mg/l except pH

- () is # of observations

mean includes data below detection limit; limit used in average

Severson Coulee Creek Chemistry

Mean and standard deviations of upstream, midstream, and downstream Severson Creek samples are tabulated in Table 5.14. Complete data are located in Appendix JJ. All values were low and were uniform upstream to downstream. The Mindoro ridge and furrow system did not adversely affect this creek.

Site Observations

During the course of this study, the following site observations were made at the Mindoro ridge and furrow: 1) wastewater distribution and solids build-up in furrows, 2) cover crop, 3) winter operation, 4) monitoring equipment performance, and 5) slope stability.

At Mindoro, only 40% of the furrows received wastewater and only 30% of the total area was loaded during the project. Dead grass accumulation has plugged furrows in the northern half of cell 1, the southern 75% of cell 2, and the southern half of cell 3. The header ditch design was not efficient as well. Since the inlet was located in cell 1, the southern half of this cell continually received wastewater. The stop gates between cells were in need of attention. Wastewater flow either went through or underneath them, causing cells 2 and 3 to also receive *z*. continuous load. An improvement of wastewater distribution at this site could considerably improve treatment by spreading the load over more area and providing more soil aeration. Since many of the furrow openings were plugged, most of the wastewater suspended solids settled in the header ditch. There was no noticeable solids accumulation in the furrows. The cover crop of canary grass at Mindoro flourished with the additional water and nutrients provided by the wastewater. During early spring, before site burning, grasses were brown and knocked down to about knee high. Thick grasses grew head high during spring and early summer in used portions of the system. In August and September, grasses were browning and knocked down by wind. Grass was completely brown and knocked down to about hip high in November. Grasses in unused portions of the system were not as thick and only knee high. Since furrows were only two feet wide, these tall grasses provided furrow protection during the winter months.

Winter operation at the Mindoro ridge and furrow was not a major problem since wastewater effluent temperatures were warm. The southern quarter of cell 1 and northern quarter of cell 3 received wastewater during subzero weather. During above-zero whether, however, most of the area normally treating wastewater was operable. Once system loading began, the ice would melt and infiltration would continue. Header ice conditions ranged from no ice near the inlet to six inches at the far north end. Furrow ice conditions ranged from no ice near the header, to one inch thick ice with wastewater or air underneath, to completely (two inches thick) frozen.

The monitoring equipment used at this site was quite adequate. The only problems which occurred concerned the Teflon lysimeters. To obtain a sample of sufficient volume (greater than 50 ml) for chemical analysis, a 20-inch (mercury) vacuum was applied to a lysimeter and a two to 24 hour vacuum period was used. This technique was successful for lysime-

-183-

ters 325 and 415 only. The background lysimeters (21, 25, 211) lost a vacuum within 15 minutes. This was most likely the result of easy air entry from the surface or low soil moisture. A high, continuous vacuum or a bentonite seal may have been necessary to obtain a sample. Lysimeters 36 and 310, located in cell 2, did not provide samples but did maintain a 20 inch (mercury) vacuum for more than a month. These units were installed in soils with a higher clay content. A high soil moisture tension, caused by low moisture content, was most likely the reason for the failure of these lysimeters to obtain samples.

Winter conditions also caused trouble with lysimeter operation. During sub-freezing weather, ice droplets would form in the Tygon tubing and the tubing itself would contract. The pressure provided by the hand pump could not overcome these blockages. Lysimeters were not used from December (1983) to March (1984) at this site.

A final site observation concerned slope stability along the southern half of the west system boundary. Site topography from well 6 to the creek drops at least 20 feet. Obvious soil creep was noticed near well 6. This slope movement should be monitored in the future so the integrity of the cell 1 berm can be maintained.

Mindoro Nitrogen Budget

A nitrogen balance was estimated for the unsaturated zone at the Mindoro ridge and furrow system. Wastewater flow readings, wastewater nitrogen data, plant uptake results, and well 5 nitrogen data-were used in this estimate. The balance was on an annual total pounds basis.

-184-

Additions to the budget came from applied wastewater nitrogen. Total TKN plus NO_3 -N averages were used assuming all N eventually entered the soil. As with the Brodhead budget, additions to the system by nitrogen fixation, rainfall, and organic soil debris mineralization were assumed negligible. Nitrogen losses in the balance were be plant burning and leaching. The difference between these additions and losses was accounted for by denitrification. It was further assumed that precipitation and evapotranspiration were approximately equal (ie. all vertical flow was from wastewater), no volatilization occurred (pH less than 9), and all soil NH₄+-N adsorption sites were saturated (ie. no soil storage). Table 5.15 tabulates the results of this budget; calculations are presented in Appendix KK.

TABLE 5.15

NITROGEN BUDGET ESTIMATE - MINDORO SITE

Addition/Loss	lb/yr	% of Applied N
Wastewater Applied N	1381	100
Plant Uptake Loss	252	18
Leaching	175	13
Denitrification	954	69

With this procedure, a denitrification loss of 69% of applied nitrogen was calculated for the Mindoro site. This figure was similar to flow path losses discussed previously in this chapter. Leaching accounted for 13% of applied N loss and plant uptake accounted for 18%.

CHAPTER 6: COMPARATIVE DISCUSSION OF THE BRODHEAD AND MINDORO RIDGE AND FURROW SITES

Brodhead and Mindoro ridge and furrow system nitrogen budgets, wastewater organic loadings, nitrogen transformations, and operation and maintenance were compared in order to discuss the treatment and performance of ridge and furrow systems.

Nitrogen Budget

On a total pounds/year basis, the Brodhead ridge and furrow received 5375 lb N/yr while Mindoro treated 1381 lb N/yr, almost four times less. If actual area loading rates are used, however, the nitrogen applied at Brodhead and Mindoro was 1144 lb N/acre/year (3.1 lb N/acre/day) and 1534 lb N/acre/year (4.2 lb N/acre/day), respectively. With improved distribution at Mindoro, the nitrogen loading rate could have been 460 lb N/acre/year (1.3 lb N/acre/day).

The nitrogen budget estimates for the Brodhead ridge and furrow sites showed that denitrification was the major sink for applied wastewater nitrogen. Denitrification accounted for 66% of wastewater-N loss at Brodhead and 69% of the applied-N loss at Mindoro. These percentages matched those cited earlier in the literature (Patrick and Goth, 1974; Tusneem and Patrick, 1971; Reddy and Graetz, 1981; Chen and Patrick, 1981; Olson et. al., 1980; and Leach and Enfield, 1983). Respective denitrification rates were 751 lb/acre/year at Brodhead and 1060 lb/acre/year at Mindoro using actual loaded areas. If the total Mindoro site area was used, a denitrification rate of 318 lb/acre/day resulted. As stated earlier, improved wastewater distribution may have increased

-186-

denitrification loss by providing soil aeration and lowering the groundwater mound. Lowering the mound would increase the unsaturated zone travel time.

Plant uptake losses accounted for 0% of applied wastewater nitrogen at Brodhead and 18% (or 280 lb N/acre/year) at Mindoro. Even though a crop nitrogen uptake of 74 lb N/acre/year was observed at Brodhead, grass nitrogen losses were zero since the crop was not burned in the spring. Whatever nitrogen that was taken up during the growing season was returned again by the dead plants resulting in a zero net loss. Spring burning of the dead grass at Mindoro removed 73% of the nitrogen contained in the crop. A nitrogen uptake of 493 lb N/acre/year, using the actual loaded area was found at Mindoro for the 1984 growing season. Uptake using the total site area was 148 lb N/acre/year. Higher crop uptakes at Mindoro were attributed to the silt loam soil. The soil's lower infiltrative capacity improved the availability of nutrients and water to the plants. Burning may also have enhanced nutrient uptake.

The remaining loss of applied wastewater nitrogen was through leaching from the unsaturated zone to the groundwater. Leaching losses accounted for 34% (392 lb N/acre/year) of applied nitrogen at Brodhead and 13% (193 lb N/acre/year) of added nitrogen at Mindoro. A rate of 58 lb N/acre/year was calculated if total site area was used for Mindoro. Higher leaching losses at Brodhead resulted from the higher travel times through the sandy soils. These faster travel times meant a shorter contact time between the percolating wastewater and the denitrifying soil bacteria.

-187-

Finally, it should be remembered that inputs of nitrogen by fixation and precipitation, losses by volatilization, and soil absorption were assumed negligible. These assumptions seemed practical and were discussed previously.

Wastewater Organic Loading Rates

As stated in the last section, Brodhead received 1144 lb N/acre/year (3.1 lb N/acre/day) while Mindoro treated 1534 lb N/acre/year (4.2 lb N/acre/day). Currently, the Wisconsin DNR does not have a nitrogen loading limit. Using the nitrogen budget results, a suggested total nitrogen loading rate was suggested for each site in order to meet a 5 mg/l total nitrogen increase above background quality in the groundwater beneath the site. The choice of 5 mg/l was arbitrary; the intent was to present two procedures in which to calculate a loading rate. Proposed loading rates were determined using both the percentage of denitrification loss and the rate of denitrification loss. It was assumed that denitrification and leaching rates would remain constant under changes in the nitrogen loading rate.

At Brodhead, 34% of the applied wastewater nitrogen leached to the groundwater and 66% of the added nitrogen was lost through denitrification. At the current loading rate, the concentrations in lysimeters 3 and 6 were 9.0 and 21.7 mg/l total nitrogen on average. These lysimeters were considered to represent unsaturated pore water upon entering the groundwater. In order to set a necessary wastewater concentration to meet a 5 mg/l increase, this value was divided by the leaching percentage, or 0.34. This was based on an earlier assumption that waste-

-188-

water flow volume and leaching flow volume were essentially equal. With this procedure, a wastewater concentration of 15 mg/l total nitrogen would be needed to meet a 5 mg/l groundwater concentration increase. Using the average wastewater flow of 39,500 gpd, this would result in a nitrogen loading rate of 384 lb N/acre/year (1.05 lb N/acre/day). This rate is about a third of the present rate. In order to meet this rate, the Brodhead ridge and furrow system would have to be expanded or wastewater pretreatment before application to the furrows would be needed.

The denitrification loss rate of 751 lb N/acre/year at Brodhead was also used to suggest a nitrogen loading rate at this site. Assuming that the average wastewater flow rate of 39,500 gpd equals the leaching flow rate, a mitrogen leaching rate of 128 lb N/acre/year would increase the lysimeter total nitrogen concentration 5 mg/l above background. The summation of these two rates (751 + 128) would account for a new suggested wastewater loading rate of 879 lb N/acre/year. This corresponds to an average wastewater total nitrogen concentration of 34 mg/l which is 75% of the current nitrogen concentration.

At Mindoro, 13% of the applied wastewater nitrogen leached to the groundwater and 69% of the added nitrogen was lost through denitrification. At the current loading rate, the highest average total nitrogen concentration in the groundwater beneath the system was 4.1 mg/l. The current rate of 1534 lb N/acre/year (or 4.2 lb N/acre/day) was adequate to meet a 5 mg/l total nitrogen groundwater concentration increase above

-189-

background. With improved wastewater distribution and proper load/rest cycling, the Mindoro site could most likely treat a higher nitrogen loading rate if necessary.

It should be remembered that these suggested rates do not apply to all ridge and furrow systems but only those with similar hydrogeological characteristics to Brodhead or Mindoro, respectively.

Using the actual site area loaded during the study, both Brodhead and Mindoro had average BOD₅ loading rates over the 100 lb/acre/day DNR limit. Brodhead's average rate was 125 lb/acre/day while Mindoro's rate was 108 lb/acre/day. Site expansion, pretreatment, and wastewater distribution improvement are all possible methods to meet this standard.

Based on project results, however, the COD concentrations (which follow BOD5 values) were greatly reduced as wastewater infiltrates from the furrows, through the unsaturated zone, and down gradient in the groundwater. Tables 6.1 and 6.2 summarize COD reductions at Brodhead and Mindoro, respectively. Both ridge and furrow systems had high COD reductions (96-99%) at the sampling locations indicated but these site decreases were achieved differently. Brodhead reductions were aided by dilution, as indicated by the chloride data (16-80% chloride dilution). Cell 1 unsaturated zone actual COD reductions, as indicated in lysimeters 2 and 3, were relatively high at 72-82%. With contined use of the 2 week/1 week load/rest cycle, the groundwater most likely will reflect these losses as well. Mindoro declines were mainly biological, with 52-89% actual COD losses. Since nitrification was observed at both

-190-

TABLE 6.1

SUMMARY OF RELATIVE COD

REDUCTIONS AT BRODHEAD

LOCATION	AVE# Cl-	AVE# COD	% C1- REDUCTION	% COD REDUCTION	% ACTUAL COD LOSS
Wastewater	930	2390	0	0	0
Lysimeter 2	690	51	26	98	72
Lysimeter 3	780	56	16	98	82
Lysimeter 5	400	32	57	99	42
Lysimeter 6	390	20	58	99	· 41
Well 15	570	57	39	98	59
Well 10A	380	17	59	99	40
Well 10B	630	24	32	99	67
Well 14	190	18	80	99	19

units: mg/l

TABLE 6.2

SUMMARY OF RELATIVE COD

REDUCTIONS AT MINDORO

LOCATION	AVE#	AVE# COD	% C1- REDUCTION	% COD REDUCTION	S ACTUAL COD LOSS
Wastewater	100	1200	0	0	0
Lysimeter 325	92	35	8	97	89
Lysimeter 415	53	10	47	99	52
Well 5	91	31	9	97	88
Well 9	92	47	8	96	88

units: mg/l

sites, it was determined that applied BOD5 (or COD) did not impart a high oxygen demand on the soil system which would have inhibited this transformation.

The present BOD₅ loadings at both sites are, therefore, not overtaxing the treatment capacity of the soil. COD reductions of 96-99% occurred at both sites and nitrogen reductions were not impaired by the applied oxygen demand. Load/rest cycle most likely aided in controlling O_2 demand, especially at Brodhead.

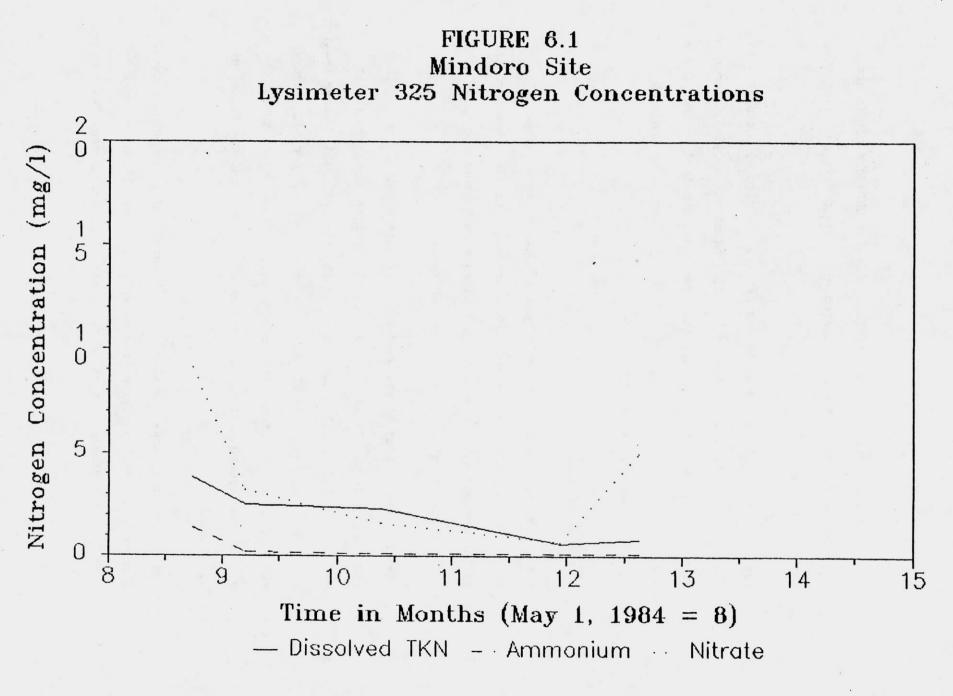
NITROGEN TRANSFORMATIONS

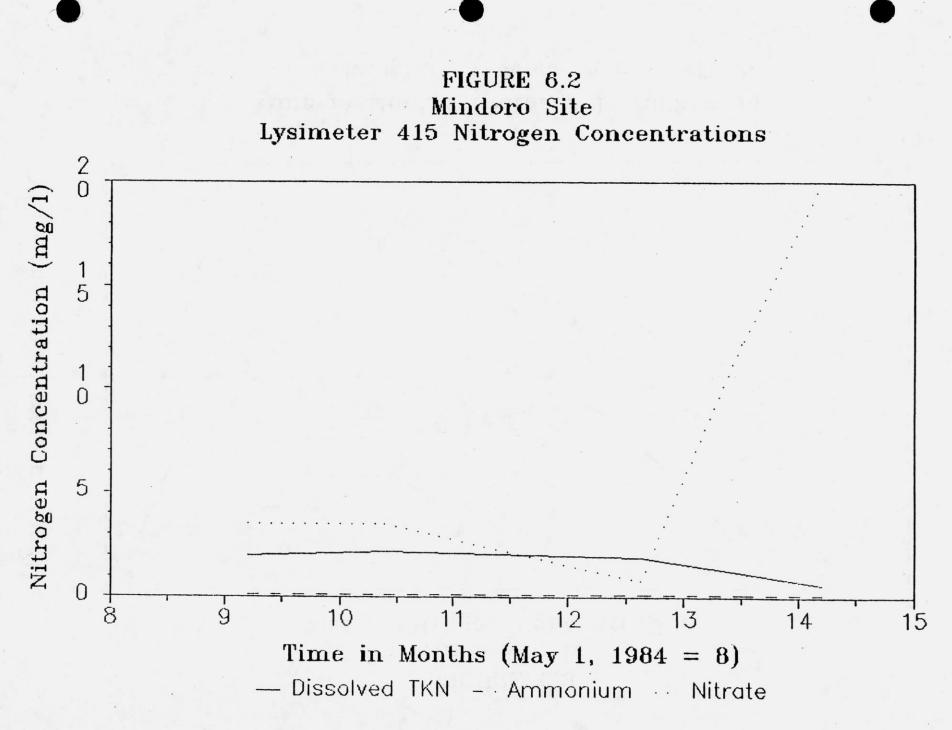
A similar nitrogen transformation pattern was observed at both the Brodhead and Mindoro ridge and furrow systems. Little ammonia rolatilization occured since wastewater pH values were less than nine. Both site wastewater samples had average pH values of 7.5.

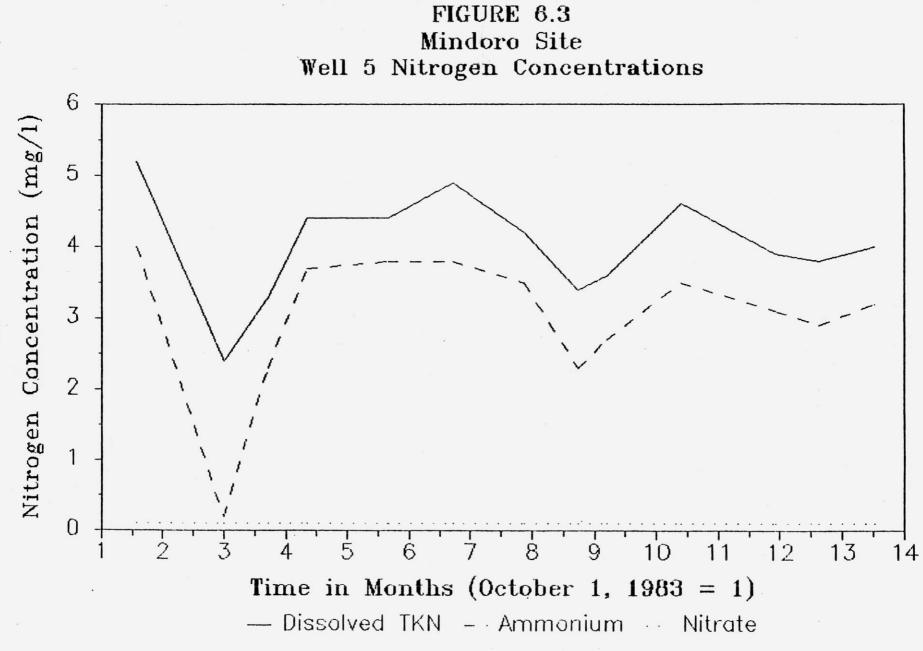
The nitrogen content of the wastewater at both sites was mainly in the organic form. On average, the total nitrogen concentration at Brodhead contained 88% organic-N while the Mindoro total N concentration contained 96% organic-N. Twenty-five to 40% of this organic nitrogen was associated with the solid form as well. These solids subsequently settled and accumulated in the furrows at Brodhead and the header ditch at Mindoro.

The organic-N in these settled solids mineralized to ammonium and this diffused into the overlying furrow water column. Dissolved organic nitrogen also mineralized to NH_4^+ -N in the furrows. Furrow TKN samples

-192-







at both sites contained 54 to 100% ammonium. The literature in Chapter 2 stated that mineralization was favored at C:N ratios below 20 to 30:1. For wastewater, this ratio was represented by the ratio BOD₅:TKN (EPA, 1975). The BOD₅:TKN ratios in this project were 42:1 and 26:1 for the Brodhead and Mindoro wastewater, respectively. The Brodhead C:N ratio was high but furrow samples had BOD₅:TKN ratios ranging from 21 to 34:1, which would promote mineralization. Furrow wastewater pH values were not condusive to ammonia volatilization.

As ammonium infiltrated the unsaturated zone, it was aerated to nitrate at both sites. This transformation occured between lysimeters 1 and 2 (or 1 and 3 feet depth) underneath cell 1 at Brodhead, between the furrow and the water table (or 0 and 5.5 feet depth) underneath cell 2 at Brodhead, and the upper few feet of the unsaturated zone at Mindoro. Tables 4.7 and 5.8 and Figures 4.26, 4.27, 4.31, 4.32, 4.33, 6.1, 6.2, and 4.12 illustrate this change.

This nitrate denitrified to nitrogen gas (NO, N₂O, or N₂) as the percolate ecountered anaerobic microzones in the unsaturated zone at both ridge and furrow sites. This transformation was verified by observing nitrate decreases between lysimeters 2 and 3 (between 3 and 4.8 feet depth) beneath cell 1 at Brodhead, between lysimeter 6 and well 15 (between 3.6 and 5.5 feet depth) in November (1984) at Brodhead, and between the shallow lysimeters (L325 and L415) and the water table at Mindoro. These changes were illustrated in Tables 4.7 and 5.8 and Figures 4.27, 4.28, 4.33, 4.12, 6.1, 6.2, and 6.3. As mentioned in

-196-

Chapter 4, well 17 nitrogen concentrations did not reflect recent well 1 nitrification or denitrification due to the slow groundwater velocities. Ammonium concentrations averaged 30 mg/l in this well.

It should finally be noted that the nitrogen transformations were quite dynamic at Brodhead. Nitrogen species in certain lysimeters or wells did not remain consistantly nitrate or consistantly ammonium during the project. Data from lysimeter 2, lysimeter 3, and well 15 were good examples (Figures 4.27, 4.28, 4.12). During the late summer (1984), nitrogen values switched from being mainly ammonium to being mainly nitrate in these lysimeters. In well 15 ammonium concentrations increased dramatically in late summer (1984). This was followed by a dramatic rise in nitrate values, with concurrent ammonium decreases in October (1984). Nitrate decreases in November (1984) subsequently followed. Such radical changes were not observed at Mindoro.

The dynamic nature of nitrogen transformations was most likely the result of the load/rest cycle. Nitrate production was encouraged by soil aeration during resting and inhibited by cell overloading. These transformations or increases/decreases could not be especially related to the load/rest cycle when comparing time concentration plots to the loading schedule, however.

Ridge and Furrow Operation and Maintenance

Operation and maintenance is an important yet often neglected aspect of a ridge and furrow treatment system. Wastewater distribution and infiltration, load/rest cycling, winter operation, and annual cover crop burning are all aspects of operation and maintenance which influence wastewater treatment and disposal.

-197-

Wastewater distribution at the Brodhead site was complete, meaning that all cell furrows received waste during a loading cycle. This allowed for even distribution of liquid and centaminants over the entire site area which improved the hydraulic (infiltrative) capacity and contaminant reductions (eg. nitrogen reduction) of the site. Due to grass overgrowth and leaky header gates, the wastewater distribution at Mindoro was poor. Even though cell loading changes were made, the same area (0.9 acres) was loaded throughout the study. As a result, ponding of wastewater occurred in the far north and south ends of the system with other regions receiving little or no wastewater. Overloading of waste in the ponded areas reduced the soils infiltrative capacity and could also have decreased the soil's ability to reduce contaminant concentrations.

Load/rest cycling of cells at a ridge and furrow site is also an important part of system operation. It alternately aerates and deaerates the soil which stimulates nitrogen reductions by nitrification and denitrification. After strictly loading cell 1 before 1980, the Brodhead system had utilized a consistant two week load/rest pattern. This was modified during the project to determine the effect of a shorter cycle (one week beginning October 1984) on wastewater treatment in the unsaturated zone. The impact of a shorter cycle was favorable beneath cell 2 as well 15 total nitrogen concentrations decreased from almost 80 mg/l to less than 10 mg/l during October (1984). Future sampling would be needed to determine if low levels would continue or if well 15 concentrations were inherintly variable.

-198-

After November 20, 1984, the Brodhead system has been on a cycle which loads cell 1 for one week and rests it for two. Cell 2 has been operated in the opposite mode. This was done to improve the soil aeration beneath cell 1 and also improve the cell's infiltrative capacities. As mentioned in Chapter 4, the cell 1 infiltration rates were slower than those in cell 2. Improved soil aeration may cause groundwater around well 17 to nitrify as those around well 15.

The Mindoro ridge and furrow has been on a one month loading, two month resting schedule for each of its three cells. The site operator has not followed this pattern strictly, however. This, combined with the slower unsaturated flow rates and poor wastewater distribution, has resulted in the observed ponding. Following the 2/1 schedule more closely would result in better soil aeration and possibly improved contaminant reduction.

Winter operation was not a problem at the ridge and furrow sites studied. Both had wastewater effluent temperatures warm enough to melt existing ice and snow and to allow percolation into the soil. Distribution of wastewater remained excellent at Brodhead and fair at Mindoro during subzero temperatures.

Annual grass burning at ridge and furrow sites is also an important maintenance procedure at some systems and could be at other sites. Besides volatilizing cover crop nitrogen, burning clears the furrows of dead grass and enhances new spring crop growth. An 18% nitrogen loss occurred during burning at Mindoro and a possible 3% nitrogen loss could have occurred at Brodhead had the operator burned at that site. Spring burning of grass is recommended for all sites where it is feasible.

-199-

System Performance

When comparing downgradient contaminant concentrations at both sites, the Mindoro ridge and furrow system appeared to impact groundwater quality less than the Brodhead system. Downstream wells at Brodhead contained total nitrogen concentrations greater then 10 mg/l on average. This nitrogen was also in NH_4^+ -N form. Nitrogen concentrations averages in downstream wells at Mindoro were consistantly below 5 mg/l. These downgradient values do not provide the complete picture however. Brodhead concentrations in wells 10A, 10B, and 14 were indicative of past use. Loading changes made during the course of this project would not be reflected in these wells until the spring of 1987 due to the travel times involved.

During this study, nitrogen and COD reductions in the unsaturated zone, as demonstrated by nitrogen budgets and data comparisons (Tables 6.1 and 6.2), were similar at Brodhead and Mindoro. The percentage of applied wastewater nitrogen leached was the major difference in the unsaturated zone treatment provided at these sites. At Brodhead, 34% of applied -N was leached; at Mindoro, 13% of applied -N was leached. This difference was attributed to the different soil types at the respective sites. At Brodhead, the sandy soil allowed for faster unsaturated travel times. This limited the available nitrification and denitrification time which allowed ammonium and nitrate to reach the groundwater only partially treated. Uniform wastewater distribution and proper load/rest operation were critical to treatment at Brodhead to keep the soil alternately aerated and deaerated. Considering the infiltrative capacities

-200-

of cell 1 and cell 2, the 2 week cell 2 loading followed by a 1 week cell 1 loading would optimize nitrogen and COD reduction at Brodhead. Higher nitrate concentrations in lysimeters 5 and 6 and well 15 could be reduced if cell 2 received a longer loading; high ammonium values in well 17 could be nitrified if cell 1 underwent a longer resting period.

The Mindoro site provided good treatment of wastewater nitrogen even though flow distribution and load/rest operation were poor. The silty loam soil, with its slower unsaturated and saturated zone travel times, overcame these operational problems. Better treatment could have been obtained with improved distribution and a strict load/rest cycle.

In general, nitrogen reductions at ridge and furrow sites are dependent on the soils infiltrative capacity, the wastewater distribution efficiency, and the load/rest cycle.

-201-

CHAPTER 7: CONCLUSIONS

The following were conclusions of the ridge and furrow study at Brodhead and Mindoro:

- Sixty-six to 69% nitrogen losses were observed in nitrogen budgets around the unsaturated zone at both sites. These were attributed to denitrification. Leaching accounted for the fate of 34% and 13% of applied nitrogen at Brodhead and Mindoro, respectively. Plant uptake losses accounted for 0% and 18% at Brodhead and Mindoro, respectively. These budgets are summarized in Table 7.1.
- 2) Since nitrification occurred in the unsaturated zone at each site, it was concluded that the BOD₅ wastewater loading did not place an oxygen demand high enough to inhibit this transformation.
- 3) Actual COD (after dilution) reduction in the unsaturated zone ranged from 41-82% at Brodhead and 52-89% at Mindoro.
- 4) Nitrogen and COD reductions were dependent on infiltrative capacity, which affected travel times, wastewater distribution efficiency, and the load/rest cycle. The distribution and load/rest cycling were more critical at Brodhead, where unsaturated zone flow times were fast (three days).
- 5) The wastewater nitrogen applied at both sites was mainly in organic-N form. Much of this organic-N was contained in the suspended solids of the wastewater and settled in furrows upon loading. This organic-N then ammonified to dissolved ammonium and diffused into the overlying furrow wastewater.

-202-

TABLE 7.1

SUMMARY OF RIDGE AND FURROW

NITROGEN BUDGETS

	BR	BRODHEAD		MINDORO		
•	lb/yr	lb/acre/yr*	¥.	lb/yr	lb/acre/yr*	z
Total N Loading	5375	1144	100	1381	1534	100
Losses Due to						
Denitrification	3529	751	66	954	1060	69
Leaching	1845	392	34	175	193	13
Plant Uptake	0	0	0	252	280	18

#Areas: Brodhead - 4.7 acres

Mindoro - 0.9 acres

- 6) Dissolved ammonium was the primary form of nitrogen in the furrow wastewater. As this percolated into the unsaturated zone, nitrification occurred within 3 feet (cell 1) and 3.6 feet (cell 2) at Brodhead and 1.5 feet at Mindoro. Denitrification of this nitrate was observed below 3 feet depth (cell 1) and 3.6 feet depth (cell 2) at Brodhead and through the entire unsaturated profile at Mindoro.
- 7) In order to meet a 5 mg/l total nitrogen arbitrary increase above background quality in the groundwater immediately beneath the ridge and furrow systems, it was determined that Brodhead's wastewater nitrogen loading rate be reduced to 384 lb N/acre/year (1.05 lb N/acre/day) on a percent basis and 879 lb N/acre/year (2.4 lb N/acre/day) on a denitrification rate basis. Mindoro's present 1534 lb N/acre/year (4.3 lb N/acre/day) was adequate.
- 8) No surface water contamination resulted from operation of these ridge and furrow systems.
- 9) Winter operation at ridge and furrow systems is possible if effluent temperatures are warm.
- 10) Spring burning of grass removed a significant portion of applied nitrogen at Mindoro (18%) and also stimulated new growth.

-204-

CHAPTER 8: RECOMMENDATIONS

The following recommendations were based on results of the ridge and furrow study at Brodhead and Mindoro:

1) Continued research at Brodhead should be done to:

- a) better determine whether loading changes impacted the groundwater quality or whether the dynamic nature of concentrations occur every year by sampling wells 10A, 10B, 14, and 15 and lysimeters 1, 2, 3, 5, and 6 twice a year,
- b) better quantify unsaturated flow times by installing tensionmeters (or some other appropriate method), and
- c) evaluate the sinking contaminant plume by installing a deep well in the location of well 14, which would be sampled twice annually.
- 2) The factories should meet the suggested nitrogen loading rates, on average, to reduce or maintain groundwater nitrogen concentrations.
- 3) Wastewater pretreatment at Brodhead would reduce solids accumulation in the furrows and lower the amount of applied nitrogen. Changes inside the plant could also be made to reduce the amount of lost product.
- 4) The chloride concentrations in the Brodhead wastewater should be reduced by brine removal in the plant to reduce groundwater chloride values.

-205-

- 5) Annual spring grass burning is suggested for all ridge and furrow systems where it is feasible.
- 6) An improved lysimetry method is necessary to obtain a more "instantaneous" sample and to allow for winter sampling.

.

7) A downstream well nest should be installed at Mindoro to better define the movement of contaminants off-site.



Appendix A1: List of References

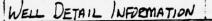
- 1. Alexander, Martin, <u>Introduction to Soil Microbiology</u>, 2nd ed. New York: John Wiley and Sons, 1977
- 2. Bohn, Hinrick L., Brian L. McNeal, and George A. O'Connor. Soil Chemistry, New York: John Wiley and Son, 1979
- 3. Bouma, Johannes, "Unsaturated Flow During Soil Treatment of Septic Tank Effluent", J. Environ Eng. Div. ASCE, 101, 967 (1975)
- 4. Chen, R. L. and W. H. Patrick, Jr., "Efficiency of Nitrogen Removal in a Simulated Overland Flow Waste Water Treatment System", <u>J. Environ, Qual.</u>, 10, 98 (1981)
- 5. Cooper, Hilton H., John D. Bredehoeft, and Istavros S. Popadopulos, "Response of a Finite-Diameter Well to an Instantaneous Charge of Water", <u>Water Resour. Res.</u>, 3(1), 263 (1967)
- 6. Cooper, Hilton H., et.al., "On the Analysis of 'Slug Test' Data", Water Resour. Res., 9(4), 1078 (1973)
- 7. EPA, Process Design Manual for Nitrogen Control (1975)
- 8. Freeze, R. Allan and John A. Cherry, <u>Groundwater</u>, Englewood Cliffs, New Jersey: Prentice-Hall, Inc., 1979
- 9. Freney, J. R., J. R. Simpson, and O. T. Denmead, "Ammonia Volatilization", <u>Terrestrial Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- 10. Kardos, L. T., et. al. "Sewage Effluent Renovated through Application to Farm and Forest Land," <u>Science and the Farmer XII</u>, <u>4</u>, 4 (1965)
- 11. Khanna, P. K., "Leaching of Nitrogen from Terrestrial Ecosystems -Patterns, Mechanisms and Ecosystem Responses', <u>Terrestrial Nitrogen</u> <u>Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- 12. King, Larry D., "Land Application of Untreated Industrial Waste Water", <u>J. Environ. Qual.</u>, <u>11(4)</u>, 638 (1982)

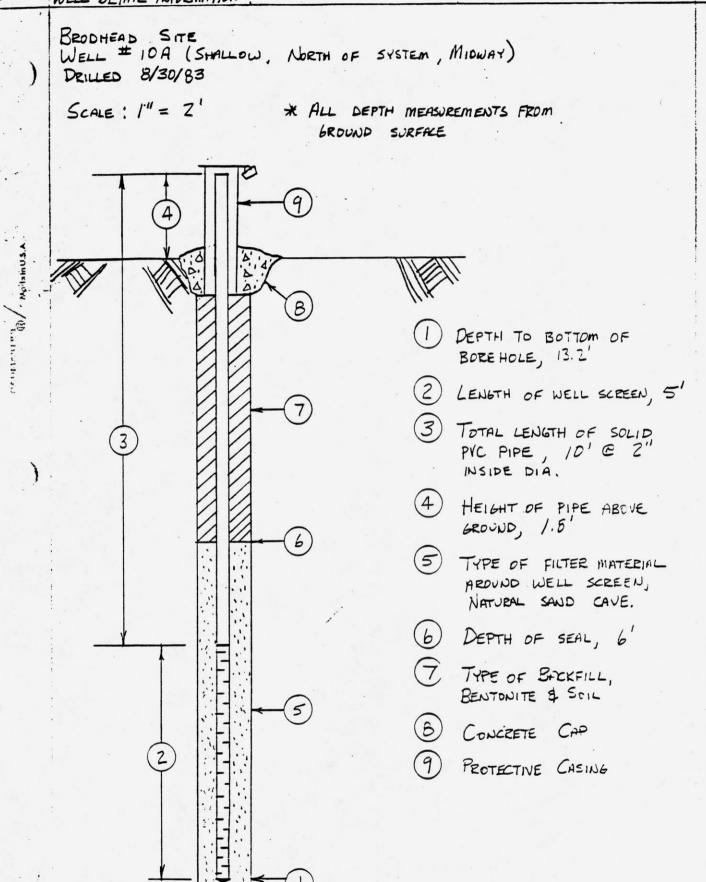
- 13. Knowles, R., "Denitrification", <u>Terrestrial Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- 14. Kudeyarov, V. N., "Mobility of Fixed Ammonium in Soil", <u>Terrestrial</u> <u>Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- Leach, Lowell E. and Carl G. Enfield, "Nitrogen Control in Domestic Wastewater Rapid Infiltration Systems", <u>JWPCF</u>, <u>55</u>, 1150 (1983)
- 16. Lund, L. J., et. al., "Nitrogen Balances for an Effluent Irrigation Area", <u>J. Environ. Qual.</u>, <u>10(3)</u>, 349 (1981)
- 17. McMichael, F. C. and J. E. McKee, "Wastewater Reclamation at Whittier Narrows", State of California, State Water Quality Control Board Pub. No. 33 (1966)
- 18. Metcalf & Eddy, Inc., <u>Wastewater Engineering: Treatment, Disposal,</u> <u>Reuse,</u> 2nd Ed. New York: McGraw-Hill Book Co., 1979.
- 19. Mitchell, James K., <u>Fundamentals of Soil Behavior</u>, New York: John Wiley and Sons, Inc., 1976
- 20. Monson, Helmer, "Disposal of Cannery Waste by Ridge and Furrow Irrigation", <u>Land Disposal of Liquid Waste</u>, a Collection of Papers Presented at the University of Wisconsin Engineering Institute on Industrial Wastes, 1956
- 21. Morrison, Robert D., "A Modified Vacuum Pressure Lysimeter for Soil Water Sampling", <u>Soil Science</u>, <u>134(3)</u>, 206, (1982)
- 22. Nommik, H., "Fixation and Biological Availability of Ammonium in Soil Clay Minerals", <u>Terrestrial Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- 23. Palazzo, A. J., "Seasonal Growth and Accumulation of Nitrogen, Phosphorous, and Potassium by Orchardgrass Irrigated with Municipal Waste Water", <u>J. Environ. Qual.</u>, <u>10(1)</u>, 64 (1981)

- 24. Patrick, W. H. and R. Wyatt, "Soil N Loss as a Result of Alternate Submergence and Drying", <u>Soil Sci. Soc. Amer. Proc., 28</u>, 647 (1964)
- 25. Patrick, W. H. and S. Gotoh, "The Role of Oxygen in N Loss from Flooded Soils", <u>Soil Sci.</u>, <u>118(2)</u>, 78 (1974)
- 26. Paul, E. A. and N. G. Juma, "Mineralization and Immobilization of Soil N by Micro-organisms", <u>Terrestrial Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- 27. Reddy, K. R. and D. A. Graetz, "Use of Shallow Reservoir and Flooded Organic Soil Systems for Waste Water Treatment: Nitrogen and Phosphorous Transformations", <u>J. Environ. Qual.</u>, <u>10(1)</u>, 113 (1981)
- 28. Rodenberg, Jerry R., "Ridge and Furrow System Design", Wisconsin Department of Natural Resources, Industrial Wastewater Section, (1980)
- 29. Ryden, J. C., et. al., "Direct Measurement of Gaseous Nitrogen Losses from an Effluent Irrigation Area", <u>JWPCF</u>, <u>53(12)</u>, 1677 (1981)
- 30. Sawyer, C. N., and P. L. McCarty, <u>Chemistry for Sanitary Engineers</u>. New York: McGraw-Hill Book Co., 1967
- 31. Schraufnogel, F. H., "Ridge and Furrow Irrigation Disposal of Milk Wastes", <u>Land Disposal of Liquid Waste</u>, A Collection of Papers Presented at the University of Wisconsin Engineering Insititute on Industrial Waste (1956)
- 32. Schraufnagel, F. H. "Waste Disposal by Ridge and Furrow Irrigation", Wisconsin Committee on Water Pollution Report No. WP-108, Engineering Experiment Station Research Report No. 20 (1962)
- 33. Sepp, E., <u>Nitrogen Cycle in Groundwater</u>, Bureau of Sanitary Engineering, State of California Department of Public Health (1970)
- 34. Soil and Plant Analysis Laboratory, University of Wisconsin Extension, <u>Wisconsin Procedures for Soil Testing</u>, Plant Analysis, and Feed and Forage Analysis (1980)

- 35. Stanford, G., R. A. Vander Pol, and S. Dziena, "Potential Denitrification Rates in Relation to Total and Extractable Carbon", <u>Soil</u> <u>Sci. Soc. Amer. Proc., 39</u>, 284 (1975)
- 36. Tiedjo, J. M., J. Sorenson, and Y-YL. Chang, "Assimilatory and Dissimilatory Nitrate Reduction: Perspectives and Methodology for Simultaneous Measurement of Several Nitrogen Cycle Processes", <u>Terrestrial Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- 37. Tusneem, M. E. and W. H. Patrick, Jr., <u>Nitrogen Transformations in</u> <u>Waterlogged Soil</u>, Louisiana State Univ., Agricultural and Mechanical College, Dept. of Agronomy, Ag. Expt. Station Bulletin No. 657 (1971)
- 38. Winneberger, J. H. T., <u>Nitrogen, Public Health and the Environment:</u> <u>Some Tools for Critical Thought</u>, Ann Arbor, Mich.: Ann Arbor Science, 1982
- 39. Wisconsin Department of Natural Resources, <u>Evaluation of Industrial</u> Wastewater Ridge and Furrow Systems, (1984)
- 40. Wisconsin State Lab of Hygiene, <u>Manual of Analytical Methods</u> -<u>Inorganic Chemistry Unit</u>, (1980)
- 41. Woodmansee, R. G. and L. S. Walloch, "Effects of Fire Regimes on Biogeochemical Cycles", <u>Terrestrial Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)
- 42. Woodmansee, R. G., et. al., "Grassland Nitrogen", <u>Terrestrial</u> <u>Nitrogen Cycles</u>, Proceedings of International Workshop arranged by SCOPE/UNEP International Nitrogen Unit of the Royal Swedish Academy of Sciences and the Commission of Research on Natural Resources of the Swedish Council for Planning and Coordination of Research at Gysinge Vardshus, Osterfarenbo, Sweden. (Sept. 16-22, 1979)

A PPENDIX A Prode In U.S.A. BRODHEAD : WELL AND LYSIMETER LOGS





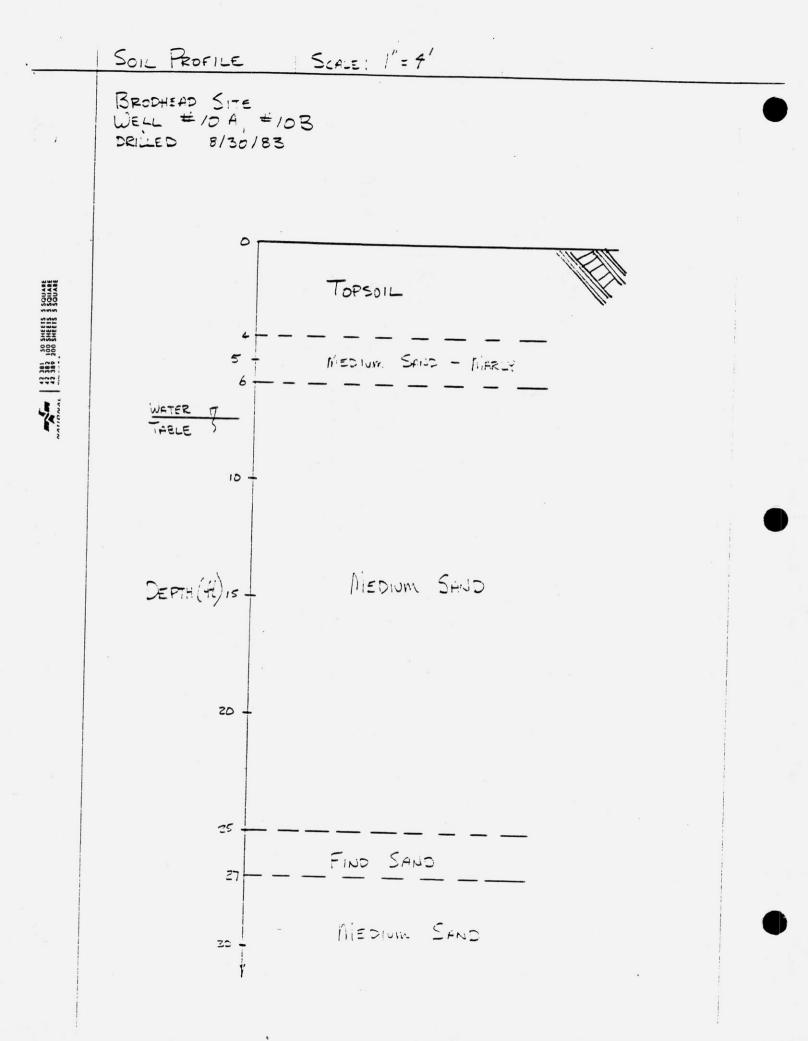
WELL DETAIL INTERNATION

BRODHEAD SITE WELL #10B (DEEP, NORTH OF SYSTEM, MIDWAY) DRILLED 8/30/83 SCALE: 1"= 4' * ALL DEPTH MEASUREMENTS FROM GROUND SURFACE 4 9 8 DEPTH TO BOTTOM OF BOREHOLE, 26.2' 6 (2)LENGTH OF WELL SCREEN, 2,5' 3 TOTAL LENGTH OF SOLID PUC PIPE, 25'@ 2' INSIDE DIA. (4) HEIGHT OF PIPE ABOVE 60000, 1.3' 3 (5) TYPE OF FILTER MATERIAL AROUND WELL SCREEN, 5 NATURAL SAND CAVE 6 DEPTH OF SEAL, 6' 7 TYPE OF BACKFILL, BENTONITE AND SOIL B CONCRETE CAP 9 PROTECTIVE CASING

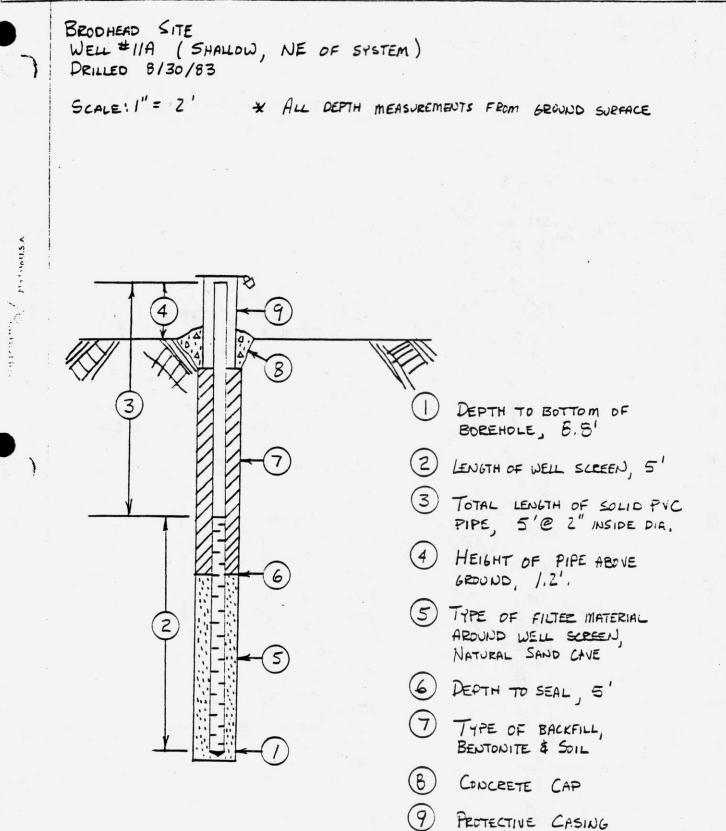
•

Alamanus.A

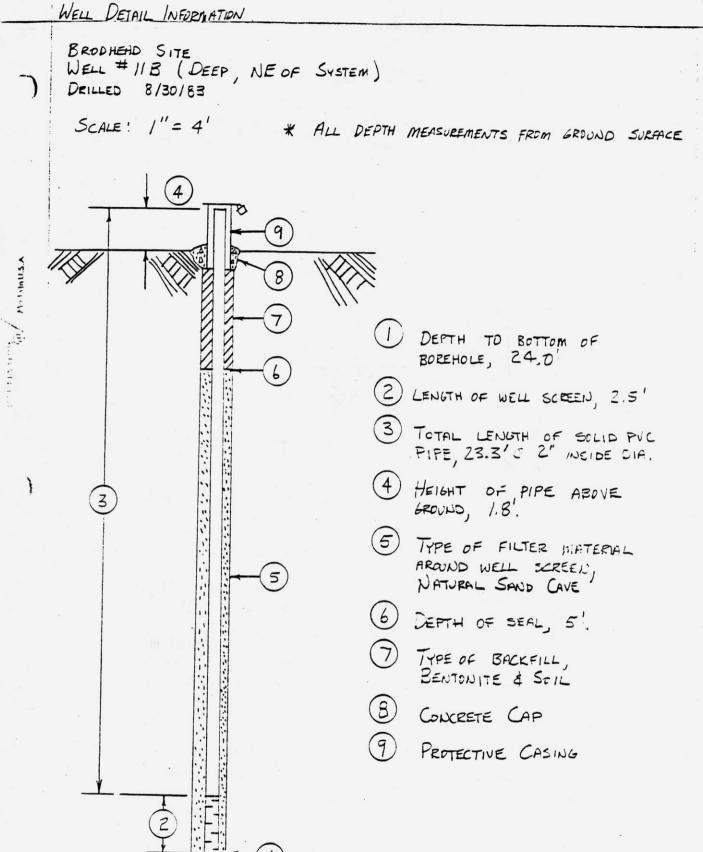
1 Case

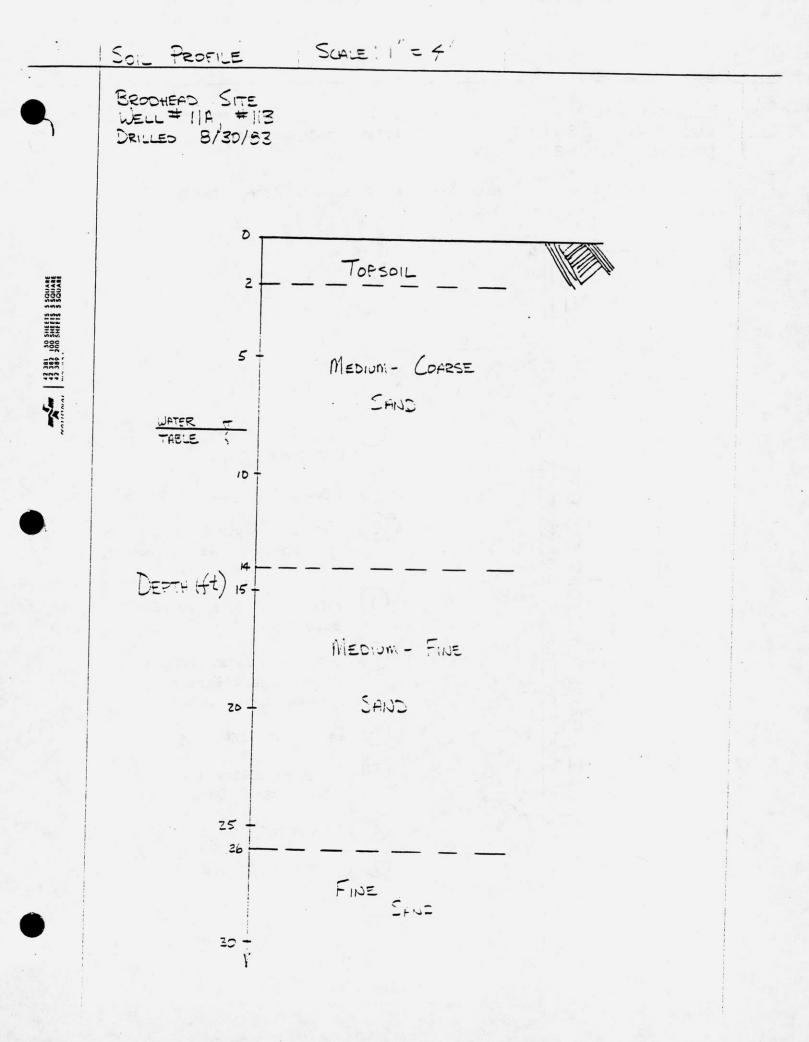


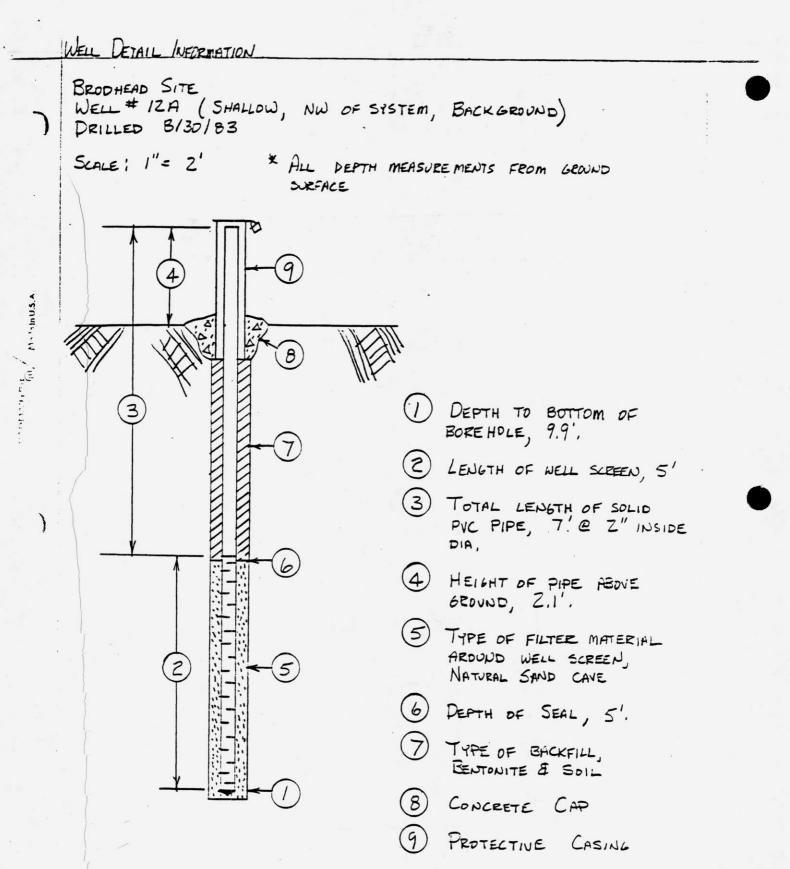
WELL DETAIL INFORMATION

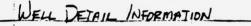


•





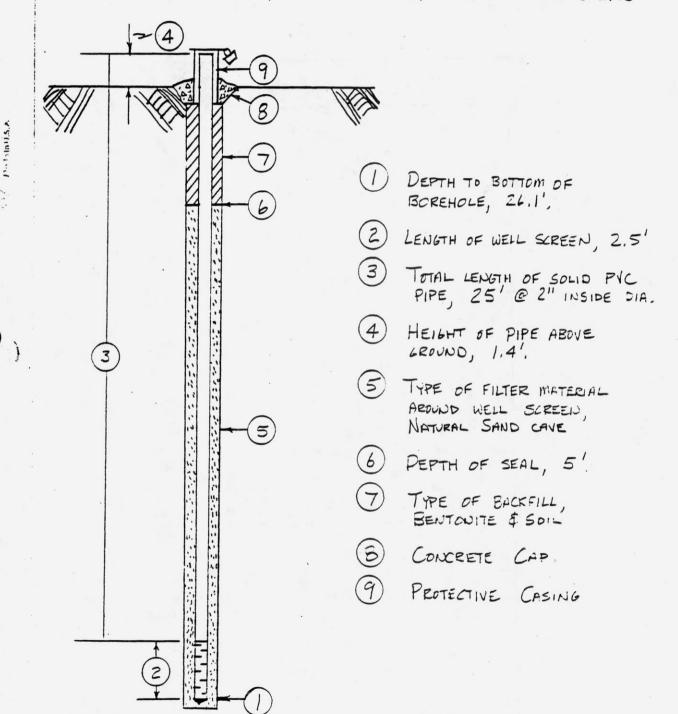




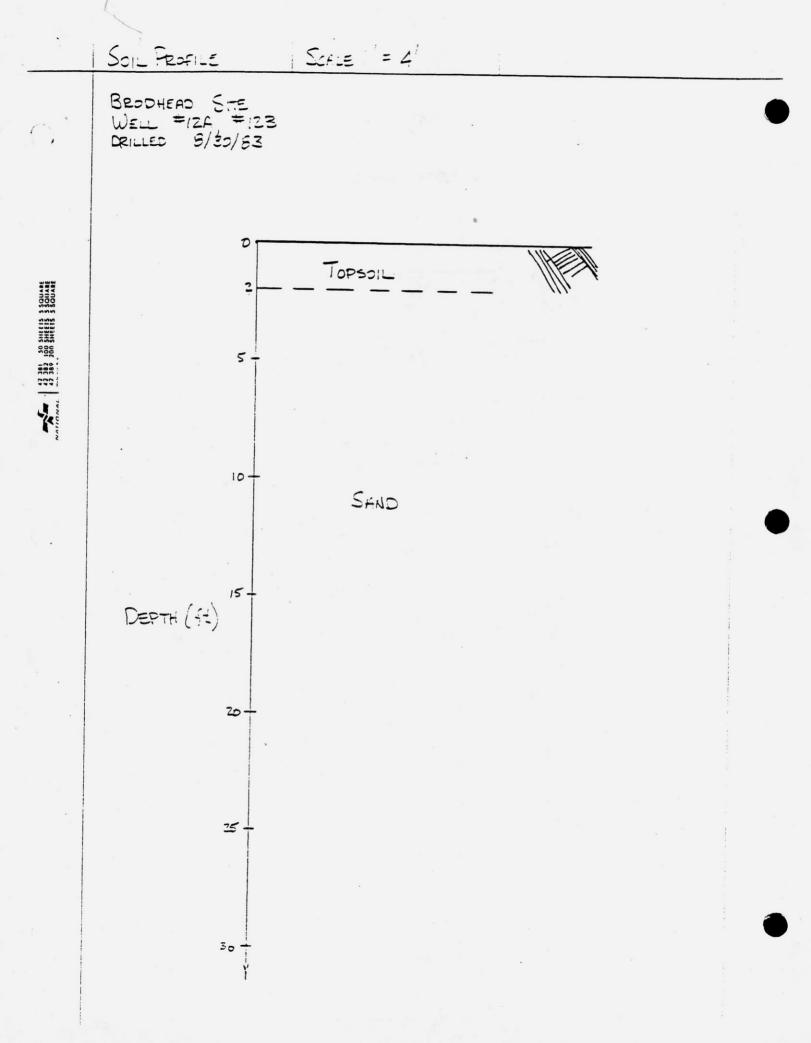
BRODHEAD SITE WELL #12B (DEEP, NW OF SYSTEM, BACKGROUND) DRILLED B130/83

SCALE: 1" = 4'

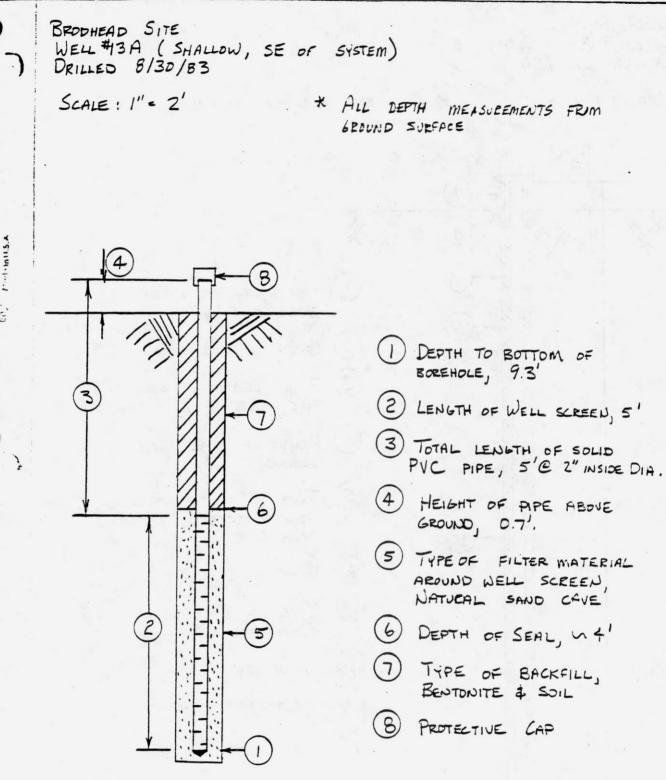
* ALL DEPTH MEASUREMENTS FROM GROWD SURFACE

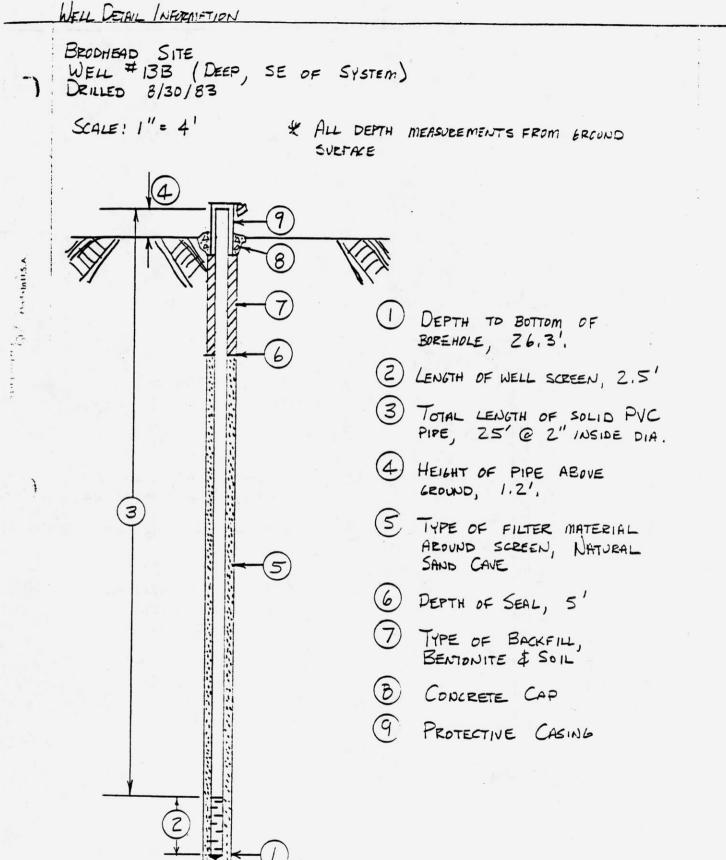


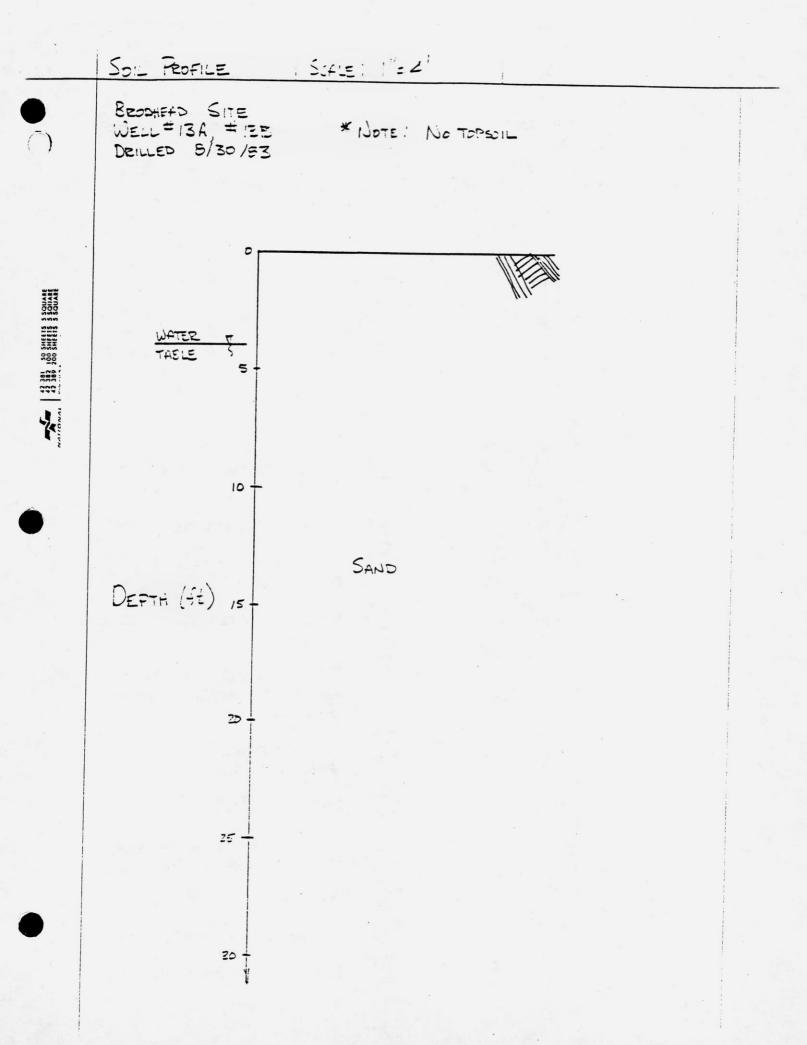
•

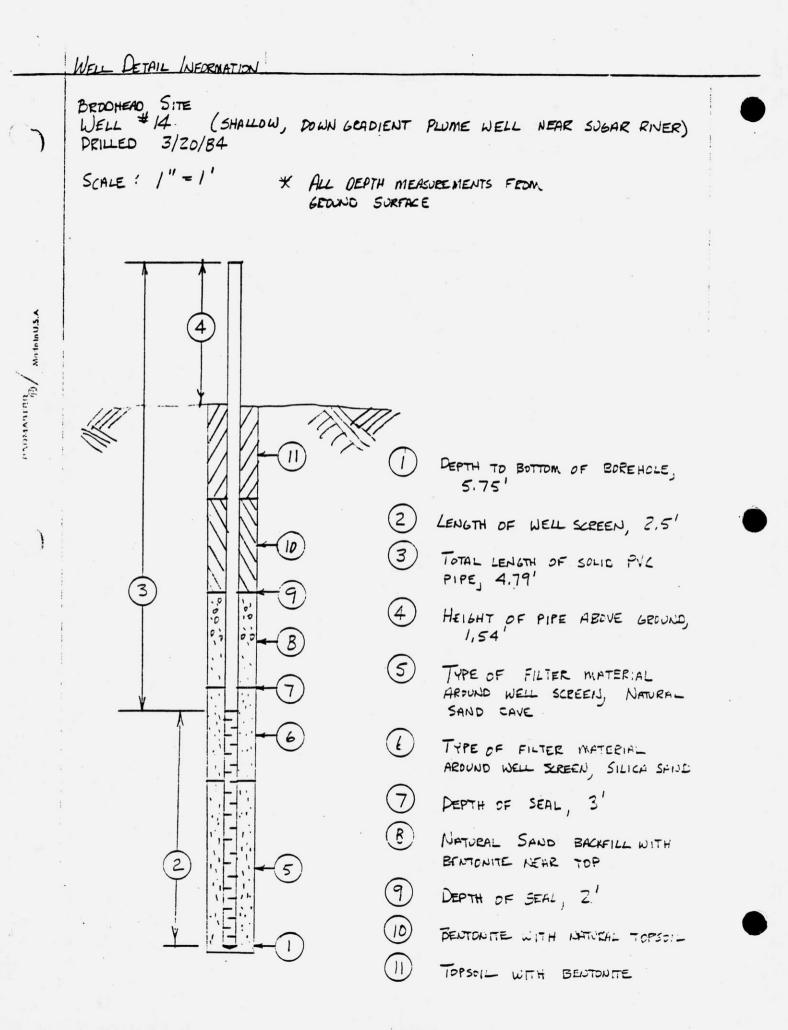


WELL DETAIL INFORMATION.

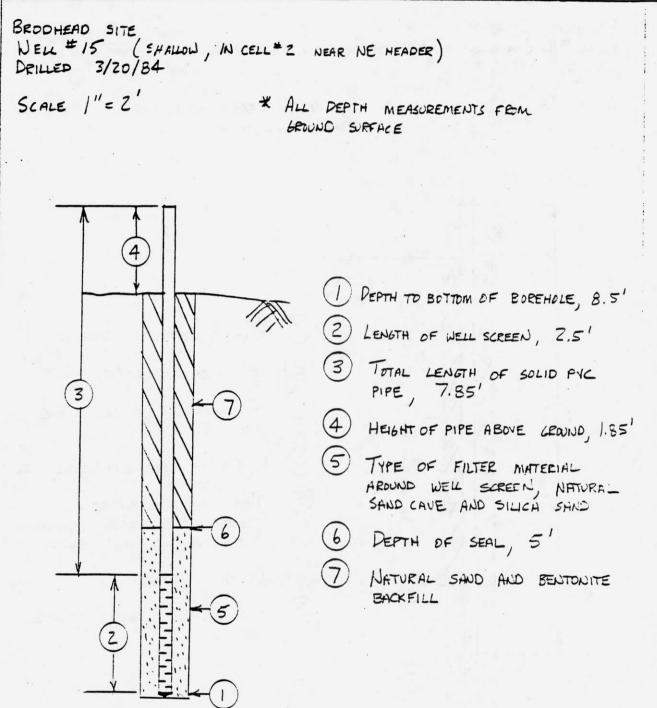






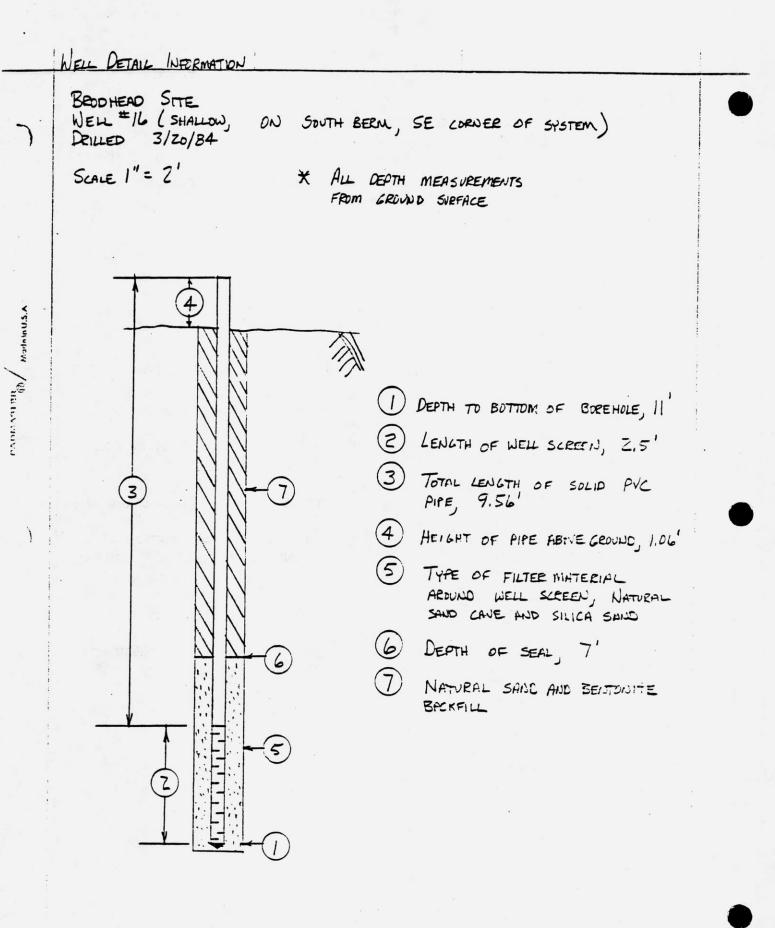


WELL DETAIL INFORMATION

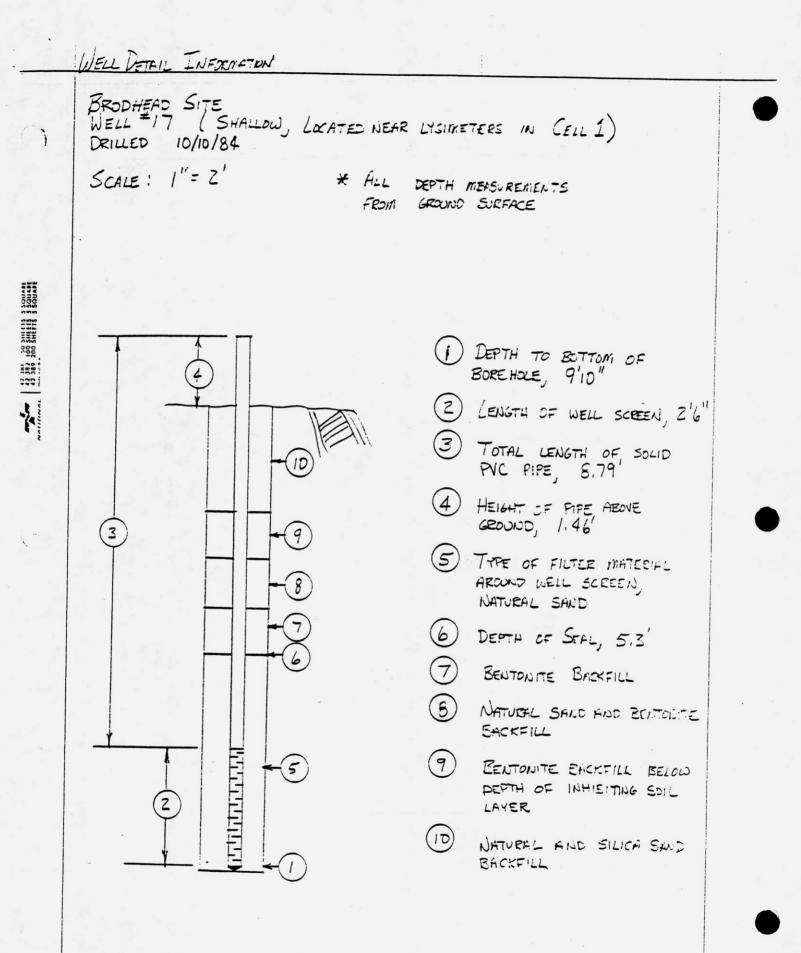


A.S.Uni-ticle 10

1 Manaration of



WENS 14, 15, 10 SOIL PROFILE SCALE: 1"=1" DRILLED 3/20/84 WELL 14 WELL 15 7 WELL 16 0 SILTY TOPSOIL 1 SILTY TOPSOIL V.S.University APPENENTING Ζ SAND 3. SAND DEPTH (41) 4 SAND TABLE S 5 6 3 7 6 ٩



SOIL PROFILES

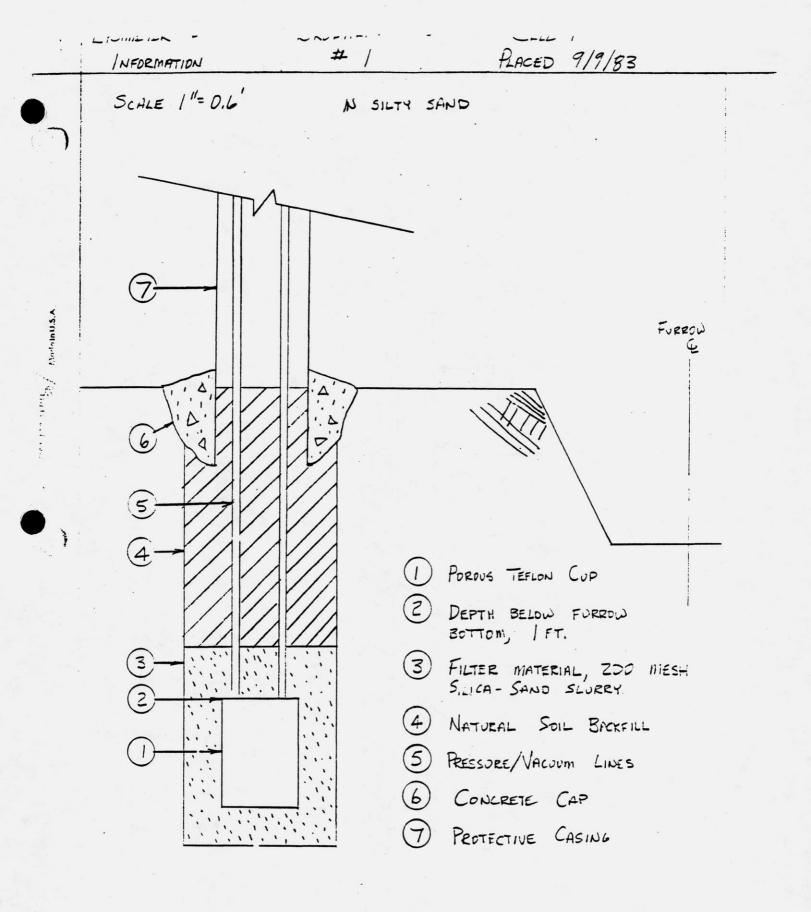
.5/10/84

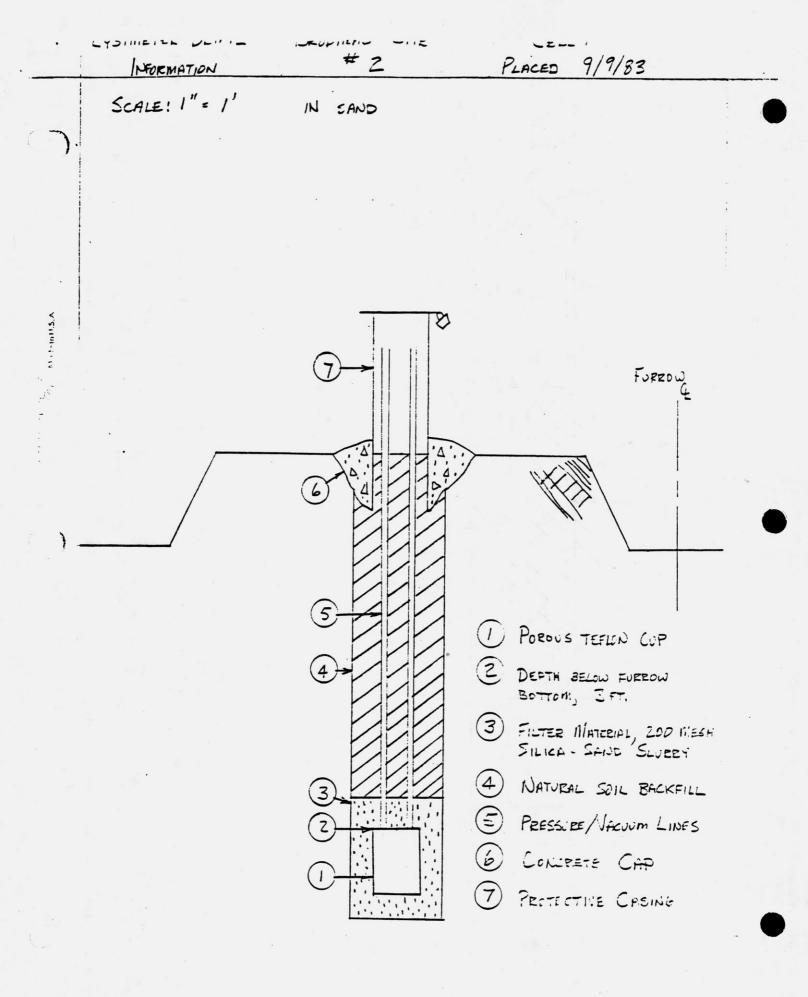
•

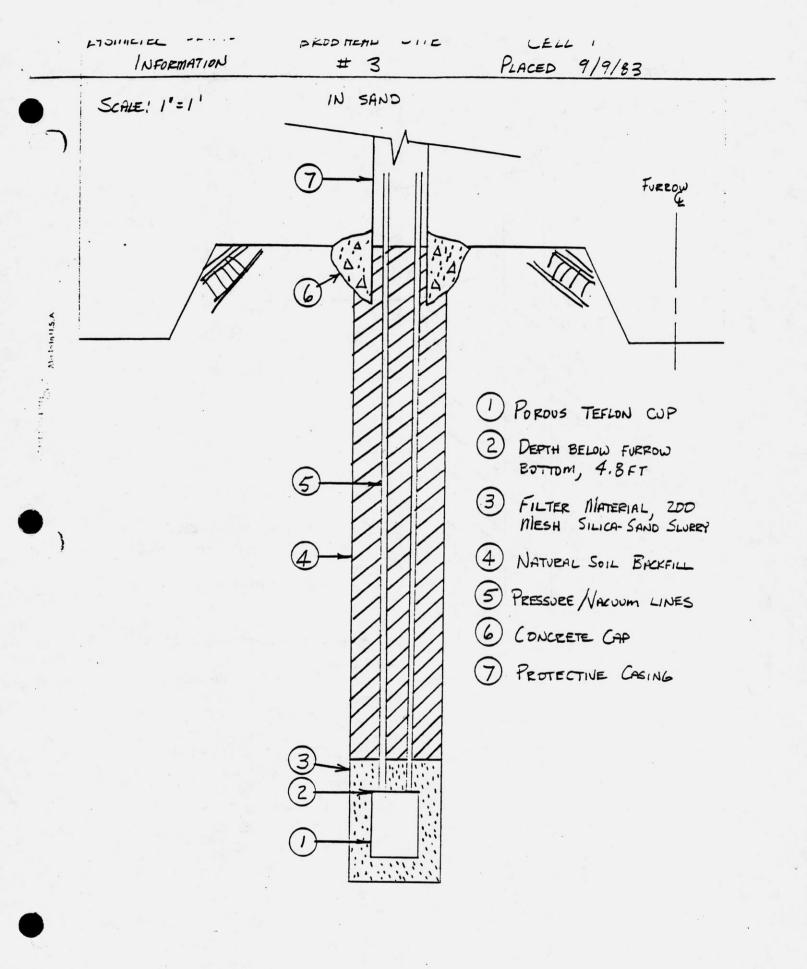
•	6 -	WELL 17	CELL I FURROW By Lysimeters	FUREOW NU IN CELLI OUTSIDE HEADER	RIDLE IN HIN CELL! AUTEIDE HEADER
()					
		DARK SATURATED			
	· 1	SAND	AIR!	AIR	REDDISH
QUARE QUARE QUARE	*				SAND
SHEELS SS	2	FUREOW BOTTOM	ORGANIC MAT	ORGANIC MAT)	
42 381 50		DARK SAND-ORGHISC INHIBITING LAYER	DARK SHND	DARKER Sand	
Marianar 13 38, 100 SHEIS SOUND	3 -	CLEAN	DARK Organic		WHITISH. SAND
		Moist	BASFIER		
•	4 +	SAND	CLEANER SAND		
	5				
	6 +	"ANAERDEIC"			
	0	DARK SAND			(and)
	7 +	7 WHTER 7 TAELE			
	в				
	9 - ¥				

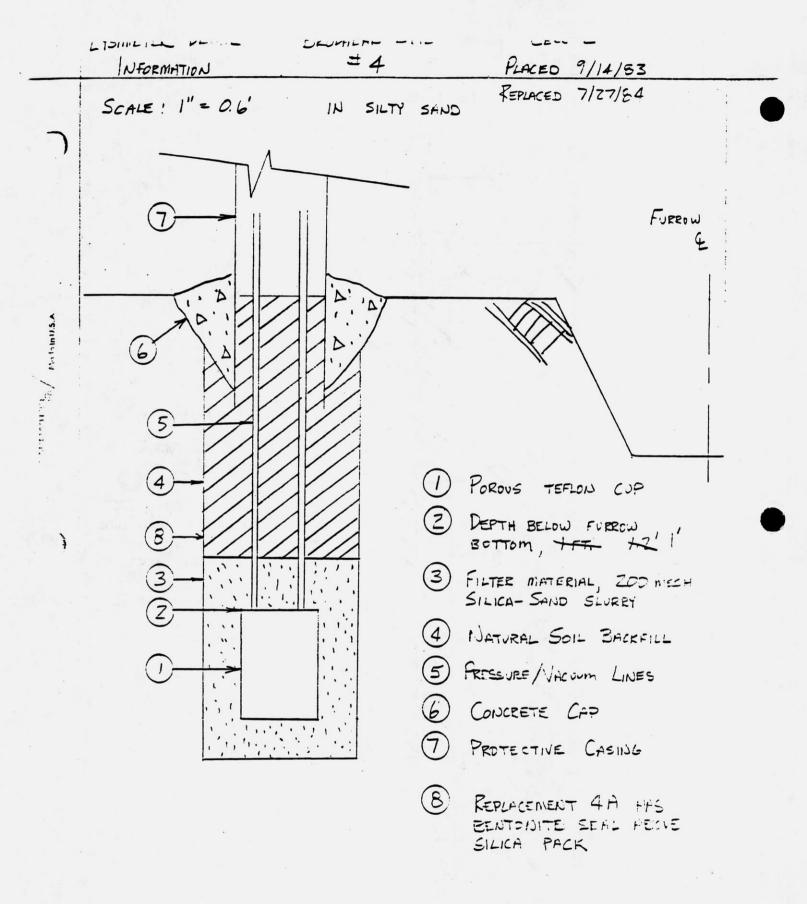
ο.	NWCELL 1 FUEDOW INSIDE HEADER	NW CELL - RIDLE INSIDE HEADER	NE CELL ZRIDGE Dutside header	CENTER CELLZ Eftween Headers	
, _	AIR!		D11.	Au	
	ORGANIC MAT)	Sand	ALL Sand	ALL Sand	
	Dark Sand	DAEK SAIJD DAERIER			
	LIGHTER Sand	LIGHTER SAND			

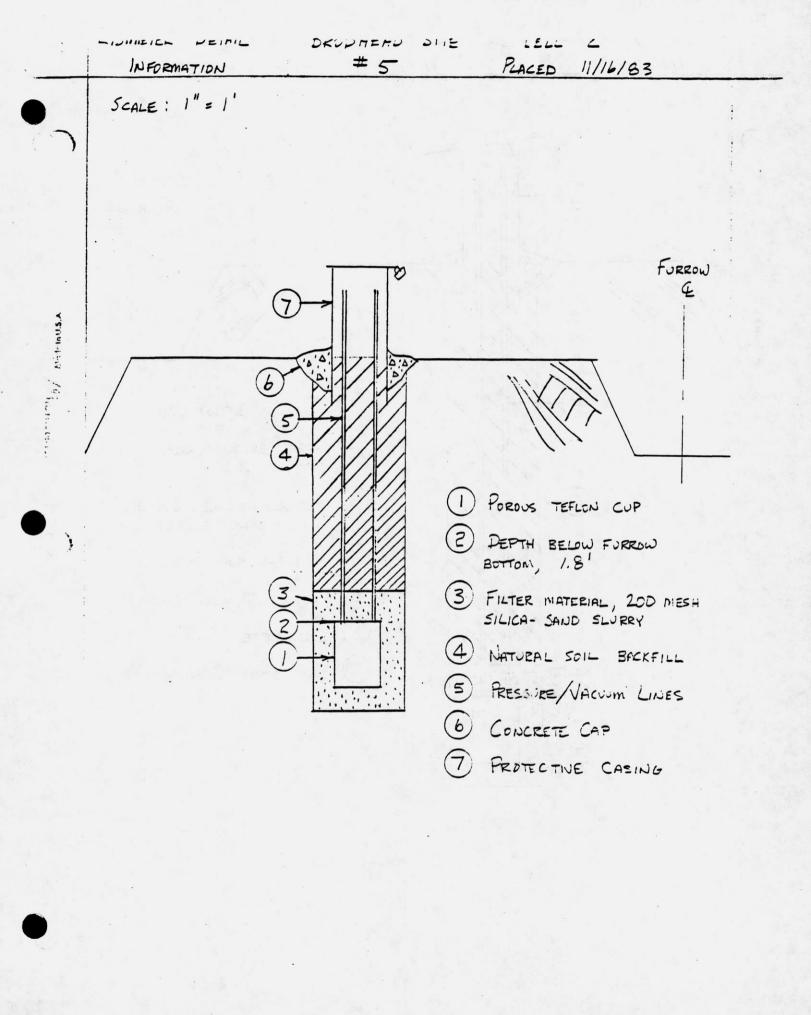
CELL Z HAS SUBSURINCE BARRIERS ONLY AROUND # 44 HEAVER AS SEEN WHEN INSTALLING LYSINGTER # 44

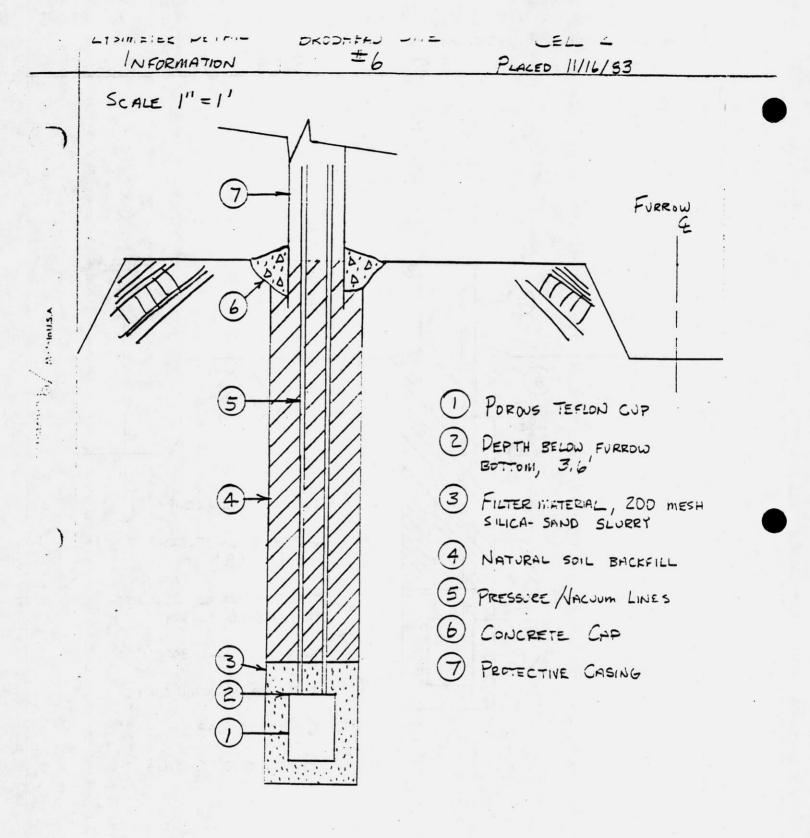


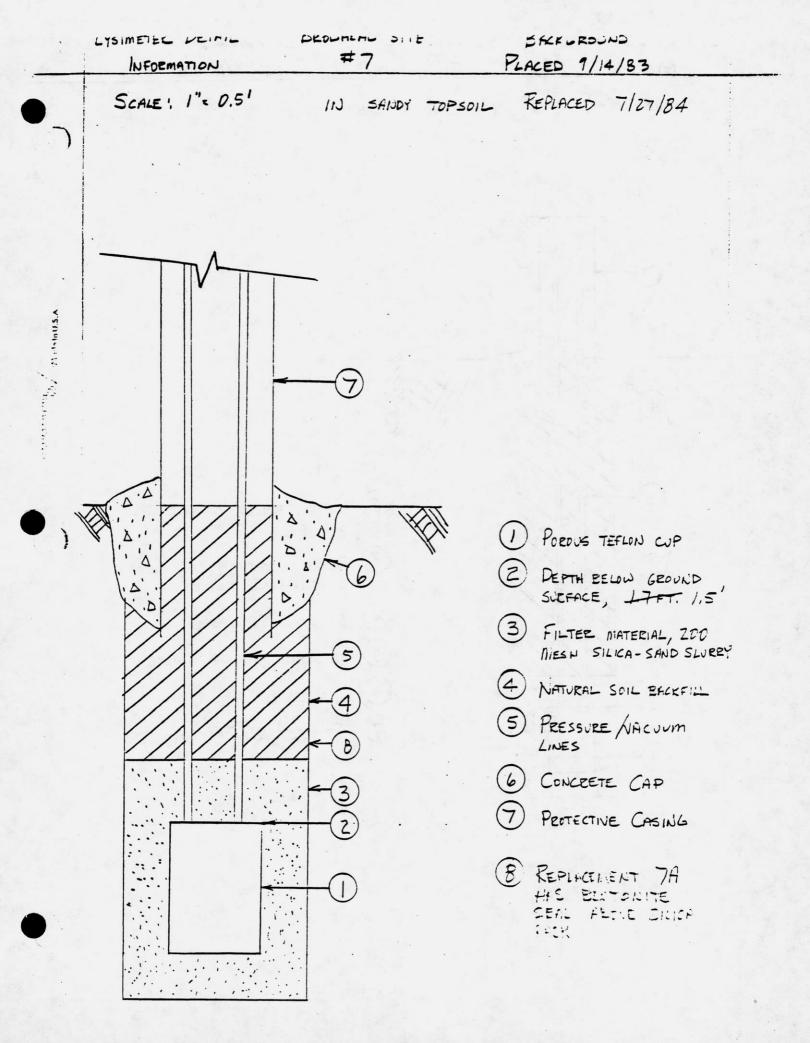


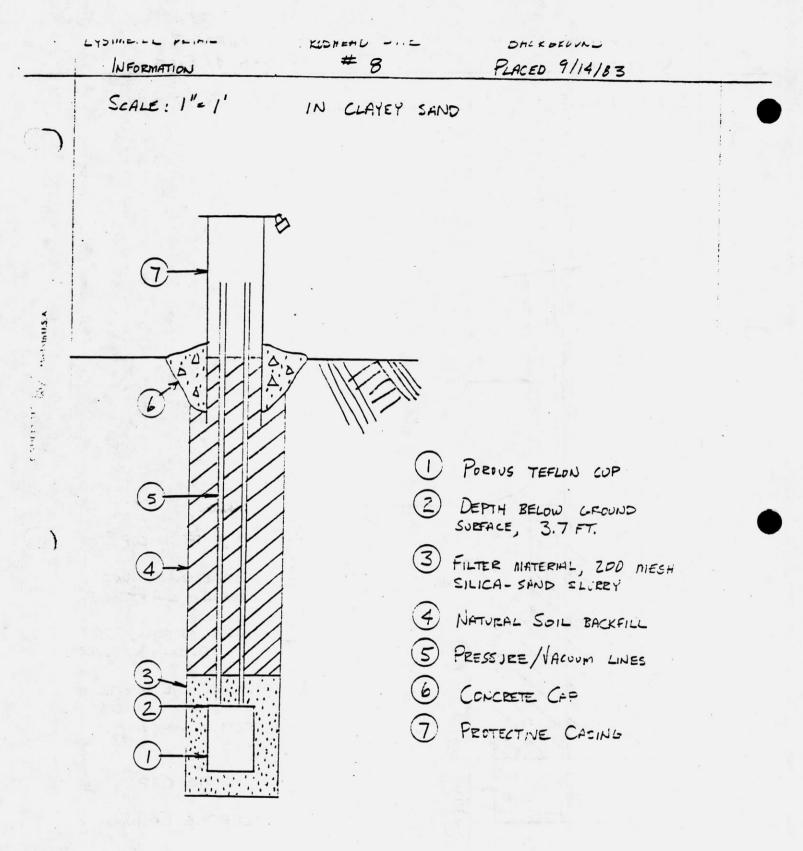


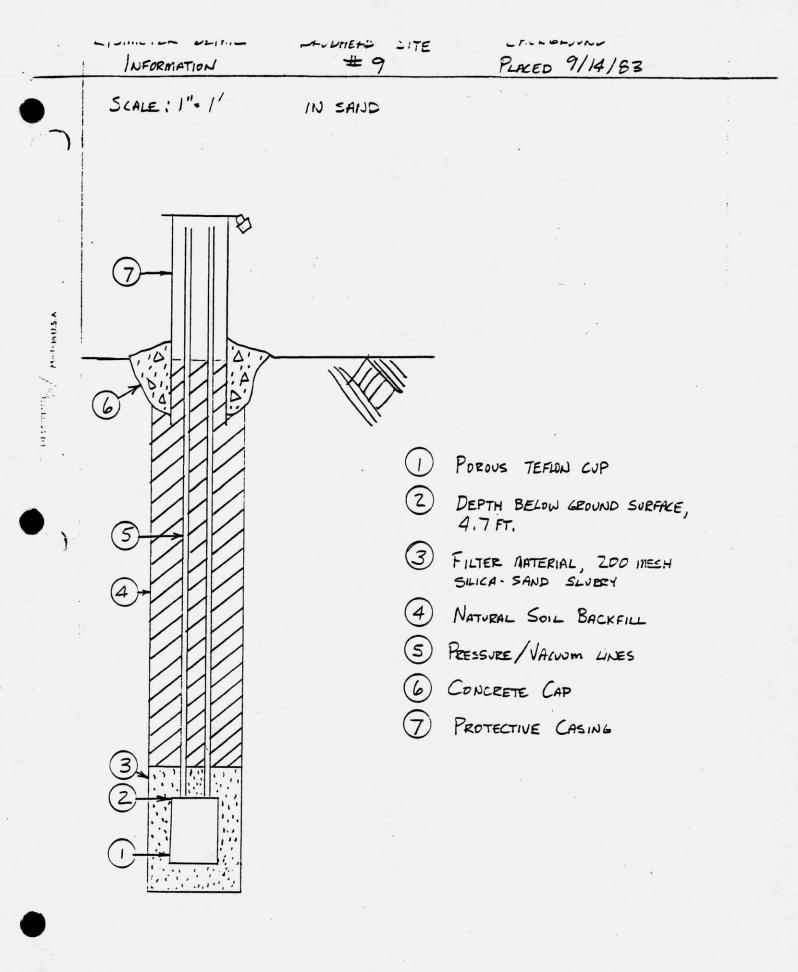


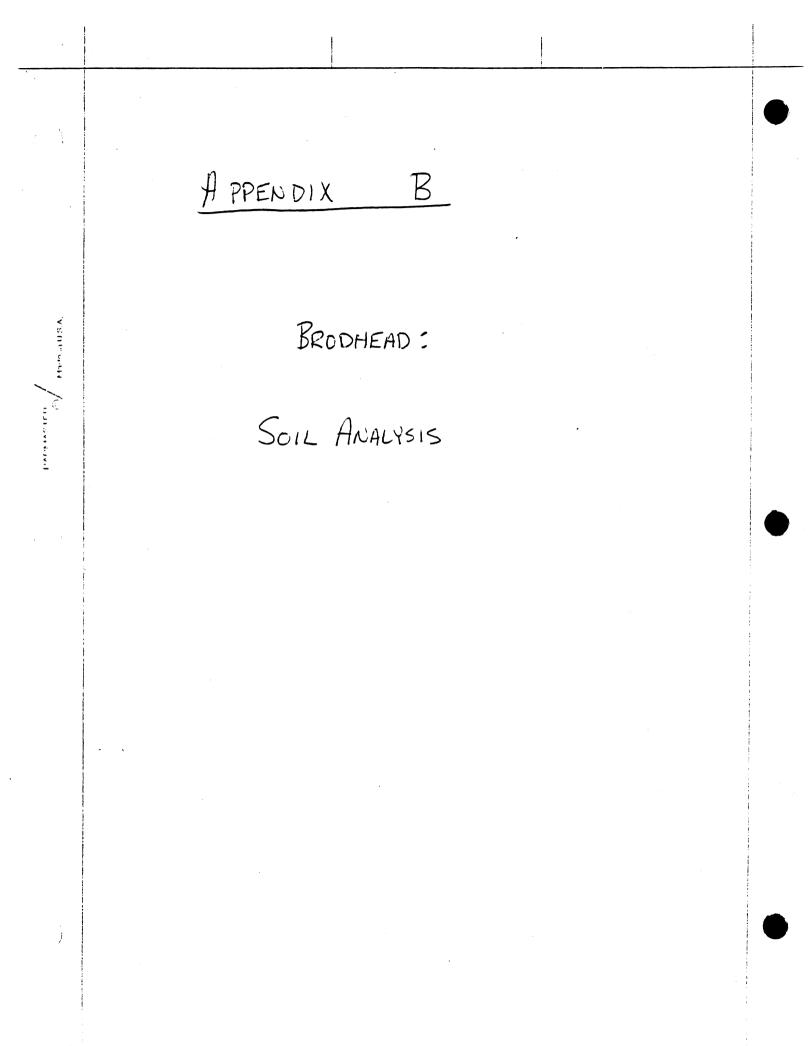












OPERATIVE EXTENSION PROGRAMS

University of Wisconsin-Extension

LUEX University of Wisconsin-Madison Chil & Plant Analysis Laboratory, 806 South Park Street, Madison, Wisconsin 53715; 608-262-4364



DEPARTMENT OF SOIL SCIENCE

October 7, 1983 Acct. 900 Lab No. 00290

MEMORANDUM

David Sauer Wis. DNR Box 7921 53707 Madison, WI

FOR BEDDHIFD

Soil/Plant Analysis Lab FROM:

5 soil samples submitted Sept. 15, 1983.

RE: Results o	r analyses on 5 s	5011 Jump -		Total
	Sand	Silt	Clay	<u>N</u>
Sample No.			-%	0.04
	91	4	5	
$\sum_{i=1}^{n-1}$		16	3	0.11
レビン 2 して 2	81		3	0.01
√ 2 2 √ 3 2 4 3	96		2	0.01
1	93	5	-	-0.01
	95	3	2	
£' 5	,,			

"_" values = less than.

All additional analyses are attached.

Your invoice for these analyses is enclosed.

Encls.

/sf

University of Wisconsin-Extension . United States Department of Agriculture . Wisconsin Counties Cooperating and Providing Equal Opportunities in Employment and Programming

IDENTIFICATION FIELD VIELD GGALS S3707 S3707 FARME ACRES NAME BOIL NAME BOIL NAME BOIL DATS CON ALFAFA DATS 70.0 LABORATORY ANALYSIS FOR EACH SAMPLE FARME Image: State	0-00290 LAS. FARM NO. 11A18 CUMITY ACCOMING WI 13 900 DATE REGIVED DATE PROCESSIO 09-21-83 09-21-83	SOIL & PLANT ANALYSIS LAB 806 S PARK MADISON WI 53715	SOIL TEST REPORT Somples Analyzed By: SOIL & PLANT ANALYSIS LAB 806 S. PARK MADISON WI	IS FOR: DAVID SAUER WIS DNR-BOX 7921	COOPERATIVE EXTENSION I UWEX University of Wistoman Soils Department, Ma WISC
ORAGE Bit Generita Bit Generit	FIELD 1 ACRES SOIL NAME GROUP XE PLOW	CORN ALFALFA	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	ANALYSIS FOR EACH SAMPLE	07 COOPE FARMER
Induman Image: Constant of the state	LIME TONS HIS ACR GRADE PH on PH USED 0.6 6.9 00-89 NONE on NONE 10 ONE MONE	CROP YEAR PLANI NUINENIS P205 CORN YIELD GOAL BU/A FIRST 0 110 61-80 SECOND 0 110 81-100	PLANT NUTRIENTS ALFALFA N $\frac{7}{20}$ s $\frac{1}{8}$ N $\frac{7}{20}$ s $\frac{1}{8}$ 100 0 20 120 0 25 140 0 30	FERTILIZER PROGRAM NUT NUTRIENTS OTHER CHOPS AND VIELD GOAL 05 K20 010 R CHOPS AND VIELD GOAL 0 75 0 75 0 150	PLANT NUTRIEN
RETEST THIS FIELD AT LEAST EVERY 2 YEARS BECAUSE OF THE LOW POTASSIUM BUFFERING CAPACITY OF ITS SOILS.	FERTILIZATION PROGRAM <u>CORRECTIVE</u> <u>MAINTENANCE</u> <u>NUTRIENT ADJUSTMENT (S TOTAL</u> FERTILIZATION RECORD DATE AND/OR METHOD OF APPLICAT	SI RATE TION RATE CRADE RATE CRADE RATE CRADE RATE CRADE RATE CRADE RATE CRADE RATE CRADE FERTILIZAT AND/OR ME FERTILIZAT AND/OR ME	TION VEAR 19 M PLANT NUTRIENT STIVE ENANCE NT ADJUSTMENT (S) TOTAL ION RECORD DATE THOD OF APPLICATION	LIZER REQUIREMENTS CROP FERTILIZATION PROGRAM CORRECTIVE MAINTENANCE ICTRIENT ADJUSTMENT (S) TOTAL FERTILIZATION RECORD DATE AND/OR METHOD OF APPLICATION	

COOPERATIVE EXTENSION PROGRAMS

LLLEX University of Wisconsin-Madison

Soil & Plant Analysis Laboratory, 806 South Park Street, Madison, Wisconsin 53715; 608-262-4364

DEPARTMENT OF SOIL SCIENCE

September 21, 1983 Acct. 900 Lab No. 00221

MEMORANDUM

Dave Sauer TO:

Wis. DNR Box 7921 Madison, WI 53707

BRODHEAD

FROM: Soil/Plant Analysis Lab

RE: Results of analyses on 7 soil samples submitted Sept. 1, 1983.

Sample No.	Sand	Silt	Clay	Total N
	*****	%		*******
1 W10 4-6	95	2	3	0.04
2 10 6-5	97	0	3	0.02
3 10 9-14	96	1 ·	3	0.03
4 WID 21-29'	99	0	1	-0.01
5 WII 2-4	95	2	3	0.04
6 WII 9-14'	99	0	1	-0.01
7 WIZ 2-4	95	0	5	0.03

Note: "-" values = less than.

If you have any questions concerning these analyses, please feel free to contact us.

Additional analyses attached.



)

I E PLANT	ANALYSIS LAB			S			Analyzed By		THIS R	EPORT					LUEX	University University	NBION P of Wisconsin of Wisconsin partmant, Ma	 Extension Madison
S PARK ISON	WI 53715				. C S.	PLAN1 PARK	ANALY	SIS LAD		UN:	И		SAUER NR BOX 7 ON	7921 537(WI 07			ISCONSIN DEPERATIVE S COPY
YIELD G	QALS		的。					LABORAT	ÓRY AN	ALYSI	Ş F	OR EA	сн şамр	LE State				
A A	_FALFA	Sample Number	Soil Texture (Code)	Est. CEC	Soil pH	Organic Matter Tons	P Phosphorus Ibs	K Potassium Ibs	Ca Culci Ibs	um	Mag	Mg gnesium Ibs	B Boron Ibs A	Mn Manganese Ibs	Zn Zinc Ibs A	s	S ultur Ibs/A	Lab Use Buffer Code
	70.0	12	1	2 1	6.5 7.1	32	15 VL 23 VL	65 L 75 L	700	M	18	M DI	WID 6.8					7.5
		3	1	22	7.3	2	14 VL 11 L	75 L 50 L	900 900	M	10		10 -1-14 10 26-21					
		5	1	3	7.4	11 2	23 VL -14 L	45 L 55 L	1250 550		25 10	0 M 0 L	WI 2.41					
					SOIL	TEST LEVE	L CODES: VL (V	(ery Low), L(Low)	, LM(Low Med	ium), M(Mə	idium) ,	HM(High N	Aodium), H(High),	VH(Very High), E	H(Excussively F	ligh)		
CORRECTIVE		1. L. A. L.		* 3			M	AINTENAN	NCE FEF	TILIZE	R P	ROGR	AM	n Fris de La Su Antoine La Su		1-7-942 		
- PLAN NUTRIE P205 ¹⁰⁵ /A		AL		P		NTS K20	ALFAI YIELD T/A	LFA GOAL		NT ENIS K ₂ O Ibs		ОТНЕ	ER CROPS A	ND YIELD	GOAL	PLANT N		NTS K20 Ibs/A
ND			<u> </u>						·									
,																		

· .

	WORKSPACE FO	R CALCULATING YOUR TOTAL ANNU	Al ^{de} fertilizer requir	IEMENTS	
	YEAR 19	C-1OP:	YEAR 19	CROP;	YEAR 19
	$\frac{PLANT NUTRIENTS}{N} \frac{P_2 O_6}{(t_0, t_0, t_0)} \frac{\kappa_2 O}{\kappa_2 O}$	FERTILIZATION PROGRAM	PLANT NUTRIENTS	FERTILIZATION PROGRAM	
[CORRECTIVE		CORRECTIVE	

	·													•
C PLANT ANALYS	IS LAU					T REP(Analyzed By: AtiALYS		This report Is for:	DAVE S	54UER	c		University of Wisconsin University of Wisconsin Soils Department, Mi	Madison 🚦
YIELD GOALS		Semple Number	Soil E feature Cl (Code)	at. Soil pH	Organic Matter Tone A	Phosphorus Ibs 17 VL	ABORATO Polessium Ibs A 60 L	Ce Culcium Ibs/A 8CJ M	ISTFOR FA	CH SAMP	Mn Manganese Ibs A	Zn Zinc Ibe	S Sultur Ibs/A	Lab Use Buller Code
7	C.C			2 6.9	4	Ti Ar	OV L							
	CORN YIELD GOA BU/A 61-8	AL 10	PLAI	NT NUTRIE P205 Ibe/A 30		ALFAL YIELD T/A	INTENAN	LM(Low Muchum), M(N CETREFTILL PLANT PLANT PLOS F205 IUS 105 105 105 105 105 105 105 105	ERIPROGR OTH	AM ER CROPS A		GOAL		
60 110 60 0	81-1 101-1		120 140	35 40	30		1-4.0	50 200						

 \mathcal{C} APPENDIX BRODHEAD: WASTEWATER CHEMISTRY DATA)

	BRODH Data Sh		INULATIVE	Chem (m)(c)		BWWNSE		IW ASTEW	TAKEOFF ATER	- Speaketert
	DATE	TOTAL BOD5	TOTAL COD	165 705	3 a TKNX	10	11 NO2-N+ NO3-NX		NOB) PH	OTHER
	0/20/53	3200		_	46.0	1.4	3.0	89D	10.4 W	_
	3/7/34	1700	2300	5961570	34	2.7	4.1	32	-	TSS 1570
	4/23/34	1200	2400	964	4)	2,5	5.8	1600	6.6	775 3775
•	5/24/84	1600	2100	805	4D	2.2	0.7	2300	7.4	7.4 IAE ph
	6/7/84	1600	>2000 ?	930	45	2,5	2.7	90	6,3	
	7/19/84	1800	2700	470	44	4,1	1.6	1300	7,1	-
	7/31/84	1700	2400	464	36	Z.8	z ,4	980	-	
	8/30/34	>1100	2100	764	32	1,5	0,4	620	7.4	
	9 25 84	Z500	3400	1110	75 TOT 37 DISS	Z.9 Z.9	5.2	94 <i>0</i> .	7,4	
	10/9/84	980	3220 0655 2100	555	28 TOT	2,6	4.2	1400	7,8	
	11/20/34	2200		1130	21 0155 38 tot 2.1 0:55	2.0	5,2	190		
			-							
			-				·			
			-							

	DATA SH			TAR	AMETERS	(mg/_)) KA	w was	TENATER
	DATE	ALKALINIT	P TOTAL	50,2-	$\int Ca^{2+}$	Nat+	Mg2+	K+	OTHER
	10/20/83	~814*	120.	14,	170	840	78-	-	_
	3/7/84	<u> </u>	29	210	69	86	3Z	35	_
	4/23/84	35Z	44	-	110	1100	49	5/	-
•••••••••••••••••••••••••••••••••••••••	5/24/84	67Z	46	/al T	89	0071	28 j	47	-
- 1. 	6/7/84	—	-		_		- 		-
7	7/19/84	-	-	-	_	-	-	_	-
1	7/31/84	_	-	_	~	_	-	-	_
	8/30/24	530	49		110	490	52	22	_
					109.6	343.2	47,8		
					1.37	36.68	0.98		
\mathbf{v}									
/									
•									
•									
					-				
1									

APPENDIX D. PARMATER / HARMUEA BRODHEAD: WASTEWATER FLOWS TO RIDGE & FURROW 30- DAY AVERAGES 24 HR PROJECT FLOWS

		_						
		U	BRO	XDHEAD TER FL	ош (баш	cns/DAt		
			YEARS	1979 -	1984			
	MONTH	1979	1980	1981	1982	1983	1984	1924
	JANUARY	26496	54740	43214	37966	35371	35289	
	FEBRUNRY	36459	55575	48502	3721Z	36024	3816Z	
	MARCH	39979	59618	51924	37/74	35347	37548	
	APRIL	19054	49288	31206	42200	43000	25056	43070
	MAY	12734	32466	27764	53220	26429	37656	50400
-	JUNE	16785	39547	28506	49348	33035	428R	42064
	JULY	43178	22780	33018		40441	44145	29167
	AUGUST	428ZI	27567	40469		25855	35664	
	SEPTEMBER	41895		50317		38578	48414	56600
	OCTOBER	54664	36849	49852	46469	4234Z	43075	45000
	NOVEMBER	55658	35616	61427	50389	37665	-	
	Decemetr	45124	41563	22339	51731	28197		
	YEARLY AVERAGE	36237	41436	40713	45082	35140	38783	44717
	Commutative Averave	36237	38723	394D6	40567	39404	39507	\times

ł

* 24-HR FLOWS DETERMINED ON SAMPLING DAYS

-)

Presenting LL C. A.

)

) APPENDIX E A S H ni ebeta BRODHEAD : GROUNDWATER ELEVATIONS AND CONTOURS

DATA SHEET ! WELL IDA / ENISTH ! 15/10 17/10

- ..

SHALLOW

	DATA	SHEET ; WEL		6TH: 15/ H.67	1.4.64	SHALLOW	·
	DATE	DEPTH TO GW	GW Elevation	Volume HzD (ft)	3 Volumes (GAL)	Volume Removed	COMMENT
	9/23/83	6.37	772.00	-			
	10/20/83	7.00	771,37	~ B	. 4	5	
	11/16/83	7.54	770.83	7. /	3,6	5	GRAY TINT, SLIGHT ODDE, NO DRAWDOWN
	12/20/B3	. 6.67	07 <i>.</i> 177	~ 8	4	4	LIGHT BROWN, SMELLY LOOKS OILY
. • •	1/12/84	7.00	771.37	7.7	3.8	4	BROWN - GBAY ODOR
A.2.UnlahrM	2/12/84	7.18	771.19	7.5	3.7	4	LIGHT - BROWN COOR
· .	3/20/84	7,40	770.97	7,2	3,6	4	CLEAR TO YELLOW COOR
саналагин 60/	4/22/84	6.34	771.53	7.8	3.9	4	LIGHT BROWN Slight oddr
40101010	5/24/84	6.73	771.64	7,9	4.0	4	LIGHT BEDWN
	6/7/84	6.85	771.5Z	7.3	3.9	4	BROWNISH - YELLOW
	7/19/84	6.35	771,99	8.3	4.)	5	YELLOW
)	8/30/£4	7.18	19.17	7.5	3.3	4	YELLOW
	9/25/84	7.06	77/.31	7.6	3.3	4	YELLOW - JOOR
	10/9/84	6.92	771.45	7,7	3,9	4	TELLOW
	11/20/84	6.59	771.73		_		
:		1					
			-				

	ے ہور ہوت میں اور	111- 14		
-	1	1	lla	

Ur

DATA SHEET : WELL IOB LENGTH : 27.5/26. 78/76 76 DEEP

	DATA :	SHEET WEL	T 108 TEN	6TH: 27.5/Cb.	78/26.76		
	DATE	DEPTH TO. GW	GW Elevation	Volume H20 (H)	3 NOLUMES (GAL)	VOLUME REMOVED	COMMENT
	9/23/83	6,87	770.99			_	_
	10/20/03	6.51	771,35	~ ZI	10.5	15	
	11/16/83	7.06	770,80	~ 19.7	9.9	15	GRAY TINT, SLIGHT ODOR, NO DRAWODWN
	12/20/83	6.21	771.65	ZD. 6	10.3	12	CLEAR, SMELLY
·	1/12/84	6.50	771.36	20.3	10.1	12	YELLOW, STEALS OOOR
∧.2. U ut -	2/12/84	6.69	771.17	20,2	10.1	12	LIGHT BROWN CDDR
/ wet	3/20/84	6.90	770.96	19,9	9,9	12	JELLOW TO GREEN COOR
	4/22/84	6.35	771.51	Z0.4	10.2	12	LIGHT BEDWN Slight Odor
	5/24/84	6.22	771.64	2D, 5	10.3	12	LIGHT BROWN SLIGHT DOOR
•	6/7/84	6.35	771.51	20.4	10.2	. 12	YELLOW - CODE
	7/19/54	5.88	771.98	ZD.9	1D.4	12	YELLOW
)	8/30/54	6.68	171,18	20.1	10.l	12	LIGHT YELLON - DOOR
	9/25/84	6,56	771,30	Z0,Z	10.1	12	LIGHT VELLOW - DOR
	10/9/84	6,43	771.43	20,3	10. Z	12	LIGHT YELLOW - DEOR
	11/20/84	6.12	771.74				
						4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	
	:						
				、			
	-						
D							

ļ	<u></u>	DEPTH TO	GW ELEVATION	VOLUME HED	3 VOLUMES (GAL)	VOLUME	
	DATE	GW		(ft)	(GTL)	REMOVED	COMMENT
	9123/83	4.15	771.65				
1	10/20/83	•	771.58	5,5	2.8	5	FAIRLY CLEAR, SLIGHT
	11/16/83	4.86	770,94	4,95	2,5	5	ODOR, NO DRAWDIN BROWN-GRAY
	IZ/20/83		771, 73	5,75	2.9	4	SLIGHT ODOR
	1/12/B4	4.38	771.42	5.4	2.7	4	BROWN- GRAY SLIGHT ODOR
	2/12/54	•	771.32	5.3	2.7	4	LIGHT BEDWD SLI GHT DOCR
	3/20/84	4.78	771.02	5.D	2,5	4	YELLOW CDOR
4	4,/22/84	4.1D	סר ודר	5-7	Z.B	4	LIGHT GREY
	5/24/84	4.06	771.74	5.7	Z, 9	4	(LEAR
1	617184	4.21	771.59	5.6	Z. 8	4	WHITISH YELLOW
:	7/19/84	3.62	771.98	6 · D	3.0	4	LIGHT YELLOW
	8/30/84	4.42	771.38	5.4	2.7	4	DOLL HITE - COOR
•	9/25/84	4.37	77/,43	5,4	2,7	4	LIGHT YELLOW - DEER
	10/9/54	4,25	771,55	5,5	Z,B	4	YELLOW
	11/20/84	3,97	771.83	-			
÷							
•							
		- 					
						1	
		40 - 4 Vieta					

DATA S	HEET	WELL IIB	LENGTH: 27.5 /25.52/25.49
--------	------	----------	---------------------------

~	DATA S	SHEET ; WELL	1/B LEN	LTH: 27.5 /25.5	2/25.49	DEEP	
	DATE	DEPTH TO 6 W	6W ELEVATION	Volume HD (52)	3 VOLUMES (GAL)	Volume Removed	COMMENT
	9/23/83	4.72	771.69		-		
	10/20/83	4,81	771.60	~ Z	10.5	15	
	11/16/83	5,45	770.96	20,1	10.0	15	FAIRLY CLEAR, SLIGHT DOR, No DRANDOWN
	12/20/83	4,66	771.75	20.86	10.4	12	CLEAR, SLIGHT ODOR
	1/12/84	4,93	771.43	20.5	10.3	12	FAIRLY CLEAR SLILHT ODOR
Ahod atn U.S. A	2/12/84	5.07	771.34	20,4	10. Z	1Z	CLEAR SJUHT DOOR
, Wate	3/20/84	5,33	77]. OB	20.2	10,1	12	DULL CLEAR
	4/22/84	4.66	771.75	20.B	10.4	12	CLEAR
4 	5/24/8A	4.62	771.79	ZD. 9	10.4	12	CLEAR
	617/84	4.77	771.64	20.7	10.4	12	FAIRLY CLEAR
	7/19/84	4.36	772.05	21.1	1D.6	12	CLEAR
	3/3c/84	4.99	771.42	ZD.6	10.3	12	DULL CLEAR
	9/25/84	4,94	771.47	20,6	10.3	12	CLEAR
	10/9/84	4,31	771.60	ZD,7	10.3	12	CLEAR
	11 /20/84	4.54	771,37			-	
	:						

```	DATA 2	SHEET: 12 1 DEPTH TO	water wa	TH: 12/12.54/1 VOLUME 12D		•	BACKGROUND	
	DATE	62	ELEVATION	(5t)	(GAL)	REMOVED	COMMENT	
$\overline{)}$	9/23/83	6.57	771.49		-		_	
	10/20/83	6.76	771.30	5.3	2.7	5		
	11/16/83	7,32	770.74	4.B	2,4	5	CLEAR, NO ODOR NO DRAWDOWN	
	12/20/BZ	6.39	771.67	5.75	Z. 9	5	BROWN, SILTY	
4	1/12/84	6.73	771.33	5.4	Ζ,7	4	YELLOW - BEDWIN	
4.5.Uuls.>	2/12/84	6.93	771.13	5.2	2.6	4	BROWN SILTY	
	3/20/84	7.11	770.95	5,0	2.5	4	CLEAR	
-	4/22/84	6,53	771.48	5.6	<i>2.</i> B	4.	CIEAR	
	5/24/84	6.43	771.63	5.7	Z. B	4	CLEAR	
	6/7/84	6.54	771.5Z	-	-	_		
	7/19/84	6.03	772.03	—	-	-	_	
	8  3)/84	6.89	71.17		-	-	-	
	9/25/84	6.75	771, 31	-	~	-	_	
	10/9/84	6.67	771,39	-	-	-	_	
	11/20/84	6.25	771.31	-	-	-	-	
	-							
					-			
•						-		

	DATE	DEPTH 70 GW	6W ELEVATION	Volume H20 (ft)	'3 Volumes (GAL)	Volume Removed	COMMENT
7	9/23/83	5.93	771,44	-	-		
	10/20/83	6.10	771.27	∽z1	10.5	15	
	11/16/83	6.65	770.72	20.Z	10.1	15	CLEAR, No DOR NO ORAW DOWN
	/2/20/83	5,72	771,65	21.15	10.6	12	LIFAR, ODOPLESS
¢	1/12/84	6.03	771.34	20, 3	10.4	12	FAIRLY CLEAR
	2/12/84	6,25	771,12	ZD.6	10.3	12	CLEAR
	3/20/84	6.43	770,94	20.4	10.3	12	CLEAR
•	4/22/84	5.92	771.45	20,9	10.5	12	CLEAR
	5/24/84	5.75	771.62	21.1	10.5	12	CLEAR
	6/7/84	5.85	771.52			. —	
1	7/19/84	5.36	772.01	<u> </u>	-	-	_
•	8/30/84	6.21	771.16	-	-	-	<u> </u>
	9/25/24	6.06	771.31	-	-	-	-
	10/9/84	5,93	771,39	-	-	-	-
	11/20/84	5.57	771.30			_	_
	•				1		

	DATE	DEPIH TD Gw	6W Elevation	Volume to D (ft)	3 VOLUMES (GAL)	VOLUME REMOVED	COMMENT
)	9/22/63	4.55	772,20		_		
	10/20/83	4.73	1772.02	~5	2.5	5	
	11/16/83	5.26	771.49	4.5	Z.3	5	LIGHT BROWN TINT, NO ODDE, No DEAWDOWN
	12/20/83	4,62	1772.13	5.2	2.6	4	SLILHTLY CLOUDY ODORLESS
	1/12/84	4.31	v771.94	5.0	2,5	4	BROWN- LEAY
	z/12/84	4,92	j771, B3	4,5	2,4	4	LIGHT BROWN
	3/20/84	5.06	1771.69	4.7	2.4	4	YELLOW TO CLEAR
	4/22/34	4,52	1772.23	5.2	Z. 6	4	LIGHT GREY
	5/24/84	4,41	? 772.34	5.4	2.7	4	CLEAR
	6/7134	4.43	2772.27	5.3	. 2.6	4	LIGHT YELLOW
ì	7/19/84	4.03	1772.72	5.8	2.9	4	CLEAR
;	8/30/84	4.79	1771.96	5.0	Z.5	4	LIGHT YELLOW
	.9/25/84	4,54	772.21	5,3	2,6	4	LIGHT YELLOW
	10/9/84	4,65	1772.10	5.1	2,6	4	LIGNT YELLEN
	11/20/24	4.08	772.66	_	_	<u> </u>	_
;							
				-			
					-	-	

1- WIETU WINNUETIVE	
DATA SHEET ! WELL ISB	LENGTH: 27.5 /26 ET/2: 00

x		SHEET ! WELL		GTH: 27.5/26.8	·····	DEEP		
	DATE	DEPTH 70 6W	6W Elevation	Volume H20 (ft)		VOLUME REMOVED	COMMENT	
$\overline{)}$	9/23/83	5.20	772.20		_			
·	10/20/83	5.42	771.98	~Z1	10.5	15		
	11/11/83	5.89	771.51	20,9	10.5	15	CLEAR, NO ODOR NO DRAWDOWN	
	12/20/83	5,23	772.17	Z].64	10.8Z	/Z	CLEAR, CODRLESS	
	1/12/54-	5.46	94.17	21.4	7.0	12	CLEAR	
Alctinu.S.A	2/12/84	5.54	771.86	21,3	10.7	12	CLEAR	
	3/20/84	5,71	771.69	21.1	10.6	12	CLEAR	
i g	4/22/84	5,14	772.26	21.7	Ю.9	12	CLEAR	
	5/24/34	5,0Z	772.33	21.8	10.9	12	CLEAR	
	6/7/84	5.12	772.23	_		-	_	
) .	7/19/154	4.66	772.74	-	-	-	_	
	3/30/84	5.42	771.95	21.5 -	7.01	12	CLEPR	
	9/25/84	5,18	772.22	21.7	10,9	12	CLIFAR	
	10/9/54	5,27	51,277	Z1.6	10.5	12	CLEAR	
	11/20/24	4.72	772.68					
;								
		-		• •				
					-			

Contraction / Contraction SA

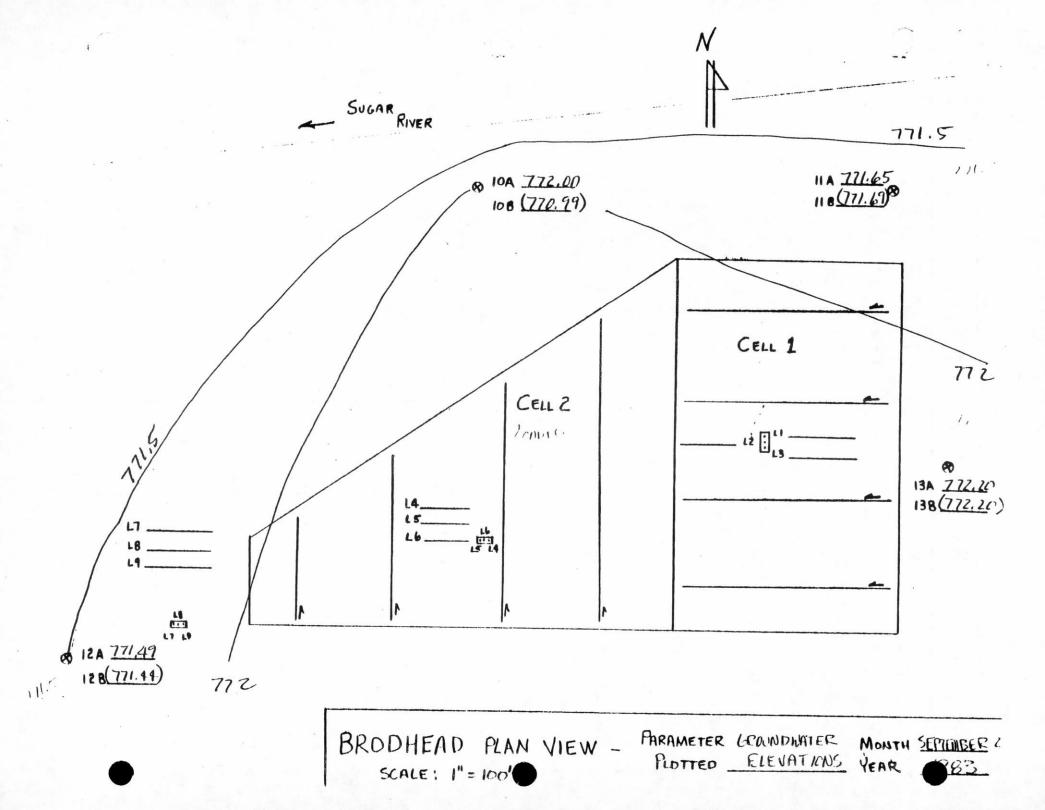
	DATA S	I		<u>стн: 7.29'/л</u>	· · · · · · · · · · · · · · · · · · ·	SHALLOW	
	DATE	DEPTH TO GW	GW ELEVATION	Volume H2D (ft)	3 VOLUMES (GAL)	VOLUME REMOVED	COMMENT
•	3/20/84	5.84	770.45	1.45	_		LIGHT BROWN
	4/22/84	5.12	רו .ודך	2.2	1.1	2	BROWN
	5/24/84	5.06	771.23	Z.Z	1.1	2	LIGHT BROWN
	6/7/84	5.23	771.06	Z.D	1.0	2	LIGHT BROWN
	7/17/84	4.83	771.46	Z.5	1.2	Z	Brown
	8/30/84	5.46	770.83	1.9	0,9	Z	Brown
	9/25/84	5.56	770.73	1,8	0.9	1	BROWN
	10/9/84	5,24	771.05	2.0	1.0	Z	LIGHT BROWN
1	1/20/34	5,08	771.20	_	-	-	-
		2					
1				-			
							·
		-					

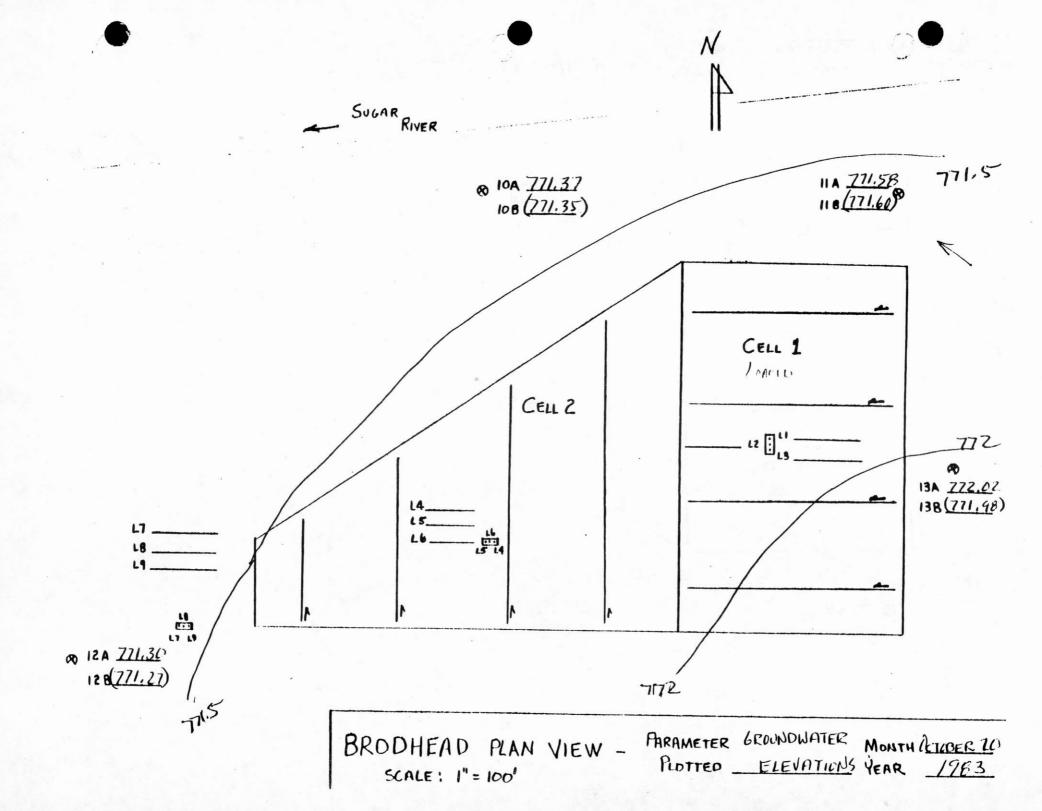
	DATE	Рергн то Бы	GW Elevation	Volume HD (St)	3 VOLUMES (GAL)	VOLUME REMOVED	COMMENT
)•	3/20/84	8.87	771.ZB	1.48	·	_	LIGHT BEOWN STEDNG OPOR
	4/22/84	8.35	771.80	2.0	1.0	Z	GEEY, STEDHE DOOR
	5/24/84	8,15	772.00	2.2	1: 1	Z	DULL CLEAR, OPOR
	6/7/84	8.Z4	771.91	Z.1	1.0	Z	WHITE - COOR
	7/17/84	7.72	·772,43	Z.6	1.3	Z	DULL - ODOR
	8/30/84	8.61	771.54	1.8	0,9	Z	Dull - COOR
	9/25/34	8,32	77/, 83	2,1	1,0	1:5	YELLAN - OCOR
	10/9/84	8.37	771.77	Z,D	1,0	1.5	FOLGY CLEAR - DOOR
	10/12/84	8.42	771,73	1,9	1,D	1,5	
	10/15/84	8.44	771.71	1.9	1.0	1.5	
	10/18/84	8.07	772.08	2.2	1.1	1,5	
	10 23/84	6.28	773.87	4,1	2 _: 0	3,5	
	10/26/84	7.67	772.43	2.7	1.3	2	
	10/30/24	7,54	772.61	z.8	1.4	2	
	11/1/84	7.07	773.08	3,3	1,6	2	-
	11/8/84	7,74	772.41	2.6	1.3	2	
	11/13/84	7.76	772.39	-	-	-	
	11/16/34	7,89	772.26	2,5	1.2	Z	
•	11/20/34	7.99	772.16	2,4	1-2	2	

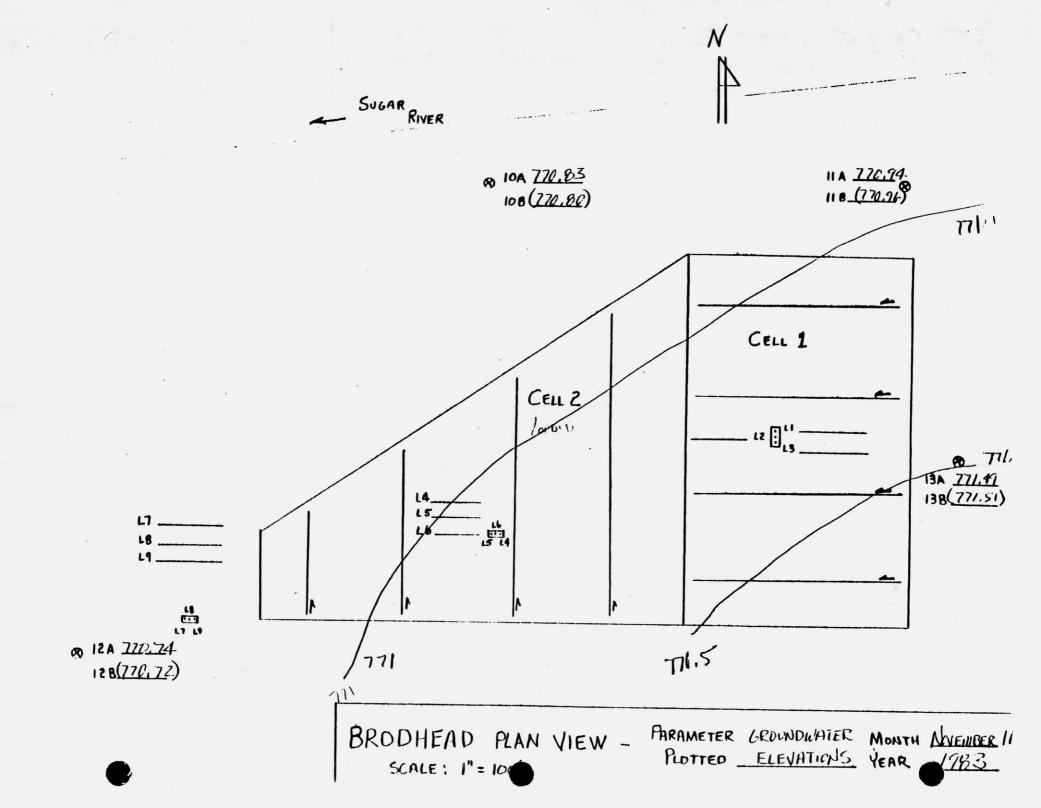
		SHEET! WE	I GW	GTH 12.06	7.1		LOW, SE BERM
	DATE	РЕРТН ТО 6W	ELEVATION	Volume H2D (57)	3 VOLUMES (GAL)	VOLUME REMOVED	COMMENT
)	3/20/84	10.06	771.99	Z,D	-	_	BROWN
	4/22/84	9.60	772.45	2,5	1.2	2	GREEN, GREY, DOOR
	5/74/84	9.43	772.62	Z.6	1.3	Z	LIGHT BROWN SLIGHT
	617184	9.49	77.2.56	2.6	/.3	Z	BEOWN, SLIGHT DOD
	7/ <i>191</i> 84	8,95	773.10	3.1	1.6	Z	Brown
Alade In U.S.A	8/30/84	- 9.78	772.27	23	1.2	Z	DULL
< l>	9/25/84	9,55	772.50	2,5	1.Z	Z	PARK GREEN
<b>0</b> ,	10/9/82	9,73	772.32	2,3	1,2	Z	Beown
	10/12/84	9,75	772,30	2,3	1.2	2	
	10/15/84	9.75	772,30	2.3	1.2	1.5	
	10/18/84	9,42	772.63	2.7	1,3	2	
}	10/23/24	7.81	774.24	4,3	2,1	3	
	10/26/82	B.71	773.34	3,4	1.7	2	
	10/20/34	B.79	773.Zb	3.3	1.6	2	
	11/1)84-	8,12	773,93	4.0	2.0	2,5	
	11/8/84	8.76	773, 29	3.3	1.7	2	
	11/20/84	9.00	773.05	-	_	-	
	•						
					. '		
-	<b>4</b>						

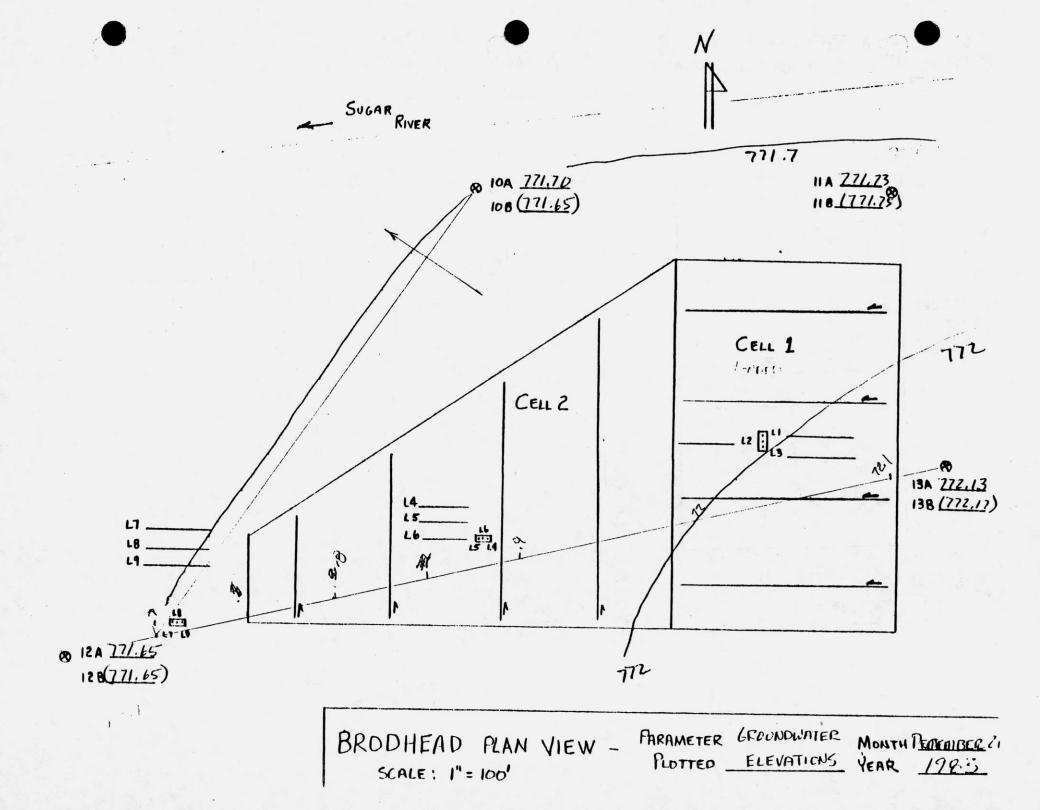
CATE	DEPTH TO GW	GW Eleverick	VELUME HZE (GE)	E VILUMES (612)	VOLUME RENEVED	COMMELT
10/12/54	8.59	772.09	3,4	1.7	6	BLACK
10/15/84	8.16	772.02	3.4	1.7	Z	<b>.</b>
Ø18/84	7,78	772,40	4,2	2.1	3	
10/23/24	6.10	774.05	5,4	2.7	4	
10/26/34	7,25	772,93	4,3	2.1	3	-
	7.25	772.93	4.3	2,1	3	
11/1/84	6.66	773.52	4.9	2.4	3	
11/8/84	7,30	772.88	4,2	2,1	3	•
11/12/84	7,39	772,30	~			
:1/16/84	7,45	772.73	4,1	2.0	Z,5	
1 /20/24	7.51	772.67	4.0	Z.0	2,5	
			•			
			·			
						•

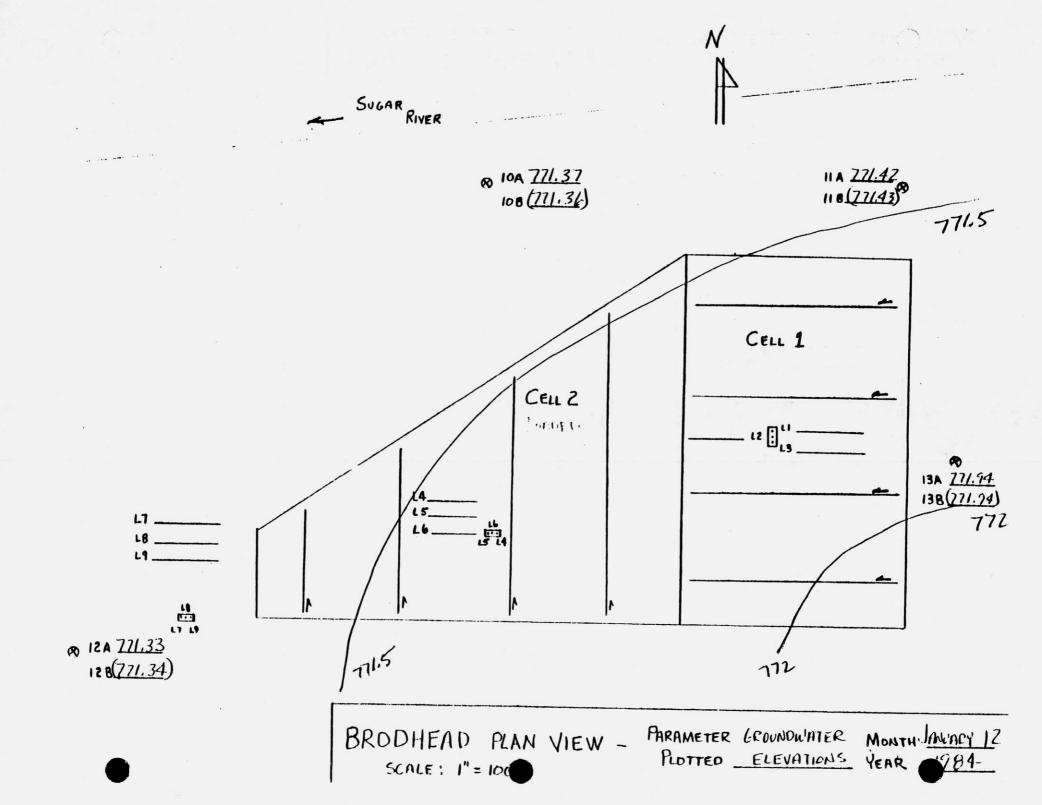
The state and surfice a source

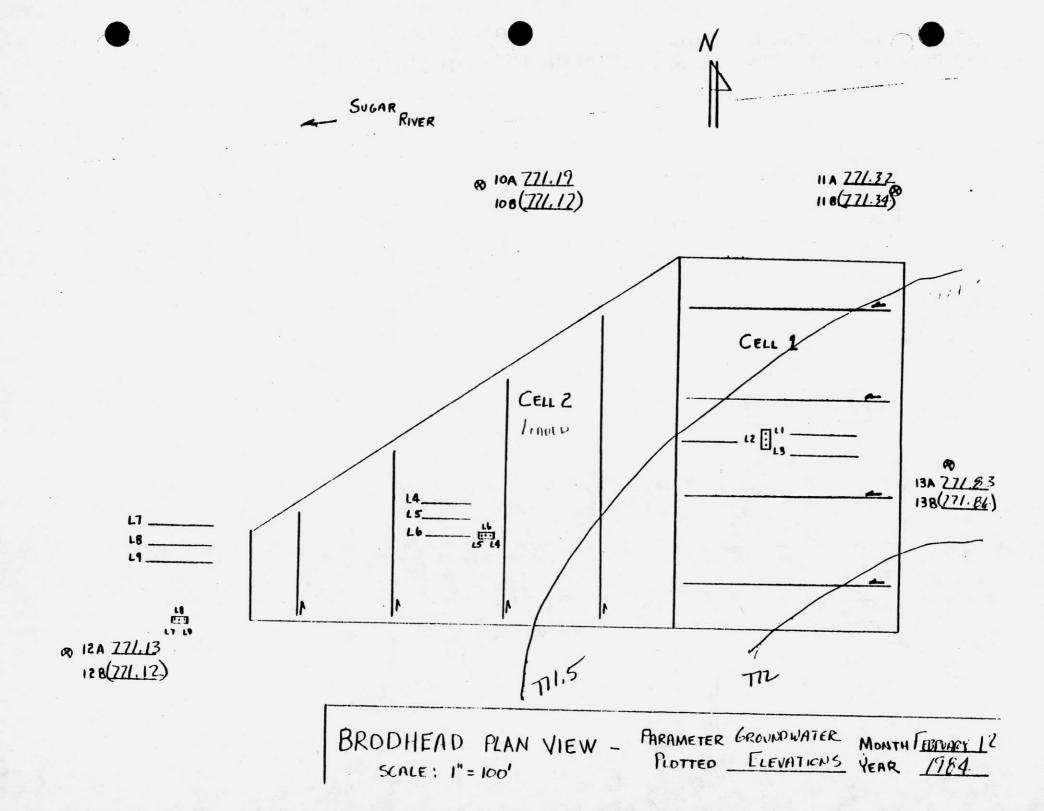


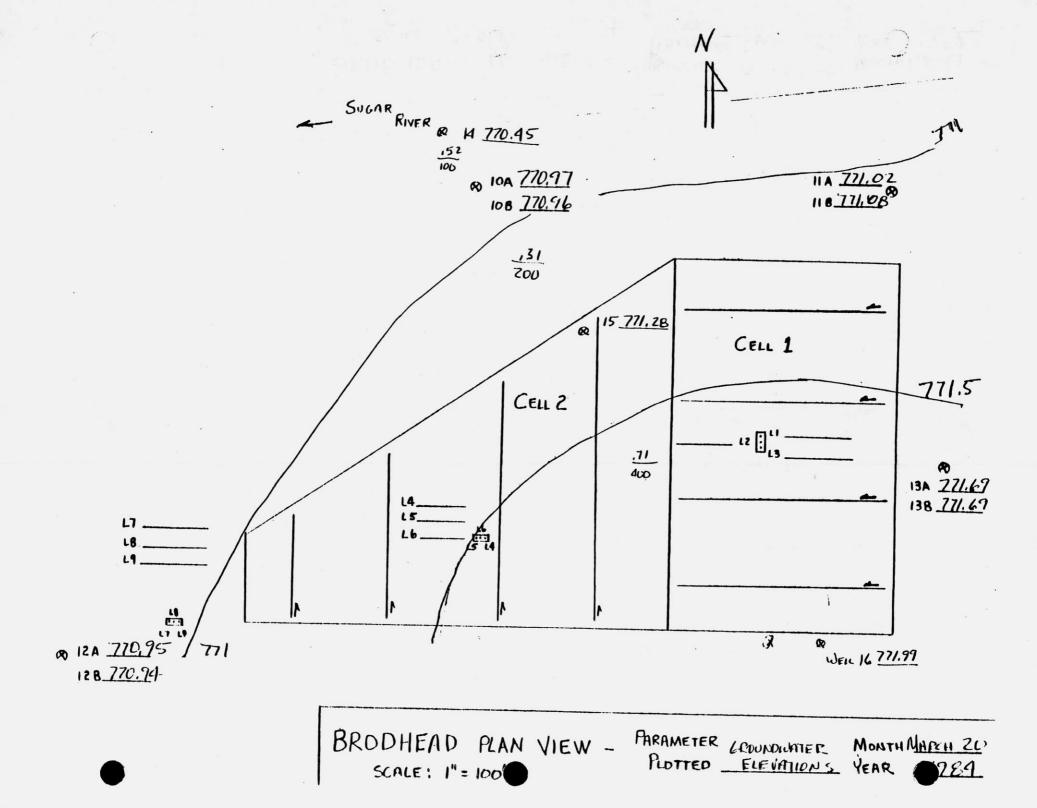


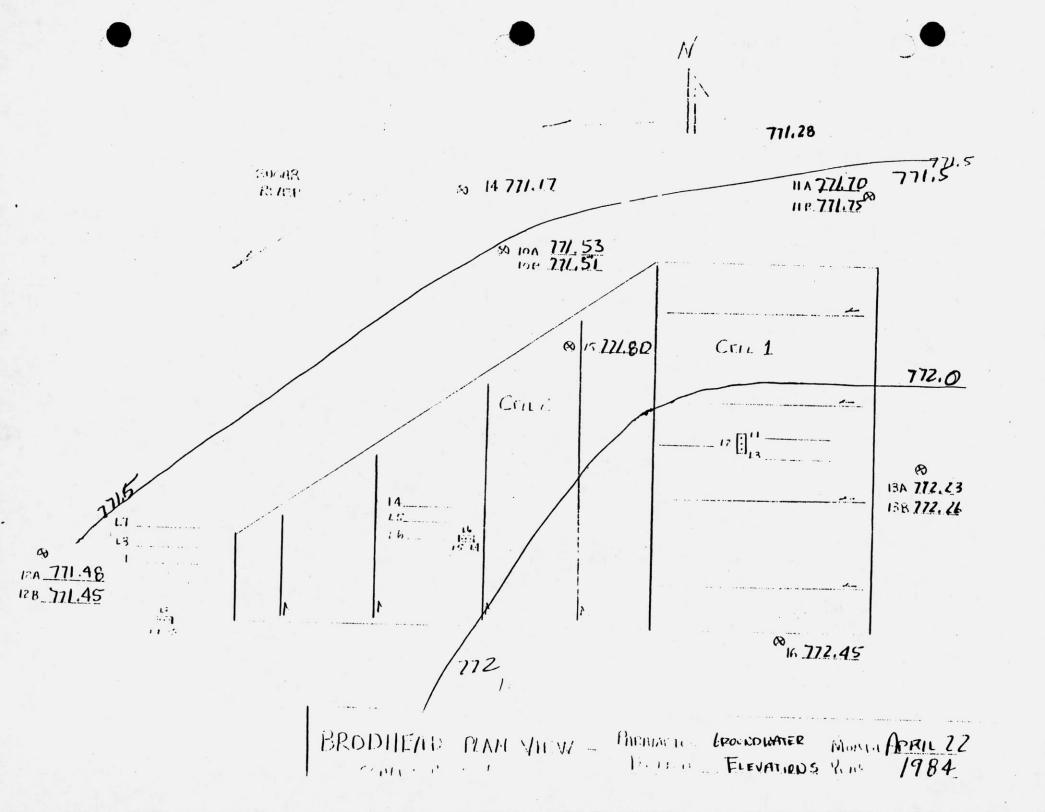


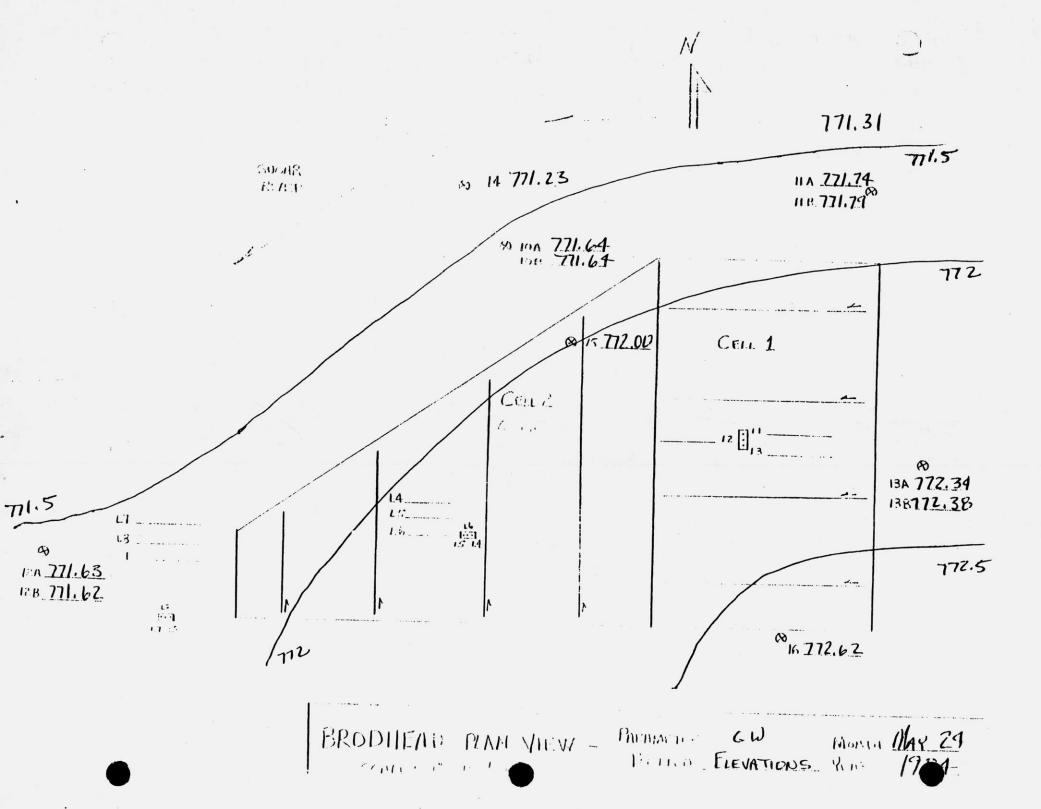


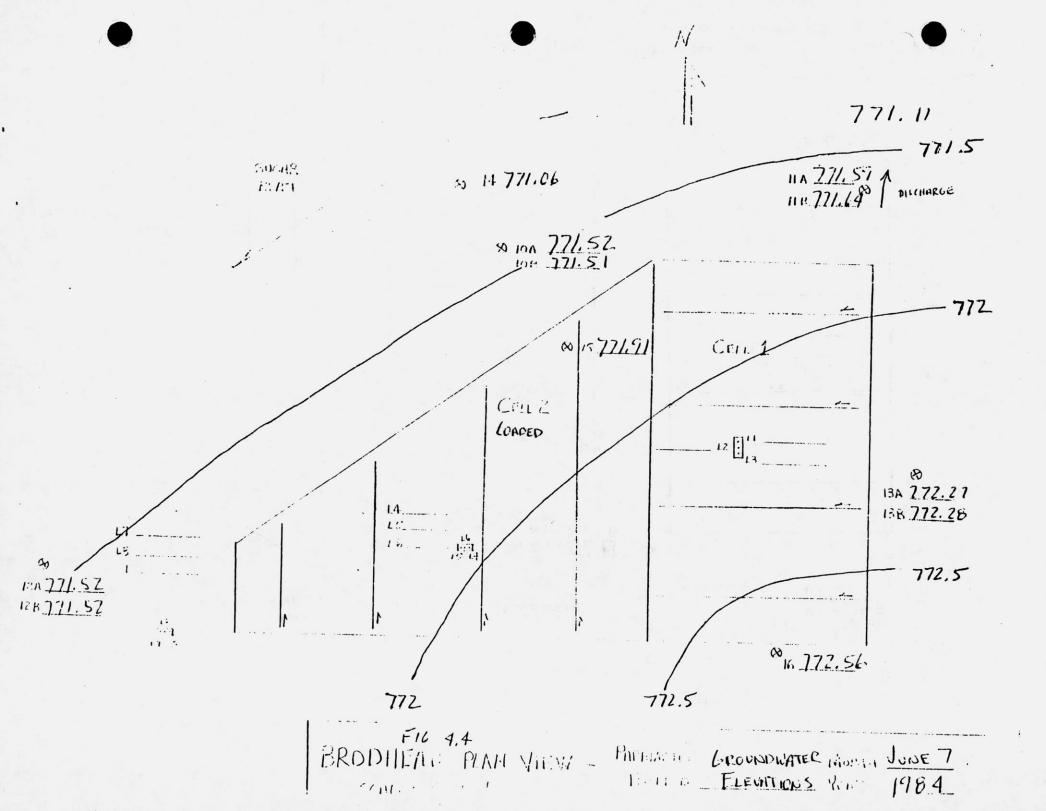


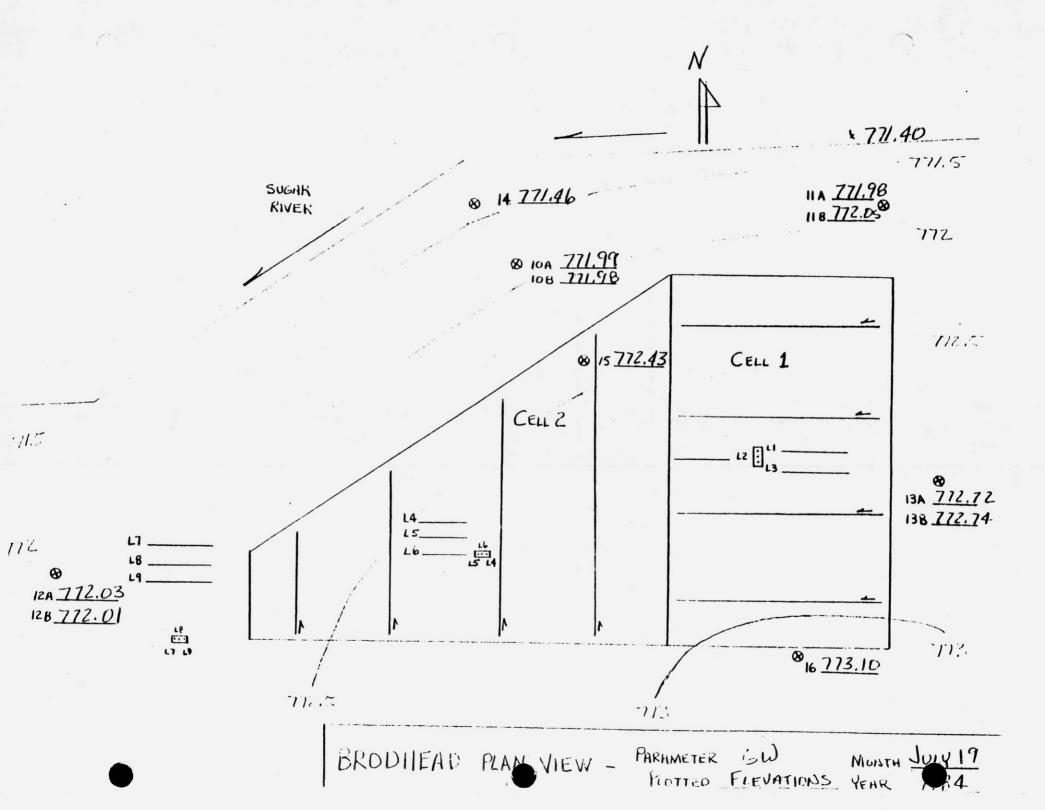


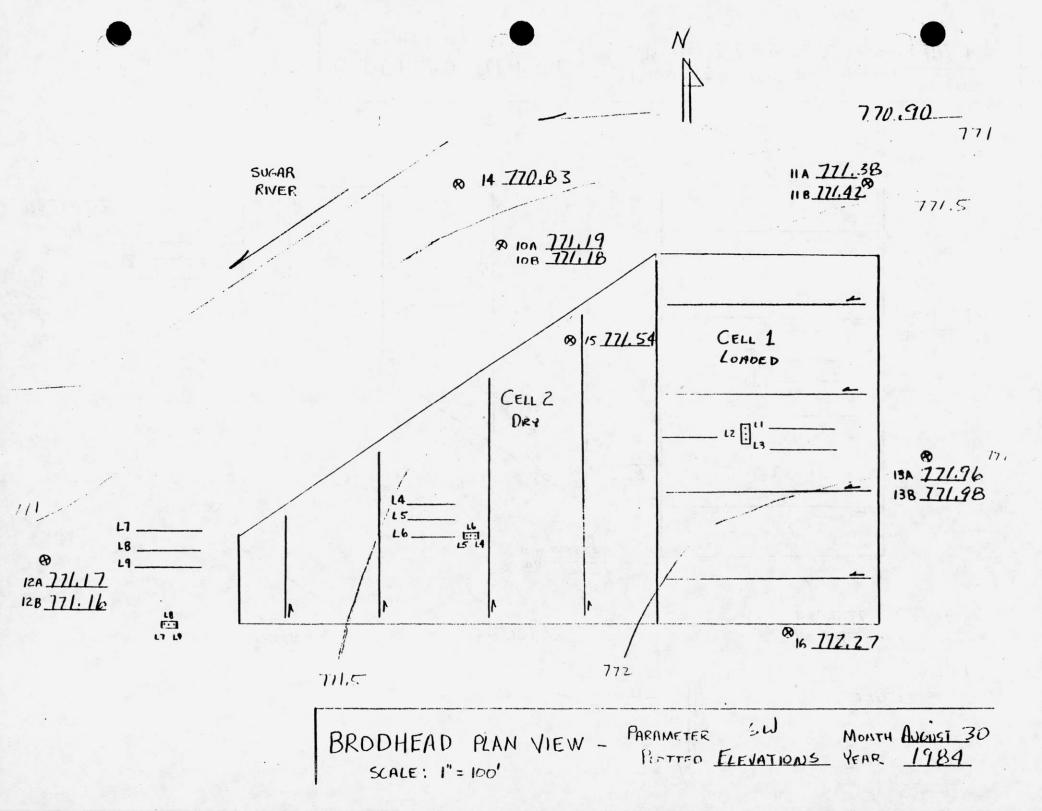


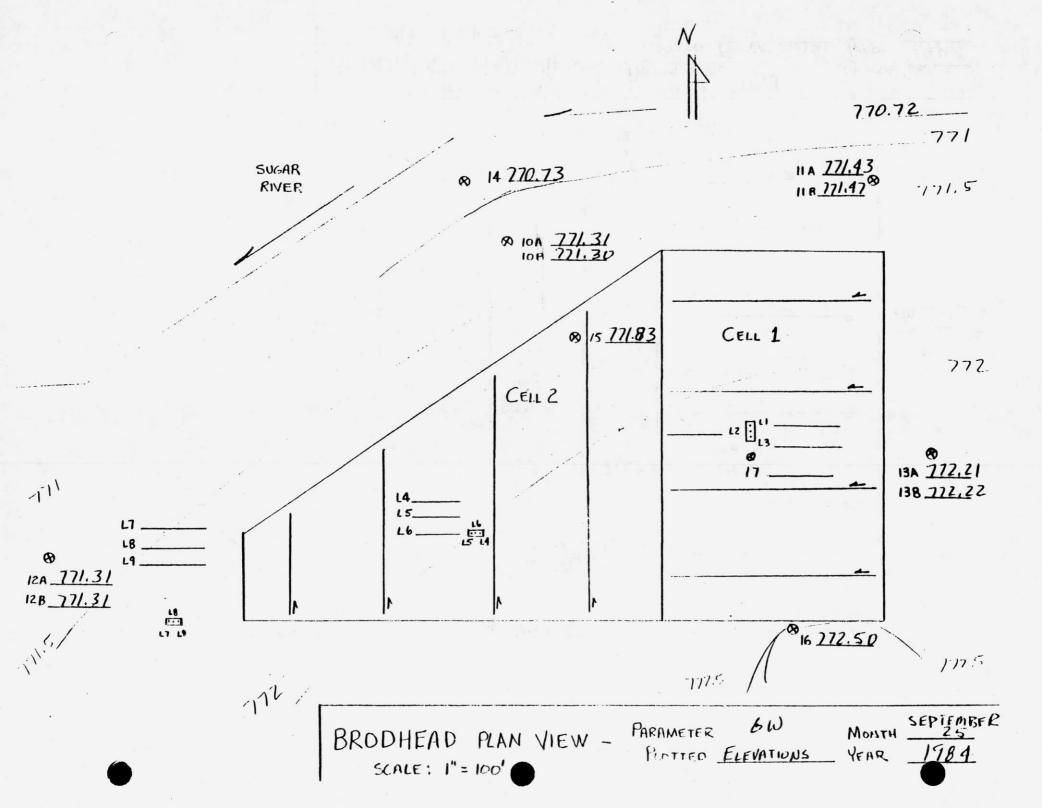


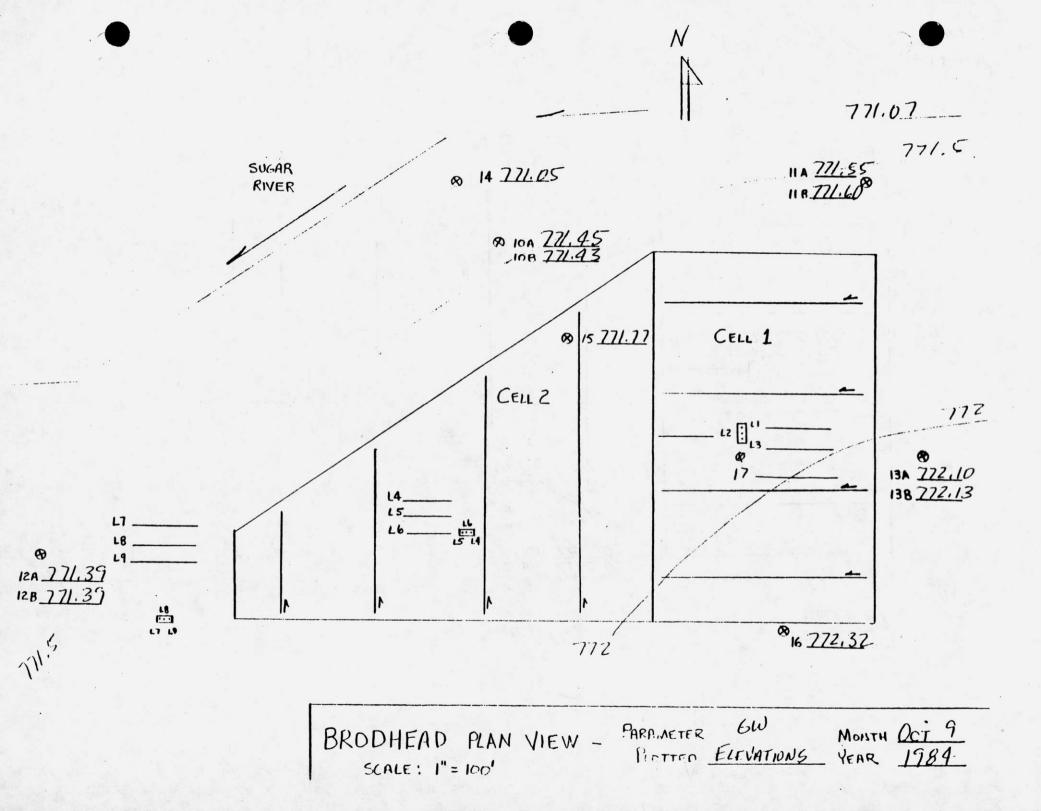


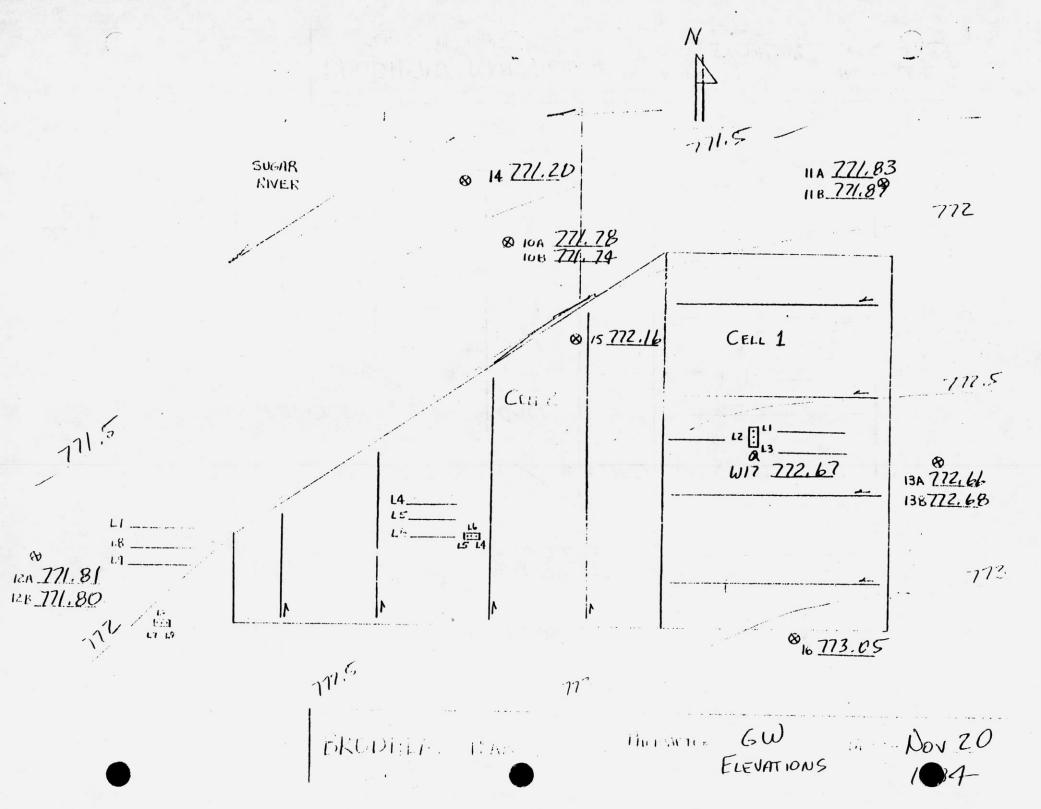












Ì APPENDIX F PAPA4ASTER (m) Hoom HUSA. BRODHEAD : GROUNDWATER CHEMISTRY DATA

	DRDD HE DATA	HD LUMA		_	METERS	(mg/]	•	L IDA	- SHAL	Low	
•	DATE	DISSOLVED BODS	DISSOLVED COD	(100°C))	DISSOLVED TKN	NH3-N	NOZ-N+		(LAB) PH	OTHER	
	<i>10/20/</i> 83	1.8	14,	-	15,	14.	<0.1	230.	8.5	-	
	11/16/83	9.8	18.	672	13.	17.	< 1	230	-		
	12/20/83	6.5	17.	922	14.	4.	Z 1.D	350	-	-	
	12/20/83	TOTAL <4	TOTAL 19	TOTAL SS 2020	TOTAL 14, -	-	-	340	_	-	
,	1/12/84	3.4	17	920	13	12	0,1	340	_	_	
Mada <b>h U.S.A</b>	2/12/84	43	16	888	12	12	0.1	310	_	_	
	3/20/84	23	15	960	11	11.0	20.1	350	6.9	-	
Non anten co	4/22/54	>13	14	1010	13	13	D, 1	360	7.1	-	
aran co	5/24/84	4,9	14	1020	15	15	0.2	380	7.0	7.9 (hb = 17	
	6/7/84	10	19	1000	14	14	D.1 -	Z90	6.9		
)	7/19/84	12	17	1220	7	16	<1.D	530	6.3	-	
1	8/30/24	6,5	20	1230	13	12	20.5	500	6.7	-	Ļ
	9/25/54	8.2	21	1160	10	10	<1.0	460	6.5	—	
	10/9/84	n	13	1180	10	10.	21.0	490	6.5	_	

<b>.</b>	DATA	SHEET	······		meters	(mg/_I)	WE	10A-	SHALLOW	
	DATE	ALKALINITY	TOTAL	504-	Ga ²⁺	Na+	Mg2+	К+	OTHER	•
	10/ZD/83	∽ 250 [≭]	<0.02	16.	5Z	130	30	_	_	1
	11/16/83	-	-	-	-	. –	-	_	_	2
	12/20/83	-	-	-	_	-	-	· -	—	4 · · ·
	1/12/84	_	<b>—</b>	-		-	_	_		
	2/12/84	-	_	-	-	_	_	_	_	3
►.C.U.n.	3/20/84	-	-	-	-	_	_	-	_	
1 Parts	4/22/54	-	~	-	~	-	-	-	_	
 	5/24/84	352	0.0Z	D.B *	45	270	23	30	-	
u Marana'ru	617184	-	-	~	-	-	_	_	~	
	7/19/84	-	-	-	—	_	-		_	
	8/30/774	406	< 0.02	<1.0	53	360	27	26	-	
	•									
•										<b>`</b> .
									•	
							-			
	* 1000									

`	DATA	SHEET		Paran	neters	( ^{mg} /_			- PEEI	2
	DATE	DISSOLVED BODSX	COD	TDS	TKNX	NH3-NY	$NC_2 - N + NC_3 - N$	CTX	PH (LAB)	CTHER
	10/20/83	2.5	9	-	6.0	5.1	< 0.1	350	8.4	_
	11/16/83	11.	27.	1390	10.	9.3	0.1	520	-	_
	12/20/83	12	24	1570	9.6	9.7	ل. ا	550	<b>—</b>	-
	)/12/84	4.9	25	1538	7.5	6.7	0.1	560	-	-
	2/12/84	23	21	1670	7.D	7.0	20,1	650	-	-
MadalnU.S.A	3/20/84-	5.3	18	1550	5,4	5,2	D,1	580	6.8	_
	4/22/34	718	13	1650	7,4	7.2	D,1	640	6,9	-
Allen Inv	5/24/84	8,9	19	1660	10	iO	0.1	690	6.8	7.5 LAEpH
•	6/7/84	77	23	0771	15	15	20.1	740	6.7	_
	7/19/84	21	25	1510	17	16	L1.0	750	6.7	-
	6/30/84	<12	33	1980	17	17	<0,5	790	6.6	-
)	9/25/84	21	21	1920	24	24	<1.0	סור	\$ . <del>5</del>	
	10/7/54	3.7	34	1730	26	26	<1.0	650	6.7	
						-				
		•								
	:									

	DR.UN	TA SHE	ET	PARAM	ETERS	(m=/_{)	WE	L IDB	- DEEP
	DATE	ALKALINITY	P TOTAL	504 ²⁻	Ca ²⁺	[ Na+	M-2+	κ+	OTHER
7	10/20/83	~ 220	<0.0Z	5,9	59	230	24	-	-
	11/16/83	-	-	-	_	-	-	-	-
	12/20/83	-	-	_	-	-	-	-	-
	1/12/84	~	-	-	-	-	-	-	
	z/12/84	-		-	_	-	_	-	-
A. 2-1 n IU. S. A	3/20/84	-	· ~	~	-	-	-	-	-
	4/22/84	-	-	_	-	_	-		-
	5/24/84	436	LD.02	1.D	80	440	37	30	-
	617184	-	-		-	-	-	-	-
	7/19/84	-	_	_		-	_	-	_
,	8/30/84	608	0.02	10	73	570	39	47	~
)								-	
					•				
-									-
-	•								
			-						
						-			
									· ·

Dat	A SHEE	T	Paran		(mg/2,			- SHALL	-ow
DATE	Dissolved, BODs	diss Cod	DISS TDS X	DISS TKN X	DISS NHJ-NX	$NC_2 - N + NC_3 - N$	CIT	PH (LAR)	OTHER
10/20/33	1.8	19	-	4.0	3.2	< 0.1	96.	8.5	-
11/16/83	<b>26</b>	13	312	2.4	Z. D	<0.1	49	-	-
12/20/83	<4	15	494	3.8	3.6	< D. 1	120		-
12/20/83	TOTAL 1B	TCTAL 26	TOTAL SS 1980	707 AL 5, 2	-	-	ΠQ	-	
1/12/84	∠3	30	64D	4.D	3,4	<0.1	160	-	_
2/12/84	23	13	514	3, Z	3.0	<0.1	DII		
3/20/84	∠3	19	476	2.6	2.1	0.1	100	6.7	
4/22/84	<3	ZD	493	3,4	2,9	0.1	110	7.D	-
5/24/84	23	22	526	3,5	3.0	0,1	110	6.9	В, I (АВр1
6/7/84	<3	25	57 <del>9</del>	3.3	Z.9	0.1	140	6.5	_
7/19/84	4.3	23	552	2.9	2,3	<0.1	130	6.9	
8 30/84	26	26	948	2.2	1.5	0.1	230	6.5	
9/25/84	4,9	30	833	3,5	2.9	<0,1	260	6.5	
10/9/84	7.7	42	640	Ξe	2,8	<0.1	170	6.7	_
1			•						
	•								
	DATE 10/20/33 11/16/83 12/20/83 12/20/83 1/12/84 2/12/84 3/20/84 4/22/84 5/24/84 6/7/84 5/24/84 6/7/84 5/24/84 8/30/84 9/25/84	DATE       Dissoured, BOD; $10/20/33$ $1.8$ $11/16/83$ $1.6$ $11/16/83$ $46$ $11/16/83$ $46$ $12/20/83$ $41$ $12/20/83$ $41$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/12/84$ $43$ $1/19/84$ $46$ $1/19/84$ $49$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	DATEDissoured BODsDiss CODDiss TDS TDS X $10/20/33$ 1.819- $11/16/83$ 1.819- $11/16/83$ <6	DATEDissalued BOD,Diss CODDiss TDS, TDS, TKN, X10/20/831.819-4.011/16/83<6	DATE         Diss alled         Diss         Dis         Dis         Diss	Drift         Diss         Diss         Diss         Diss         Diss         Diss         Diss         Diss $ND_2 - N$ DATE         BODs         COD         TDS         TKN         NH2-N         ND2-N         ND2-N           10/20/83         1.8         19         -         4.0         3.2         <0.1	DATE         Disc         Diss         Dis         Diss         Diss <th< td=""><td>$\begin{array}{c c c c c c c c c c c c c c c c c c c$</td></th<>	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

`	DAT	a Shee	Τ	Paran	netees	(me/s)	WEL	/H -	StIALLOW
	DATE	ALKALINITY	TOTAL	5042-	G2+	1.b+	Aig 2+	К <b>†</b>	OTHER
)	10/20/83	- 191	0.72	43.	47	ଚା	20-		
	11/16/53	-	-	-	-	-	-	-	~
	12/20/83	-	~	-	-	-	-	-	
	1/12/84	-	~	-	-	-	-	-	
	2/12/84	-	_	.—	-	-	-	-	
	3/20/84			-	-	-	-	-	-
	4/22/84	-	-	~	-	-	-	_	-
	5/24/64	270	D <b>.</b> 39	24	57	95	24	5	-
	6/7/24	-	-	-	-	-	-	-	-
	7/19/84	-		_	-		-		
`	8   30/84	358	1.07	5,Z	74	ZID	30	6	
)	1 r								
	•				4 4 4 4				
	· ·								
•	•			_				N	
			-						
						1			
					Ĩ				

	DAT	DISSOLVED		0155	DISS	DISS NHZ-NY	NO2-13+	LIB	- L	
-	DATE	BODS		TDS	A			<b>`</b>		
)	10/20/53		<5		D.4	0.2	0,1	94. ⁻	e. 2	
	11/16/83	¢ ک	5	326	0.Z	0. ]	0.1	29	-	-
	12/20/83	< 4	<5	308	0.Z	0.1	<i>∠0.</i> 1	12		-
	1/12/84		. <5	316	0.3	D.1	<b>≺</b> D.1	12		-
۲.	z/1z/84	23	<5	304	0.2	0.1	20.1	12		
A.S.Unitera	3/20/84	<3	25	304	0.2	0.2	0,1	15	7.2	_
	4/22/84	23	×5	30 B	0.Z	0,1	0,1	13	7.2	
	5/24184	<3	45	336	D. Z	0.1	D, 1	17	7.3	8.2 LABAH
	6/7/34	18	62	75 2	0.2	D.1	20.1	190	7.D	
	7/19/84	15	17	1140	1.2	0.5	0.1	390	6.9	<b>.</b>
``	8 30/84	×12	م ا	324	0.6	0.5	0.1	46	6,9	<b>—</b>
)	9/5/84	<4	<5	358	0.6	0.5	20,1	34	7,0	 - •
	10/9/84	23	B.	432	0,8	0.6	<i>20,1</i>	75	7.0	
•									-	1 - 1
					• •					•
				4 4 4						
				• • • • • •	) 1					
				1						
				:	- - - - - -				-	
					k		-			
•				•	:					
	· .			:						
				•	•	•	- - - - -	•		• • •

`	DA	TA SHE		Parai	neters	(mg/_()	WEL	<u>л // В</u> .	
	DATE		TOTAL	504 ²⁻	G ²⁺	Na ⁺	Nig 2+	К†	STHER
)	10/20/83	w21Z	0.07	23.	70	47	35	-	
	11/16/83	-	-	-	-	-	-	-	-
	12/20/83	-	-	-	-	-	-	-	—
	1/12/84	~	<u> </u>	~	-	~	~	-	-
	2/12/84	-	-		-	_	-	-	-
	3/20/84	-	~	-	-	~	-	-	-
	4/22/84	-	~	~	<u> </u>	-	_	_	_
	5/24/84	222	<0.02	31	61	5	27	Z	_
	6/7/84	-	~	-	-	-	_	-	~
	7/19/84	-	-	-	-	-	-	-	
	<i>2 30</i> /84	252	<0.0Z	26	66	34	24	6	_
*	•								
			-						
		-							
						-			
•									
								-	

1.6 2.3 4.9 2.3 2.3 2.3 2.3 2.3	0155 20 7 9 14 8 25 25 25	DISS TDSX 356 470 378 378 346	DISS TKN _X 0. Z 0. Z 0. 7 0. 3 2.0	NHz-N <0.1 <0.1 0.1 20.1	122-N+ 1303-NX 12.0 8.3 8.7 7.9	CI-K 6524 29 33 29		
23 4,9 23 23	9 14 8 25	470 378 346	0.Z 0.7 D <b>.3</b>	<0,1 0,1 20,1	12.0 8.3 8.7	Z9 33	B.4 - -	-
4,9 23 23 23	14 8 25	470 378 346	0.7 D <b>.3</b>	0,1 20,1	6,7	33	-	-
23 23 23	8 25	378 346	D.3	20,1			-	-
23 23	25	346			7.9	29	-	
23			2.0	/			-	
	25			∠D.1	7.6	Z9 ·	_	_
< 3		343	0.Z	0.1	7,3	25	6.5	- ·
	<5	34 D	0.Z	0.1	7.6	2.3	7.1	_
23	25	372	C. Z	0,1	9,7	23	7.0	<i>G.3 lae</i> pH
~		~		-	<b>—</b>	-	-	-
-	· -	_	_	-	-	-	-	-
-	<b></b> .	-	_	-	-		_	
								4 • •
			·					•
								* : :
			-					
						•		
-								
				-			-	
								- - - -

•

DATE	ALKALINIT	TOTAL P	5042-	G ²⁺	Nat	Mg ²⁺	Κ <b>†</b> [	DTHER
10/20/53		0.02	17.	65	14	30	-	·
11/16/83		-	-	-	-	-	-	-
12/20/83	-	-	_	-	-	-	-	. —
1/12/84	-	~		-	-	-	_	
2/12/84	-		-	-	-	-	-	
3/20/84	-	-	-	-	-	_	-	
4/22/84	-	-	~	-	-	-	-	-
5/24/84	214	20.02	ZD	61	12	ZB	Z	-
617/84	-	-	-	-		-	-	-
7/19/84		-	-	-	-	-	-	
8/30/34		-		-	-	-		_
								f i i -
							-	
							•	
								· ·
•								
		:						

	Дата	SHEE	10-ri , 10-	PARA		(m;/,)	WELL 0155- ND2-N+,	128-	DEEP, E	36
	DATE	Dissolved BOD5	0135 COD	TDS	CISS	DISS NH3-N	NO2-N+ NO3-N	-۱۷	CH (LAB)	DTHER
)	10/20/83	1.6	< 5	<b>—</b>	0.2	<0.1	10.1	39.	5.7	
	11/16/83	< 3	6	330	<0.Z	<0.1	8.4	40	-	
	12/20/83	23	L5	396	0.Z	<0.1	13.Z	37	-	
	1/12/84 SPLIT SAMPLE	23 23	76	432 438	0.2 20,2	<0.1 <0.1	13.6 13.2	36 36		-
1.5.A	2/12/8 <b>4</b>	23	25	414	. 20,2	20.1	12.7	37	-	
A.2.11 n1 - 1-1-1	3/20/84	23	25	368	0.2	0.1	9.4	39	7.4	-
	4/22/84	23	<5	35D	< 0, 2	0,1	8,3	33	7.5	
1.1.5.7.7.	5/24/84	23	25	335	<0.2	0.1	6.9	33	7.6	5.3 LABPH
1. 1	6/7/24	-	-	-	-	~	-	-		-
	7/19/84	-	-	-			-	-	-	~
)	8/30/54	-	-	-	-	-	-	-		-
	•					-				
•										
		: : !								
			1							
		-								
	•									
•			• 1 -							
		4 	: : 4				•	1 1		•
									•	

	DA1	iu Winin Ta Shee	T	Param	ICTERS	(ب_رومه)	WEL	L 12B-1	Deep, 35	
X.	DATE	REKA LINTY	TOTAL	5042-	Ca ²⁺	Na ⁺	Nig 2+	κ <b>+</b>	dther	
	10/20/83	~214	0.04	23,	56	24	32	-		
	11/16/83	_	-	-	-	-	-	_	-	
·	12/20/83	-	_	-	-	-	-	-	~	
	1/12/84	-	-		-	-	~	~	—	
	Z/12/84	-	_	-		-	-	-	_	
Alada <b>InU.S</b> .A	3/20/84	_	-	-	-	-			-	
/ Mad	4/12/84	-	-	-	-	-	_	~	-	
/ Barro	5/24/84	166	0,02	24	42	21	Z6	2	-	
a + Ptharster	617/84	-	-		-	-		_		
	7/19/84	-	-	-	-	-	-		_	
•	8/30/84	-	-	_	-		-	-		

	t Sheet	,	PARAN	NETERS	(m;/2)	- piss	L 13A-		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
ATE	DISSOLVED. BOD5	Diss Cod	TDSX	DISS	DISS NH3-NX	Noz-N+	CIX	(LAE) Ptl	OTHER
120/83	0.3	20		1.D	0.1	1.6	Z,4	8.6	—
/16/83	< 3	18	ZIZ	0.6	D.1	<0.1	2.8	_	-
120/63	<3	17	Z16	0.6	0.1	20.1	2.1	-	-
2/20/83	TOTAL < 3	TOTAL 18	total 55 334	0.9		-	2.0	~	
/12/84	< 3	17	222	0.6	0.1	0.1	1.5	-	-
/12/84	23	16	236	0.6	0.2	<i>40.1</i>	1.6	-	-
/20/84	∠3	21	286	0.7	0.2	0,2	2.3	6.7	-
12 /84	23	22	320	0,3	0.2	D,1	3.D	7,1	
124/84	<b>43</b>	25	384	0.9	0.3	0.2	4,3	7.0	E.DLAZ;H
,/7/54	23	31	438	.1.D	D.Z	0.2	5,5	7.D	-
/19/84	< 3	36	476	1.2	0,2	0.1	6,1	7,3	
3 /30/84	13	33	432	1.2	0.3	1.3	6,4	6.9	
25/84	23	20	280	0,3	0.Z	0.7			
c/9/84	イス	21	254	0,3	0.2	0.1	6.1	7.1	
			• •						
		1		•					
	20/63 111/83 120/63 120/63 120/63 120/63 120/64 120/84 120/84 120/84 121/84 121/84 121/84 121/84 121/84 130/84 130/84 125/84	20/63  0, 6  11/83  < 3  20/63  < 3  20/63  < 3  20/63  < 3  20/63  < 3  11/84  < 3  12/84  < 3  20/84  < 3  20/84  < 3  20/84  < 3  21/84  < 3  24/84  < 3  24/84  < 3  21/84  < 3  25/84  < 3  25/84  < 3	20/83 $0, 9$ $20$ $116/83$ $< 3$ $18$ $120/63$ $< 3$ $17$ $120/63$ $< 3$ $17$ $120/63$ $< 3$ $17$ $120/63$ $< 3$ $17$ $120/63$ $< 3$ $17$ $120/83$ $< 3$ $17$ $120/83$ $< 3$ $17$ $12/84$ $< 3$ $17$ $12/84$ $< 3$ $16$ $120/84$ $< 3$ $21$ $12/84$ $< 3$ $22$ $121/84$ $< 3$ $22$ $121/84$ $< 3$ $22$ $121/84$ $< 3$ $22$ $121/84$ $< 3$ $25$ $121/84$ $< 3$ $31$ $121/84$ $< 3$ $35$ $121/84$ $< 3$ $35$ $17/64$ $< 3$ $36$ $130/84$ $4 3$ $33$ $125/64$ $23$ $20$	20/83 $0.9$ $20$ $ 116/83$ $< 3$ $18$ $212$ $120/63$ $< 3$ $17$ $216$ $120/63$ $< 3$ $17$ $216$ $120/63$ $< 3$ $17$ $216$ $120/63$ $< 3$ $17$ $216$ $120/63$ $< 3$ $17$ $222$ $12/84$ $< 3$ $17$ $222$ $12/84$ $< 3$ $17$ $222$ $12/84$ $< 3$ $16$ $236$ $120/84$ $< 3$ $21$ $286$ $120/84$ $< 3$ $22$ $320$ $121/84$ $< 3$ $25$ $364$ $121/84$ $< 3$ $25$ $364$ $121/84$ $< 3$ $25$ $364$ $17/64$ $< 3$ $31$ $425$ $19/84$ $< 3$ $36$ $476$ $130/84$ $43$ $36$ $432$ $125/64$ $23$ $20$ $280$	120/63 $0.6$ $20$ $ 1.0$ $116/83$ $< 3$ $18$ $212$ $0.6$ $116/83$ $< 3$ $17$ $216$ $0.6$ $120/63$ $< 3$ $17$ $216$ $0.6$ $120/63$ $< 3$ $17$ $216$ $0.6$ $120/63$ $< 3$ $17$ $216$ $0.6$ $120/63$ $< 3$ $17$ $222$ $0.6$ $120/83$ $< 3$ $17$ $222$ $0.6$ $111/84$ $< 3$ $17$ $222$ $0.6$ $112/84$ $< 3$ $17$ $222$ $0.6$ $120/84$ $< 3$ $21$ $286$ $0.7$ $121/84$ $< 3$ $22$ $320$ $0.3$ $121/84$ $< 3$ $25$ $384$ $0.9$ $1/21/84$ $< 3$ $25$ $384$ $0.9$ $1/21/84$ $< 3$ $31$ $4256$ $1.0$ $1/9/84$ $< 3$ $36$ $476$ $1.2$ <	120/63 $0.6$ $20$ $ 1.0$ $0.1$ $116/83$ $< 3$ $18$ $212$ $0.6$ $0.1$ $116/83$ $< 3$ $18$ $212$ $0.6$ $0.1$ $120/63$ $< 3$ $17$ $216$ $0.6$ $0.1$ $120/63$ $< 3$ $17$ $216$ $0.6$ $0.1$ $120/63$ $< 3$ $17$ $216$ $0.6$ $0.1$ $120/63$ $< 3$ $17$ $222$ $0.6$ $0.1$ $120/83$ $< 3$ $17$ $222$ $0.6$ $0.1$ $11/84$ $< 3$ $17$ $222$ $0.6$ $0.1$ $112/84$ $< 3$ $16$ $236$ $0.7$ $0.2$ $120/84$ $< 3$ $21$ $286$ $0.7$ $0.2$ $121/84$ $< 3$ $25$ $384$ $0.9$ $0.3$ $17/64$ $< 3$ $31$ $435$ $1.0$ $0.2$ $19/84$ $< 3$ $36$ $476$ $1.2$	1.0 $1.0$ $0.1$ $1.6$ $11.93$ $3$ $18$ $212$ $0.6$ $0.1$ $4.6$ $11.93$ $3$ $18$ $212$ $0.6$ $0.1$ $4.6$ $11.93$ $3$ $17$ $216$ $0.6$ $0.1$ $4.6$ $120/63$ $43$ $17$ $216$ $0.6$ $0.1$ $4.6$ $120/63$ $43$ $17$ $216$ $0.6$ $0.1$ $4.6$ $120/63$ $43$ $17$ $222$ $0.6$ $0.1$ $0.1$ $120/63$ $43$ $17$ $222$ $0.6$ $0.1$ $0.1$ $120/64$ $43$ $17$ $222$ $0.6$ $0.1$ $0.1$ $121/84$ $43$ $21$ $286$ $0.7$ $0.2$ $0.2$ $1/20/84$ $43$ $25$ $364$ $0.9$ $0.3$ $0.72$ $0.7$ $1/24/84$ $43$ $34$ $476$ $1.2$ $0.2$ $0.7$ $1/1/84$ $43$ $33$	1.0 $1.0$ $0.1$ $1.6$ $7.4$ $20/63$ $0.6$ $20$ $ 1.0$ $0.1$ $1.6$ $7.4$ $11.183$ $< 3$ $18$ $212$ $0.6$ $0.1$ $<0.1$ $2.4$ $11.183$ $< 3$ $18$ $212$ $0.6$ $0.1$ $<0.1$ $2.5$ $120/63$ $< 3$ $17$ $216$ $0.6$ $0.1$ $20.1$ $2.1$ $120/63$ $< 3$ $17$ $216$ $0.6$ $0.1$ $20.1$ $2.1$ $120/63$ $< 3$ $17$ $216$ $0.6$ $0.1$ $0.1$ $2.1$ $120/83$ $< 3$ $17$ $222$ $0.6$ $0.1$ $0.1$ $1.5$ $11/84$ $< 3$ $16$ $236$ $0.6$ $0.2$ $2.3$ $0.2$ $2.3$ $11/2/84$ $23$ $21$ $286$ $0.7$ $0.2$ $0.2$ $2.3$ $1/2/84$ $< 3$ $21$ $232$ $326$ $0.7$ $0.2$ $0.7$ <t< td=""><td>$\begin{array}{c ccccccccccccccccccccccccccccccccccc$</td></t<>	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

	DA	th Shee	-	PARAN	neters	[no]s	) WEL	L 13H-	SHALLOW
	DATE	ALKALINITY	TOTAL	5042-	G2+	No.+	Nig 2+	K+	OTHER
·	17/20/83	~196	0.14	33	48	З	27	. –	
	11/16/83	-	-	-	-		-	-	-
	12/20/83	-	-	-	-	-	-	-	-
	1/12/84	-	~		-	-	-		-
	2/12/84	-	-	-	-	-	-	_	_
Mad • In U.S.A	3/20/84	-	-	-	-	-	-	-	
	4/22/84	-	-	-	-	-	-	-	-
	5/24/84	244	0.03	71	67	3	35	Z	_
	6/7/84		-	-	-	-	-	_	
	7/19/84		NSS POLEP	-	-	~	-	-	_
	8/30/84	250	0.04	100	-	4	39	3	-
)									
	•								
								-	
	•								
				:		1 1 1 1 1			
						•			

			Para	METERS	(mg/1)	ALS C	L 13B-		>
DATE	DISSOLVED. BOD5	DISS	TDS	DISS TKN	DISS NH3-N	NOz-N+	CI-	۲ŧ۹	OTHER
10/20/83	0.3	25	_	20.2	<0.1	4.6	32.	8.6	-
11/16/83	23	25	300	0.Z	<0.1	4.7	33	-	-
12/20/83	< 3	45	292	0.2	20,1	4.6	36	-	-
1/12/84	<3	6	32Z	0.2	20.1	4,3	40	_	-
2/12/84	23	25	316	0.2	<0.1	4.5	35	-	-
3/20/84	23	<5	314	0.Z	0.1	4.2	33	7.1	-
4/22/84	23	25	3/2	0.2	0,1	3.9	33	7,4	-
5/24/84	23	<5	344	20.2	0.1	3.7	34	7.0	8.2 LACF 4
6/7/84	-	-	-	-		-	_	. –	-
7/19/34	-	-	-	-	_	_	-	-	-
8/30/34	. 23	<5	316	0.2	0,1	3.9	34	7.3	
9/25/24	23	<5	339	0.2	20.1	.4.1	35	7,4	
10/9/84	23	5	316	QZ	K0 1	4,2	37 -	5,7	
		÷							
								- -	
-									
	DATE DATE 10/20/83 11/16/83 12/20/83 1/12/84 2/12/84 3/20/84 3/20/84 3/20/84 5/24/84 5/24/84 5/24/84 5/24/84 5/24/84 5/24/84 5/24/84 5/24/84 9/25/84	DATA         SHEE           DATE         Dissolved           DONTE         BODs           10/20/83         0.8           11/16/83         23           12/20/63         3           12/20/63         3           2/12/84         23           3/20/84         23           3/20/84         23           4/22/84         23           5/24/84         23           6/1/84         23           7/19/54         -           8/30/34         23           9/25/84         23	10/20/93 $0.8$ $45$ $11/1L/83$ $43$ $45$ $11/1L/83$ $43$ $45$ $12/20/63$ $43$ $45$ $1/12/84$ $43$ $45$ $1/12/84$ $43$ $45$ $2/12/84$ $43$ $45$ $3/20/84$ $43$ $45$ $3/20/84$ $43$ $45$ $4/122/184$ $43$ $45$ $5/24/84$ $43$ $45$ $6/7/124$ $  7/19/54$ $  8/30/34$ $43$ $45$ $9/25/74$ $43$ $45$	DATA         SHEET         PARAI           DATE $BOD_5$ $COD$ TDS $In/Zo/83$ $O.8$ $45$ $ II/IL/83$ $43$ $45$ $ II/IL/83$ $43$ $45$ $ II/IL/83$ $43$ $45$ $292$ $I/IZ/84$ $43$ $45$ $292$ $I/IZ/84$ $43$ $45$ $312$ $2/IZ/84$ $43$ $45$ $314$ $4/22/84$ $43$ $45$ $314$ $4/22/84$ $43$ $45$ $344$ $6/7/84$ $   7/19/54$ $   8/30/34$ $43$ $45$ $316$ $9/25/64$ $43$ $45$ $339$	DATA SHEETPARAMETERSDATEDISSOLVED, BOD5DISS CODTDSTKN10/20/030.8 $< 5$ - $< 0.2$ 11/11/183 $< 3$ $< 5$ - $< 0.2$ 11/11/183 $< 3$ $< 5$ $< 0.2$ 11/11/183 $< 3$ $< 5$ $< 0.2$ 12/20/63 $< 3$ $< 5$ $2.92$ $0.2$ 11/12/184 $< 3$ $< 5$ $2.92$ $0.2$ 11/12/184 $< 3$ $< 5$ $3.16$ $0.2$ 2/12/184 $< 3$ $< 5$ $3.14$ $0.2$ 3/20/184 $< 3$ $< 5$ $3.14$ $0.2$ 4/122/184 $< 3$ $< 5$ $3.44$ $2.2$ $< 1/21/184$ $   < 1/19/54$ $   < 1/19/54$ $   < 1/19/54$ $< 3$ $< 5$ $3.16$ $< 1/25/184$ $< 3$ $< 5$ $3.39$ $0.2$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	DATA SHEET       PARAMETERS $(mg/1)$ WEL         DATE       DISSOLVED,       DISS       TDS       TKN       NH3-N       ND3-N         DATE       BODs       COD       TDS       TKN       NH3-N       ND3-N         In/20/93       O.B $4.5$ $ 20.2$ $20.1$ $4.6$ 11/1L/83 $4.3$ $4.5$ $ 20.2$ $20.1$ $4.6$ 11/1L/83 $4.3$ $4.5$ $2.92$ $0.2$ $20.1$ $4.7$ 12/20/63 $4.3$ $4.5$ $2.92$ $0.2$ $20.1$ $4.3$ $2.5$ $2/12/84$ $2.3$ $4.5$ $314$ $0.2$ $0.1$ $4.2$ $4/12/84$ $4.3$ $4.5$ $314$ $0.2$ $0.1$ $3.7$ $5/24/84$ $4.3$	DATA         SHEET         PARAMETERS $(mg/1)$ WELL 13:5- biss           DATE         BODs         COD         TDS         TKN         NH3-N         ND2-N+           DATE         BODs         COD         TDS         TKN         NH3-N         ND2-N+           10/20/03         0.3 $45$ - $20.2$ $20.1$ $4.6$ $32.$ 11/16/83 $43$ $25$ $300$ $0.2$ $20.1$ $4.6$ $32.$ 11/16/83 $43$ $45$ $292$ $0.2$ $20.1$ $4.6$ $32.$ 11/16/83 $43$ $45$ $292$ $0.2$ $20.1$ $4.6$ $35$ 12/20/63 $43$ $45$ $292$ $0.2$ $20.1$ $4.3$ $40$ 2/20/64 $43$ $45$ $316$ $0.2$ $0.1$ $4.3$ $40$ 2/2/84 $43$ $45$ $314$ $0.2$ $0.1$ $3.7$ $34$ $4/22/84$ $43$ $45$ $314$ $0.2$ <td>$\begin{array}{c c c c c c c c c c c c c c c c c c c$</td>	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

	DATE	ALKALINTY	P	504 ²⁻	Ca ²⁺	Nat	Mg ²⁺	K+	OTHER	
)	10/20/83	∽ <i>1</i> 78	<0.02	25.	52	17	Z 2	-	_	•
	11/16/83	-	-	-	-	_	-	-	_	
	12/20/83	-	~	-	-	-	-	-		• • •
	1/12/84	-	-	-	-	-	-	-		- -
	z/14/84	-	-	-	_	-	-	-	-	•
A.3.1.4.4.1.5.A	3/20/84	-	-	-	-	-	-	• 🗕		
- F1 - F	4/22/84	-	-	-	-		-	—		
1 (L)	5/24/84	193	27.02	25	49	21	23	6	-	
	617/84		-	-	-	-	-	_	-	
	7/19/84	-	-	-	-	-	-		-	
Ň	8 30/84	. 193	20.02	24	54	23	22	8	-	
)										
•								•		
	-									
	•									
	: !						-			
-										

·		HD CUNIM ATA SHE			mg/l)	(1)L'I ERS	WE	14 -		DW
	DATE	DISS BOD5X	PISS CDD	TDSX	DISS TKN X	DISS NH3-N	D155 NO2-N+ NO3-NX	CIT	PH	OTHER
)	3/20/84	4.6	15	638	1.4	1.]	0.4	180	(118) 7.2	~
	4/22/84	1B	30	634	1.3	0.3	1,5	160	7.D	-
	5/24/84	23	14	646	1.3	1.0	4,5	150	7.0	Bil LABEN
	6/7/84	24	14	634	1.4	1.0	2.4	150	7.1	-
	7/19/84	23	30	410	1.2	D. Z	1.8	80	7.2	
	8/30/34	23	13	017	2.1	Z.0	0.1	250	0.7	-
	9/25/84	23	רו	774	Z.3	2.6	0.1	230	71	
	10/9/84	23	/3	742	2.6	2.3	0.1	270	7.1	
	, ,									
, , ,										1
	а - С									
	•									
	:						-			
					-					
							:			
	:									
	, -									
	•									
				-			• .			
					-					
	1						•			

		HL CONN SHEE		FARET	eteks –		VJE	14 -	SFFICEW
	DRITE	ALKALINITY	TOTAL	55 ²⁻	G2+	í Ja:	116	<u>K</u> +	ÛTHE R
、 [	5/24/84	238	acz	45	43	120	29	9	-
	6/7/84	-	~	-	-	-	-	-	-
	7/19/84	-	-	-	-	_	_	-	-
	8/30/84	Z30	0.05	/9	48	180	23	11	-

. . •

	DATE			+	(mg/_	· · · · · · · · · · · · · · · · · · ·				V, CELL 2
	DATE	DISS BODS X	Diss	TDSX	DISS TKNX	D155 NH3-N3	DISS NO2-N+ NO2-NY	Ċľ,×	FIELD	OTHER
Ĵ.	3/20/84	160	170	1390	19	19	D, 1	420	6.4	-
	4/22/84	120	130	1360	18	17	21.0	280	6.4	-
	5/24/84	<15	28	1620	6.8	6.3	1.6	690	6.2	7.2 LHBCH
	6/7/84	230	30	1560	Z,5	1.7	0.1	660	6.3	_
	7/17/84	12	42	161D	27	26	0.1	600	6.6	-
	7/31/84	12	34	1570	18	17	0.1	620	-	_
	8/3/84	30	51	1660	26	25	<1.0	640	-	-
	3/7/84	87	120	1800	36	34	21,0	650	-	-
	5/10/84	120	150	1500	46	42	<1.0	670	-	~
	8  30 /84	11	100	1910	78	74	11	660	6.5	
` `	9/25/54	7,8	23	1320	61	60	7.0	540	6.5	_
1	10/9/84	11	31	1960	4.0	4,0	48	740	64.	
	10/12/84	-	27	-	7,3	1.2	74	560		_
	10/15/84		27	-	1.1	0.4	37	470		-
:	10/13/54	$\sim$	25	-	1.2	<1.0	17,8	500	-	-
•	10/23/64	-	21	_	1.0	0.2	44	500	-	
	10/26/84	_	25		1.0	×1.0	53	600		
;	is 3:/E4		23	-	1.0	21.0	33.	440		
;	11/1/85	-	24	-	D, 3	<1.0	30	490	_	-
	11/5/84	-	-	-	1,5	0,3	G, B	£30	_	-
	11/13/84				2.0	1.0	10	530	_	· _ ·
•	11/16/84	-			Z. B	2.1	7.2	620		
	11/20/34	-	-	·	4.Z	3.6	6.2	680	_	

DATE	HIKALINTY	TOTAL	5042-	G24	1.0	11/02+	K *	S7HER
5/24/84		15	F 0	92	350	51	26	
617/84	-	-	~	~	-	-		<b>—</b>
7/19/84	_	-	-	-	-	-	-	-
7/31/84	-	-	-	-	_	-	<u> </u>	_
3/3/84		-	-	-		-	_	_
8/7/84		-	-	-	-	-	-	-
6/10/34	-	-	-		-	-	(	-
8/30/84	563	22	190	43	540	26	43	
								· · ·
	: : :							
		6 		2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2				
					-			· · ·
	à	2	• • •					
		-						

•

		DISS	ASS .	<del></del>	(mc/)	DISS	D155 NO3-NT	1	FIELD	1
	DATE	BODSY	CO D	TDSX	TKNX	NH3-N	NOZ-NT	Cix	p +1	OTHER
r.	3/20/84	7.7	34	550	1.6	0.8	0.1.	100	445) 7.0	-
	4/22/84	743	160	1650	1,6	0,5	0,1	720	6.4	-
	5/24/84	83	130	1500	3,D ⁻	1,9	0.1	630	6.4	7.4 LAEF-
	617184	ID	38	1140	Z.D	1.Z	0,1	470	6.4	
,	7/17/84	140	200	1370	3.3	1.9	0,1	720	6.6	
C.C.D.UI EDOM	7/31/84	42	36	1990	3.4	2.0	0.1	Bġd		_
	8/3/84	73	130	2030	3.0	1.7	0.1	370	-	
	8/7/34	35	92	1900	3,1	1.6	0.1	740	_	<b>—</b> ',
	3/10/84	190	ZAD	2010	3.6	1,7	0,1	720	-	
	8/30/34	39	100	1460	3,8	2,6	0.1	510	6.4	_
ì	9/25/24	9.2	39	1350	5.2	4.4	20.1	510	6.4	
j	10/9/24		43	1730	7.8	6.9	20,1	700	6.6	
	10/12/24	_	41	_	7.2	6.3	20.1	550		_
	10/15/24	·	41	_	6.4	5.7	6.1	480		_
	17/18/54	-	34	<u> </u>	6.4	5,4	20.1	460	·	_
	10/23/24	-	41	-	9,2	7,8	0.1	550	<b>—</b> .	-
	10/2-154		40	—	9,4	<del>،</del> ت	0.1	600	_	
	0/3:124		48	<u> </u>	8.Z	6.9	0.1	560	_	-
	11/1/34		41		6.3	5,3	0.1	430	<del>.</del>	_
	11/2/84	—	-	-	5,4	4.0	0.1	320		-
		<b>1</b>						, , , , , , , , , , , , , , , , , , ,		
							1			
				.			-	•		

	D4.7	م جوينا ج	-	CTEA. FFR-		ر اجراح	)	• _ /io -	SE EFFI	
	DATE	HLKELINT		5022-	2 ⁴	Nat	î lir	\$. <b>4</b>	JTHER.	
	5/24/34	329	14	100	5	450	20	24	-	
	617/84	-	-	-	_	_	· · · · · · · · · · · · · · · · · · ·		_	
	7/19/84	-	-	~	_	-		-	-	
	7/31/84	-	-	-	-	-	_		_	
S SQUARE S SQUARE	813/84	-	~	-	-	-	-	_		
	8/1/84	-	-	-	~	-	-	_	-	
42 391 10 301 10 300 400 42 347 100 300 10 10 10 300 40 41 10 100 300 10 10 10 10 10 10 10 10 10 10 10 10 1	6/10/24	-	-	-		-	-	-	-	
WATIONAL	8/35/84	433	11	7,4	44	460	25	27	~	

	UNUITE DATA	SHEET	v v. p. v -	(7× 11)	برزم بر			12 17	· CE11	1	i. I
	PATE	0155 BOD5	D155 (07	TPS	DISS	015 S NH3	2155 124-122	CI ⁻	ρH	CTHER	
	10/12/24	23	55	2090	ΞZ	20	< 2.1	750			
	10/15/84		49		30	29	<1.0	סור	~	-	
	10/18/84	< 61	63	3000	30	30	<1.0	700	. <u> </u>		
	10/25/84	_	60		33	30	<1.0	650			
S SQUARE S SQUARE S SQUARE	10/26/84	_	50		30	30	<1.0	680		_	
	10/30/84	_	51		36	31	<1,5	690		-	
42 381 50 SHEETS 42 382 100 SHEETS 42 382 200 SHEETS	11/1/84	_	43		34	22	<1.0	700	-	-	
WALLOWAL	11/3/34		-	~	34	34	<1.0	660	_	-	
• • • •	11/13/84			-	30	30	<1.0	590	_	-	
	11/16/54		-		29	29	<1.0	540	_	_	
	11/25/84	-	_		27	Z7	<1.0	500	_	-	
					· · · · · · · · · · · · · · · · · · ·	,					
			-								
			-								
			:								
		-			- - - - - - - - - - 			- - -			
						. ··					
·				-							

6/83 20/83 2/84	< 3 < 3 < 3	<5 6 (?) 25	6 4 8	<0.2 <0.2 <0.2	NH3-N <0.1 20.02 <b>20,1</b>	NO3-N D. 1 20.02 0.1	<0,3 <0,3 <0,3	-
			1					-
2/84	< 3	25	8	<0,2	£0,1	0.1	<0,3	
				-				
							1	1
			·					
			•					
				-				
			۲.					

APPENDIX G Plade in U.S.A. BRODHEAD : ระกษณะเรย โต UNSATURATED ZONE REFERENCES - BOUMA, 1975 )

System no.		tension (cm) sidewall	s for twelve subsurface R (days)		Calculated flow in gals/sq ft/day (cm/day)		
			bottom	sidewall	bottom	sidewall	
	23	35	6.9	35	1.8(7.5)	0.4(1.7)	
1 2	23	24	6.5	6.5	1.4(5.8)	1.4(5.8)	
3	25	21	4.6	3.2	1.6(6.6)	2.2(9.2)	
4	25	28	7.1	9	1.8(7.5)	1.1(4.6)	
5	80	60	267	74	0.09(0.4)	0.2(0.8)	
6	120	100	9000	4000	0.007(0.03)	0.011(0.04	
7	65		34		0.45(1.9)		
8	34	37	73	82	0.16(0.67)	0.14(0.6)	
9	n.d.	20	n.d.	97	n.d.	0.16(0.7)	
10	20	20	52	47	0.22(0.92)	0.20(0.8)	
11	6	6	115	115	0.04(0.17)	0.04(0.17)	
12	15	. 20	28	30	0.18(0.75)	0.15(0.62	
13	4		20		0.15(0.62)		

Table 2. Monitoring data obtained in situ with tensiometry and derived performance characteristics for twelve subsurface seepage systems.

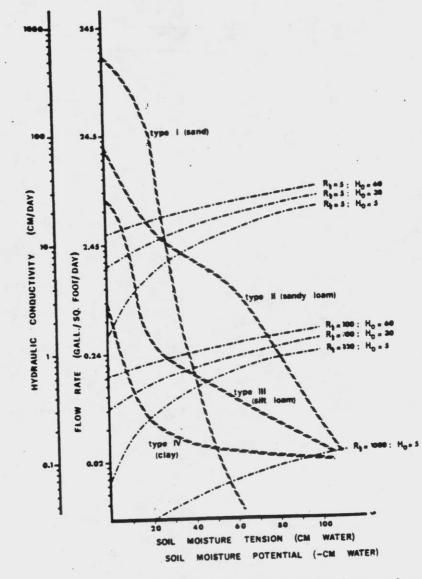


Fig. 1. Hydraulic conductivity curves for four major types of soil and curves expressing the hydraulic effects of impeding barriers and different hydraulic heads (see text).

APPENDIX H BRCOHEAD : FURROW WASTEWATER AND LY SIMETER CHEMISTEY DATA

1	DATA SAFET		1 2.4	(my/d)		TURRICO CRUE 1 CONSTRUCTOR CRUE 1			42 <u></u>
DATE	TOTAL BOL:	DIES COD	DISS	10155 10113-10	DISS NOZTNOZ		P.H	בד+רג ER	STHER
7/9/34	DISS 400	460	16	14	0.1	740		705	
7/13/34	0155 50	300	32	22	<1.0	570		حت ما ا	
7/10/34		150	39	35	21.0				
7/19/34	DISS 130	160	60	53	<1.0	550	7,5	705 1610	_
7/27/34	DISS 710	810	13	5.3	0,1	820	-	_	_
7/31/54	1000	870	23	/3	0.1	720	_	_	_
8/2/54	0155 750	740	てつ	15	41.0	900	_	_	-
9/7/34	110	230	22	15	<1.0	570	_	-	
8/10/54	300	370	Z2	15	<1.0	240		-	_
8/37/34	707AL 590	710	11	8.7	0.1	660	6.7		
9/4/84	707 1300	1000	ZZ	12	0.1	640	_		_
10/12/82	$\overline{}$	<u> </u>			-	-			_
10/15/34	<b>-</b>	⊷		•			-		_
15/18/54		790	25 DISS 43 757	N Z	1,4	700			
12/23/84	_	390	14 707 7.4 D165	5.5	<u></u>	3ZD		_	-
10/26/84	_	470	16 DISS 59 TOT	13	0.1	460	_	_	_
4 ت/زيز/ر	-	3E D	EZTOT 160155	12	5.1	350	_	_	_
1:11:24	<u> </u>	87	5.8 Tot 2.9 0155	1.6	0.1	160	_	_	
1/8/84	-	-	12 0155	9,3	0.1	540	-	-	_
112/22	_		27 707 14 DKS	12	2,1	4ED	_	-	_
110107			20 TOT	7.7	20,1	570			_
1=0/34	_	<u> </u>	32 TOT 13 C:55	(B),T	<0.1	60	_	_	_

WAYKEWAL 12 301 500 SHEETS 3 SOUARE

(⁻)

.

`

		974 Shee	······		11.11 - 17 (11-12	)	TUK.	li li TEWATE E	- (E4	2	
	DATE	TOTAL BOLS	DISS CDD	Diss TKN	Diss NH3-N	6155 NO2+NO2	c1-	¢Н	ether	CTHER	
	7/11/84	DISS 700	1100	61	26	<1.0	580	-	705 2540	_	
	7/13/84	DISS >450	1300	ZS	13	<1.0	970		TD5 2720	_	
	7/16/84	_	570	5.7	0.9	0.1	_	_	-	-	
and And And And And And And And And And A	7/19/84	2400	430	24	7.4	0.1	540	6.7	-		
42 301 500 SHEETS 5 SOULAR 42 302 300 SHEETS 5 SOULAR 42 303 300 SHEETS 5 SOULAR	7/24/84	dis 5 580	2300	42	39	0.1	510	_	_	_	
200 SHE 200 SHE 200 SHE	15/12/84	_	850	-= 015 50 - T	13	6	-60	_	_		
	10/15/84	—	600	16. 5155 47 TOT	6.1	0,1	300	. —	—	_	
untional	10/18/24		830	24 DISS	21	4.0	600 .	<u> </u>		_	
	10/23/84	_	1200	60 tot 36 dis6	14.	Z1.0	670	_	-	-	
	10/20/24	-	800	32 TOT 18 DISS	5.2	0,1	640	_	-	-	
•	10/30/84	-	710	29 TOT 19 DISS	15	<1.0	520		-	_	
	11/1/84	-	Z30	14 TOT 5.0 DISS	ک ک	5.1	200	-	_	_	
	11/8/24	-	-	2270T 227155	3.2	0.1	720	-	_	_	
	11/1=/24	-		54 TOT 12 DISS	5.3	0,1	610	_	-	<u> </u>	
	11/16/24	~	-	370,TOT 33 0155	27	21.0	870	-	-		
	177:184	-	-	57 TOT 27 DISS	13	11.0	Edo				
						-					
		-									
•											
			a a a a a a a a a a a a a a a a a a a								
			ţ					i	i		

	DA-	TA SHEE	T	- ۳۱۲۲ ا	(mg/1)	)		SIMETE	r =   -	CELLI
	DATE	BODS	600	TDS	DISS	DHSOLVE	01550LVED NO2-N+ 1303-N		ρH	OTHER
$\overline{)}$	11/16/83	-	LOW LEVEL	d	1.3	20.1	D,]	-	~	-
	2/12/84	-	TOTAL 760	JET NE	[7.2]	4,4	20.1	-	-	
•	3/20/84	-	scivele 51	-	2,5	1.1	20.1	_	6.8	_
-	4/24/84	<b>—</b>	LIWLEVEL 61 A	W -	70TAL 2.4	0,1	<0.1		~	—
	5/24/84	_	-974L	-	4.3	0.6	D. 1	-	12	_
Madaln <b>U.S.A</b>	7/31/ <del>3</del> 4	-	501.0BLE 76		9.2	7.5	0.1	1100	_	-
	8/3/84	-	62	_	zÓ	12	20.1	_	-	-
6	3/7/ <i>8</i> 4		67	-	ZO	· ۲۱	0.2	900	~	-
	8 /10/34		502	ZZ60	24	ZD	Di 1.	360	—	
	8/30/84	_	5° L 63	-	24	20	<1.0	770	7,1	_
)	9/4/84	-	50L 70	_	Z6 ·	23	<1.0	630		<u> </u>
	ic/12/84	_	50	<u> </u>	35	30	0.1	720	-	-
	10/15/84		43	·	3Z	3)	×1.0	707		~
	10/18/84	_	44	—	31	30	21.0	720	-	_
	10/23/84	-	42	Z <del>2</del> 00	30	29 .	<i>≺1,</i> 0	710		
•	10/26/84	-	40	_	29	27	<1.0	690	-	_
	10/30/84		47	—	27	25 -	<1.D	730		_
	11/1/84	_	46	-	26	Z3	<i>∠</i> 1.0	710	-	_
	11/8/84	-	-	-	ZI	20	<1,0	610	-	-
-	11/13/84	-	-		16	15	<1.0	480	-	
	11/14/84	<u> </u>	-	_	14	13	21.0	440	-	
•	11) 25  84	-	—		13	12	<i>∠1.</i> 0	420	_	
						-				
					1	4		1		

	DATE			<u>رت الم</u>	(mg/-1)	-7 2 200	_ • 5	METER		ELL 1
	DATE	0155 3005	D:55 (20)	TDS	TKI	(1540-050 1012-1. K		N CI	F H	STHER
	Z/12/B4	-	410	-	TOTAL	2.2	20.1	_		
	3/20/84		5.76	_	4.0	5.2	0,1	530	6.7	
	4/24/84	_	TOTAL 62	-	Z.B	0.1	< 2.1	_	-	
	5/24/84	-	62	-	TOTHL G.D	I.D	0.1	-	7,4	_
	6/11/34		49	-	14	14	0.0	_	-	_
- - 	6/18/84	-	42	~	18	16	20.1	-	-	
	7/9/84	4,1	43	1910	ZI	20	<0,1	750	_	:
-	4 <i>ב וו</i>  ר	-	53		22	-20	0.1	-	_	<b>—</b>
	7/13/84	_	51	2130	22	21	0.1	_		
	7/16/84		37	_	16	16	<i>0.</i> Z	-	-	
	7/19/84	-	49	-	24	22	21.0	-	7,4	_
	7/24/34		58	-	24	23	<1.D	-		
	7/27/84		52		25	. 22	<1,0	·	_	-
	7/3/154	_	52	-	24	23	<1,D	-	—	
	8/3/84	-	54	-	25	Z5	<i><b>L</b>1.D</i>	-	_	_
	8/7/84	-	56		25	23	2.4	300		<u>-</u>
	<i>2/10/</i> 94	-	62		25	Z4	3,2	840	-	
	8/30/84	_		-	11	8,3	15.6	500	7.2	
	9/4/34	_	63	-	6.2	4,1	ZD	740	-	—
	10/12/34	-	52	-	2.0	21.0		730	-	-
	10/15/24	<u> </u>	55	-	1.9	<1.0	22	750		
	10/19/84		46		1.8	-1.0	2/	720	-	_
	125/54		45		1.6	<1.D	21	710	-	-
	12/20/24	_	42		1.10	D.1	20.	690	·	_

i

	DATA	SHEET		<i>ت</i> من ا	1111 (71 -	VACANESE	245	INNETER	= Z, C	E21)	Z
	ÛATE	DIES BODS	diss COD	TOS	OKS TKN	DISS NH3	0155 NOz+NO2		ρH	OTHER	
Ì	10/30/84		IS	_	1.6	0.1	18.4	660		_	
	11/1/ê <del>4</del>		50		1.9	2.1	16,5	630		_	
	11/5/84	_	_	_	1.6	D.1	16.0	550	-		
	11/13/84		-	-	1.5	0,1	16.6	550	_	_	
S SQUARE	n/16/54	-	-	-	1.3	D.1	17.3	590	<u></u>	-	
6 200 SHEETS 3 SQUARE	11/20/84				1.6	20.1	20	600		-	
								-			
WATERAL											
				-							
										-	
:											
	•	•		I	1	I		i	i		

-

. • •

	DATE	BODS	COD	TDS	TKN	<b>I</b>	NCZ-N+	101-	pH	OTHER
7	11/16/83		LOWLEVEL 44		TOTAL	15	5,0			
2	2/12/84	~	TOTAL 52D		TOT 4	9.5	1.8			
	3/23/84		190		2.6	<0,1	0,1			
	6/18/84	_	_		2,3					
	7/9/84	-	53		2,9	0.6	0,5	610		
[	7/19/84		_		3.0	0.3	Z.D		7.6	-
1	7/27/84	<u>`</u>	_		2.0	0.1	2,4	-	_	_
	8/3/84		59	_	1,8	0.1	Z,S '	_	_	_
	8/10/84		~		2.5	0.1	4.4	_		
	9/4/84				IS	0.1	18.Z	1000	_	· · ·
	10/12/84	_			IS	1,9	6.4	850		
	10/15/84	_		_	Is	2.3	7.4	92 <b>0</b>		
Ľ	10/15/84		No	SAMPLE			-			•
	10/23/24	_	15		15	3.6	8,7	750		_
	10/26/84	_	15	_	IS	2.0 1	8.9	730	-	-
Ì	10/30/84	_	_	_	IS	0.9	ê.Z	750	_	<u> </u>
X	11/134		No	SAMPLE						
	11/8/84	-	-	-	IS	1.7	10.0	730	-	·
	11/11/24		-		4,4	1.6	10.2	760		-
1	11/20/84	-	-	-	IS	1.6	10.9	720	-	
1							1 1 1			
			-				f			
:				-						

	Driff	DISS			(me/l 0155	DISS	24	IME TER	#5-C	F12 Z
	DATE	BODS	COD	TOS	TKN	11H3-1		S CI-	φH	C7 HER
Ì	-,/9/54	<3	24	· -	0.9	0,1	1.3	710		
	7/11/84	_	24	-	0.5	20,1	2,1	_	_	_
	7/13/84		27	_	0,9	0,1	2.5	580		-
	7/16/54		23	_	0,6	D.1	2.7		_	_
	7/19/84	-	30	_	1.1	0,1	5,8	_	7,4	_
	7/24/84	_	40	_	1,Z	0.1	2,5		-	
	7/27/84	_	53		1,4	0.1	0.1			-
	7/3//84	-	51	<b>_</b>	116	0.1	0.1	-	_	-
	5/3/84	_	4 E	-	1,4	0.1	5.1.	_	-	-
	8/30/24				1.2	0.1	20			
	9/4/24		39	_	1.2	0.1	11.5	590		
	10/12/84	_	—	-	0,7	0.1	16,1	500		_
	10/15/84	<u> </u>	31		D, 3	0.1	21	400		-
	10/18/34	_	24		0,3	<1.0	23	390		_
	10/23/26	<u> </u>	23		D, 3	21.0	24	420	-	
	10/26/84	_	19	_	0.7	<1.0	24	440		
	:5/3:/34		27	—	0,6	21.0		420	-	. —
	/:/ <del>:</del> /:/		23	_	2,3	401	19,9	290		
	11/8/31	-	-		0.6	0.1	16.3	Z30	-	-
	11/13/54		-	-	0.5	<0,1	12	220		-
	11/16/24		-	-	0.6	20.1	6.6	160	-	-
	1:/20/34		. —	_	0.6	0.1	7.0	110	-	-
				-						

A2 349 200 SHEETS 3 SQUARE

×.	A	T	ee -		r		<u>_</u>	E METER	= - (	Ell Z
	DATE	5155 2025	2:55 C0D	TDS	Dies TKN	, DISS 1343-N	D155	C1-	pH	CT PSE
/ `\ •	6/11/34	23	/9	1230	0.6	0.1	24	400		
	6/13/84	2,5	19	1490	0.4	-20.1	22	56D	-	_
	7/9/34	23	17	1680	0.6	0.1	17.5	652	-	_
	7/11/84	23	13	-	0,3	0.1	$ 1\rangle$			-
SQUARE	7/13/34	<4	14	1210	0.Ь	0.1	.9.6	410	_	-
0 SHEE15 1 0 SHEE15 1	-116/84	23	/3	כרוו	0.6	0.1	4,5	360		_
42.382 100 SHEELS 5 SQUARE	7/19/34	۷٢.	24	1400	С, Э	0,1	37	415	6.3	-
NALITY N	7/24/84	4,9	30	ل 147 ک	1.0	2.1	46	420	_	-
<b>4</b> - <del>2</del> - 2	7/27/5.4	24	24	1540	0,5	<1.0	35	430		_
-	7/31/84	23	22	1430	0.5	<1,0	24	450		-
	S/3/E4	<3	17	1370	0.6	D. 1	15.5	270	_	-
	5/30/84		_	_	1.2	<1.D	23			
-	9/4/84				IS	∠1.0	29	470		
	10/12/84									
	10/15/24									
	10/18/84					-				
	ic/2=/34_		<u>-</u> کر		15	0.1	27	520		-
	10/20/24	-	20	<u> </u>	0.6	<1.0	27	500	_	
	/5/30/34	-	21	-	0.6	<1.0	23.0	420		
	11/1/54		ZO		0.5	20.1	18.4	310	-	-
	11/E /¿4	-	-	-	D.6	0.1	14,1	220		-
	11/13/84			-	C.6	20,1	10	150		-
	16/24	-	-		C.6	Z D.1	7,5	140	—	-
	11/20/24	_	-	<u> </u>	0,6	<0,1	6,2	110	-	-

		a <u>Shife</u> Diss	DISS		(m/) D155	DISS	2155		5-5	
	DATE	EODs	600	TES	7KN		hig-high	<i>C</i> 1-	P :-/	£7,⊬ER
)	7/24/84	3,4	14	236	0.4	0,1	0.1	1.2	-	-
	7/27/84	24	12	276	0.4	<0,1	0,1	1,4	-	_
	7/31/84	<3	D	254	0,4	<i>D.1</i>	0.1	0,6	-	_
	8/3/84	23	12	Z50	0,4	0.1	0,1	0.9	-	-
5 SQUARE	2/30/84	-	-		0.6	0.1	0,5		. –	
DO SHEETS	10/12/84		14	273	04	0,1	D, 1	1.3		-
42 389 200 SHEETS	10/15/84		[]	-	5.4	<i>حال.</i> ا	20.1	1.5		_
NA110NA1	10/18/94		12	-	0.4	451	<0.1	1.5		
2	10/23/84	-	12	243	0,4	20.1	0,1	1,4		_
	13/3-/34		15	-	6.4	2.1	3.1	1.3		_
	10/30/34	-	-		IS	2.1	<01	1.5		_
	11/1/202		-	-	IS	<0.1	2.1	IS	-	_
	n/8/84	-	-	-	D,4	0.1	<0,1	1. D	-	-
		-								
	-									
	•			-						

	DATE	Diss Bods	COD	TDS	OISS TKN	NH3-N	1 ND3 - N+	CT	IPH	OTHER
$ \mathbf{r} $	11/16/83	. <b></b>	LOW LEVEL	-	TOTAL 0,7	20.1	0.2	-	-	_
i	3/23/84	∠3	JOTAL ZO	360	0.5	0.1	0,1	2,4	_	
	4/24/84	LB	TOTAL	_	тотац 0,4	DI	0,1	1,4	-	
	5/24/84	_	15		0.4	20.1	0.1	-	6.7	C.03 615
	6/11/84	13	ZI	320	0.5	<0,1	0:1	/.3		
	6/18/184	3.1	25	330	0.6	0.1	0,1	1.2	_	-
i	7/19/34	<3	20	-	0.6	20.1	0,1	1.4	6,3	_
	7/24/84	23	25	296	0,6	0.1	0,1	1.2	_	-
	7/27/34	24	23	320	0,6	<0.1	0,1	1.4		-
:	3/3/84	23	Z6	356	0,6	0.1	0,1	1.7		-
	8  30/84	-	-	-	1,2	0.1	0.1			
}	10/12/54	_	25	304	С.Ь	0.1	0,1	1.2		_
	10/15/34	· _	23	-	0.6	20.)	201	1.3	-	-
	10/18/54	_	19		ما، 0	20,1	20.1	1.2	-	_
1	10/23/54	_	24	296	0,7	20,1	0,1	1.1		
; ) ;	10/26/84		27	-	0.6	9.1	0.1	1,0		_
1	10/30/84	)	28		D.6	0.1	(0.1	1.0		_
	:/1/84	-	28	_	0,3	20,1	0,1	1.1	-	_
1	18/84	-	-	-	0.7	20.1	20,1	1,3	-	
-		-							an anna 1	. · ·
i										
i			!	-	-				1	•

••		-
(* )	APPENDIX I	
174644451En / Mada In U.S.A.	BRUDHEAD: CROP UPTAKE ANALYSES AND CALCULATIONS	

19611 -

SALIER

UNEX SOIL AND PLANT ANALYSIS LAB 5711 MINERNU FOUNT FOAD MADIOON WI FEBROS

• • •

•

-

SITE OF ANALYSIS: 5/23/34

Phase

				%						opei			
SAM	PLE	P	ĸ	CA	MG	S	ZN	5	М	FE	55	-1	NA -
				<u> </u>									
1	1A	0.130	0.154	0.417	0.099	0.163	45.71	6.421	165.1	915.3	5.778	856.8	( 23.2
2	18	0.203	0.165	0.544	0.110	3.175	46.37	7.501	173.0	323.2	5.538	1032	( 64.6
3	2	0.149	0.335	0.300	0.877	0.153	22.22	5.303	34.83	230.2	4.390	239.7	< al.7

1A	1.56	
2	1.44	
Ashed		
1B	0.42	
2	0.26	•

X N of A	sh		•	•
	2.49		•	-
2	2.64			•

Weight	of oven	dried	sample	(grams)		Z Ash
1.4	87.3					
18	67.2					17.1
2	311.3				•	10.1

IA, IB -> MINDORD 2 -> EREDHEAD

APRIL SAMPLES OF GRASS

SAUER ASHED SAMPLES KSL

448 5010 AND FLANT ANALLEDIS 148 5701 MINESAL FOINT 7040 - -401100 TA ADDIS

CATE OF ANALYSIS: 7/6/84

	<u> </u>					254						
SAMPLE	P	ĸ	CA	MG	S	ZN	8	N	- FE	CU	÷.	N <del>'</del> A
	<u> </u>						<del></del>					
1 18	0.137	0.149	9.495	0.096	0.071	35.51	5.330	157.6	638.2	4.711	662.2	< 61.3
2 28	0.140	0.352	0.278	0.072	0.072	20.33	3.249	32.65	243.4	4.448	243.8	( 62.5

#### COOPERATIVE EXTENSION PROGRAMS University of Wisconsin-Extension University of Wisconsin-Madison

Soil & Plant Analysis Laboratory, 5711 Mineral Point Road, Madison, Wisconsin 53705; 608-262-4364

#### DEPARTMENT OF SOIL SCIENCE

September 14, 1984 Acct. 900 Lab No. S0035



#### MEMORANDUM

TO: Dave Sauer--DNR 101 S. Webster, Box 7921 Madison, WI 53707

FROM: Soil/Plant Analysis Lab

RE: Results of analyses on 4 canary (+grass) samples submitted July 24, 1984.

Sample Identification		Sample Weight	Ash	Nitrogen of Tissue	Nitrogen of Ash
		grams	%	%	3/ /g
Mindoro	1	87		2.15	
Brodhead	1	134		1.28	
Mindoro	2	121	8.5		0.64
Brodhead	2	110	5.9		0.48

Additional analyses are attached.

If you have any questions concerning these analyses, please feel free to contact either Todd Kaehler or Ita Steingraeber at 262-4364.

Encl.

/ss

S35 DAVE SALER DNR

UNEX SOIL AND PLANT AVALISIS LAB 5711 MINERAL POINT ROAD MADISON HI 53705

JATE OF ANALYSIS: 9/14/84

<u> </u>									PPM			
SAMPLE	Ρ	K	CA	MG	S	ZN	8	HN	FE	CU	AL	NA
1 MIN.1	0.255	2.443	0.314	0.216	0.252	12.73	5.361	43.93	60.01	3.843	< 36.7	< 63.8
2 BR0.1	0.229	1.632	0.235	0.128	0.144	12.43	3.736	54.69	42.23	3.828	< 35.8	294.7
3 HIN.2	0.327	2.632	0.298	0.214	0.249	16.72	4.708	88.24	56.14	4.626	< 36.4	< 63.3
4 BR0.2	0.306	1.953	0.300	0.180	0.182	20.50	4.709	40.22	62.70	3.724	< 36.2	296.0

5 DAVE SALER (ASHED)

UNEX SOIL AND PLANT ANALYSIS LAB 5711 MINERAL POINT ROAD MADISON HI 53705

INIE OF ANALYSIS: 9/14/84

X					PPM							
SAMPLE	P	K	CA	MG	S	ZN	8	MN	FE	CU	AL	NA
			', <u> </u>									
1 MIN.2	0.318	2.509	0.296	0.204	0.145	16.88	5.042	85.15	63.52	6.397	< 36.1	99.24
2 BR0.2	0.293	1.813	0.289	0.165	0.056	18.51	4.301	39.88	70.66	3.827	< 36.0	295.8

1985

S134 DAVE SAUER

act samples

THE OF ANALYSIS: 12/28/84

				%						PFM				
SAY	IPLE	P	K	CA	MG	Ş	ZN	8	Mi	FE	CU	AL	141	
	<del></del>				<del></del>									
1	BROD.1	0.264	2.091	0.277	0.142	0.199	15.03	4.999	47.97	79.14	4.834	59.96	219.6	
2	BRCD.2	0.300	1.909	0.323	0.209	0.231	14.14	4.598	96.42	66.52	5.836	46.27	118.7	
1	BROD.1 ash	0.258	1.819	0.273	0.139	0.073	15.98	4.959	47.48	79.84	4.631	71.31	426.4	
2	BROD.2 ash	0.300	1.808	0.329	0.209	0.071	14.44	3.662	95.97	70.06	5.610	50.63	404.0	

Sample Id.	Sample Wt. grams	%Ash	N of Tissue
Brodhead Cell 1	64.2	5.5	2.70
Brodhead Cell 2	67.5	5.6	3.50

* Results for 2N of Ash will follow in several days.

235 DAVE SAUER Dec Somples

TE OF ANALYSIS: 12/28/84

## UNEX SOIL AND PLANT ANALYSIS LAB 5711 MINERAL POINT ROAD MADISON WI 53705

!									PFM				
4	PLE	P	К	CA	MG	S	ZN	В	MN	FE	CU	ĤL	N#1
							<del></del>						
1	BROD.1	0.189	0.651	0.201	0.092	0.158	14.20	4.326	56.47	87.73	3.712	65.37	835.3
2	BROD.2	0.197	0.491	0.146	0.136	0.117	40.26	3.183	79.76	57.89	4.060	42.81	430.5
-	MIND.1	0.139	0.497	0.230	0.094	0.086	23.59	4.866	139.9	106.3	4.450	102.9	< 61.0
- 4	MIND.2	0.172	0.549	0.177	0.097	0.158	28.68	4.610	70.91	79.80	3.719	64.15	184.6
1	BROD.1 ash	0.183	0.603	0.194	0.038	0.065	11.36	3.085	48.61	82.87	3.032	68.71	992.6
ż	BROD.2 ash	0.197	0.472	0.142	0.135	0.046	38.19	2.927	74.83	55.44	3.273	43.73	555.2
3	MIND.1 ash	0.135	0.431	0.225	0.039	0.048	28.92	3.336	128.6	103.0	4.723		< 59.3
4	MIND.2 ash	0.165	0.480	0.175	0.092	9.077	25.42	< 3.56	67.28	81.79	4.276	89,32	242.9

Sample Id.	Sample Wt. grams	%Ash	WN of Tissue
Brodhead Cell 1	54.5	3.9	1.59
Brodhead Cell 2	61.0	3.1	1.4i
⊮ Mindore Cell 1	88.7	9.2	0.58
Mindoro Cell 2	219.8	8.6	1.42

*Results for  $\mathbb{A}N$  of Ash will follow in several days.

#### COOPERATIVE EXTENSION PROGRAMS University of Wisconsin—Extension University of Wisconsin—Madison

Soll & Plant Analysis Laboratory, 5711 Mineral Point Road, Madison, Wisconsin 53705, 608-262 4364

### DEPARTMENT OF SOIL SCIENCE

January 7, 1985 Acct. No. 900 Lab Nos. S134; S235

#### MEMORANDUM

TO: Dave Sauer Wis. Dept. of Natural Resources Box 7921 Madison, WI 53707

FROM: Soil/Plant Analysis Lab

RE: Results of %N of Ash on 5 samples. All other analyses have been reported.

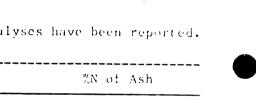
Sample Identificar	on	%N of Ash
(S134) CC ^A SH MMS BRODHEAD CELL 1	1	0.72
BRCDHEAD CELL (S235) Control BRODHEAD CELL	2	0.50
BRODHEAD CELL	1	0.45
BRODHEAD CELL	3 -	0.49
MINDORO CELL	1	0.15
MINDORO CELL	2	0.47

If you have any questions concerning these analyses, please feel tree to contact us.

The invoice for all of the tissue analyses is enclosed.

/ 55

Encl.



BRODHEAD - GRASS NITEOGEN CALCULATIONS APRIL 21, 1984 CELLZ (AREA - 110,000 ft2) AREA OF SAMPLE = 3'x4' = 12 ft² WEIGHT OF SAMPLE = 311.39 (OVEN DRIED) TOTAL WEIGHT OF GRASS - CEIL Z =  $\frac{311.39}{17.572} \times 110,000 \text{ ft}^2 \times \frac{16}{453.89} = 6288 \text{ lb}$ 70 N (DRY WEIGHT BASIS) = 1.44 AMOUNT OF N IN CELL Z GRASS PRICE TO BURNING = 6288 (0.0144) = 90.6 1b 20 ASH AFTER BURNING = 10.1 WEIGHT OF ASH - CELL Z = 6288(D.101) = 635 16 JON OF ASH = 2.64 AMOUNT OF AGHNON CELL Z AFTER BURNING = 635(0.02.4) = 16.8 16 EXTRAPOLATE VALUES TO TETAL SITE AREA TOTAL AREA CELL 2 ASEA = 1.85 TOTAL WEIGHT OF GRASS = 6288 × 1.85 = 11.646 15 A MOUNT OF GRASS N PRICE TO BURNING = 90.6 (1.85)= 168 16 AMOUNT OF ASH NON SITE AFTER BURNING = 15.8(1.85) = 31.0 90 N LOST = 168-31 (100) = 81.5%

► ∠``)

$$\underbrace{\bigcup_{U \in Y} 13, 1984}$$
AREA OF GRASS SAMPLES - CELL 1 = 1'x2' = 251²  
CELL 2 = 1'x3' = 351²  
WEIGHT OF GRASS SAMPLES - CELL 1 = 134g  
CELL 2 = 110g  
TOTAL LUEIGHT OF GRASS ON SITE = NOTE: CELL 1 is  
 $4670 \text{ or}$   
TETL HEEA;  
 $(246 (4.7 \text{ MEES})(43500 \frac{31\%}{\text{MARX}})(134g)$   
 $453.8 \frac{9}{16} \times 251^{2}$   
+  $\frac{0.54(4.7)(43560)(110)}{453.8(3)} = 22,847 \text{ Ib}$   
 $2 = N (DRY WEIGHT BASIS) = 1.28$   
And WEIGHT OF N ON SITE PRICE TO BURDING = 22,847(0.0126) = 29216  
 $20 \text{ ASH AFTER BURDING = 5.7}$   
WEIGHT OF ASH ON SITE =  $22,647(0.057) = 1347 \text{ Ib}$   
 $20 \text{ N OF ASH = 0.48}$   
Ancident OF N ON SITE AFTER BURDING = 1347(0.0026)  
 $= 6.5 \text{ Ib}$   
 $20 \text{ N LOST BY BURDING =  $\frac{292-6.5}{292}$$ 

)

PAREASTER / 1140 M.15 A.

SEPTEMBER 25, 1984 AREA OF GRASS SAMPLES - CELL 1 = ZO"X 17" = Z,36 ft"  $CELL 2 = 24" \times 28" = 4.67 ft^2$ WEICHT OF GRASS SAMPLES - CELL 1 = 64.29 CELL 2 = 67.59 TOTAL WEIGHT OF GRASS ON SITE (0.46)(4.7)(43560)(64.2), (0.54)(4.7)(43560)(67.5)453.8 (2.36) 453,8(4,67) = 5648,0 + 3522,6 = 9171 1b 90 N (DEY WEIGHT BASIS) - CELL 1 = 2,70 CEIL 2 = 3.50 AMOUNT OF NON SITE PRICE TO BURNING = 5648 (0.0270) + 3522,8 (0.0350) = 276 16 20 ASH AFTER BURNING - CELL 1 = 5,5 (EIL Z= 5.6 WEIGH OF ASHEN SITE = 5648(0,055) + 3522,8(0.056) = 507.9 K 20 NOF ASH → CELL 1 = 0.72 CELL 2 = 0.50 AMOUNT OF NON SITE AFTER BURNING = 5648(0.055)(0.0072) + 3522.8(0.056)(0.0050) = 3,2 16 9° N LOST BY BRNING = 276-3.2 (100) = 98.970

A 2.0 al chert

1. HULLAN MINIT

NOVEMBER 20, 1984

)

v Sill të spred

1

AREA OF GRASS SAMPLES - CEIL 1 =  $14" \times 11" = 1.07 ft^2$ CEIL 2 =  $12" \times 14" = 1.17 5t^2$ WEIGHT OF GRASS SAMPLES - CEIL 1 =  $54.5_q$ CEIL 2 =  $61.C_q$ 

TOTAL LUEIGHT OF GRASS ON SITE =

- $\frac{(0.46)(4.7)(43560)(54.5)}{453.8(1.07)} + \frac{(0.54)(4.7)(43560)(61.5)}{453.8(1.17)}$ = 10,575 + 12,707 =
- 90 N (DRY WEIGHT BASIS) = CEIL 1 = 1.59 CEIL 2 = 1.41 AMOUNT OF NON SITE PRIOR TO BURUNG = 10,575(0.0159) + 12,707(0.0141) = 347 16 70 ASH AFTER BURNING  $\Rightarrow$  CEIL 1 = 3.9 CEIL 2 = 3.1 WEIGHT OF ASH ON SITE = 10,575(0.037) + 12,707(0.031) = 412 + 394 = E06 16 70 N OF ASH - CEIL 1 = 0.45 CEIL 2 = 0.49 AMUCUNT OF N ON SITE AFTER BURNING = 412(0.0045) + 394(0.0047) = 3.7 16 90 N BURNING LOSS =  $\frac{347 - 3.7}{347}$  [100] 70 PE,97.

APPENDIX J

Place in U.S.A.

L'UNASIEN

BRODHEAD : SUGAR RIVER CHEMISTRY DATA

	:	SHEET		_====	•		LUGAR FRER JPSTREAM
•	DATE	TOTAL BDD5	Total Cod	C1 ⁻	TKN	755	OTHER
$\langle \rangle$	12/20/83	41	470	25	22	2600	-
	4/22/84	23	9	21	0,9	382	pH - B.D NO2+NO2 - 3.7 No12 - 0.4
•	8/30/84	3.7	18	23	1.2		TALK - Z60
							pH = 7.8 $TM_0 = 3L$
J.S.A							$T_{K} - 260$ $T_{G} + - 66$ $P_{H} - 7.8$ $T_{M_{g}} + - 36$ $D_{ND_{2}} + ND_{2} - 3.8$ $D_{T} KN - 0.6$ $T_{N_{0}} + - 12$ $T_{SD_{4}} - 22$ $T_{SD_{4}} - 22$ $T_{FD_{4}} - 0.32 (CP - 0.24)$ $D_{NH_{3}} - 0.2$ $T_{K} + - 3$
Ala:1+ In U.S.A							$TN_{0}^{+} - 12$ $T50_{4}^{+} - 22$
() () () () () () () () () () () () () (							$7 PO_2 - D.32 (CP - 0.24)$ D $NH_3 - 0.2$ T $V + 0.2$
					-		
۳.							
·							
2 7 8 8 8 8				-			
Ĩ							
•							
ļ							
•							
-							
	-						
•							
					1		
	•	i	<b>I</b>			i i	•

		SHEET TOTAL	TOTAL		(-{) +	1-11	MIDSTREAM	
	DATE	BOD	COD	C/ <b>-</b>	TKN	TSS	OTHER	
· )	12/20/83	15	150	33	6.4	510	-	
	4/22/64	3.1	]]	20	0.3	380	ρH- B.D "	az-203-3,6 NHz-0,2
	8/30/84	3,1	18	23	0,8		TALK - ZGZ	
							TCa# - 66 pH - 7.8	•
	-						TG# - 66 pH - 7.8 TM,# - 36 DNO2+NO3 - 2.5 DTKN - 0.4	
Alada in U.S.A							$17 N_{p}^{+} - 12$	1
							$T SO_4^2 - 21$ $T PO_4 - 0.281$ $D NH_3^4 - 0.1$	(CP-0,20)
					•		$DNH_{3}^{4} - O_{1}$ TK ⁺ - 3	
VDH1A8								•
Ċ						•		
-								
)								
								•
:								
1			-				<b>`</b>	
.							•	
•								
:								
)   		-					•	
			- 					

	DATA	Sheet		CHERICA.	L IFRHILET IL)	er>	SUBAR KIVER Downstream	
	DATE	TOTAL BODS	TOTAL COD		TKN	T55	OTHER	
	12/20/83	6.8	Z9	30	1.6	13 2 :	- 2	6 5.
	4/22/84	23	ID	13	0.6	366		•
Ma†nInU.S.A	8/30/ 84	24	17	20	DISS D.4 TOTAL ID		$TALK - Zb2  TG + - b6  pH - 7,9  TMg + - 36  DISS N0_{2} + N0_{3} - 3.4 TNa + - 70:TS0_{4} - 21T P0_{4} - 0.26 (@P - 0.17)$	)
							T PO4 - 0.26 (@P-D.17) NH2 - 0.1 T K+ - 2	
						- -		
-								
•						-		
:								
-								
:							-	

A PPENDIX K BRODHEAD : Production II.S.A. NITROGEN BUDGET CALCULATIONS

$$\begin{split} & \bigcup_{ASTE WATER ADDITION} (EACH CELL RECEIVED ½ OF Finit) \\ & CELL 1 = \frac{19750}{10^{6}} \frac{g_{Ay}}{g_{Ay}} \times \frac{44.7}{M} \frac{mg \cdot N}{R} \times 8.34 \times \frac{3650 \text{ M}^2}{72} = \\ & = 26.87 \frac{10}{M} \frac{1}{72} \times \frac{10}{10^{6}} \times \frac{10}{1$$

 $\langle \rangle$ 

J

DENITRIFICATION LOSS

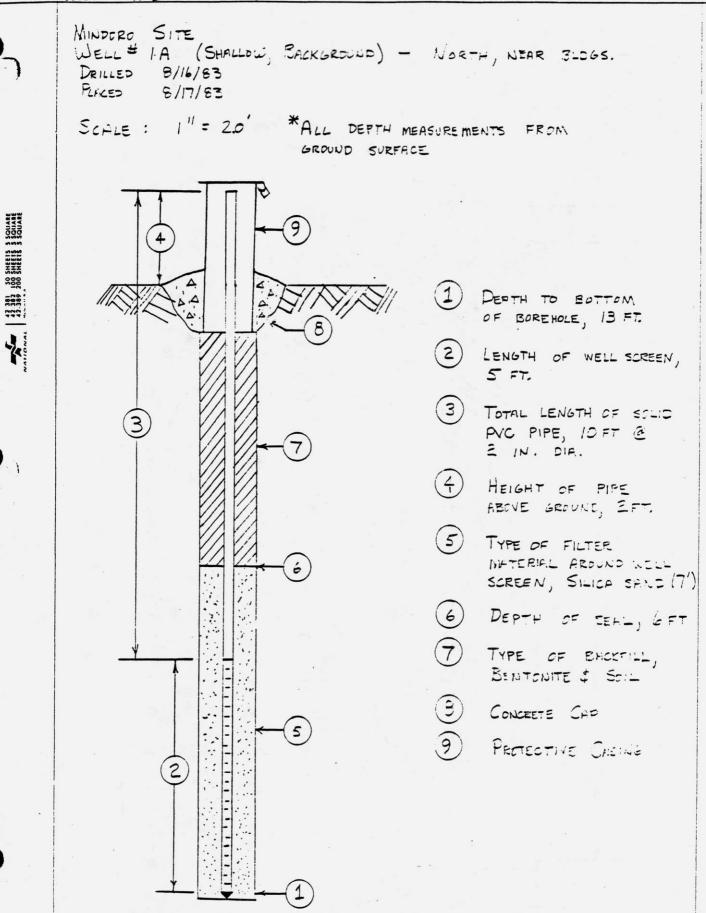
Marte Ig U S.A.

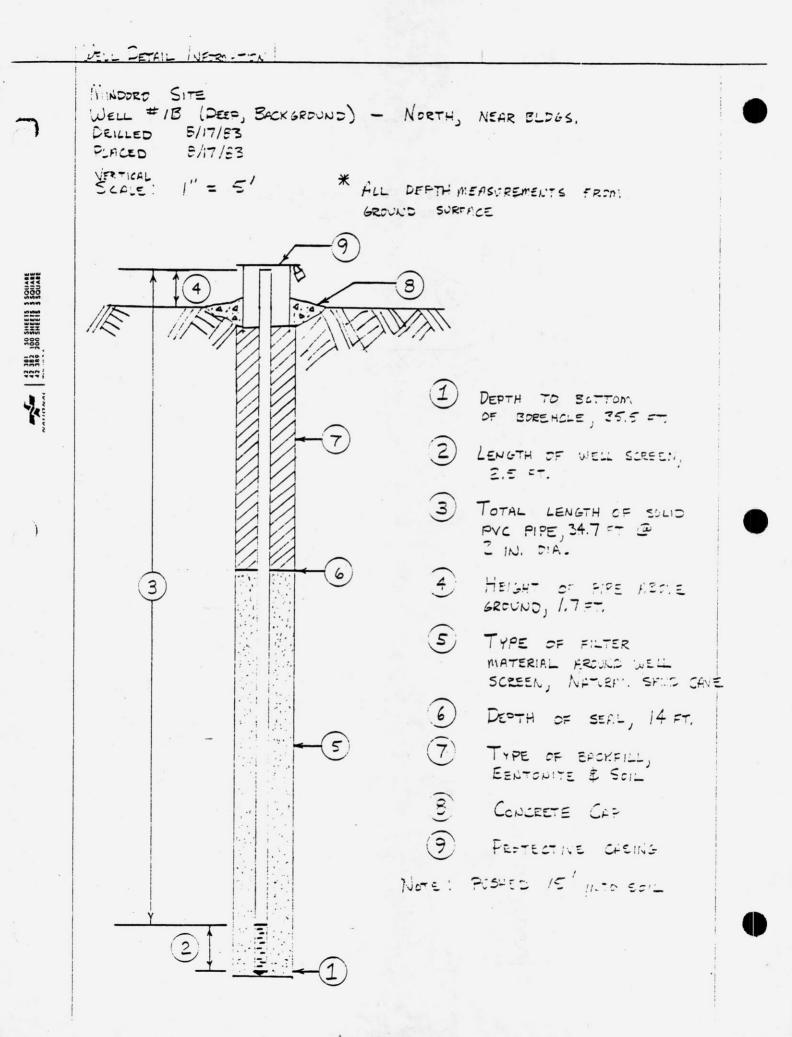
คงกพงจรรรก /

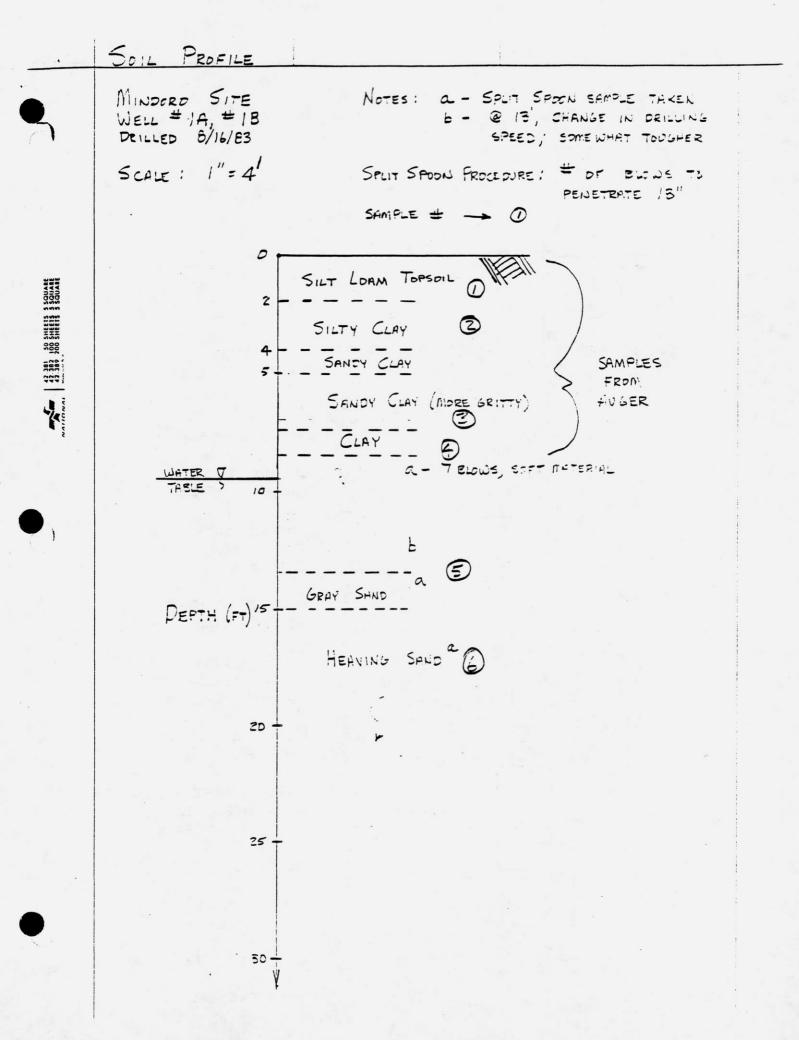
 $\begin{array}{rcl} (E111 = 2687 - 541 = 2146 \frac{b}{yr} (or E0% cF \\ CELL 1 APPLIEON) \\ CELL 2 = 2687 - 1304 = 1383 \frac{b}{yr} (or 51% oF CELL 2 \\ APPLIED - N) \\ TOTAL = CELL 1 + CELL 2 = 3529 \frac{b}{yr} (or 66% cF \\ TOTAL APPLIED - N) \end{array}$ 

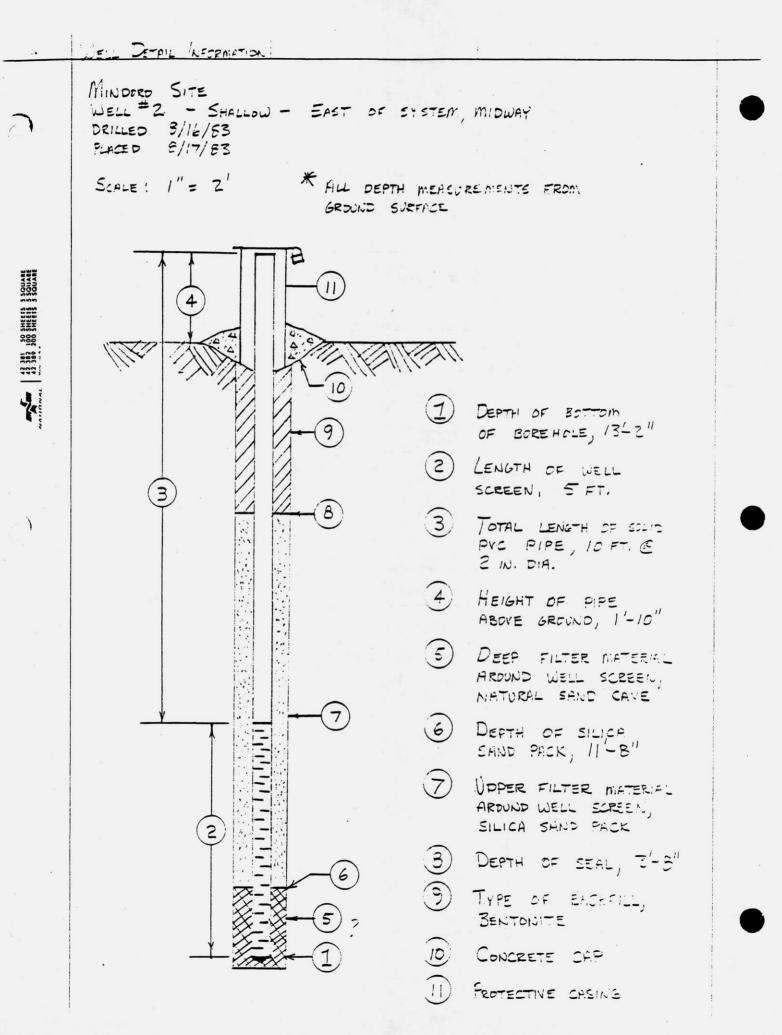
APPENDIX AA MINDORO : EACHAASTER (iii)/ HANDAINUSA WELL AND LYSIMETER LOGS

# WELL DETAIL INFORMATION.



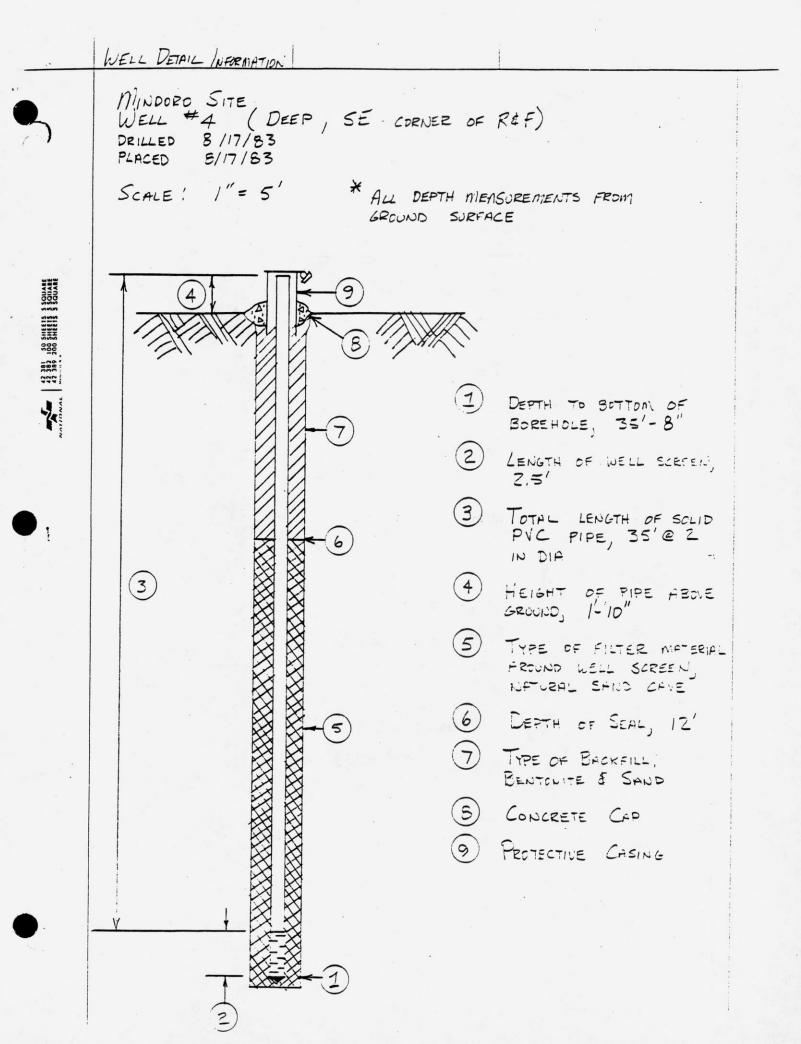




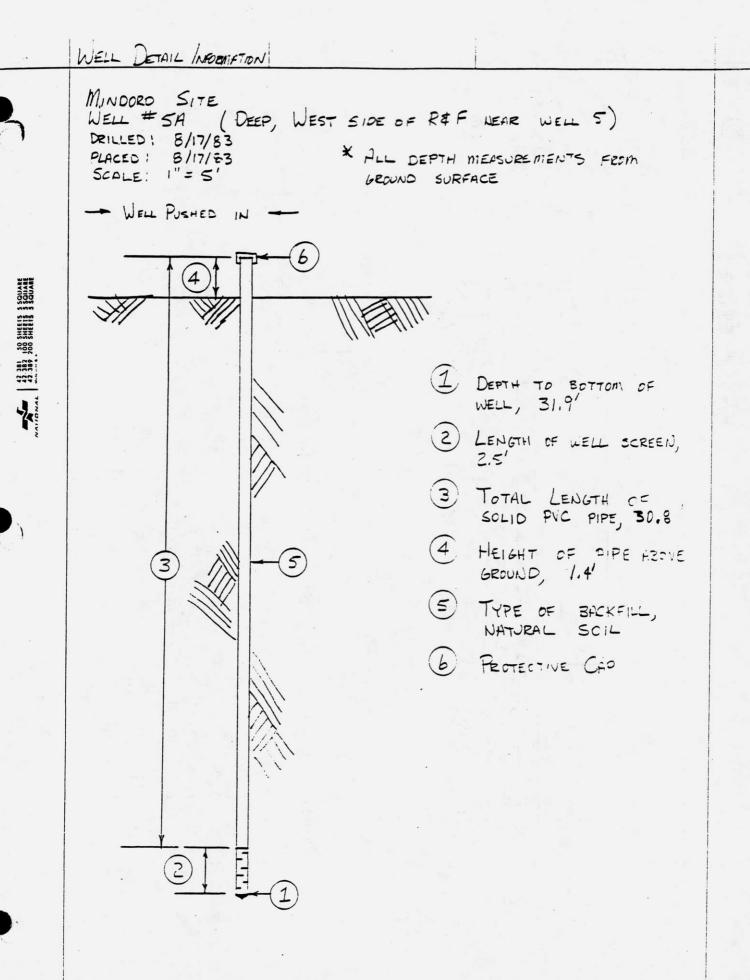


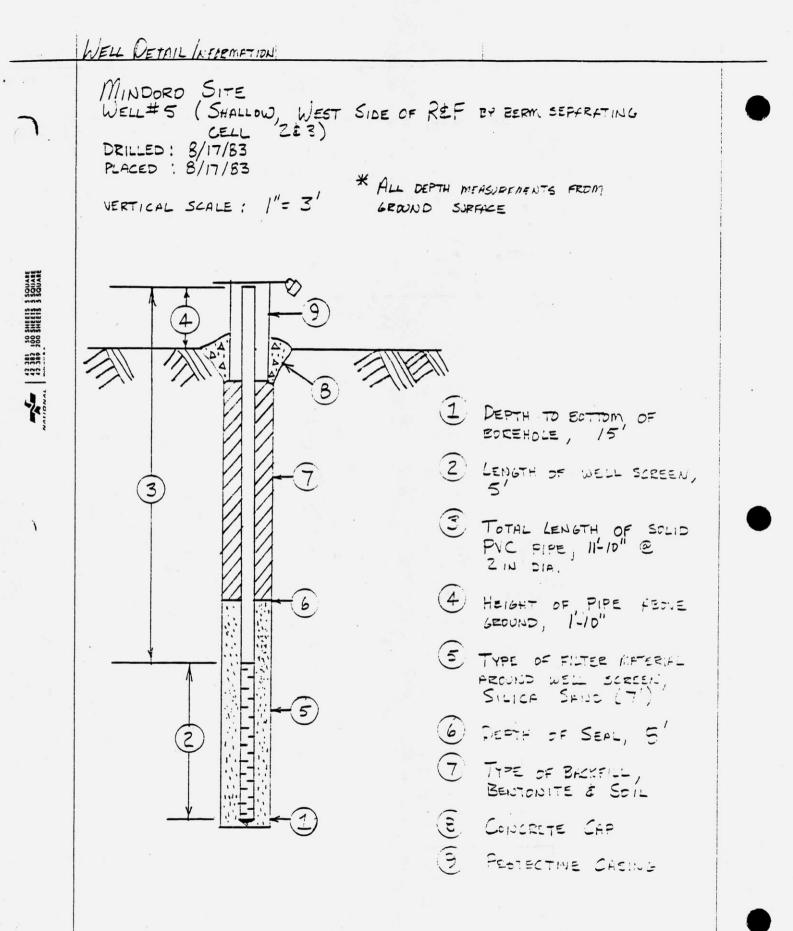
SOIL PROFILE MINDORO SITE Well#2 Drilled B/16/83 VERTICAL : 1"= 2' SCALE : 1"= 2' D 175,811 SILT LOAM TOPSCIL 3 5 DARK BEOWN CLAY DEPTH (FT) 8 LIGHT BROWN CLAY 10 11 SAND AND CLAY 13 GREENISH SAND ( UNIFORM WELL ROUNDED BRANE) 15 .

	WELL DETRIL INFORMETION	
A	MINDORD SITE Well #3 (SHALLOW, SE CORNER OF R&F) DRILLED B/17/83 PLACED B/17/83	•
	SCALE: 1" = 3' * ALL DEFTH MEASUREMENTS FROM GROUND SURFACE	
100 SHEETS 5 SQUARE 100 SHEETS 5 SQUARE		
42.388		
	(1) DEPTH TO BATTOM OF BOREHOLE, 15'-2"	
	3 2 LENGTH OF WELL SCREEN, SFT.	
)	B TOTAL LENGTH OF FILD PVC PIPE, 11-10"E ZIN DIA.	•
	HEIGHT OF PIFE REIVE GROUND, 1'8" FT.	
	TYPE OF FILTER MATERIAL AROUND LOWER WELL SCREEN, NATURAL SHALL CALE	
(e	6 DEPTH OF SAND PACK, 12 FT	
	2 TYPE OF FILTER WATERIAL PROVINE UPPER WILL	
	ECREEN, SHLICA SAND PROK	
	9 TYPE OF EXERCISE DENTROLITE	
	10 CONCRETE CAP	
	T FROTECTIVE CHSING	٠

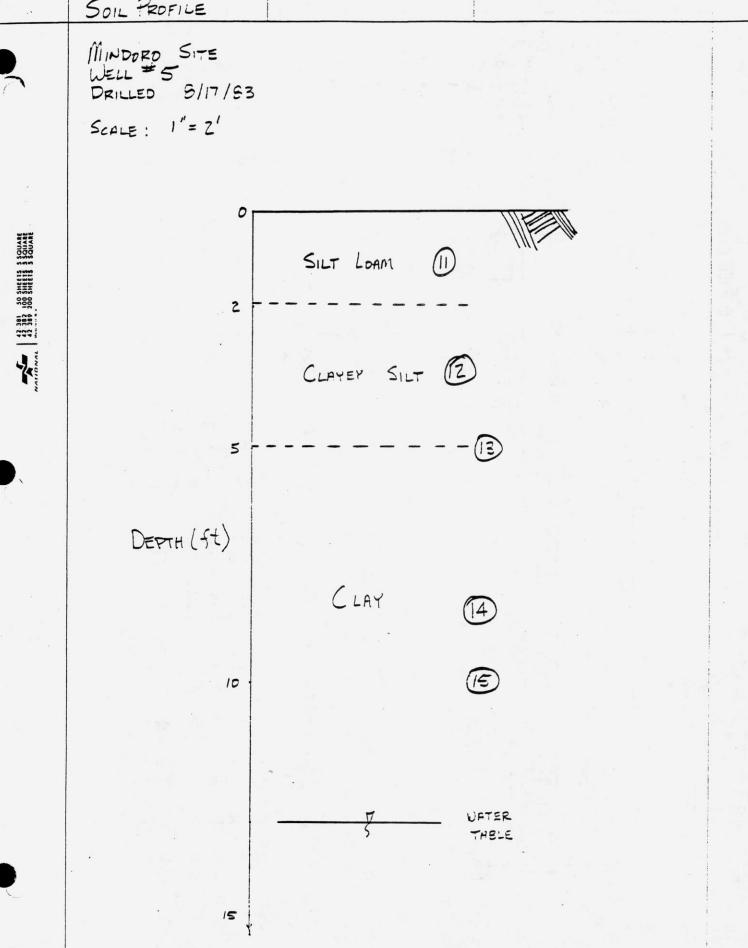


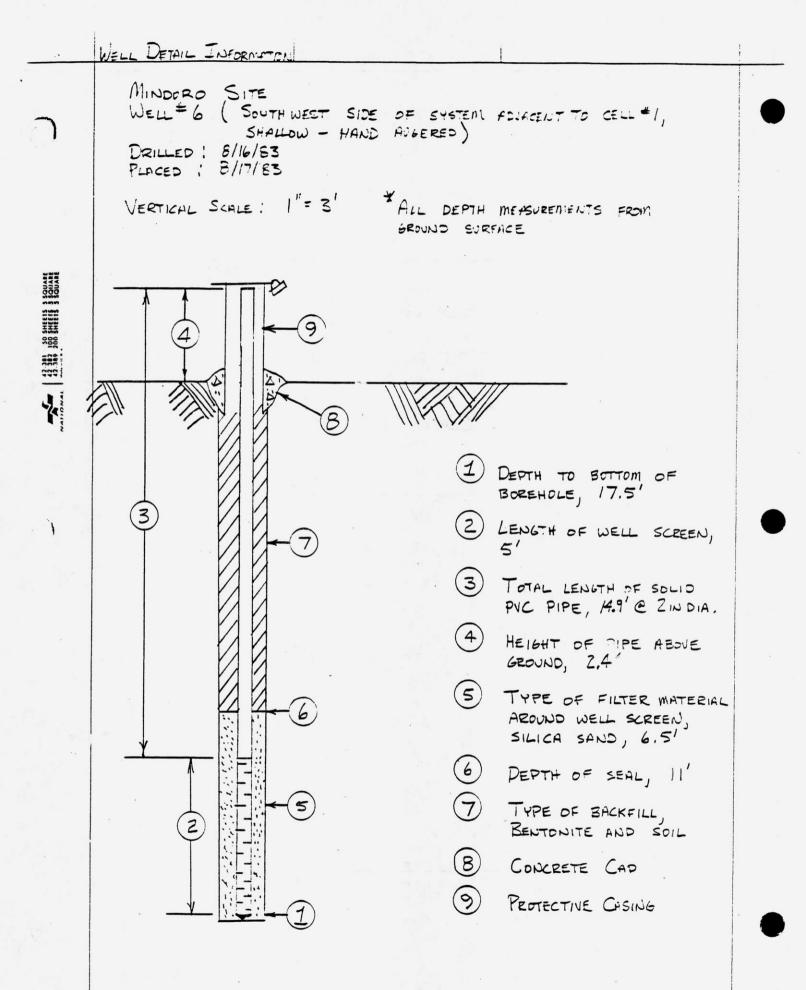
SOIL PROFILE Mindoro Site Well #3 & #4 8/17/83 DRILLED SCALE ! 1"= 3" 0 22 301 300 SHEETS 3 SOULAR SILT LOAM TOPSOIL 2 BROWN CLAY Ð 5 BEDWN- GRAY CLAY 3 GRAY CLAY 0 ) DEPTH (52)10 11 BLUE - GRAY CLAY 10 13 WATER 5 TABLE LIGHT BROWN -BLUE SAND 15-(SATURATES) 20 V

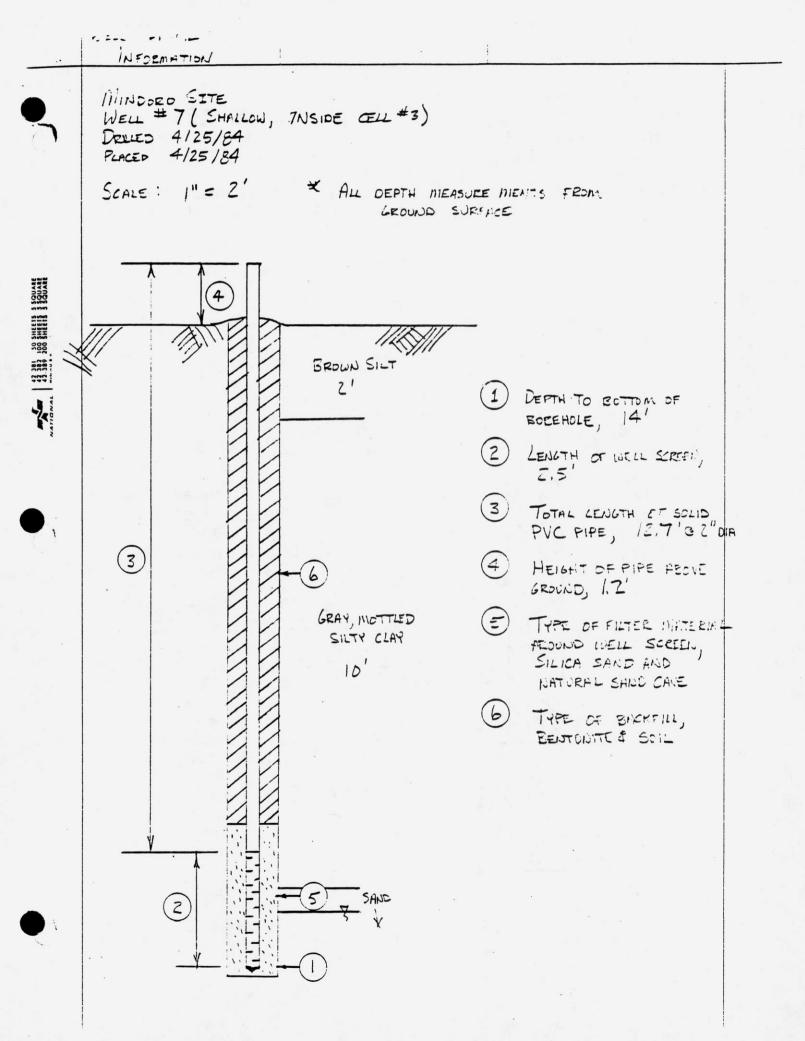


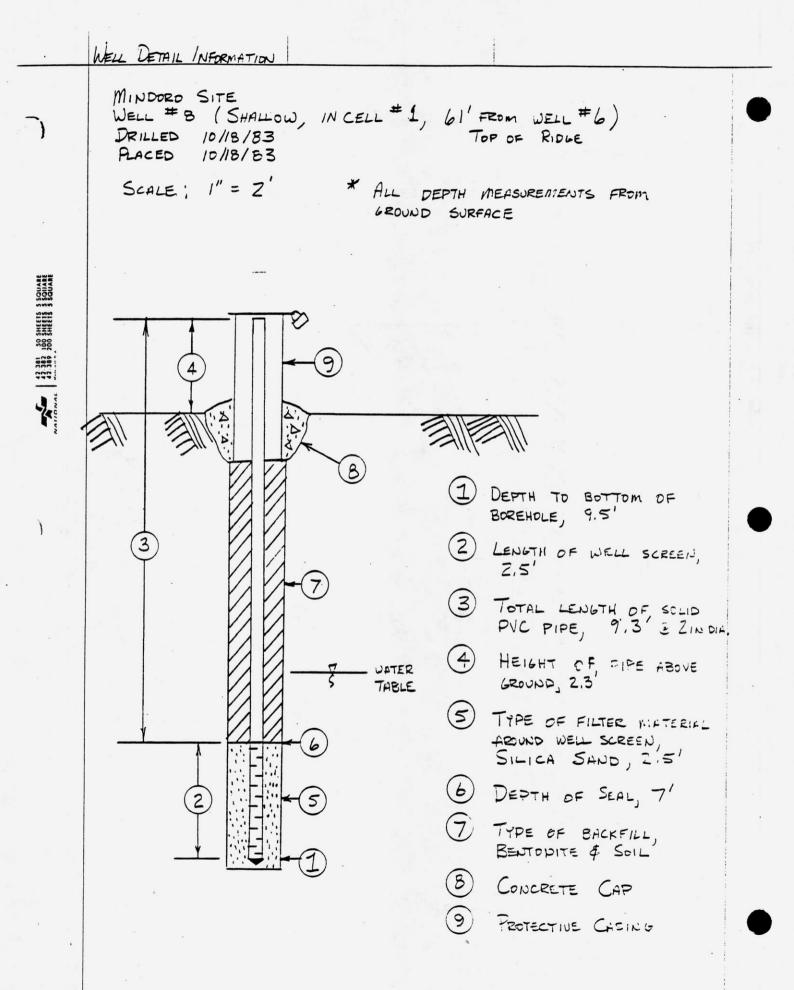


# SOIL PROFILE

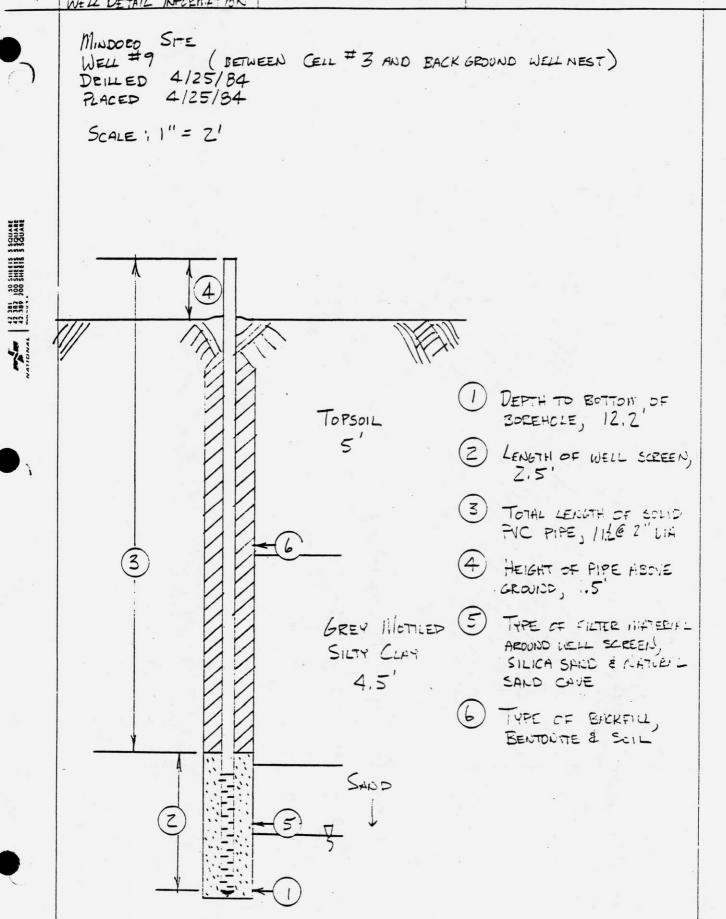


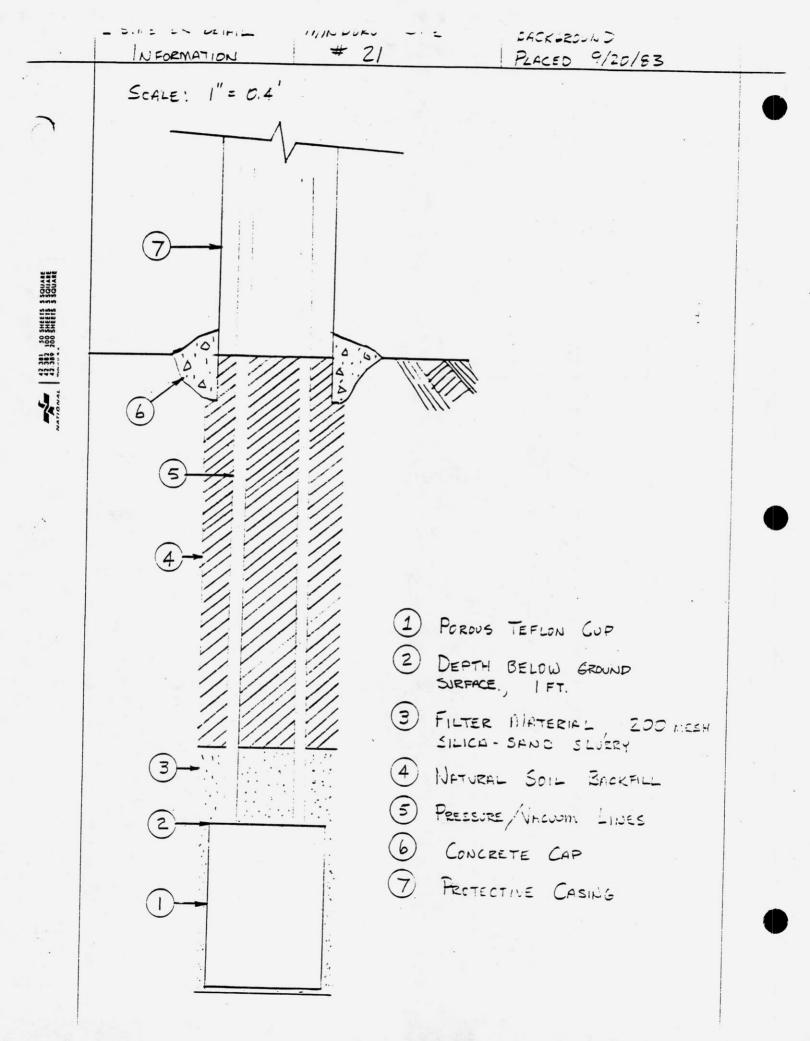


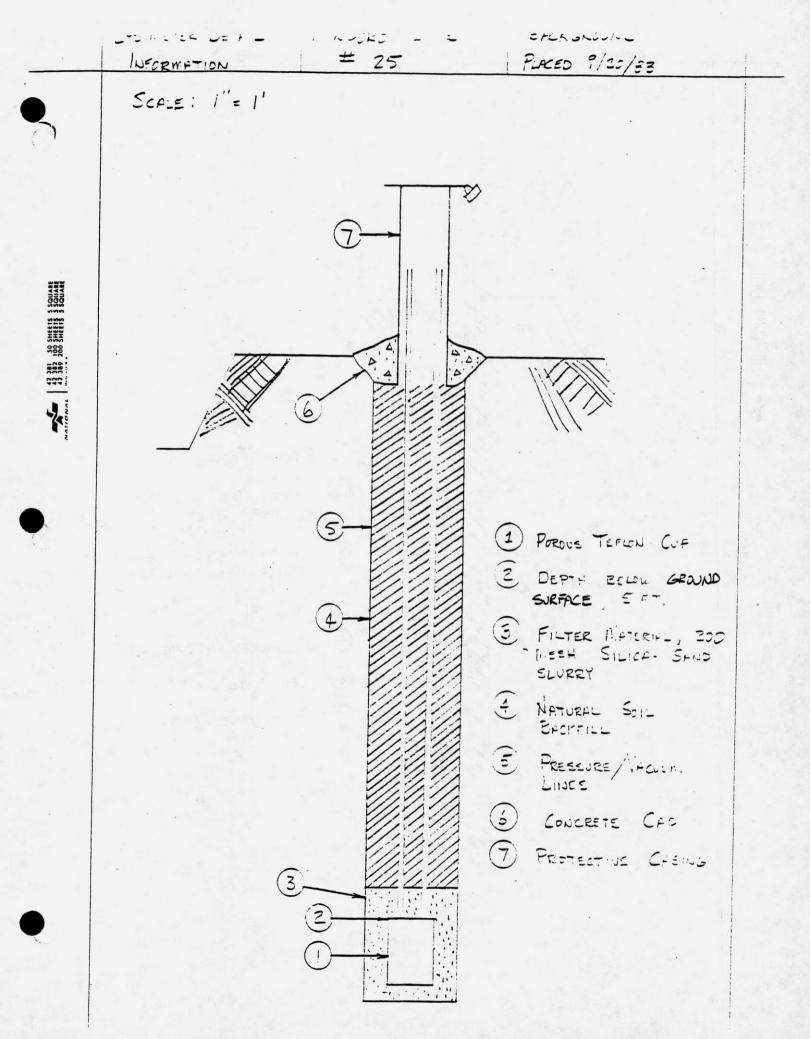


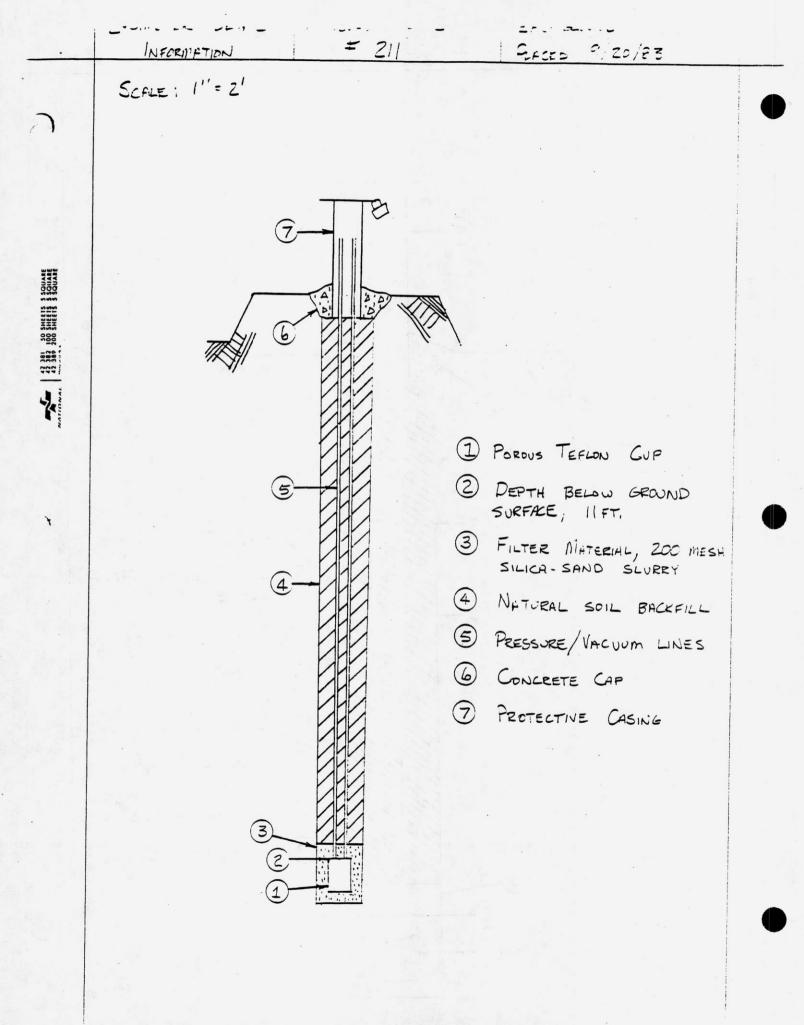


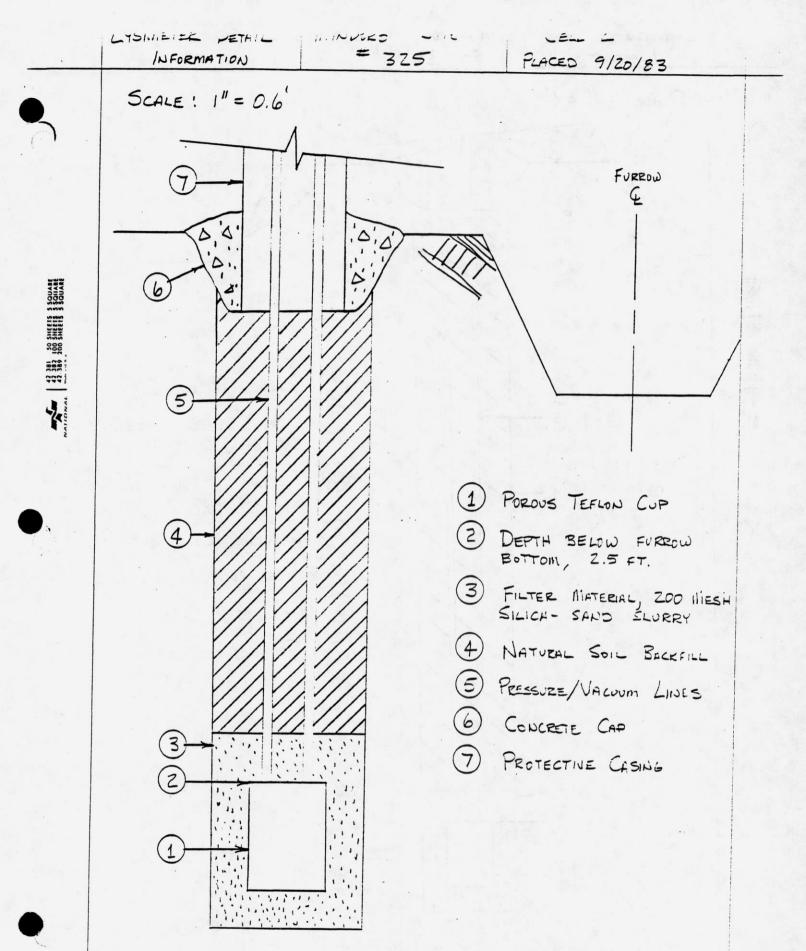
WELL DETAIL INFORMETION

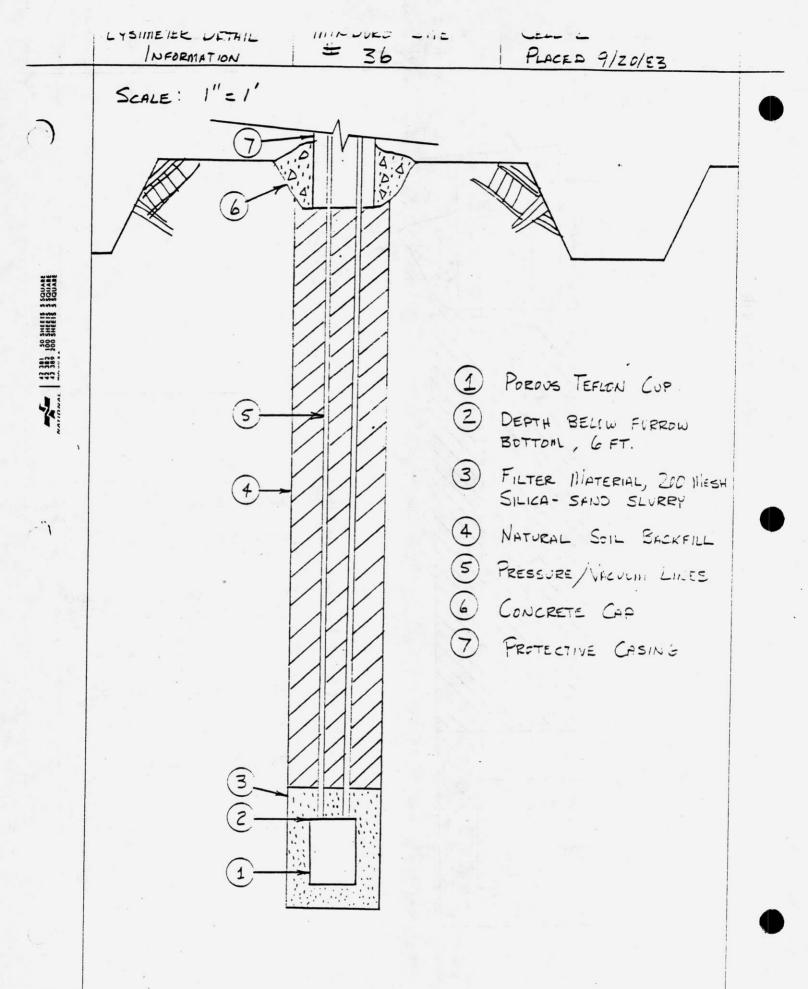


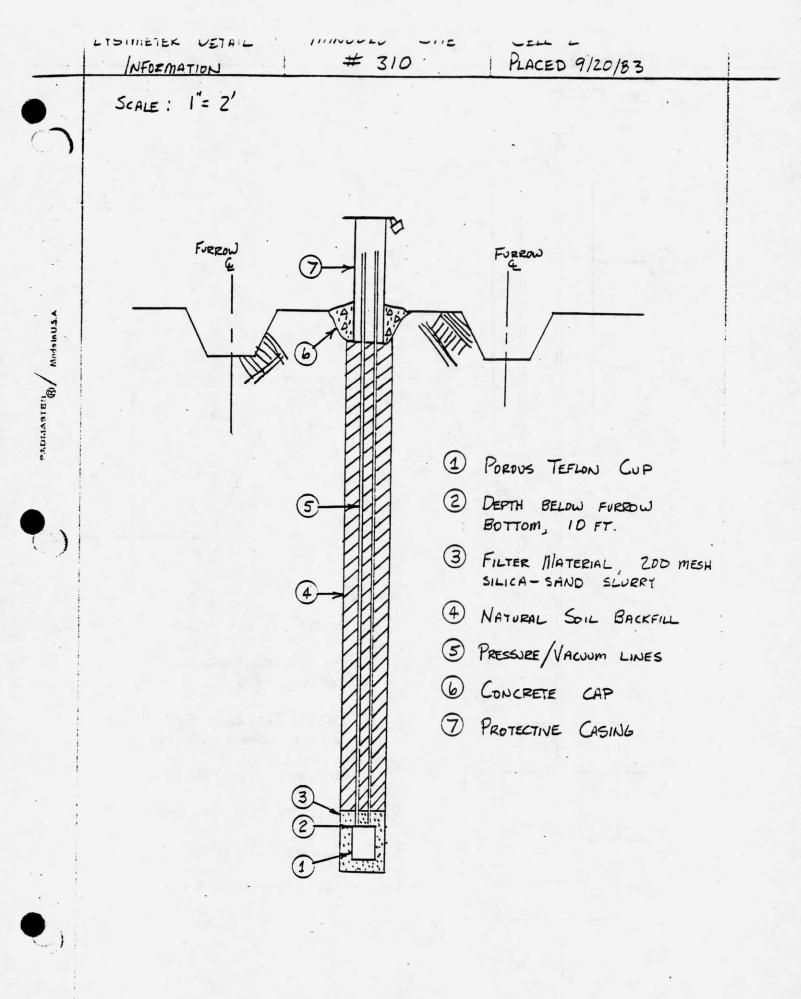


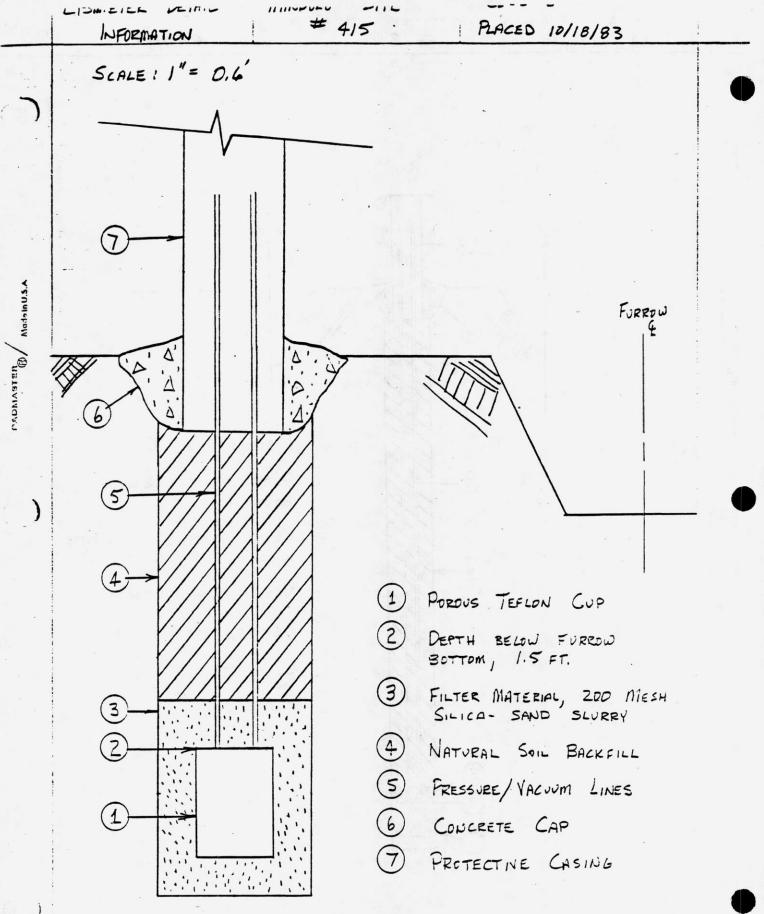


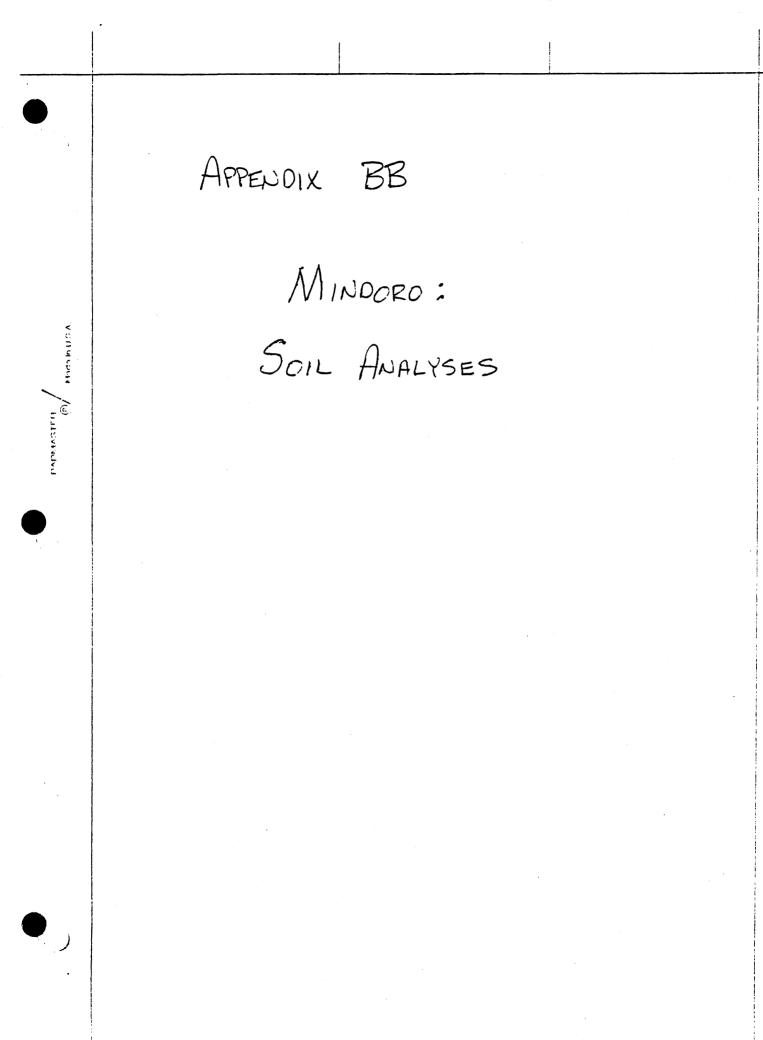












COOPERATIVE EXTENSION PROGRAMS University of Wisconsin-Extension University of Wisconsin-Madison

Soil & Plant Analysis Laboratory, 806 South Park Street, Madison, Wisconsin 53715; 608-262-4364 Street, Madison, Wisconsin 53715; 608-262-4364

## DEPARTMENT OF SOIL SCIENCE

August 31, 1983 Acct. 900 Lab No. S0052

#### MEMORANDUM

<u>TO</u>: Dave Sauer Wis. DNR - Box 7921 Madison, WI 53707

FROM: Soil/Plant Analysis Lab

RE: Results of analyses on 15 soil samples submitted August 19, 1983.

Sample No.	рH	SMP	0.M.	Р	к	 Ca	 Mg	Est	 Total
ICPSCIL 1 CLAY-SILT 2	8.0× 8.0× 7.2 7.5 7.7 7.7 6.0	<pre></pre>	T/A 55 6 5 4 1 1 7 3 4 10 55 17 4 5 4	54 30 34 5 8 x 17 36 91 40 6 40 13 52 7 5	155 230 190 205 105× 155 200 245 240 285 155 200 160 250 225	4050 4050 3000 3450 3000 <i>x</i> 2000 4200 3600 3450 7000 4500 4900 4000 3550 3450	880 800 800 1250 300 × 280 980 1120 1090 1080 1020 1120 1090 1220 1180	CEC 14 14 11 11 9× 6× 15 14 13 22 16 17 15 14 14 14 14	N % 0.31 0.03 0.02 0.05 0.01 × 0.01 + 0.07 0.01 0.02 0.04 0.28 0.10 0.02 0.01 0.02 0.01 0.02 0.01 0.02 0.01

Physical analyses will follow.

If you have any questions concerning these analyses, please feel free to contact us.

Encl.

/sf



University of Wisconsin—Extension • United States Department of Agriculture • Wisconsin Counties Cooperating and Providing Equal Opportunities in Employment and Programming COPERATIVE EXTENSION PROGRAMS University of Wisconsin-Extension University of Wisconsin-Madison

uil & Plant Analysis Laboratory, 806 South Park Street, Madison, Wisconsin 53715; 608-262-4364



## DEPARTMENT OF SOIL SCIENCE

October 7, 1983 Acct 900 Lab No. 00341

### MEMORANDUM

:01	David Sauer	
	Wis. DNR	
	Box 7921	
	Madison, WI	53707

FORM: Soil/Plant Analysis Lab

**<u>RE</u>**: Results of analyses on  $\beta$  soil samples submitted Sept. 22, 1983.

			-	•	
Sample	No.	Sand	Silt	Clay	Total N
				%	
LYSIMIETER 36 1 @ 1.5'	BEOWD SILT	11	64	25	0.11
LYSIMETER 36, 2 Z.5	GREY/ZLUE CLAY (SMELLY)	13	68	19	0.07
LYL INIETER 310 3 @ 9.5'	GREY/ZLUE CLAY (SMELLY) 11	15	64	21	0.03

All additional analyses are attached.

You invoice for these analyses is enclosed.

Encls.

/sf

0-00341 STATE COUNTY ACCOUNT NO WI 32 900 BATE MECHANIN 09-28-83 09-28-83	SOIL & PLANT ANALYSIS LAB Bog S Park Madison Wi B3715	SOIL TEST REPORT Samples Analyzed By: SOIL & PLANT ANALYSIS LAB BOG S. PARK MADISON WI 53715	THIS REPORT IS FOR: DAVE SAUER WIS DNR BOX 71 MADISON	COOPERATIVE EXTENSION PRODUCT UMEX University of University of University Solids Department, Medican, Wis WISCONSIN WISCONSIN WI 53707 FARMER COPY
FIELD 1	CORN ALFALFA		Contain Magning Borgan I	APLI           4
ACRES SOIL NAME GROUP XA	0ATS 70.0 2	2 12 6.9 4 126 EH 310 H 2 14 7.6 4 B L 280 H	3000 M 1000 M 4REY/ 3	LUE CATY I C 25
PLOW DEPTH	SON 1	EST LEVEL CODES VL Nory Low, I Rowl, IM R	ow Madium), M (Madium), HM Bligh Madium), H	Brligh), VH (Very High), EH Encassively High).
LIME PROGRAM			E FERTILIZEA PROGRAM	PLANT NUTRIENTS
GRADE PH or PH USED 6.0 0 6.9 0-09 NONE NONE	VEAR P205 K20 VIELD GOAL BU/A BU/A	N P205 K20 N 1A N1A N1A 140 40 30 3.1-4.0	P205         K20         OTHER CROPS AND V           mi1A         mi1A         01A           50         200         61-90	

250

300

65

75

4.1-5.0

8.1-6.0

CROP:	_	YEAR		CROP:			YEAR	19	CROP:			YEAR	19
ERTILIZATION PROGRAM	PLAN MIA			FERTILIZAT PROGRAM		PLAN PLAN	TNUTH		FERTILI	ZATION GRAM	PLAN	TNUTA	ILENIS
CORRECTIVE				CORREC	TIVE				CO	RRECTIVE			1
MAINTENANCE				MAINTE	NANCE				MA	INTENANCE			
NUTRIENT ADJUSTMENT (S)				NUTRIE	NT ADJUSTMENT (S)	1			NU	RIENT ADJUSTMENT (S)			
TOTAL					TOTAL					TOTAL			1
ERTILIZATION RECORD DATE AND/OR METHOD OF APPLICATION	RATE h)/A		ADE 05 K20	FERTILIZAT	ON RECORD DATE	RATE HILA		ADE 05 K20		ZATION RECORD DATE METHOD OF APPLICATION	RATE hi/A		
					· · · · · · · · · · · · · · · · · · ·					· · · · · · · · · · · · · · · · · · ·			

45

50

160

190

35

40

0

0

121-140

141-180

0

0

SECOND

THIRD

" NONE

NONE

FOR OTHER LIME GRADES SEE LIME SECTION ON BACK

80-89

APPENDIX CC MINDORO : PAPETER // HIVIN IN U.S.A. WASTEWATER CHEMISTRY DATA

	······· [	) ATA SH	HEET	_	METERS	(mg/f)	Kai	~ WAS	TEWATER	2
	DATE	TOTAL BODS	COD	TSS	+TKN*	NH3-N	NO2-N+ NO3-N	+CI-*	(LAE)	OTHER
	10/18/83	< 600	_	110	17.	1.7	0.1	96.	6.9	_
	11/30/63		1300	332	40	1.3	0.2	210	9.Z	
	12/22/83	430	660	80	14,	_	-	70,	-	
	1/11/84	470	077	110	20	_	_	ଟଠ		
	Z/19/84	1300	1600	440	52	_	_	86	9.3	-
Matinu.S.A	3/22/84	7 <b>30</b>	1300	192	40	-	-	120	(FIELD) B.Z	-
~	4/26/34	920	1000	616	40	(0155)	0,5	43 N	9,5	p = 7.3143
Agener.	5/23/34	920	1400	472	34	(C155) 0,9	1.6	95	8.6	_
1445-11	6/6/84	890	1300	404	33	D.6	D. 1	110	8.1	<b>—</b>
	7/12/84	550	850	192	23	0,1	0.1	87	6.4	<b>—</b>
$\langle \cdot \rangle$	B/29/34	93D	1300	212	35	0,4	$\mathcal{D}_{i,1}$	110	5.1	
· )	9/19/34	640	1000	165	25	0.1	<0.1	- 86	6.5	
	10/16/84	1200	1900	149	31 0:55 40 tst	C, 9	0,3	100	7.5	•
	11/6/84	830		195	Z9 TOT 16 DISS	1,4	0.6	97	. —	
:										
• • •			-		-					
:							:			

.

	DATE	TO TAL	TOTAL	504	Ca ²⁺	Na ⁺	Mg Z+	κ+	OTHER	
<b>`</b> )	10/18/23	394	17.	100.	73	160	38	-	-	
	11/30/83	294	34	~	_	-	-	-	-	
	12/22/83	_	-	-	-		-	-	_	- - -
	1/11/84	. –	-	-	-	-	-	-	_	
	2/19/84	-	-	-	-	-			-	
Ma1940.5.A	3/22/84		_	_			.—	-		•
	4/26/84	446	37	57	95	ZCO	23	Z3		
Currentine	.5/23/84	~	_	-	-	-	-	. —	-	•
	6/6/84	~	-		-	-	-	~	-	
	7/12/84	-	-		-	-	-	-	<b>—</b>	
	8/ <i>2</i> 9/84	Z66	45	. —	62	130	35	36		
	•				53.3	163,3	57, 2			
	:				1,02	7.15	0.7E			
-										
		_								
			-						•	

APPENDIX DD

P1114 In U S A.

PACHARTER (

MINDORO :

WASTEWATER FLOWS TO RIDGE AND FURROW

- 30 DAY AVERAGES - 24-HR ON SAMPLING DAYS

MINDORD WASTEWATER FLOW (GALLONS/OAY)

MONTH	1982	1983	1984	1984*	1983*
JANUARY	13200.	11220	13200	13012	
FEBRUARY	13860	10560	13011	18175	
MARCH	13860	13860	13294	13099	
APRIL	17160	16500	13482	13921	
MAY	19140	16594	13011	16155	
UNE	18480	16688	14614	15613	
JULY	17820	15934	12540	13086	
AUCUST	19140	15557	11597	8582	
SEPTEMBER	145ZD	13105	12917	98CB	
OCTOBER	13860	10227		15766	
NOVEMBER	11220	6006		15379	9978
DECEMBER	13860	12634			10263
YEARLY AVERAGE	15510	13244	13074	+ 13295	X
C.MMVLATINE Averabe	15510	)4 377	14022	$\times$	X

* 24 HR FLOWS DETERMINED ON SAMPLING DAYS

PROJECT Z4 - MR FLOW AVERAGE

Plade in U.S.A.

PACMASTER (P)

ر 🖓

APPENDIX EE

MINDORO :

というこう

/ (B) 1000 4000

)

GROUNDWATER ELEVATIONS AND CONTOURS SLUG TEST DATA AND CALCULATIONS

	DATE	DEPTH TO GW	GW ELEVATION	Volume HzJ (St)	Z Volumes (GAL)	Volume Removed	COMMENT
)	9/20/83	10.40	771. :9'	_			
	10/18/53	14.49	767.10	0.14	0.07	0.02	
	11/27/63 11/30/83		767.31 767,02	0.35	0.2	ALL	€ 2:30 PM TOU DRY € 10.30 AM TOU SAMPLE
SOUARE SOUARE SOUARE	12/21/83 12/22/83	14.59 14.61	767.CO 766,98	0.05	0.02	NONE	C 12:18 Fr. Too Day C 9:55 HT. To SAMIPLE
42 391 50 SHEETS 5 SOULAR 42 392 100 SHEETS 5 SOULAR 44 399 200 SHEETS 5 SOULAR	1/10/64 1/11/84	DRY Dey	-		-		E 11:55AM TOU DRY E 9:11AM TO SAMPLE
	7/18/84 Z/19/84	13.89 14,02	767.70 767.57	0,73	0,36	l volume	ЦЬНТ ВЕОШЛ (10:40) (9:25)
	3/21/84 3/22/84	14,50 14.55	767.09 767.01	0.12	0.06	NONE	TOO DET TOO SAMPLE
	4/25/34	DRY	_	-	-	-	
	5/22/34	dry	-	-	-		<b>-</b> *
	6/5/34	DRY	-	~	~	~	-
	7/11/84	Dry	-	-	_	_	-
	8/25/84	DRY	-	-	-		
	9/18/84	DP.Y	_	-		_	-
	10/15/34	dry	_	-	-	-	-
	11/5/32	14,50	767.09				<u> </u>
							-

ę. ę.		DATA	Sheet : We	ELLIB LEN	GTH : 37.2	/ 37.13/37.11	DEEP,	BRCKLRDUND
•		DATE	DEPTH TO SW	GWI ELEVATION	Nolume HD (ft)	3 Volumes (GAL)	Volume Renoved	COMMENT
		9/20/83	1.55'	769.51	_	_		
	J	10/15/53	14.69	766,67	22.6	11.3	-	—
	ſ	11/29/83 11/30/83	4.36 14.53	767.00 766,83	22.77	11.4	- 15	CIZ:40 PM CIC:40 AM LIGHT BROWN
SQUARE SQUARE SQUARE	1	12/21/83 12/22/83	-	766.34 766.22	22.10	11.05	12	C 12:20 PM. C 9:57 AN NO SMELL
42.381 50 SHEETS 5 SOUARE 42.382 100 SHEETS 5 SOUARE 42.387 200 SHEETS 5 SOUARE	! √	1/10/B4 1/11/84	15.30 15.29	766.06. 766.07	21.87	10.97	12	E 11:56 AM C9:11 AM CLEAR
1933		Z/18/84 Z/19/84	13.85 13.80	767.51 767.56	23.3	11.67	17)	(12:40) LIGHT BROWN (9:30)
T ex		3/21/84 3/22/84	14.64 14.77	766.72 766.59	ZZ. <del>.</del> 5	11. 2	12	CLEAR
	Ţ	4/25 <i>/</i> 34	14.87	766.49	22,2	)].]	12	_
`	1	5/22/34	14.89	766.47	-	-	-	fict same is
	•	6,5,34	15.03	766.23	<b>—</b>	· _	-	• 7
	t	7/11/84	14,66	766.70	-	~	-	ι <b>ι</b>
	ʻj	8/2 <i>5/</i> 84 3129/84	5.34  5.37	766.02 765.99	Z1. B	10.9	12)	CLEFE
	~	9/18/84	15.31	766.05	-		-	·
	<b>;</b> *	10/15/84	15,19	76.6.17	<u> </u>	-		
	{	11/5/34	14.55	766.81	·	_		<u> </u>
						-		
		1						

*	LAIA -	HEET ! WEL	GW	6TH ; 15/4; VOLUME H2D		VOLUME	1
	DATE	64	ELEVATION	(+2)	(GAL)	REMOVED	COMMIEN-
	9/20/83	9.30'	774.31				
	10/16/63	11.90	17.10	3.0	1.5	-	
	11/ <i>2</i> 1/83 11/30/83	10.76 12.73	772,85 772,85	A 3.97	42	ALL	EZ:50 EID:50 BLACK, TUREID
	12/21/83 12/22/83	13.22 13.69	770.29 769.92	1.39	0.7	-	E 1:00 PM E 10:12 AM GRY NO SME
2000 SHEETS 2000 S	1/10/84 1/11/84	13.85 13,83	769.76 769.78	0.8	0.4	VOLUME	E 12:04 PM E 9.49 ANL GEAY
*	2/18/84 2/19/84	12.48 12,48	771, 13 171, 13	2.2	1.1	1 VOLUME	(10:53) ВР-ШМ (10:01)
	3/21/84 3/ <i>2</i> 2/84	13.47 13.55	770.14 770.06	1.2	0.6	/ VCLUME	LIGHT BROWN
	4/2 <i>5/</i> 34 4/26/34		770.CZ 770.71	1.1	0.6	I V CLUME	
· )	5/22/34 5/22/84	13.62 13.65	769,99 769,96	).	0.5	VOLUNE -	ZIGHT EFOUND
	615154 616184	13.72 13.75	769,89 769.56	1.0	0,5	/vsiume —	LIGHT ESCLAR
	7/11/84 7/12/54	13,56 13,63	770.05 769.98	1.1	0.6	l volune —	LIGHT BROWN
	5/28/34 8/29/84	14.02 14.02	769,59 769,59	6,7	0.4	lyelume	Reown
	9/18/84	14.01	769,60	0.7	<i>C</i> .4	/ VOLUM: E	LIGHT SROWN
	10/15/34	13.52	770.09	1.2	0.6	IVAUME	BEOWN
-	11/5/84	13.40	770,21			_	
					-		

DATE	DEPTH TO GW	GW ELEVATION	VELOME HZD (ft)	3 VOLUMES (GAL)	NOLUME REMOVED	COMMENT
9/20/83	10.95'	774.37				
10/18/83	13.16	772.16	3.52	1.76		_
11/29/83 11/30/83	13:01 13:08	772.31 772.24	3,5	1.75	√5 _	C 3:25 C10:55 BROWN TURBID
12/21/E3 12/22/E3		771.95 771.92	3.12	1,56	WELL Volume	C 1:15 PM LIGHT BROWN C 10:25 MI NO SHELL
1/10/84 1/11/84		771.86 771.85	3,0	1.5	WELL VOLUME	C 12:19AN WELL CAP OFF C 10:00 AM LIGHT BROW
Z/18/64 2/19/84	12.37 12.36	772.95 772.96	4.1	2.1	/ VOLUME	(11:05) (10:20) BRDWN
3/21/84 3/22/84		772.17 772.08	3.3	1,7	/ VOLUME	LIGHT BROWN
4/25/54 4/26/84		772. 13 772. 13	5.3 L	1.6	I VOLUME	
5/22/84 5/23/84	13.28 13.32	772.04 772.00	3.2	1.6	Ivalunt	CLERE
615/84 6/6/84		771.96 96,177	3.1	1,5	I VOLUME	LIGHT BEOULL
7/11/84 7/17./5 <u>4</u>	13,44 1343	771.85 771.34	3.0	1,5	IVOLUME	LIGHT RECUN
3/28/54 E/29/24		771.62 771.62	z,8	1.4	lve: une	LIGHT BEACH
9/18/34	13.33	771,99	32	1.6	IVOLUME	VERY LIGHT BROWN
10/15/84	12.97	772.35	3,5	1.8	I VOLUME	LIGHT BROWN
11/5/84	3.  4	772,18				-

	GATE	DEPTH TO GW	6W ELEVATION	$Volume H_2 O$ (-ft)	3 VOLUMES (GAL)	Volume Removed	COMMENT
5	9/20/83	10,93	774.35				
-	10/18/63	13.09	772.19	17.7	5.85	-	-
	11 <i>/29/83</i> 11/30/83	12,9 <del>8</del> 13.01	772.30 772.27	23.55	u12 	- 15	С 3:25 С 11.05 УЕЦСИ- Верин Тле
	12/21/33 12/22/83	3.29  3.31	771.99 771,97	22.22	11.61	12	G 1:17PM 3 10 26 ft clear 200 sach
42 389 200 SHEETS 3 SQUARE	1/10/84 1/11/84	3.38  3,37	771.90 1,91,91	23.1	11.5	- 12	E 12:21 PM E 10:01 AM CLEAR
	2/18/84 2/19/84	12.35	772.93 772.97	- 24.1	12, 1	12+	(11:05) BEOUN (16:21)
	2/21/E4 3/22/84	13.07 12.13	772.21 772.15	23.4	11.7	12	BROWN
	4/25/84 4/26/84		772.17 772.16	 23.4	11.7	_ 1Z	
	5/22/84 5123/84	13.20 13.23	22,25 2,05	23.3	11.6	12	LITT BROWN
	6/5/34 6/6/E4	13.27 13.27	772.01 772.01	22.2	11.6	- 12	List Besurd
	7/11/8 4 7/12/84	13.34 13.39	771,94 90,177	23.1	11.6	12	LIGHT BROWN
	8/28/54 8/29/84	13.63 13.53	771.65 771.70	72,8	11.4	12	CLIPE
	9/18/94 9/19/84	13.58 13.55	771.70 771.73	23.0	11.5	 12	CLEAR
	10/15/84	12.80	772.48	23.7	11.8	12	BRWRISH CLEAR
	11/5/84	13,16	772,12				

	DATE	DEPTH TO GW	GU) ELEVATION	Volume HD (fl)	3 VOLUMES (GAL)	Volume Removed	Comment
)	9/20/83	12.32'	770.31		· · · · ·	-	_
	10/18/23	<i> </i> 4,21	765,42	Z.5	1.25	-	_
	11/ <i>29/8</i> 3 11/30/83		769,10 765:60	3	1,5	ALL	@4:10 @11.5D TAIRLY LLEAR
SQUARE	12/21 /83 12/22 /93	4,53  4.63	768.10 762.00	2	<u>/</u>	3	© 2:45 PM © 11:18 AM ERDUN - GRA ST.ELLY
42.309 200 SHEEIS 5 50UARE	1/10/84 1/11/84	14.77 14.73	767.86 767.90	1.8	0.9	2 -	С 1:05 PM С 11:03 Am Вешил-серч
17.30 	2/13/84 2/ <i>P</i> /84	13.62 13.51	769.01 769.12	2,9	1.4	4	(11:35) BROWN - GEEEN (11:25) OPOR
	3/21/84 3/22/84	14,33 14,35	768,30 768,25	2.2	1.1	- 1.5	FLECK - GREEF N SVOR
	4/25/84 4/26/84		765.16 768.18	2.0	1.0		
	5/22 <i>/</i> 84 3/23/84	14.26 14.40	728.27 768.23	2.1	1.1	- 2	DJLI CLEVE OP SE
	6/5/34 6/6/34	14,55 19,55	763.03 768.05	1.9	1.0	- 2	SEEN TERMIN - SOOR
	7/11/84 7/12/34	14,27 14,34	763.36 768,29	2,2	. i	1,5	.HEISH, ODIR
	8/28/24 8/29/24	4.65	767.95	1.9	9.9 -	— i	JELLON, OPOR
	9/13/84 9/17/84	14,75 14,73	767,88 767,90	1.8	C.9	1	YELLOW · DOOR
	10/15/24	14.33	765.30	2,2	1.1	1.5	LIGHT TELLOUISH DEDIND
	11/5/94	14.22	768.41		-		
		· · · · · · · · · · · · · · · · · · ·					
		:					

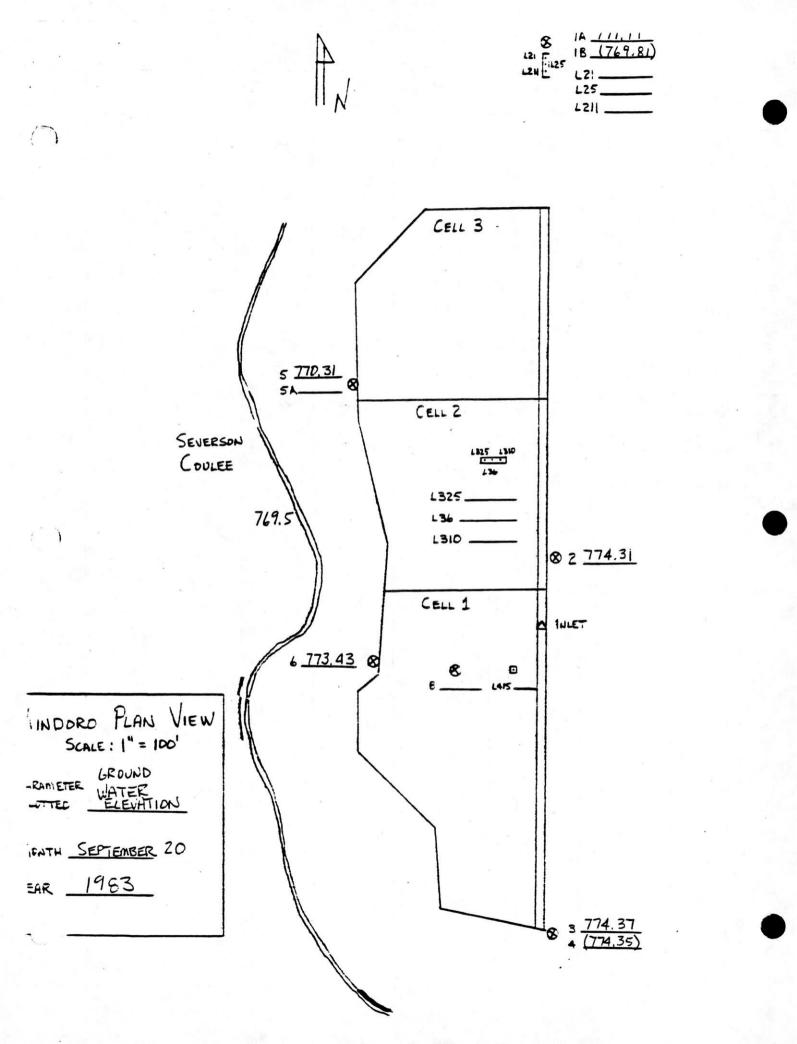
	DATE	DEPTH TO GW	GW ELEVATION	6TH 133.3/52.2 Volume H20 (ft)		Vowme Removed	COMMENT	
)	-							
	10/18/83	13.97	768.34	V.8.2	9.1.	-		
	11/30/83	13.87	768.44					
	12/21/83	14.25	768.56				C Z: 4 - P/r	
	1/10/84	14.41	767.90				C 1:0LPM	
	Z/18/84	13.07	769.24			/	(11:35)	
	3/21/34	13,99	763.32					
TENGILEN	4/25/54	4,13	763.13				_	
***	5/22/84	14.16	768.15					
	613/34	14.24	763.07	V			-	
	7/11/84	13.98	768,33		$\setminus$ /			
	5/2.8/E4	4,41	767.90		X		-	
	9/18/84-	14.43	767.88					and the to
	10/15/34	14.05	768.26		/			
	11/5/34	13.90	768.41					
						-	• • • • • • •	
					V			
						\ \.		
					·			
				- 4 - 4 - 7 - 7 - 7				- Marine -

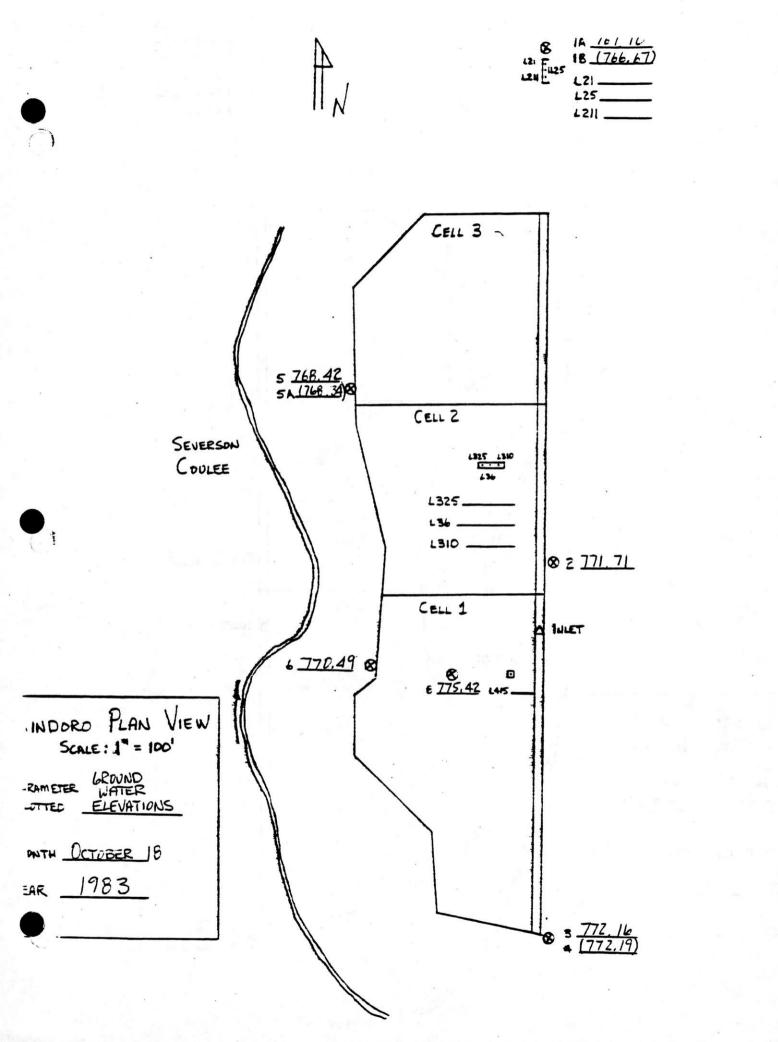
DATE	DEPTH TO GW	6W ELEVATION	Volume Hz D (ft)	3 VOLUMES (GFL)	Volume Removed	COMMENT
9/20/83	12.26	773.43				
10/18/83	<i>15.2</i> 0	770.49	5.0	2.5	_	·
<i>  29/83</i>    <i> 30/</i> 83		770.41 770.31	4,2	2.1	~5 -	@ 4:00 @ 11:40 FAIRLY CLEAR
IZ/21/83 IZ/2Z/83		770,07 776,05	3.50	1.93	4	& 2110 PM BLACK SILT C ID:58 AM LITLE CALL
1/10/87 1/11/84-	15.72 15.71	769.97 769.98	3.76	1.9	4	@ 12:45 PM @ 10:54 AM BLA(KIS H
<i>Z/18/<del>04</del> 21/17/84</i>		770,95 771.01	4.7	 Z.4	4	(11:30) BLACK (11:10) ODOR
3/21/54 3/22/84		770.29 770.22	4.1	2.0	- 4	GREY CDOR
4/25/84 4/26/84		770.19 770.19	- 4, <del>c</del>	 Z . D	/ _(v)	_
5/22/84 5/23/34	1	770,12 770,05		1.9	_ Z	GREY
6/5/84 6/6/54		770.06 770.55	3.5	1.9	- ~,	GCEY
7/11/84 7/12/84		770.10 770.03	<u> </u>	; <del>9</del>	- 2	DARK GREY
8/28/E4 E/29/E4		769,82 769,82	3.6	1,8	- 2	LEEY
9/13/34 9/19/84	15.85 15.82	769,84 769,87	3.7	1,8	Z	LIGHT GREY
10/15/84	14,93	770.71	4.5	2.2	3	GREVISH
11 <i> 5 </i> 84	15.41	770,28	—		-	
	•					

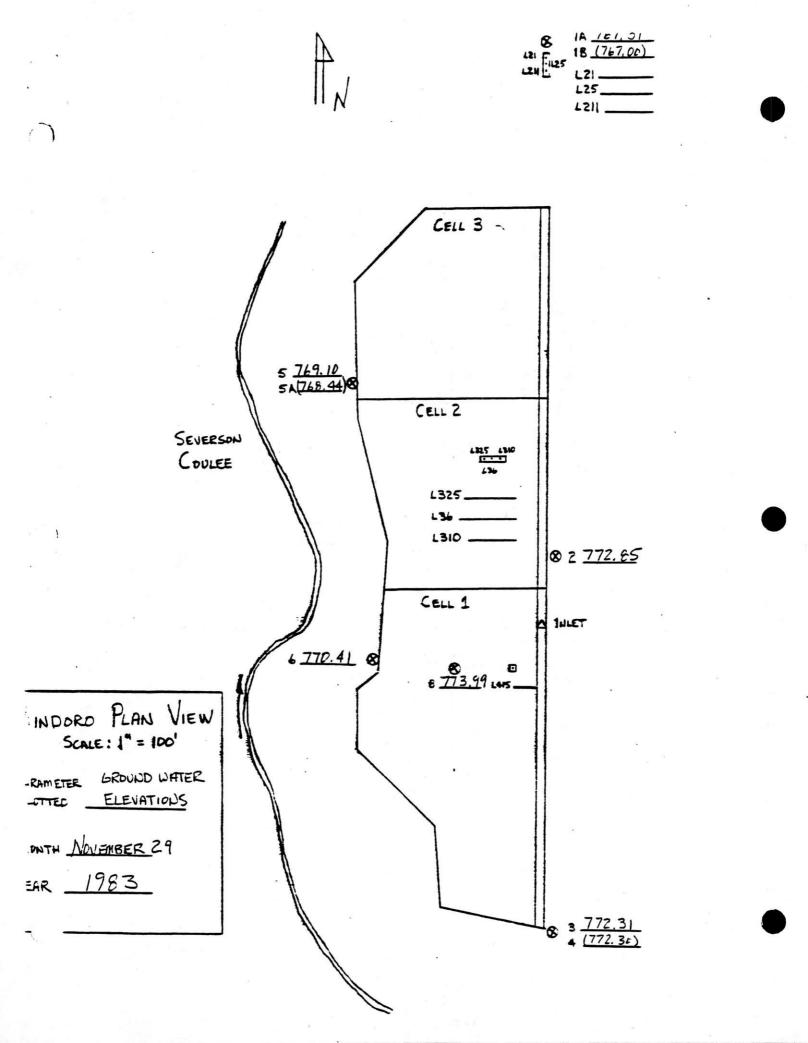
	L7 LEI	JGTH : 15.22		يند ٢ ٢٠٢٢ -	Ce41 5	
E DEPTH TO	GW ELEVATION	Volume Hz D (ft)	E VOLUMES (GRL)	Voume Renoved	Comment	
84 13,63	768,37	1.6	0.5	Z	-	
	1		0.B	1.5	DULL CLEAR ODDR	
	768.22 768.20	1.0	0,5	1.5	GREY EROWN	
	768,53 769,44	/,B	0,9	 1.D	LIGHT BROWN	
	768,00 768,00	1.3	0.6	1	FELLOW - BROWN, DOOR	
34 13.99 84 13.99	768.01 765.01	/,3	0.6	1	LIGHT YELLOW	
13.67	768.33	1.6	0.5	1.5	YELLOW EEDWAL	
[3,33	768.66					
	184 13,63 164 13,64 164 13,68 13,68 13,78 13,78 13,78 13,78 13,78 13,78 13,78 13,78 13,79 14,00 184 13,99 13,99 13,99 13,99 13,99 13,99	13, 63 $768, 37$ $164$ $13.64$ $768, 36$ $164$ $13.68$ $768, 36$ $164$ $13.68$ $768, 32$ $14$ $13.68$ $768, 32$ $14$ $13.78$ $768, 22$ $14$ $13.78$ $768, 22$ $14$ $13.78$ $768, 22$ $14$ $13.78$ $768, 22$ $14$ $13.78$ $768, 22$ $14$ $13.78$ $768, 20$ $14$ $13.78$ $768, 53$ $14$ $13.99$ $768, 00$ $768, 01$ $768, 01$ $768, 01$ $84$ $13.99$ $768, 01$ $84$ $13.99$ $768, 01$ $768, 33$ $768, 33$ $768, 33$	13.63 $768.37$ $1.6$ $164$ $13.64$ $768.36$ $ 164$ $13.68$ $768.32$ $1.6$ $14$ $13.68$ $768.32$ $1.6$ $14$ $13.68$ $768.32$ $1.6$ $14$ $13.78$ $768.22$ $ 13.78$ $768.20$ $1.0$ $13.30$ $768.20$ $1.0$ $13.47$ $768.53$ $ 13.47$ $768.53$ $ 13.47$ $768.53$ $ 14.00$ $768.00$ $1.3$ $184$ $13.99$ $768.00$ $1.3$ $13.99$ $768.01$ $1.3$ $13.99$ $768.01$ $1.3$ $13.99$ $768.33$ $1.6$ <td>13.63 $768.37$ $1.6$ $0.8$ $154$ $13.64$ $768.36$ $154$ $13.64$ $768.36$ $154$ $13.68$ $768.32$ $1.6$ $0.8$ $154$ $13.68$ $768.32$ $1.6$ $0.8$ $13.47$ $768.22$ $13.30$ $768.20$ $1.0$ $0.5$ $84$ $13.47$ $768.53$ $84$ $13.99$ $768.00$ $1.3$ $0.6$ $184$ $13.99$ $768.01$ $84$ $13.99$ $768.01$ $1.3$ $0.6$ $84$ $13.99$ $768.01$ $1.3$ $0.6$ $84$ $13.99$ $768.01$ $1.3$ $0.6$ $84$ $13.99$ $768.23$ $1.6$ $0.8$ $0.6$ $85$ $13.67$ $768.33$ $1.6$ $0.8$ $0.6$ $0.8$</td> <td>13.63 $768.37$ $1.6$ $0.5$ $2$ $164$ $13.64$ $768.36$ $164$ $13.68$ $768.36$ $164$ $13.68$ $768.32$ $1.6$ $0.8$ $1.5$ $-$<!--</td--><td>13.63 $768.37$ $1.6$ $0.6$ $Z$ $164$ $13.64$ $768.36$ $-$</td></td>	13.63 $768.37$ $1.6$ $0.8$ $154$ $13.64$ $768.36$ $  154$ $13.64$ $768.36$ $  154$ $13.68$ $768.32$ $1.6$ $0.8$ $154$ $13.68$ $768.32$ $1.6$ $0.8$ $13.47$ $768.22$ $   13.30$ $768.20$ $1.0$ $0.5$ $84$ $13.47$ $768.53$ $  84$ $13.99$ $768.00$ $1.3$ $0.6$ $184$ $13.99$ $768.01$ $  84$ $13.99$ $768.01$ $1.3$ $0.6$ $84$ $13.99$ $768.01$ $1.3$ $0.6$ $84$ $13.99$ $768.01$ $1.3$ $0.6$ $84$ $13.99$ $768.23$ $1.6$ $0.8$ $0.6$ $85$ $13.67$ $768.33$ $1.6$ $0.8$ $0.6$ $0.8$	13.63 $768.37$ $1.6$ $0.5$ $2$ $164$ $13.64$ $768.36$ $    164$ $13.68$ $768.36$ $     164$ $13.68$ $768.32$ $1.6$ $0.8$ $1.5$ $                                                                          -$ </td <td>13.63 $768.37$ $1.6$ $0.6$ $Z$ $164$ $13.64$ $768.36$ $-$</td>	13.63 $768.37$ $1.6$ $0.6$ $Z$ $ 164$ $13.64$ $768.36$ $                                                                                              -$

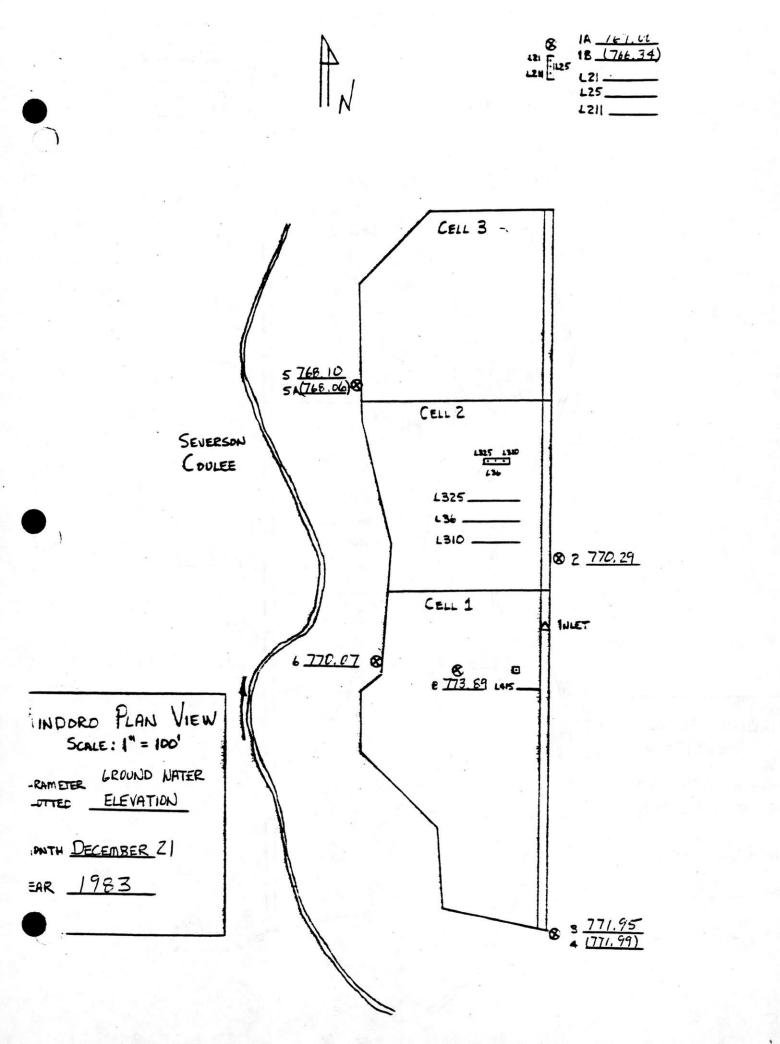
	DATE	DEPTH TO GW	GW ELEVATION	VOLUME HZD (ft)	3 VOLUMES (GAL)	VOLUME REMOVED	COMMENT
					(ene)	NERIUYEL	
	<i>10  </i> 19/83	8,43	775.42	3.2	1.6		
•	11/27/83 11/30/83	9.86 10.23	773.99 773.62	1.8	0.9	ALL -	© 3:50 © 11:35 LIGHT BROWN
SQUARE SQUARE SQUARE	12/21/83 12/22/83	9.96 10.35	773. E9 773. 50	1,72	5.56	WELL Volume	E 1:50 PM G 12:46 ANI LIGHT BEACH No small
	1 <i>/10/84</i> 1/11/84	9.99 9.66	773.36 774,19	1.7	0.8	UELL VOLUME	@ 12:36 PM EIL OVERNITE INCREASE - LIG
	2/13/84 2/19/84	8.74 9.12	775.11 774.73	2.9	1.5	VOLUME	(11:20) (11:00) BROWN
<b>L</b>	3/21/84 3/22/84	8,94 8,99	774.91 774.86	2.7	1.4	1 VOLUME	LIGHT BEDWIL
X	4/25/E4 4/26/E4	9,33 9.76	774.47 774.59	2.3	.	VOLUME	- -
·	5/22/EL 5/22/EL	10.15 10.54	773.70 773.21	1,5	0.5	NOLUNE	Light others
·	6/5/52 6/6/54	11.35 11.50	772.50 771.25	0.3	0.2	VELUNIE —	Too 253
	フルをチ	DRY	· -		-	-	_
	8/75/54	DRY			-	-	-
	9/13/54	Dey	-	-	-	-	-
	10/15/54	DRY	-	-	-	_	_
	11/5/84	10.04	773.81	-	_	~	-
				-		-	
• •							
	· · ·				•		

	DATA S	HEET I WELL	9 LEN	6741 [3.7]		5442000,	NORTH OF CELL BERN	- 
È	DATE	DEPTH TO GW	GW Elevation	Volume H20 (ft)	3 volumes (GAL)	Volumet Remisved	Comment	
	4/26/84	12.48	767.20	1.2	0.6	IJIUME	_	
	5/22/3 <b>4</b> 5/23/8 <b>4</b>		767.15 767.10	1.2	0.6	VOLUME	BROWN	
	615/84 616/84	12,69	766.99	1.0	0,5	VELVME	RUSTY	
EIS SQUARE	7/11/8A 7/12/64		767.3 <del>8</del> 767.26	1.4	7.0	1.0	RUSTY	
43 311 200 SHEETS \$ 500 ANT	8/25/24 8/29/64	12,95 12,97	766,73 766,73	0.8	0.4	IVELOME	Rusy	
naviour.	9/18/84	12,89	766.79	0,9	0.4	) VOLUME	Rusty	
	10/15/84	12.73	766.95	1.0	0.5	/ VOLUME	RUSTY	
	11/5/84	12.18	767,50	-	-	-	-	
								-
							•	
	-							
•								



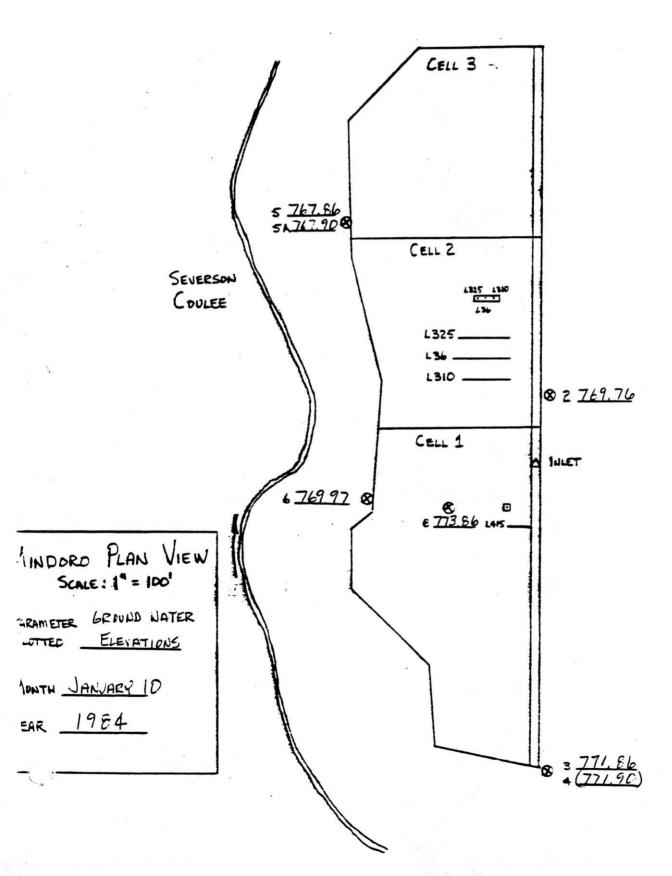


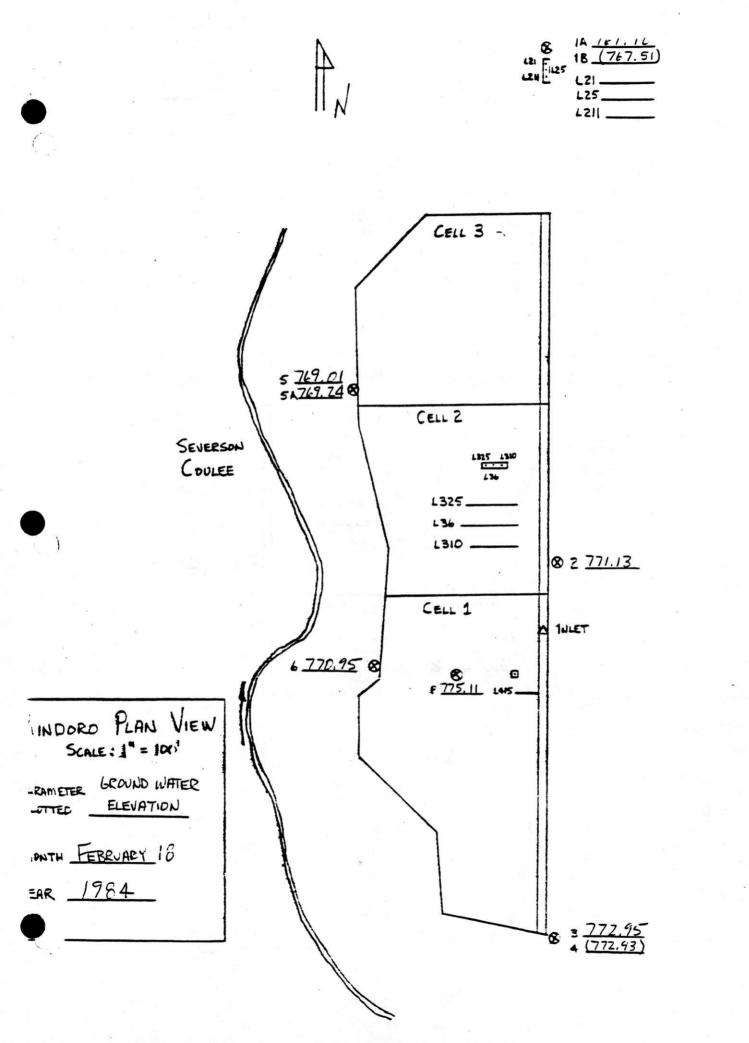


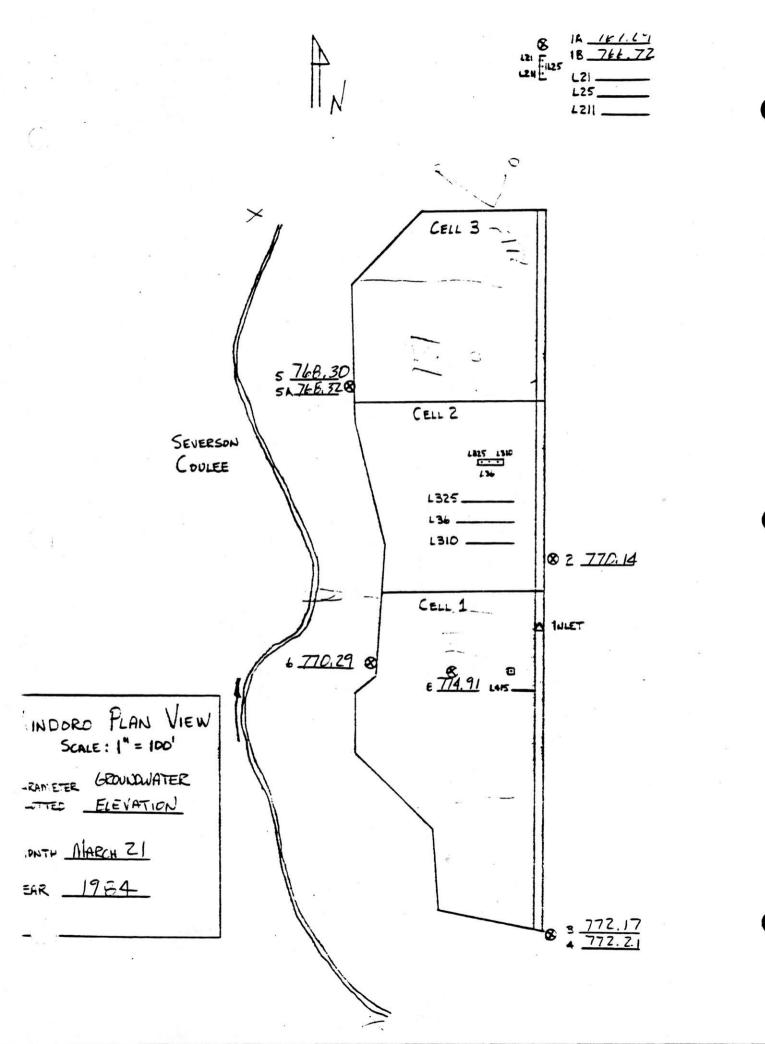


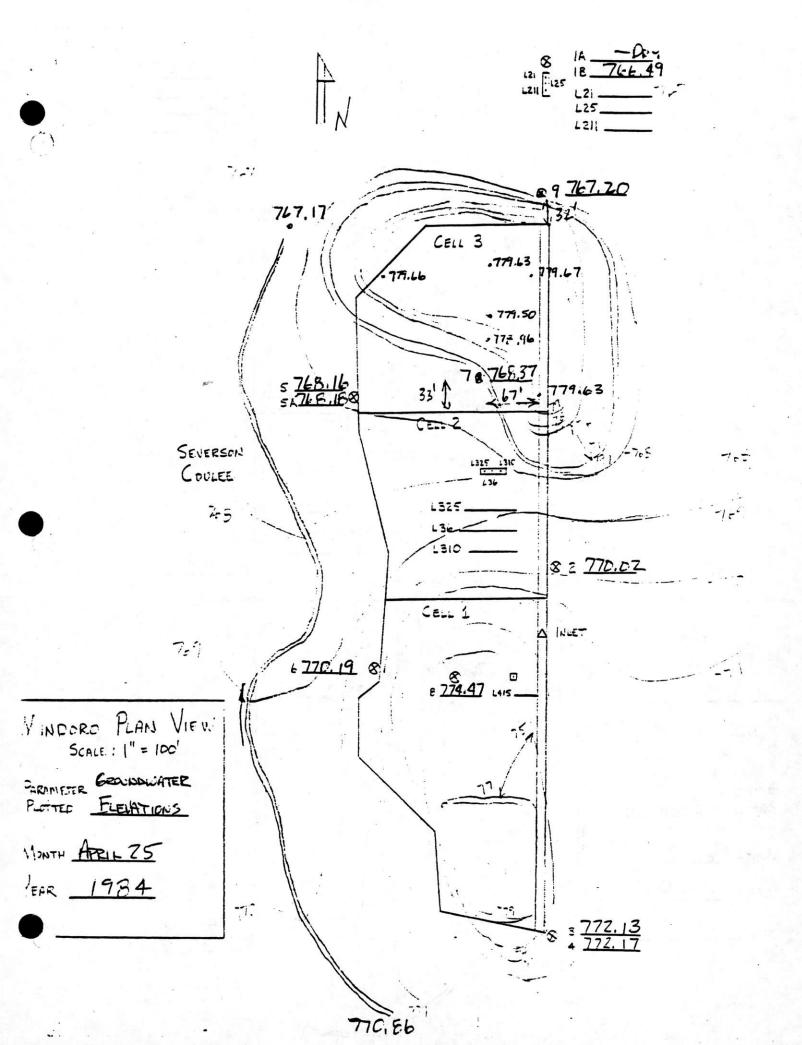
Π_N

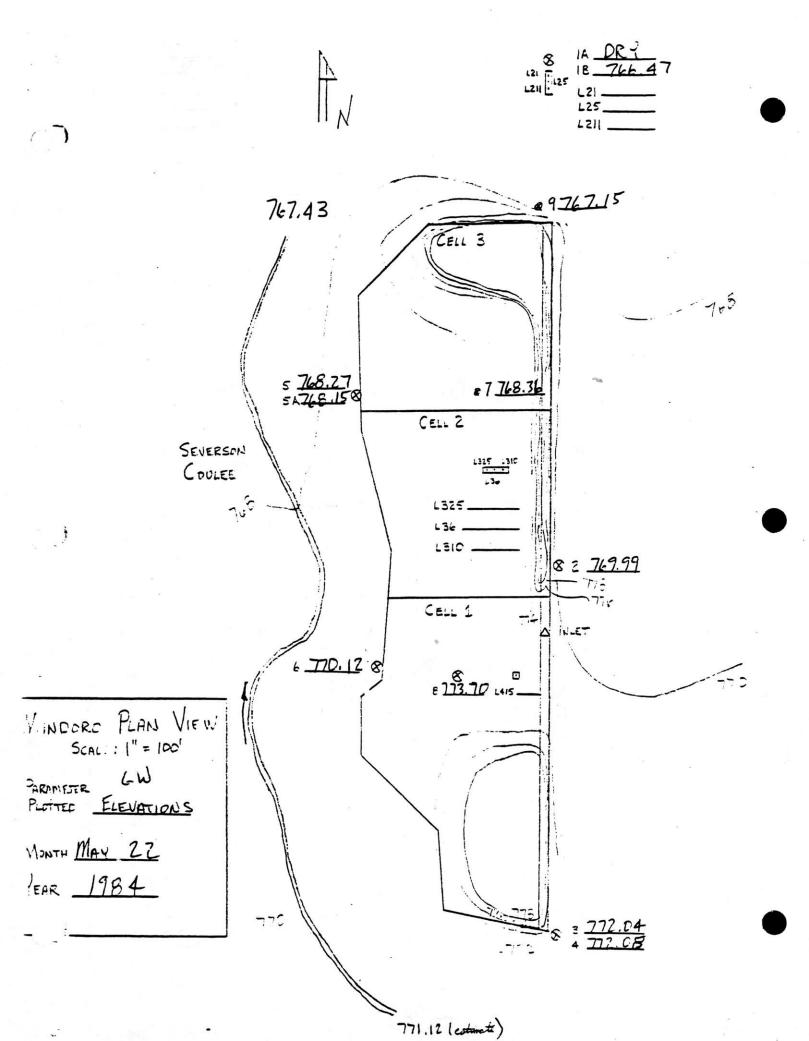
18 1766.06) 121 -1125 L21 -L25_ 1211 _

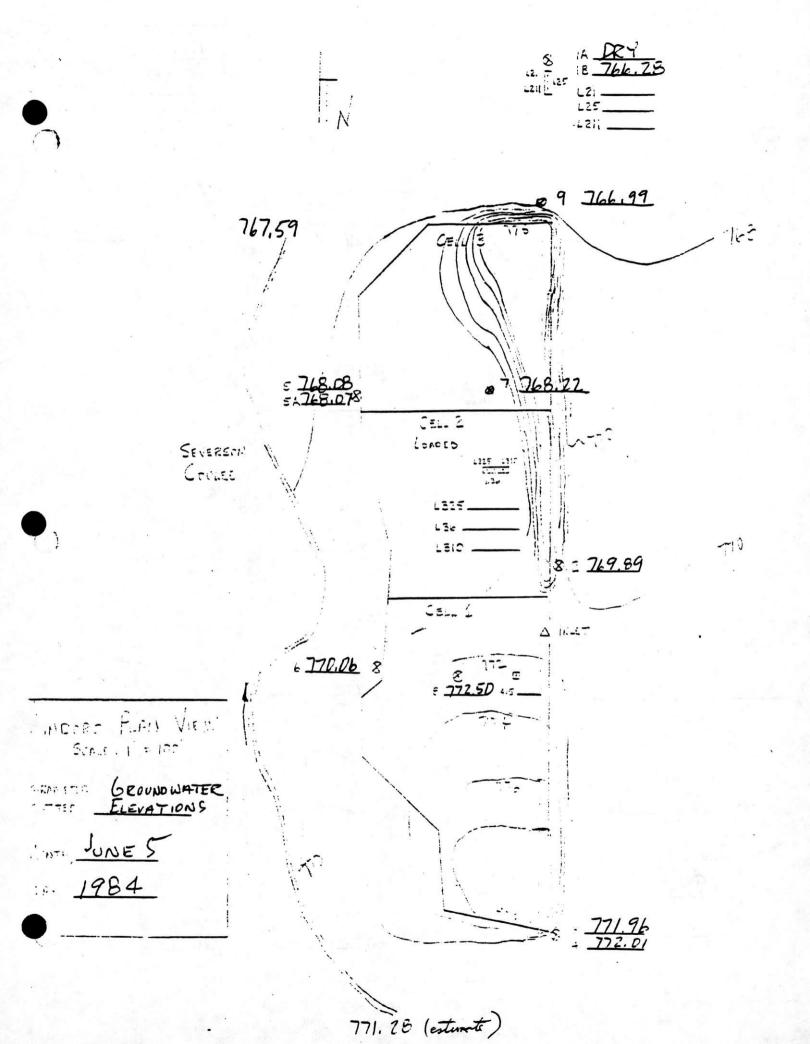


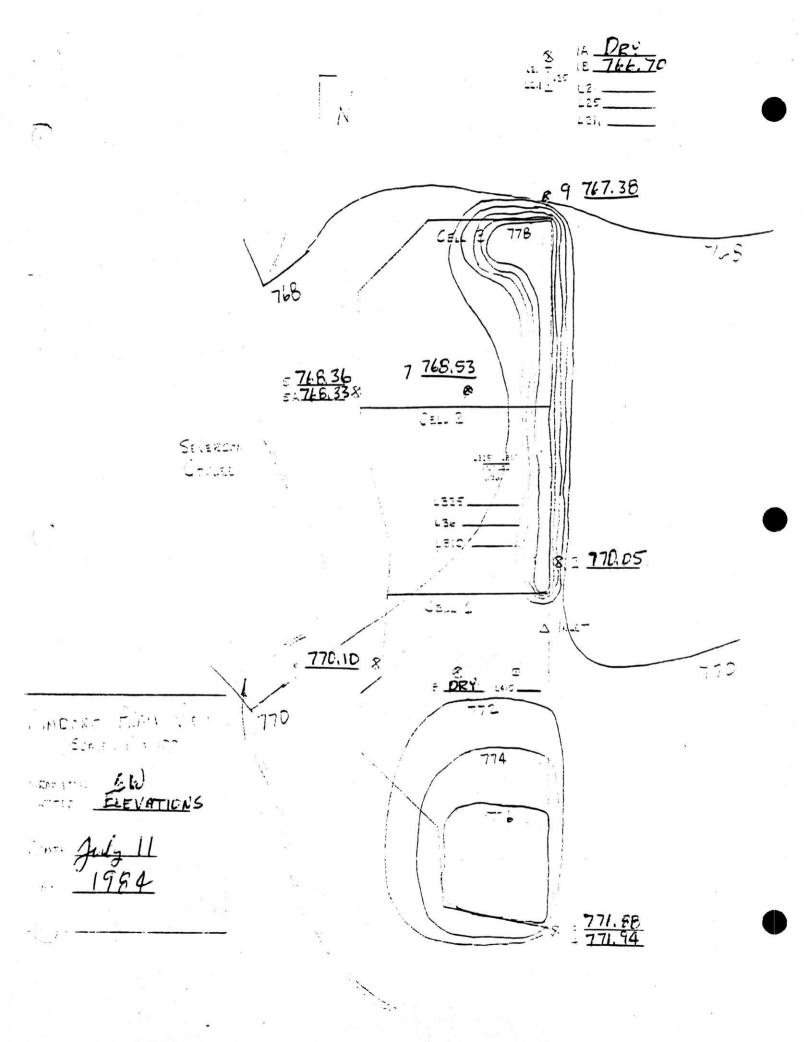


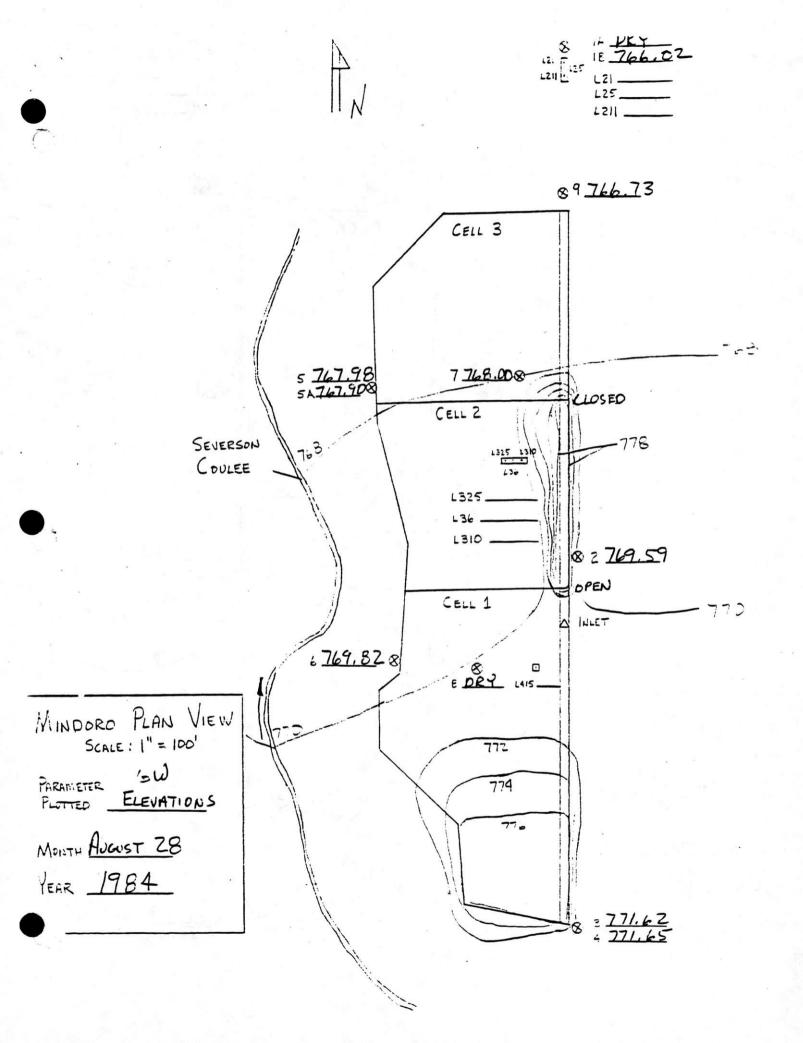


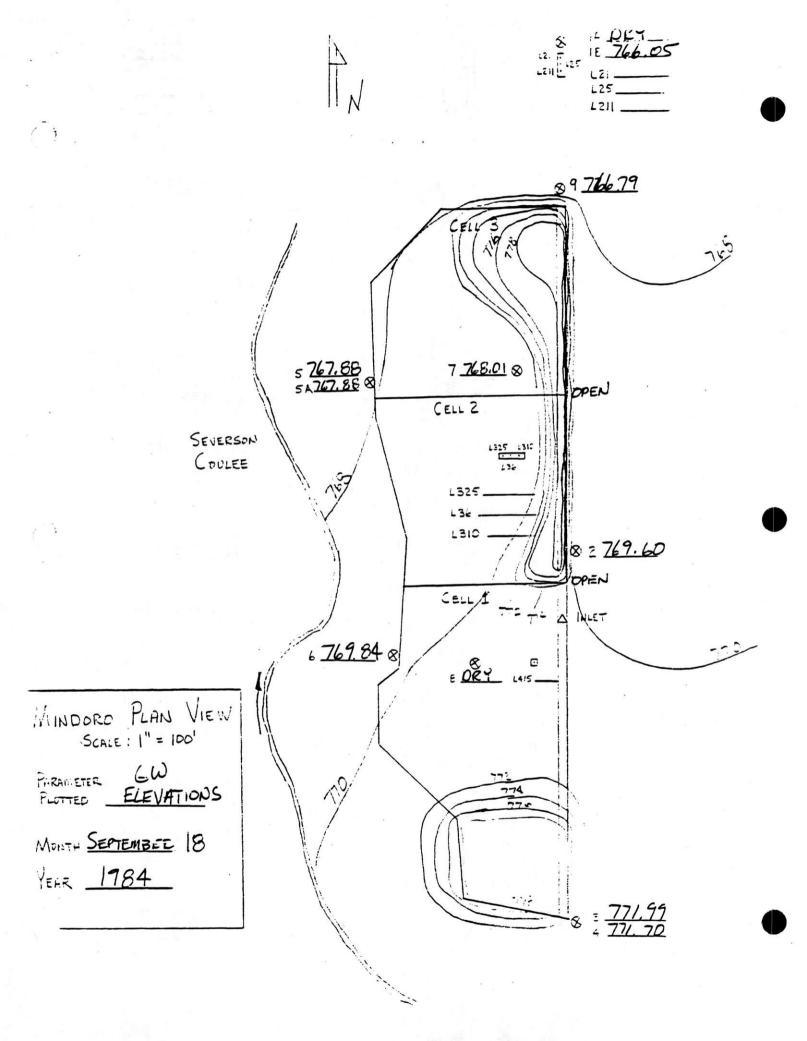


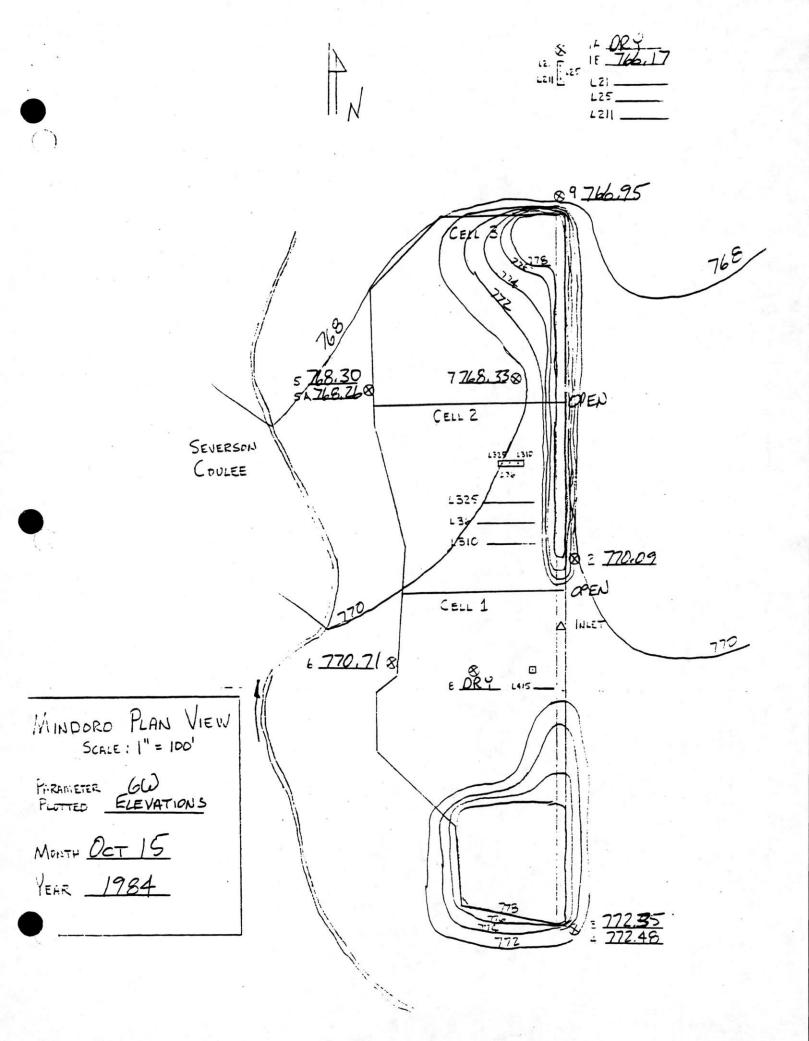












гопынатен 🦯 ниса ина А. SLUG TEST RESOLTS

· · · · · · · · · · · · · · · · · · ·	TEST	5	SITE: MIN	DORO	he	ELL:	2		
		NITIAL DE	РТН ТО (	_ : لمار	<u>13.48</u> Te	()	Ho) Rous	12	
		DEPTH (t)	H(+)	H/H_0 1.00	t	D(t).		H/HD 0,37	-
	#1 5		11.75 1 <b>¢</b> .50	0.96 0. <b>94</b>	188 201 212		4.25	0.35	
	9 15 21		11.25 10.92 10.50	0.9Z 0.50 0.86	225 24 I		3.75 3.50	0.31 0.29	
	31 36		10.00 9.75	0.82 0.80	254 273		3.25	0.27	
•	40 45 50		9,50 9.25 9, <b>0</b> 0	0.78 0.76 0.74	290 308 327		2.75 2.50 2.25	0,22 0.20 0.18	
	57 62 68		8,75 8.50 8.25	0.72 6.70 0.68	360 376	•	2.00 1.75	0.16 0.14	
	75 80 91		8.00 7.75 7.42	0.66 0.64 0.61	425 466		1.50 1.25	0,12 0.10	•
	98 106		7.08 6.75	0.58 0.55	504 556 642		1.00 1.83 0.50	0.08 0.07 0.64	
	11 Z 120		6,50 6,25	0.51	981 1220		0.25 0.08	0,02 0,01	
	129 136 142		6,20 5,75 5.50		1356		0.04	D 100	
	154 161	-	5.25 5.00	0.43 0.41					
	169		4.75	0.39					

		• • • • •	SITE : MII				0	
	TEST	· · · · · · · · · · · · · · · · · · ·			- WI		2	-
		INITIAL DE	PTH TO (	- : Lu	14-E	<u>8</u> (1	$t_o)$	
	-	RIAL 1			13.4 To	B Liar Z		
				1				
	£ (sec)	DEPTH (t)	H(t)	H/H	t	D(±).	H(+)	H/H0
·		13.48	0	0	182		4.25	0.39
	0 i4	2,65 3,40	$H_0 = 10.83$	1.00	189		4.08	0.38
			10.08	0.93	197		3.92	0.36
	18 22		9.83	0.91	205		3.75	0.35
			9,58 9,25	0.88	215		3.58	0.33
	27			0.85	224		3.4Z	0.32
*	32		9.00	0,83			-	
τ	37		8.75	0.81	Z30		3.25	0.30
	42		8,50	0.78	245		3.17	0.29
	45		8.25	0.76	251		3.03	0.28
	65		7,58	0.70	255		3,00	0.28
	75		7.25	0.67	S.		2.92	0.27
	85		6.92	0.A	260			
	92		6.58	0.61	263		2.83	0.26 .
	95		6.42		268		2.75	0.25
	118		5.75	0.53	Z73		2.67	0.25
				0.51	280		2.58	0.24
	126		5,50		315		2,25	0.21
	13)		5.33	0.49	338		2.00	0.18
	136			0.48	366		1.75	0.16
	140			0,48	398			0,14
	145		5.00	0.16	454			0.12
,	153			0.44	502 582		1	0,09 0,07
	162		4.67	V75 N	646		· ·	0.05
	168		4.50	0,42	750	1		0,04
	176		4.33	9,40	870		0.29	0,03
		•		T.	1000			~~~

.

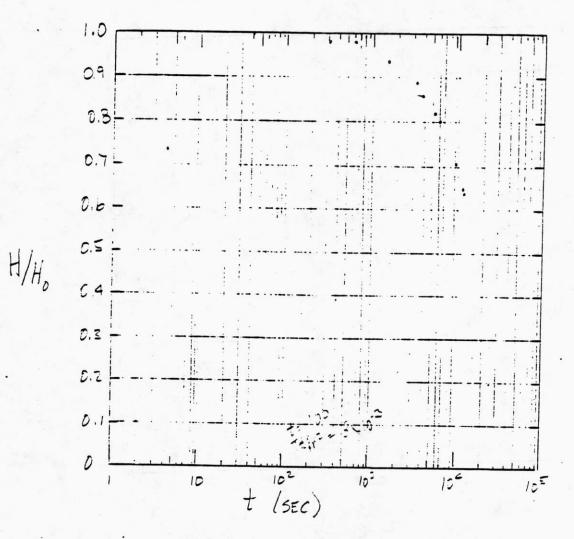
Ę

,

	TEST	15	SITE: MI	JOORO	W	ELL :	1A	·	
<b>N</b>				ςω: _			40)		
	T	RIAL 1			Te		1 cnt.		
	Ł (sēc)	DEPTH (t)	( H(+)	H/H	t	D(±).	 	H/H=	
	-1	4,50	0	0	553		8.52	0.62	
	0	0.73	H ₀ ← 13.77	1.00	604		8.27	0.60	
	10		13.60	0.99	663		8.02	0.58	
	21		13.44	0.98	719		7.77	0.56	
	31		13.27	0.96	790		7,52	0.55	
	46		13.02	0.94	851		1.27	0,53	-
	64		12.77	0.93	941		7.02	0.51	
	77		12.52	0.91	1018		6.77	0,49	
	98		12.27	0.89	1136		6.52	0.17	
	123		12.02	0,87	1243		6.27	0.46	
	146		11.77	0.85	1379		6.02	0.44	
	162		11.52	0.84	1514		5.77	0.92	
	184		11.27	0.82	2024		5.06	0.37	•
	216		11.02	0.80	2936		4.19	0.30	
	235		רד.01	0.78	3358		4.10	0.30	
	258		10,52	0.76				#12	
	295		10.27		1145		0.17	0.02	ı
	321		10.02	077	1256		0,0B	0.01	
	367		9,69	0.70				-	
	384 .		9.52	0.69	182D		0.04	0.00	
	431		9.27	0.67					
	475		9,02	0.66					
	524		8.73	0.63					

. <u> </u>	TES	сн. Т <b>S</b>	SITE: MI	NOORA	W	ELL:	8		
· •• •		INITIAL (	Дертн то			()	\$		
( ⁻ )		RIAL 1				CIAL Z			
	t (sec)	<b>ДЕРТН</b> (	t) Η(t)		t	D(±).	H( <del>{</del> )	H/HD	
	0		$H_0 = 10.04$	1.00					
	300		9.89	0.48					
	600		9.80	0.98					
	1680	•	9 AD	0.94					
	3210		8.90	0.89					
	3869		8.68	0.86					
	5400		8.23	0.82					
	6060		8.04	0.80					
	9540		7.15	0.71					_
	12,360		6.52	0.65					
	12,780		6.44	0.64					
								•	
						:			
-									
									٠
	3								
								-	

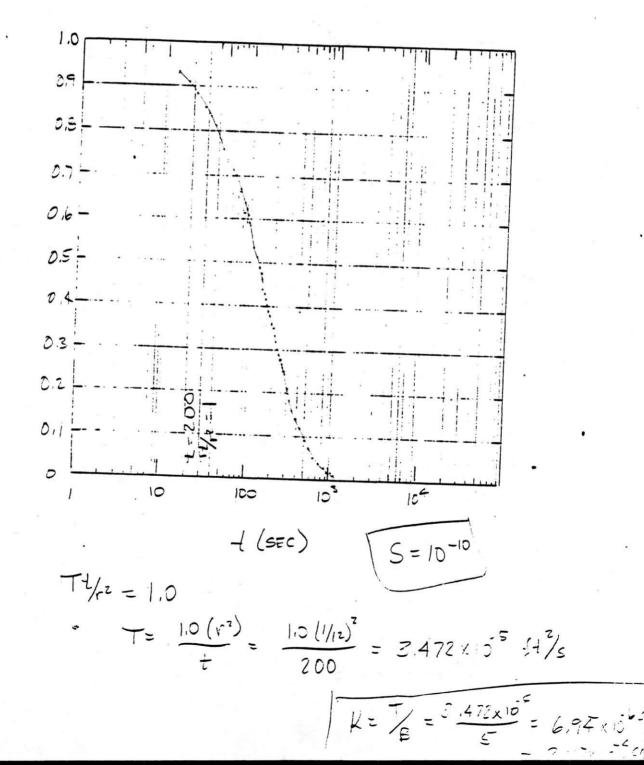
Shing Test Will 8 H/Ho vst



 $S = 10^{-4}$ 

T-1/r= 8×10-2  $T = \frac{2 \times 10^{-7} (V_e)}{100} = 5.5 \times 10^{-7} f_{e}^{2}/s$ K= 7/2 = 7/2 = 2.2×15 72/5 cm/s

Slug Test H/ ust Well 2 Round 1

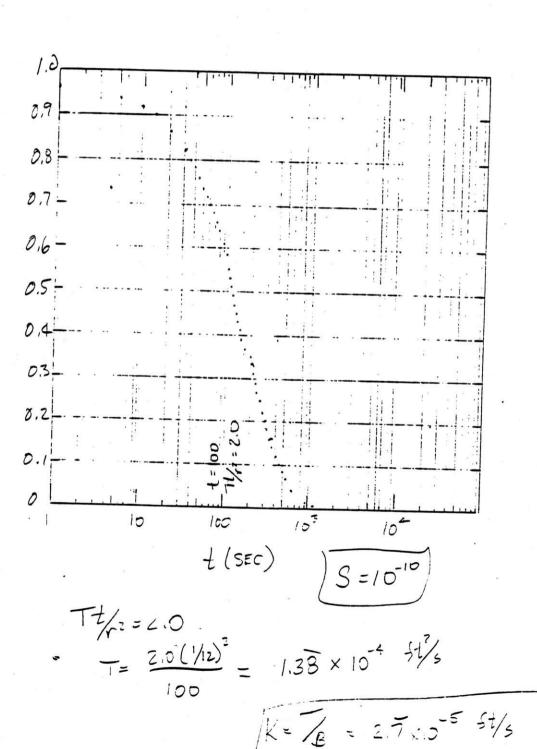


H/H.

H/Hors t Mindozo IA ()1.0 0.9 0.5 0.7-0.6 H/Ho 04 0.3 100 0.1 D 10 10 100 1000 screen length 5' ft x 17: x 2,54 cm. t (sec) 5=10-1 T=KE Tt/r2 = 5x10-2  $T = \frac{5 \times 10^{2} (r_{c}^{2})}{t} = \frac{5 \times 10^{2} (V_{12})^{2}}{100} = 3.472 \times 10^{10} \frac{1}{5}$  $K = \frac{3.472 \times 10^{\circ}}{5} = 6.944 \times 10^{7} \frac{51}{5}$ 

Slug Test H/H vst Well 2 Round 2

8,46×0° cm/2



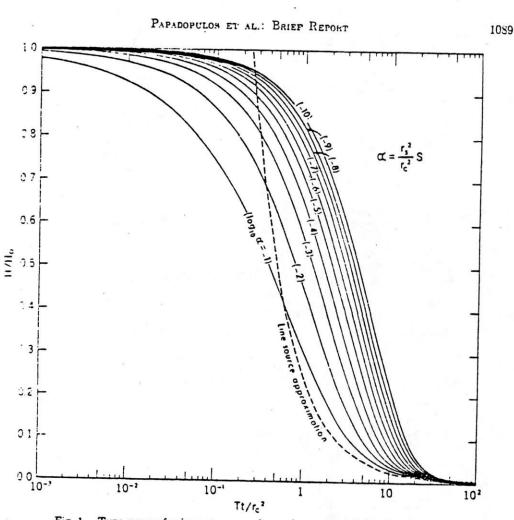


Fig. 1. Type curves for ins antaneous change in a well of finite diameter.

mination of S by this method has questionable reliability'; reliability becomes even more questionable when  $\alpha$  is smaller than 10⁻⁰ Of course, the similarity of the type curves in this ange of  $\alpha$  also affects the determinations of transmissivity. Even the most carefully and accurately collected test data could easily be marinel with more than one of the type curves. The the could expect is to be within one or two orders of magnitude of the actual  $\alpha$ . An analysis in the range  $\alpha < 10^{-4}$  indicates that, if the  $\cdots$  is a for the chosen type curve is within two entries of magnitude of its actual value, the the determined T would be less than above gree. This possible error should be kept in in i were one is making use of transmissivities 

## REFERENCES

- Cooper, H. H., Jr., J. D. Bredehoeft, and I. S. Papadopulos, Response of a finite diameter well to an instantaneous charge of water, Water Resour. Res., 3(1), 263-269, 1967.
- Ferris, J. G., and D. B. Knowles, The slug test for estimating transmissibility, U.S. Geol. Surv. Ground Water Note 26, 1-7, 1954.
- Ferris, J. G., D. B. Knowles, R. H. Brown, and R. W. Stallman, Theory of aquifer tests, U.S. Geol. Surv. Water Supply Pap. 1536-E, 104-105, 1962.

Kohlhaas, C. A., A method for analyzing pressures measured during drillstem-test flow periods, J. Petrol. Technol. 24, 1278-1282, 1972.

(Received February 23, 1973.)

APPENDIX FF MINDORO : rvns4451617 / Hacein113.4 GROUNDWATER CHEMISTRY DATA TABLE 5.6 SAMPLE CALCULATION

DATE	BOD5	Diss Cod	TDS	TKN	NHJ-N	NO3-N+	<1-	P(LAE)	OTHER
10/18/83	DRY								
11/30/83									
12/22/83									
1/11/84	DEY								
2/19/E4		TOTAL 10	_	4,5	0.1	1,9		-	-
3/22/64								- >	
4/26/84									
5/23/84									
6/6/84									
7/12/84			~						
8/29/24	1								-
9/19/E4		_ <del>````</del>	•						
10/16/84		<del>_</del>							
:					-				
•									
1									
					-				
	-								
		-							

DATE	ALKALINITY	TOTAL P	5042-	G2+	Na ⁺	Mg Z+	≮†	CTHER
10/18/83	dry							:
11/30/83	DRY							
12/22/83	DRY							· ·
1/11/84	dry -		>					
Z/19/84	-	_	-	-	_	-	-	
3/22/84	Dey				й. 			
4/26/84	DEY				-			
5/23/84				-				
			>					
7/12/84	DRY		$\rightarrow$					
:								
				-		-		
·								
		<u>.</u>						:
	10/18/83 11/30/83 12J2Z/83 1/11/84 2/19/84 3/22/84 4/26/84 7/12/84 7/12/84	10/18/83 DRY 11/30/83 DRY 12/22/83 DRY 1/11/84 DRY 2/19/84 - 3/22/84 DRY 4/26/84 DRY 5/23/84 DRY 7/12/84 DRY 7/12/84 DRY	10/18/83       DRY	10/18/83       DRY         11/30/63       DRY         12/22/83       DRY         1/11/84       DRY         2/11/184       DRY         3/22/84       DRY         3/22/84       DRY         4/26/84       DRY         5/23/84       DRY         5/23/84       DRY         7/12/84       DRY	10/18/83 DRY 11/30/83 DRY 12/22/83 DRY 1/11/84 DRY 2/19/84 3/22/84 DRY 4/26/84 DRY 5/23/84 DRY 6/6/84 DRY 7/12/84 DRY	10/18/83 DRY 11/30/63 DRY 12/12/63 DRY 1/11/84 DEY 2/19/84 3/22/84 DEY 4/24/84 DEY 5/23/84 DEY 5/23/84 DEY 7/12/84 DEY 7/12/84 DEY	10/18/83 DRY 11/30/63 DRY 12/12/84 DRY 2/19/84 3/22/84 DRY 4/26/84 DRY 5/23/84 DRY 7/12/84 DRY 7/12/84 DRY	10/18/83       DRY         11/30/03       DRY         12/12/83       DRY         1/11/84       DRY         2/14/84       DRY         3/22/84       DRY <td< td=""></td<>

	Dr	TA SHE			METERS	(براجس)	DISSACRED	1B-	PEEP,	
	Date	DISSOLVED BOD5	PISS	+ TDS	DK SOLVED TKN	NH2-N	NO2-N- 11003-N		PH (LAE)	OTHER
	10/18/83	Z.4	7	336	0.4	D.	Z.3	8.9	8.5	-
	11/30/83	<6	<5	340	<0,Z	<0.)	Z.8	9.7	-	-
	12/zz/83	< 3	SOLUBLE <5	340	< 0, Z	D.1	3.5	9.7	. —	-
	1 <i>/11/</i> 84	≺3	45	358	<0.2	20.1	3.8	9.8		
	2/19/ <del>84</del> -		<5	366	<i>40.2</i>	<0.1	3.5	8.9	FIELD)	-
	3/22/ <del>84</del>		<i></i>	362	< D. 2	0.1	3.2	3,3	7,2	-
	4/26/84	<3	<5	346	20.Z	0,1	3.1	9.4	7,5	PHIAE 5
	5/22/34	-	_			-	-	-	-	_
	6/6/84		-		-		_	-	(	
	7/12/84	~			-	-	_	_	-	-
)	3/29/64	23	<5	352	0.2	0,1	4,3	9.7	7,4	
				-						
	: :									
	:									• • • • •
	-									
			1							
		-						117 <b>Manufactura</b> 1 - Manufactura		
							va di Milinia Anglia			
	•		i I I							

•	DAT	a Shée		PARA	METERS	(mg/l)	WE	UL 18-	DEEP, BG
	DATE	ALKALWITY	P	5042-	Ca ²⁺	Na ⁺	Mg 2+	K+	OTHER
	10/18/83	254	0.06	43.	65	6	33	<b>-</b> .	-
	11 /30/83	_	-	-	-	-	-	_	-
	12/22/83	—	_	_	-	_	-	-	
	1/11/84	-	-	-			-	-	1
	2/19/8 <del>4</del>	-	-	-	-	-	-	-	-
Alvelaln <b>U.S.A</b>	3/22/84	-	-	-	_	_	-	-	-
/ w.w.	4/26/84	262	0.06	37	72	6	32	1	-
	5/23/34	_	_	_		-	-	-	
	6/6/84	-	-	_	-	-	_	-	_
	7/12/34	-	-	-	-	-	-	-	_
`	3   29   84-	250	0.06	48	70	7	34	1	-
)	;								
					-				
						-			
		-							

	DATA	SHEE		PARA	METERS	(mg/£)	DISS.	u Z-	SHALLO	ω	
	DATE	, Dissolved BODS	DISS COD	TDSX	TKN X	DISS	NO2-N+		P(LAR)	OTHER	
$\left( \begin{array}{c} \end{array} \right)$	10/18/83	49	76	674	3.8	D. I	0.1	66.	8.4	-	
	11/30/83	7.4	22	583	0.6	0.1	20,1	30 70	-	-	
	12/22/83	-	LOW LEVEL	_	TOTAL 12.	0.4	0,1	-	_	-	
	1/n/84	-	TOTAL 96	_	TOTAL 6.5	0.6	0.1	-	_	-	
	2/19/84	<3	16	686	2.Z	0,4	D.J	75	-	-	
Alada <b>inU.S.A</b>	3/22/84-	-	10	-	1.2	0.1	D,Z	-	(FIELD) 7.3	_	
	4/26/E4	×3	7	603	1.Z	0.1	0.3	70	7.5	-	
	5/23/84	<3	12	608	1,5	0,1	<0.1	72	73	-	
i tri v u	616184	4.D	11	584	1.B	0.3	.0.1	74	7.7		
	7/12/84	4,9	13	516	1,5	0.4	0.2	65	7.6	-	
	B/29/E4	3.7	1D	416	1.2	0.7	D.1	55	7.3		
()	9/19/84	23	9	354	D.B	0.5	D.1	31	7,4		
	10/16/84	JD	71	336	1.0	0.4	0.1	20	7,3		
				429c	44.4%	4.55		73%			
	: ·										
	• •								χ.		
					-	-					
-											

	IIINULK DA	U LUMMU TA SHEE		PARA	METERS	(m)/l)	WE	L Z -	SHALLOW	
	DATE	Ажалыц	TOTAL P	504	G ²⁺	Nh+	Mg ^{z+}	κ+	OTHER	
	10/18/83	40Z	0.04	76.	87	110	32		_	•
	11/30/83	-	-	-	-	-	-	_	-	
	1Z/2Z/B3	-	-	-	<b>—</b> .	-	_		-	
	1/11/84		-	-	-	-	-	-	-	
,	2/19/84	-	-	-	-	-	-	-	_	
A.1.11.5.A	3/22/84	-	-	-	-	-	·	-		
	4/26/84	415	0.02	120	_	-	_	_	-	
С. 1-1-1-	5/23/24	-	-	· _	-	-	_	-	-	
	6/6/84	-	~	~	<b>-</b> .	-	-	-	_	
	7/12/B4	-	-	_	-	-	_	-	_	
)	- -									
1									-	
	-									
	:									
					•					
	,						•			
		-								-
1 N										
·	•									

`	Dr	TA SHE	57	PARA	METERS	["]]])	WE DIES	- 2 11	SHALL	$\omega$
	DATE	BOD5	COD	TDS	TKN	NHJ-N	NO2-N+ NO3-N	CIT	PH (LAR)	OTHER
	10/18/83	1.6	11	704 .	0.6	D.1	0.1	78.	8.1	-
	11/30/83	23	16	660	0.6	0.1	< 0. J	30	-	-
	12/22/83	23	SOLUBLE 9	730	0,4	0.1	<i>20.1</i>	BD	-	-
	1/11/84-	<u>ل</u>	SULJELE 10	750	0.4	0.1	٢٥،١	76	-	
	Z/19/84	23	10	750	0.4	0.1	<0.1	75	-	-
	3/22/84	۷3	10	716	0.3	2,1	< D. 1	75	(FIELC) 6.5	-
	4/24/83	23	7	700	0.6	0.1	9.1	72	7.1	LABFH 7.4
	5/25/84	23	10	7:4	0,5	0.1	0,1	73	6.6	
	6/6/84	23	ID	714	0.6	D.1	0.1	76	6.9	-
	7/12/84	<3	3	704	C.4	0,1	0,1	75	7.0	_
	5/29/84	3.3	9	642	0.6	0.1	0.1	76	7.0	
)	9/19/84	23	15	532	1.0	0.3	0.3	99	5.3	
	10/16/84	4.3	17	620	0,3.	0,1	0.1	.67.	7.1	
	•		-							
							8			
							4			

$\begin{array}{c c c c c c c c c c c c c c c c c c c $	`	DAT	A SHE			nca L Meters	(mjj)	WE	43-	SHALLOW
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		DATE	ALKALIN TH	TOTAL P	50+2-	[ Ca ²⁺	Na ⁺	Mg Z+	K†	CTHER
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	<b>F</b> )	10/18/83	430	0.02	48.	100	120	44.		-
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		11/30/83	-	-	-	-	-	-	-	-
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		12/22/83	—	-	<b>—</b>	_	_	_	-	-
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		1/11/84	-	-		-	-	-	-	-
6/6/84		2/19/84	-	-	-	-	-	_	-	-
6/6/84	<b>K.C.</b> D. W.C.	3/22/84	-	_		_	_	-	-	_
6/6/84	7 Mar	4/26/84	473	20.CZ	62	100	סון	3.5	1	-
7/12/84		5/23/34	_	-	-	-	-	-	-	-
		6/6/84	_	-	_		_	-	_	_
3/29/34 472 0.02 63 95 120 38 1		7/12/34	-	_	_	-	-	-	-	_
	3	3/29/34	472	0.02	63	95	120	35	1	
	)									
									• .	
			-							
				-						
	2					<i>.</i>				

	DATE	A SHEE , DKSOLVED BODS		TDS	AMETER DISS TKN	S (19)/, DISS NH3-N	NO3-N+		PH (LAR)	OTHER	
`)	10/18/83	3,1	<5	270	0.4	< 0.1	0.3	Z.6	8.5	-	
	11/30/83	23	<5	273	<0.Z	<0.1	D. I	2.0	-		
	12/22/83	23	Soluble <5	276	ZD.Z	20.1	0.2	2.1	-	-	•
	1/11/84	23	<b>&lt;5</b>	284	<0.2	20,1	0.Z	1,9			
	z/19/84	23	25	272	<0.Z	20.1	0,3	Z. D		-	
	3/ZZ/84	< 3	<5	276	< D. Z	0,1	0,4	2.1	(FIELD) 7.2	-	
	4/26/84	23	<5	280	40.2	0.1	<i>D</i> 1.3	2.1	7.5	pHiae 8.2	
	5/23/84	< 3	<5	284	0. Z	0.1	0.3	2.4	7.0		
	616184	23	5	252	D. Z	∠D.1	0.3	<i>Z</i> , Z	7.3		
	7/12/84	<3	15	278	0.1	<i>≺</i> 0,1	0.Z	Z.0	7.5		
`	3/29/84	<4	<b>~</b> 5	270	0.2	<0.)	D.)	2.1	7.5		
)	9/19/24	<3	15	286	0.2	20.1	C.2	2.0	7.1		
	10/16/84	<3	5	274	0,1	0,1	8.3	Z.0	7,4		
		. 9									
					-						
. 1				ат. Фила Вил Вил Вил Вил Вил Вил Вил Вил Вил Вил							
•						1 					
1					• • •						
•					•						
-											
:			•	•							

	DA	TA SHEE		PARA	METERS	(كرادس	WEL	L4 -	DEEP	:	
	DATE	ALKALINITY	TOTAL P	5042	Ca ²⁺	Nat	111g 2+	K+	other		
	10/18/83	240	D. O.B	21.	55	3	ZB	-	J		
	11/30/83	~	-	~	~	<b>~</b>	-	-	-		
	12/22/83	-		-	-	-	_	_			
	1/11/84	~	-	-	~	-	-		-	1944 - 1944 1944 - 1944 - 1944 1944 - 1944 - 1944 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1944 - 1	
÷	z/19/84	-	-	~	-	-		_	-		
	3/22/84	-	_	_	-	-	-	-	_		
	4/26/84	242	0.06	22	55	2	25	/	_		
	5/23/84	-	~	-	-		-	-	-		
	616/84	-	-	-	-	-	~	-	-		
	7/12/84	-			-	-	-	-	· —		
ì	8/29/84	246	5.06	23	60	3	30	1			
,	•										
	· · · · · · · · · · · · · · · · · · ·										
•	• • •										
	•		4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4								
	:										
						-					·
		-									
· · · · ·			-								
					:						
	. 1					3 4					

	DAT			LHEN. PARA	iichl Meters	(mg/)			<i>Shall</i> ou	1
	DATE	, Ossowed BODSX	DISS	TDSX	DISS TKN	DISS NH3-NX	NCz-N+	CITX	PH (LAB)	OTHER
	10/18/83	3.1		664	5.2	4.0	< 0.1	39.	8.4	_
	11/30/83	12	30	605	2.4	0.Z	20.1	92	-	-
	12/22/93	15	scuere 23	708	3.3	2.3	<0,1	87	-	-
	1/11/84	12	36	726	4.4	3.7	20,1	91	_	
	Z/19/84	3.7	25	716	4.4	3.B	20.1	90	-	_
Mada In U.S.A	3/22/8 <del>4</del>	<3	25	77Z	4.9	3,3	<0.1	94	(FIELD) 6.7	-
	4/26/84	23	22	694	4.2	3,5	0.1	EB	6,9	pti437.
	5/23/84	12	33	762	3,4	2,3	0,1	95	6.6	-
	6/6/84	32	59	784	3.6	2.7	0.1	92	6.3	-
	7/12/84	13	30	764	4.6	3,5	0,1	39	6.3	_
)	8/29/84	<12	27	734	3,9	3.)	0.1	92	6.3	
)	9/19/84	<12	23	750	3,3	2,9	20.1	91	6,5	
	10/16/84	9	31	733	4,0	5,2	<0.1	93	6.6	
·					-					
						• •.				
	•									
:										
-										
							•			

	DAT	ia She		PARA	METERS	(""/[)	WEL	1 5 .	SHALLOW
	DATE	ALKALINITY	TOTAL P	5042-	Ca ²⁺	Nat Nat	Mg2+ (	K+	OTHER
	10/18/83	49 Z	0.02	2.3	76	130	34	)	-
	n/30/83	-	-	-	-	-	-	-	—
	12/22/83	-	-	-	-	_	-	-	
	1/11/84	-	-	-		-	-	-	_
•	z/19/84	_	-	_	-	_	—	_	_
	3/22/8A	-	-	-	—	_	-	_	_
	4/26/84	452	20,52	30	77	IID	32	2	-
	5/23/84	-	-		-	_	-	-	-
	616134	_	-	~	-		_	_	
	7/12/54	_	—		-	_		-	—
	8/29/34	532	20.02	3,3	87	40	36	Z	
ł									
	•		-						
	•								
	-								
	•								
									•
		<b>-</b> .							
۰.	-								

	DATE	DISSOLVED BODS	COD	TDS	TKNV	NH3-N	NOZ-N+ NOZ-N	CI-V	pH (LAE)	OTHER
)	10/18/83	4,6	-	684	3.2	1.4	20.1	5/,	B. Z	-
	11/30/83	<6	34	750	3.Z	2.2	20.1	33	-	-
	12/22/B3	∠3	5010 <b>8-E</b> 29	724	1.8	1.1	20.1	80	. —	
	1/11/84	3.(	24	077	1.6	0.8	<0.1	53	-	-
	z/19/84	∠3	27	778	1.6	0.9	<0.1	79	_	
Alada in U.S.A	3/22/84	< 3	26	750	1.6	0.3	20,1	78	(FIELD) 6.8	-
	4/26/84	23	23	716	2.4	0.3	0.1	72	7.D	(АВр 47.4
	5/23/84	<3	25	710	1.4	0,5	<0. l	77	6.6	
/ <b>6</b> 1	616/84	3.1	23	720	1.4	<i>D</i> , B	20.1	75	6.3	-
·	7/12/84	<6	ZZ	736	1,4	0.9	D,1	73	7.0	-
	8/29/84	<1D	23	724	1.5	0.9	0,1	74	6,9	
	9/19/84	27	21	760	1,4	2,3	20.1	71	6.3	
	10/16/84	4,1	26	756	1.5	C ,9	20.1	75	6.9	
									-	
				•						
										· · · ·
		-								

	DATE	ALKALINITY	P	5042-	Ca 21	Nat 1	Mg 2+	κ+ (	OTHER	
)	10/18/83		2,3	49.	BB	120	38			
	11/30/83	_	-	_	_	_	-	-	<del>_</del> .	
	12/22/83	-	-	_		_	-	-		
	1/11/84		-	-	-	-	-	_	_	
	2/19/84	-	_	_	_	-	_	-	_	
	3/ZZ/84	-	-	-	-	-	-	-	-	
	4/26/84	466	<0.02	56	89	IID	36	l	-	- - -
	5/23/24	-	-	-	-	-		-	<b>—</b> ·	1
	616184	-	-	-	-	-	-	-	_	
	7/12/84		_	-	-	-	-	-		
ì	3/29/54	476	20.02	77	93	120	40	)	-	
j.	:									
							-			
										•
				-						
	•									
				· · · · · · · · · · · · · · · · · · ·						
								·		
	1 			: :						
	•			•	1 2 1	;				

	11 iscaes		2	-772 P.		(~/-()	i he		SHRILDU, C	CELL 3	
	DATA	SHEE - Diss	D.55		CTEES Diss TKN	0155	D155	CI ⁻	FIELE 6 17	CTHER	
	DATE 4/21/84	3005 >23	(1) (4)	738	ZS	IJHIZ-N	0.1	79		1 - LAE 1.5	
· }	=/23/34		16	EZD	2.8	2.3	<i>L</i> D.1	93	6.5	-	
	616/84		1.2	680	3.0	2.4	<0.1	33	6.6	-	
	7/12/84	26	14	764	3,0	Z.3	<0,1	86	6.3		
S SOUARE 5 SOUARE 5 SOUARE	3/29/84	<6	14	574	Z.D	1.7	0.1	77	63		
200 SHEETS 35	9/19/84	24	11	605	1.8	1.4	-22.1		;4		
42 181 50 42 182 100 42 189 200	13/16/24	6.5	23	644	2.2	1.6	201	90	6.5		
MAN WALL											
-											
•											
•					、						
									: : :		
					: ;		! !	· · · · · · · · · · · · · · · · · · ·		•	

•	IIIINLORD DATE	CUTTIN JU SHEET	£7/7£ ►	L-En Para	METERS	(mo/	. JE	22 7-	SHFULDU CELL 3	
		ALKALINITT	TETAL	50 <del>2</del> -	6 ²⁺	Nat	Nig ^{Z+}	¦ ⊀+	DTHER	
$\langle \gamma \rangle$	4/26/84	572	0.02	./9	120	95	46	1	_	
	5/25/84	-	-	_	-	-	-	-	-	
	616184	~	-	-	-	· •	-	-	-	
	7/12/34	-	-	-	-	-	-	-	-	
42         141         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50         50	8/29/34	376	£0.02	3.2	5 <u>-</u>	73	35	< )		
anionar		•								
				•						

	DA			Paeai	METERS		) WE NO3-NH			~
	DATE	DISGOLVED BOD5	DISS. COD	TDSX	DISS. TKNX	0155. 1 JH3-NX	NOZ-N	$C\Gamma_{X}$	pH (LARE)	OTHER
	10/15/23	4.1	34	373	4.4	0.6	0.Z	69.	8.Z	-
	11/30/83	<3	20	654	1.4	0.1	Q.5	81	-	_
	12/22/83	-	LOW LEVEL 55	~	TOTAL 3.B	0.2	<0.1	-		_
	1/11/84	23	SOLUBLE 25	888	1.4	0.2	20,1	8Z	-	-
•	2/19/84	< 3	Z4	970	1.6	0.1	20.1	84-	_	_
Madninu.s.A	3/22/84	23	25	394	2.5	0,3	LD,1.	89	(FIELD) b.b	
	4/26/84	ζ3	26	336	3.Z	D.3	D.1	94	7.0	_
14	5123134	23	23	793	2,8	6,3	<0,1	90	6.7	-
	616/84	DRY	$\rightarrow$			•				
	7/12/84	Dey -	$\longrightarrow$							
	- - -									
}								-		
									-	
	4									
•										
	•			-						
	•									
	1 1 1 1 1								-	
						-				
							-			
•	- - - -									

	,D	TA SHE		PARA	metees	(mg/1	) WE	11 3 -	SHALLOW	
	DATE	ALKALINITT	P	5042-	G ²⁴	1Ja ⁺	Mg Z+	K+	OTHER	
÷.)	10/18/83	130	0.04	62,	N.B.	N.B.	N.B.	N.B,	-	
	11/30/83	-	-	~	-	-	—	-		
	12/ZZ/83	-	-	-	-		-			
	1/11/84	-	-	-	-	-	-	_	_	
	Z/19/84	-	-			-	-	_		
Madoln <b>U.S.A</b>	3/22/84		-	_	-	-	-	-		
· ·	4/24/34	372	20.02	130	_		_	-	-	- - -
Abun wennen	5/23/89	-	-	-	-	-	-	-	-	· · ·
PRANK I	6/6/84		>							
	7/12/54	Det.	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~							
) 1 (			: •							
			•							
	- - - -		· ·							
			•	1						
			•							
				•						
				:						
	: : :	•								
	•	-		- 						
	:			4 • •						
•									1	۲
	: - :								: : : :	

	DATA	SHEET	• •		CH- ETFES	ra(1)		29-31	Hill Ding K	log sen II
Č.	DATE	Diss BCDs	Friss COD	TDS	DISS	Diss 1JH3-N	1 DB5 - 1153-1552	CIT	Fiere	DTHER
	4/26/84	>23	76	568	4.2	1.4	5.2	76	6.9	1- at 7 C
	5/23/84	4,3	20	540	2.2	1.	C. I	3Z	6.6	-
	616/84	>22	93	456	2.4	1.1	0.1	35	6,7	-
	7/12/84	38	46	680	2.2	7.1	0.1	9Z	6,8	
S SQUARE 5 SQUARE 5 SQUARE	8/29/84	≺7	20	622	2,5	1.3	<i>D.</i> 1	97	6,3	
200 SHEETS	9/19/24	4,5	26	696	2.1	1.9	5.1	10	6.6	
42 381	10/16/54	12	47	636	(1) (1)	1.3	5,1	100	6.5	
ž										
						-				
		-								
	4								•	

	DATE	ALKALINITY	TOTAL P	SC4	G ^{z+}	Not	11:0	1 K	UTHER	
)		:	0.02		_		_	-		-
	5/23/34	-	-	-	-	-	-	-	-	
	6/6/84	-		-	-	-	_	-		
	7/12/34	-	-	-	-	-	-	-	-	
	B <i> 29 5</i> 4	-	-	<u> </u>		-	-	_	-	
				9 .						
							-			
								•		
		-	:							
						• :				
						:				
							-			
		-								
							1			

<b>x</b>	DAT	A SHEE	T	1	(mg/_	l)	L	I WA		
	DATE	BODS	SOLUBLE COD	C1-	DISS. NO2-N+ NO2-N	Di <del>ss</del> TKN	TOS	0155. NH3-N	OTHER	
3	12/22/83	23	<5	<0,3		<i>40</i> , Z	10	20,1	<u> </u>	
	1/11/84	دع	<5	<0.3	40.1	<0.2	12	×0,1	-	
				-						
Alada In U.S.A										
~										
агал ())										
MUTERMORY										
)						÷				
			-							
							· .			
-				-						
			•							
		-								
			~				-			
·										

$$T_{ABLE} 5.6 CALCULATION$$
EXAMPLE - WELL 5
$$CI^{-} CONCENTRATION = 91 \text{ mg/l}$$

$$TOTAL N CONCENTRATION = 4.1 \text{ mg/l}$$

$$\int WASTEWATER HAD \quad CI^{-} = 100 \text{ mg/l}$$

$$TOTAL N = 32.2 \text{ mg/l}$$

$$7c \text{ REDUCTION (DILUTION) of CHLORIDE AT INELL 5 = (FROM WAITE)$$

$$\frac{100 - 91}{100} \times 100 = 923$$

$$\frac{32.2 - 4.1}{32.2} = 87.90$$
ACTUAL 70 J LOSSES AT WELL 5 = 87.90

.

)

ļ

APPENDIX HH

РАРИАСТЕП / ННА НЦЕА.

MINDOED :

FURROW WASTEWATER AND LYSIMETER CHEMICAL DATA

*	•	cumul. Ta Shee			ст. 1 л mg /J)	L7111.2 - 2 - 3	FURE WAST	ci- Tenater	· (£4	Ē
	DATE	diss Edl _s	DIES Coc	C1-	PH	ries Ricz+nicz		TCS	12155 12Hz-12	
	ד/ג/34/	220	320	99	6.6	<1.0	(6	796	16	_
42 342 200 SHEFTS 5 SQUARE	10/16/84	тст) 94	160	TURROW 76	WW - 6.6	Сец З 0,1	4,¤∺≦ [7] 70T		12,	_
200 States 12:382 200 St	11/6/84			78		<i>∠1.</i> ∂	22 TOT 16 0155		15	
MATTINAL			· · · · · · · · · · · · · · · · · · ·							
	15/16/34	(187) 1400	2600	<u>+ Eno-1</u> 82	670	<u>GELL Z</u> 0.1	21.0 108707	_	)/	-
	11/6/24	_	-	120	6.2	0.1	63 707 23 diks	_	_	-
	10/10/24	(707) 250	1200	HEADER B.Z.	шш- 6,4	<u>(ru 3</u> 9.4	17. Dons		. 4	
	.i/6/84	-	-	81	~	×1.0	59 707 24 TOT 13 DISS	_	10	
·										
								•		
								·		

<i>د</i>	the second s	SHEF		TAKI	いいていたら	1914	145	erter -	325 - CEL	<u> </u>
	DATE	CISS COD	AISS	TKN	10153 1043	i diss · Bode		ςH	TES	
- ? <i>5</i> } <b>*</b>	5/23/84	94	9.1	1	14		- - - - - - - - - - - - - - - - - - -			
<del>14</del> 4			3.2	2,5	D.Z	<3	65	6.4	612	
/rº X	7/11/24	2×	4)3	1,5	<0/1	-	73	-	-	
	7/12/84		1.6	2,3	0.1	-	~	6,8		
10000 C C	8/29/24	15	0.7	0,6	0.1	24	110	6.6	564	
بر مر	9/1E/54	N.	)Xr	346	-3×1		IDD		15	
1 7 7 7 7	9/19/54	24	5.0	0,6		_	IS	6.7		
221										
•.										
			- 19 - 19 - 19 - 19 - 19 - 19 - 19 - 19							
							•	-		
				• • • • • • • • • • • • • • • • • • •			:			
			~			•				
						1				
				-				1		

`		DATE	SHEET	•	THA AVI	eter: (	"5/2)	6751	hir *r I	415.0	. 541
		DATE	7155 Bodg	DISS COD	C1-	ρH	0 55 1 NO2-NO3	TKN	TDS	DISS NH3-N	OTHER
	) 1 m	5/23/84	->≵	Ŕ	82	鋏	6.6	Ż∕∕₹	522	3/1	-
		6/6/84	4	11	81	6.1	3,5	Z.D	_	0.1	
	Xan	7/11/34	Va	-	•	-	. M	24	-	XI	_
	ى ،	7/1Z/84	∠3	9	73	6.3	3.5	2.2	460	0.1	-
SQUARE	10	10/15/84		int	36	_	Ì,∕\$	NŞ	_	X1	
O SHEETS	6	10/16/84		10	59.0	6.3	12.4	1.2	_	0,1	
42.389 20	eş	9/19/54	24	2	122	6.3	5.3	1.9	15	2.1	_
hand	₹ ؟	11/15/84 12/11/54 9/19/54 11/5/84 11/6/24	_		39	-	X	NÓ		<ک×ز	
	leng	11/6/24		-	42	_	ZD	0.6		20,1	_
			-								
			-	1							
			-								
			:								
		1				ł			-		
	-										

 $\left( \begin{array}{c} \\ \end{array} \right)^{*}$ APPENDIX TI MIDORO : คายหนะระกุ/ คาวสายเปรา CROP NITROGEN UPTAKE AND CALCULATIONS

UNEX SOIL AND PLANT ANALYSIS LAB TOTAL MINETAL FOUNT FOR MANISTRY ALL FORME

CON

M

15.1

:73.0

34.83

....

B

6.421

7.501

5.303

ZN ·

45.71

46.97

22.22

FE

815.3

329.2

230.2

. .

NA

255.8 ( 68.2

1032 ( 64.6

299.7 ( 81.7

. . -

-

30

5.778

5.598

4,390

••••

mass

S

0.163

0.175

0.153

MG

6.099

0.110

3.877

CA

0.417

0.544

0.300

0.154

0.166

0.335

0.190

0.203

0.149

SAMPLE P K

SALER

non-A:

_	
1A	
2	
	1A 1B 2

ZN (on d	ry weight basis)	
1A 2	1.56 1.44	
Ashed 1B 2	0.42 0.26	

<u>z n</u>	of	Ash
	1B 2	2.49 2.64

	e over drie	d sample (grams)	7 Ash
Weight of	UVE		
	87.3	· · ·	17.1
14	67.2	•	10.1
18	311.3		•
2			

IA, IB -> MINDORD 2 -> BREDHEAD

APRIL SAMPLES OF GRASS

SAVER ASHED SAMPLES ASh

UNEX SOLL AND PLANT ANALYSIS LAB STLL MINERAL POINT POAR MADIODN AL SERVES

DATE OF ANALYSIS: 7/6/84

			z						DDM			
SAMPLE	P	K	CA	MG	S	ZN	B	MN	T FE	3	AL.	NA
1 1B 2 2B	0.187 0.140	0.149 0.352	0.495 0.278	0.096 0.072	0.071 0.072	38.51 20.93	5.330 3.249	157.6 32.65	638.2 243.4	4.711 4.448	662.2 243.8	< 61.3 < 62.5



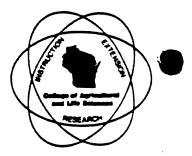
COOPERATIVE EXTENSION PROGRAMS University of Wieconsin-Extension

University of Wisconsin-Medicon

Soil & Plant Analysis Laboratory, 5711 Mineral Point Road, Madison, Wisconsin 53705; 608-262-4364

## DEPARTMENT OF SOIL SCIENCE

September 14, 1984 Acct. 900 Lab No. S0035





MEMORANDUM

Dave Sauer--DNR TO: 101 S. Webster, Box 7921 Madison, WI 53707

Soil/Plant Analysis Lab FROM:

Results of analyses on 4 canary (+grass) samples submitted July 24, 1984. RE:

Sample Identification		Sample Weight	· Ash	Nitrogen of Tissue	Nitrogen of Ash
		grams	%	%	%
Mindoro	1	87		2.15	
Brodhead	1	134		1.28	
Mindoro	2	121	8.5		0.64
Brodhead	2	110	5.9		0.48

Additional analyses are attached.

If you have any questions concerning these analyses, please feel free to contact either Todd Kaehler or Ita Steingraeber at 262-4364.

Encl.

· /ss

535 DAVE SALER DAR

UNEX SOIL AND PLANT ANALYSIS LAB 5711 HINERAL POINT ROAD MADISON HI 53705

JATE OF ANALYSIS: 9/14/84

			*						PPM			
SAMPLE	P	K	CA	MG	S	ZN	8	HN	FE	CU	AL	NA
			·									
1 MIN.1	0.255	2.443	0.314	8.216	0.252	12.73	5.361	43.93	60.01	3.843	( 36.7	< 63.8
2 BR0.1	0.229	1.632	0.235	0.128	0.144	12.43	3.736	54.69	42.23	3.828	< 35.8	294.7
3 MIN.2	0.327	2.632	0.298	0.214	0.249	16.72	4.708	88.24	56.14	4.626	< 36.4	< 63.3
4 BR0.2	0.306	1.953	0.300	0.180	0.182	20.50	4.709	40.22	62.70	3.724	< 36.2	296.0



DAVE SAVER (ASHED)

# UHEX SOIL AND PLANT ANALYSIS LAB 5711 MINERAL POINT ROAD MADISON HI 53705

OF ANALYSIS: 9/14/84

5

-	<b>x</b>						PPM							
£	P	K	CA	MG	S	ZN	B	ĦN	FE	α	AL	NA		
-			·											
1IN.2	0.318	2.509	0.296	0.204	8.145	16.88	5.042	85.15	<b>ଣ</b> .52	6.397	< 36.1	<b>99.24</b>		
3R0.2	0.293	1.813	0.289	0.165	0.056	18.51	4.301	39.88	70.66	3.827	< 36.0	295.8		

5235 DAVE SAUER Dec Somples

ATE OF ANALYSIS: 12/28/84

# UNEX SOIL AND PLANT ANALYSIS LAB 5711 MINERAL POINT ROAD MADISON WI 53705

				07 70						PPM			
SAM	PLE	P	К	CA	MG	S	ZN	В	MN	FE	CU	AL.	NA
				<del></del>									
1	BROD.1	0.189	0.651	0.201	0.092	0.158	14.20	4.326	56.47	87.73	3.712	65.37	886.3
2	BROD.2	0.197	0.491	0.146	0.136	0.117	40.26	3.188	79.76	57.89	4.060	42.81	430.5
₩ <b>3</b>	MIND.1	0.139	0.497	0.230	0.094	0.086	28.59	4.866	139.9	106.8	4.450	102.9	< 61.0
٠4	MIND.2	0.172	0.549	0.177	0.097	0.158	28.68	4.610	70.91	79.80	3.719	64.15	184.6
1	BROD.1 ash	0.183	0.603	0.194	0.038	0.065	11.36	3.085	48.61	82.87	3.032	68.71	992.6
2	BROD.2 ash	0.197	0.472	0.142	0.135	0.046	38.19	2.927	74.83	55.44	3.273	43.73	555.2
. 3	MIND.1 ash	0.135	0.431	0.225	0.089	0.048	28.92	3.336	128.6	103.0	4.723	126.2	< 59.3
· 4	MIND.2 ash	0.165	0.480	0.175	0.092	0.077	25.42	< 3.56	67.28	81.79	4.276	89.32	242.9

Sample Id.	Sample Wt. grams	%Ash	%N of Tissue
Brodhead Cell 1	54.5	3.9	1.59
Brodhead Cell 2	61.0	3.1	1.41
✓ Mindoro Cell 1	88.7	9.2	0.58
✓ Mindoro Cell 2	219.8	8.6	1.42

*Results for TN of Ash will follow in several days.

# COOPERATIVE EXTENSION PROGRAMS

University of Wisconsin-Extension

University of Wisconsin-Madison

Soil & Plant Analysis Laboratory, 5711 Mineral Point Road, Madison, Wisconsin 53705, 608-261-4364

#### DEPARTMENT OF SOIL SCIENCE

January 7, 1985 Acct. No. 900 Lab Nos. S134; S235

### MEMORANDUM

Dave Sauer TO: Wis. Dept. of Natural Resources Box 7921 Madison, WI 53707

Soil/Plant Analysis Lab FROM:

Results of %N of Ash on 5 samples. All other analyses have been reported. RE:

Sample Identificat	%N of Ash	
(S134) Cこう Sameras BRODHEAD CELL 1	1	0.72
BRODHEAD CELL	2	0.50
(S235) BRODHEAD CELL	1	0.45
BRODHEAD CELL	2	0.49
MINDORO CELL	1	0.15
MINDORO CELL	2	0.47

If you have any questions concerning these analyses, please feel free to contact us.

The invoice for all of the tissue analyses is enclosed.

/ss

Encl.

APRIL 26, 1984 WEICHT 9 AREA OF SAMPLES => CELL 3 = 1×1'= 1512 87.3  $CELL 3 = 1'x' = 1ft^{2}$ 67.2 AVERALE WEIGHT OF = 87.3+67.2 = 77.25gGRASS SAMPLE = 2SYSTEM AREA = BACRE × 43560 5t² = 130,680 ft² TOTAL WEIGHT OF GRASS ON SITE =  $\frac{116}{1.5t^2} \times 130,6805t^2 \times \frac{116}{453.8q} = 22,2551b$ TON (DRY WEIGHT BASIS) = 1.56 AMONT OF NON SITE PRICE TO BURNING = 22,255 (0.0156) = 397 16 26 ASH AFTER BURNING = 17,1 WEIGHT OF ASH ON SITE = 22,255 (0.171) = 3806 1/2 2 N OF ASH = 2,49 AMOUNT OF N ON SITE AFTER BURNING = 3506(0.0249) 95 lbs 2 20 N LOST BY BURNING =  $\frac{347-95}{247}(100) = 72.6\%$ 

Plade In U.S.A.

La Contrated

JULY 12, 1984 WEICHT 9 AREA OF SAMPLES -> CELL Z= 1'x1'= 15t2 87 CELL Z= 1'x1'= 15t2 121 AVERAGE WEIGHT OF =  $\frac{87+121}{2} = 10.4g$ TOTAL WEIGHT OF GRASS =  $\frac{104_{7}}{15t^{2}} \times 130,680 \text{ ft}^{2} \times \frac{116}{453.8_{5}}$ = 29,962 lb 70 N(DRY WEICHT BASIS) = 2.15 AMOUNT OF NON SITE = 29,962 (0.0215) = 644 16 (PRICE TO BLENING) 7. ASH AFTER BURNING = 8.5 WEIGHT OF ASH ON SITE = 29,962 (0.085) = 2547 15 70 NOF ASH = 0.64 WEIGHT OF NON SITE FFTER BURNING = 2547 (0.0064) = 16.3 16 70 N Loss BY BURNING = 97.570  $= \frac{644 - 16.3}{(.44)} (100)$ 

PTIM IN U.S.A.

NOVEMBER 6, 1984 AREA OF GRASS SAMPLES - CELL Z = 1'X1'=15t2 58.7 (EIL Z = 1'x1'= 15t 219.8 AVE. Wt. OF GDASS SAMPLE = 88.7 + 219.8 = 154.2 TOTAL WE. OF SITE GRASS = 154.2 (130,680) = 44,439 16  $\mathcal{D}_{r} \mathcal{N}(024 \text{ wt. Basis}) = \frac{1.42 + 0.58}{7} = 1.0$ AMOUNT OF NON SITE PERR TO = 44,439 (0.01) = 444 16 BURNING 90 ASH AFTER BURNING = 9.2 + 8.6 = 8.9 Wt OFASH ON SITE = 44,439(0.089) = 3955 16 % N OF ASH = 0.15+0.41 = 0.31 AMOUNT OF NON SITE AFTER = 3955 (0.0031) = 12.215 BURNING  $9 \cdot N \text{ LOST BY BURNING} = \frac{444 - 12.7}{44/1} (100) = 97.270$ 

( ____)

A 211 M PERA (6)

APPENDIX MINDORO : SEVERSON COULEE CREEK CHEMISTRY DATA

•	DAT	a Shee			(mg/1)	1"IEI EK >	<u>i</u>	REEK UPSTREAM	1
	DATE	BODS	LOWLENEL COD	C1 ⁻	DISSOLVED NOZ-N+ NOZ-N	TKN	DISSAULED NH3-N		
	11/30/83	23	6	7,9	0.7	0.Z	0.1	-	
	B/29/94	23	6	6.9	0.8	0,2 7155 0.2 70 <del>.</del>	0.1	ALK = 250 $C_0 = 53$ $M_0 = 20$ PH = 8.0 $N_0 = 4$ $SQ_1 = 14$ $T_0TP = 0.06$	te a constant a constant de semanante e a constant a
การกระการนี้มี/ การกระการระ								K ⁺ = Z	
							-		
		-							

	DATA	Shee			(mg/l)			EEK MIOWAY	
	DATE	BODS	LONLEVEL COD		DISSOLVED NOZ-N+ NOZ-N	TKN	DISSOLVED NH3-N	CTHER	
)	11/30/83	<3	25	3.6	0.8	0.2	0.1	~	
	8/29/84	∠3	25	6.9	0.8	D.Z DISS	0.1	ALK = 252	
						0.2 757		Ca#= 59 CI = 6.9	
								$m_{g} = 30$ $\rho H = 8.0$	
								Not = 4 50 = - 14 Tot P = 0.06	
Mad•InU.S.A								K+= 2	
							-		
18									
				-					
						•			
						- -			
•	• • •								· · ·
	2	-							
	: : :								
	: :					•			
ì									

	Dat	A SHEE	LOWLENEL		(mg/l)	)	PISSOLVED	REEK DOWNSTREAM	
	DATE	BOD5	00	<u>C</u> 1-	DISSOLVED NG-N NG-N	TKN	NH3-N	DTHER	
	11/30/83	23	7	8.0	98	0.3	0.1	-	
	8/29/84	∠3	7	7,2	0.8	0.3 0455 0.4 TOT	0,1	$ALK = 250$ $Ca^{\#} = 58$ $PH = 8.0$ $M_{0}^{\#} = 30$ $Na^{\#} = 4$ $S0_{4}^{*} = 15$ $To 7 P = 0.06$	
N.S.N		LOW SE	POFI	RY WELL				K+= 2	
PADRARIEN AndeinUSA	3/29/84	23	8	7.2	0.8	0.2 DISS 0.2 TeT	0.1	$f_{ik} = 252$ $G_{+} = G_{3}$ $p_{i} = 7.9$ $n_{i} = 30$ $n_{i} = 4$ $SO_{+} = 15$ $ToT P = 0.15(visc)$ $= 0.17(707)$ $K^{+} = 2$	
( )							-		
• *									
								-	
						-			
<u>,</u>			-						
						•			

f A PPENDIX KK MINDORD : 'ade In U S A. NITROGEN BUDGET CALCULATIONS 

WASTEWATER ADDITION (APPLIED)  $\frac{16}{yr} = \frac{14,000}{10^{6}} \frac{941}{x} \times 32.4 \frac{mg-N}{e} \times 8.34 \times 365 \frac{04y}{ye} =$ = 1381 16/yr V STER SEA PLANT UPTAKE LOSS 252 10/yr per Discussion in CHAPTER 5 ุ่ม เกษณะสายก____ (or 1820 OF APPLIED-N) LEACHING LOSS (BASED ON WELL 5 TOTAL-N CONCENTRATION AVERACE)  $\frac{16}{3r} = \frac{14,000 \text{ g/day}}{166} \times 4.1 \frac{m_2 N}{4} \times 8.34 \times 365 \frac{DAY}{YR} = 175 \frac{16}{3r}$ (UR 13900F APPLIED N) DENITRIFICATION LOSS 1b/yr = 1381 - 252 - 175 = 954 1b/yr (or 6950 CF APPLIED - N)



